

Supplement of Atmos. Chem. Phys., 18, 12715–12734, 2018
<https://doi.org/10.5194/acp-18-12715-2018-supplement>
© Author(s) 2018. This work is distributed under
the Creative Commons Attribution 4.0 License.



Supplement of

Biomass burning emission disturbances of isoprene oxidation in a tropical forest

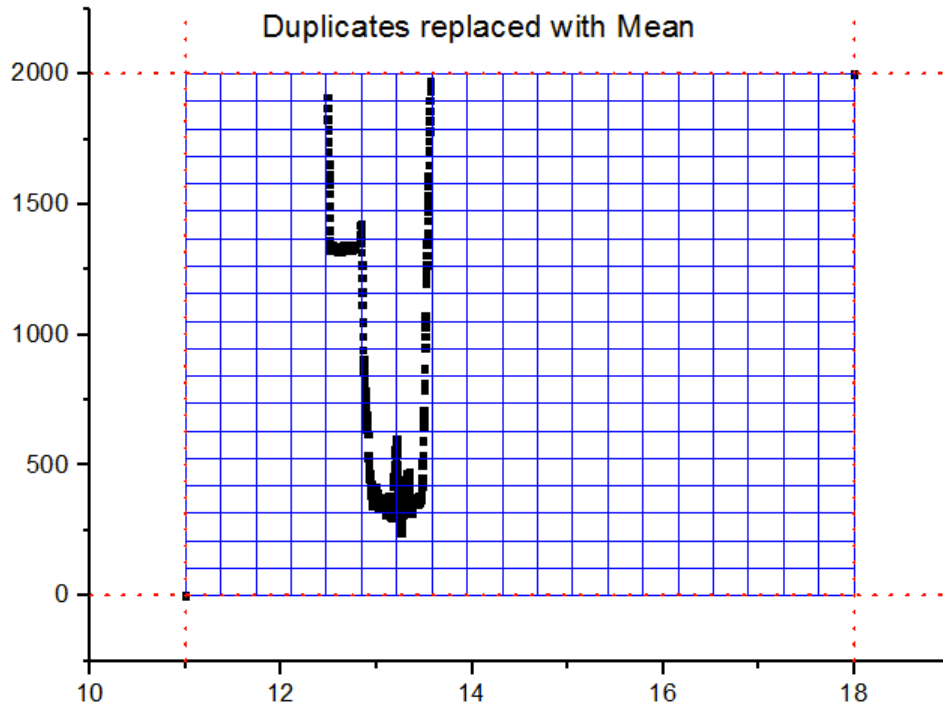
Fernando Santos et al.

Correspondence to: Fernando C. dos Santos (santos.f@mail.com)

The copyright of individual parts of the supplement might differ from the CC BY 4.0 License.

1 **S1.** Interpolation grid used in the Figures 3 and 5.

2



3

4 Figure 1. Interpolation grid (0 - 2000m and 11 - 18h)

5

6

7

8

9

10

11

12 **S2.** The kinetic rate constant measurements for OH + ISOPOOH (1,2- and 4,3- ISOPOOH), at 297 K, is $7.5 \times 10^{-11} \text{ cm}^3$
13 $\text{molecule}^{-1} \text{ s}^{-1}$ for (1,2)-ISOPOOH and $1.18 \times 10^{-10} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$ for (4,3)-ISOPOOH (St Clair et al., 2015). The kinetic
14 rate constant of MVK + OH = $1.88 \times 10^{-11} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$ and MACR + OH = $3.35 \times 10^{-11} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$ (Apel,
15 2002).

16 K_{prod} Average kinetic rate constant = 6.1325×10^{-11}

17 $K_{\text{iso}} - K_{\text{prod}} = (1.1 \times 10^{-10}) - (6.1325 \times 10^{-11}) = 4.8675 \times 10^{-11}$

18

19

20

21

22

23

24

25

26

27

28

29

30

31

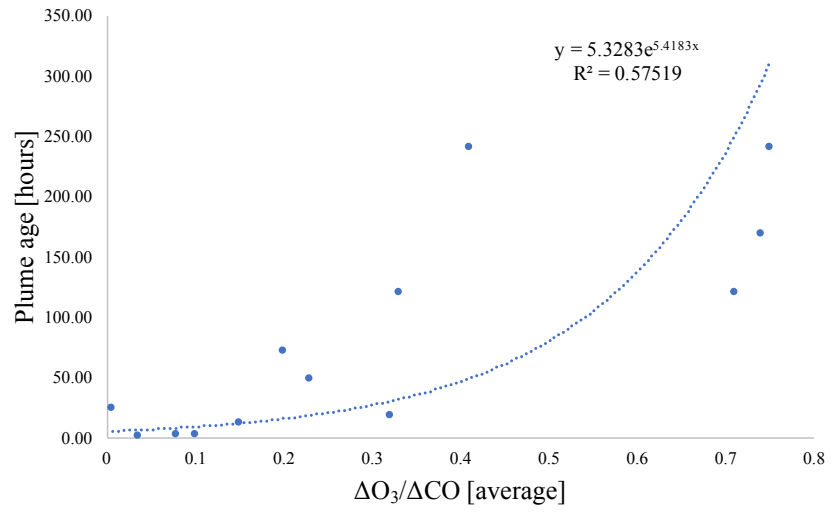
32

33

34

35 S3.

36



37

38

Figure 2. Observations of the ratio $\Delta O_3/\Delta CO$ as a function of plume age in tropical and subtropical sites.

39

40

41

42

43

44

45

46

47

48

49

50

51

52 **S4.**

53 Simulated convective velocity and planetary boundary layer from WRF-Chem in the Forest Management Station ZF-2 (02°
54 36'S and 60° 12'W), 60 km north of Manaus (Dasa Gu, personal communication, June 2015).

55

56 Table 1. Convective velocity and planetary boundary layer used to calculate OH density

57 following Karl et al. (2007) approach.

Time interval (t)	Convective velocity (W)	PBL (Zi)
<i>h</i>	<i>ms⁻¹</i>	<i>m</i>
11-12	0.58	267
12-13	1.01	462
13-14	1.32	777
14-15	1.40	1075
15-16	1.40	1237
16-17	1.43	1209
17-18	1.31	1124

58

59

60

61

62

63

64

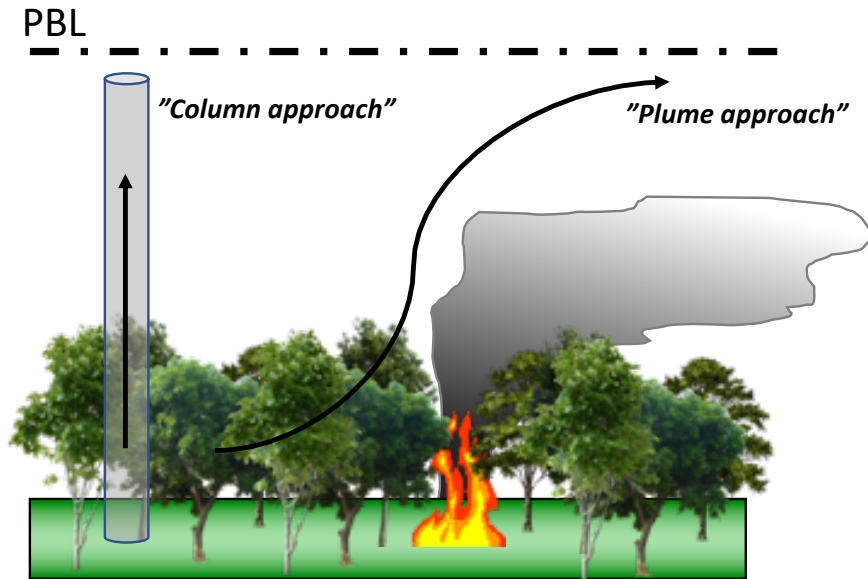
65

66

67

68 S5.

69



70

71

72 Figure 3. Schematic approaches used in OH calculation based on the sequential reaction
73 model: (1) Column approach and (2) Plume approach, which considers vertical and
74 horizontal transport for both biomass burning regimes and background environment.

74

75

76

77

78

79

80

81

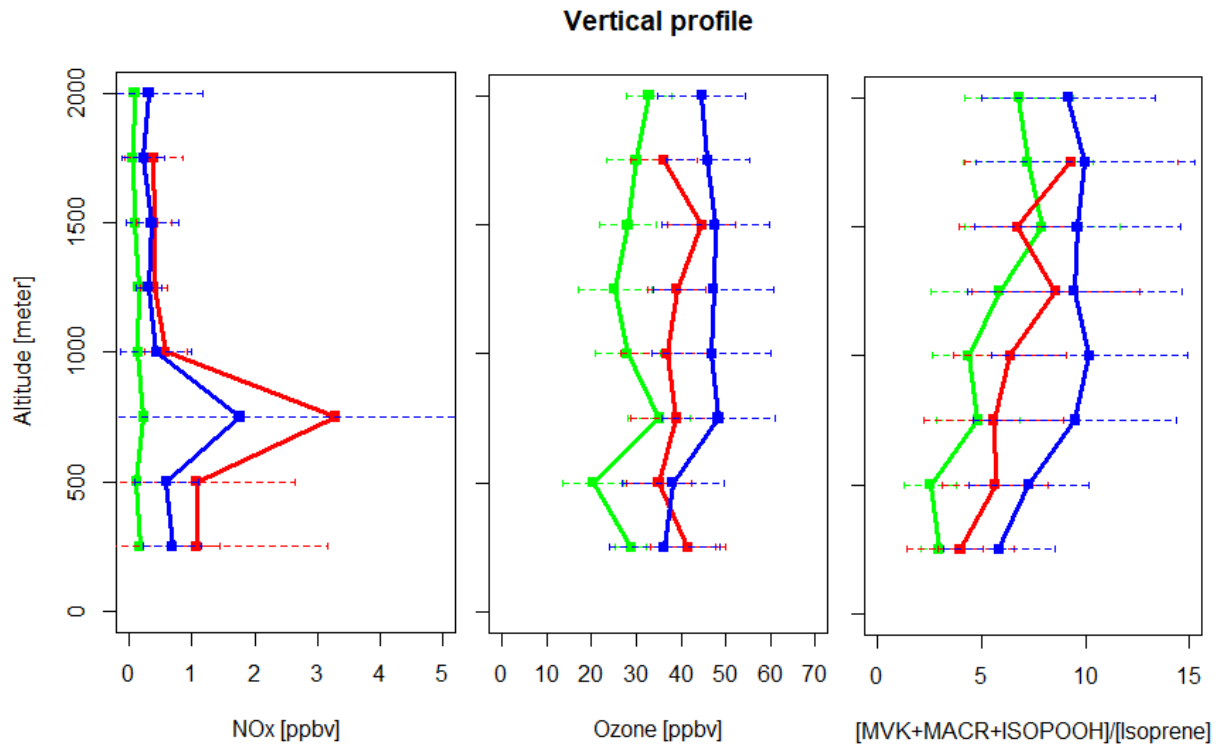
82

83

84

85 S6.

86



87

88 Figure 4. Vertical profile for the ratio $[MVK+MACR+ISOPOOH]/[Isoprene]$, NOx and Ozone mixing ratios

89 during SAMBBA campaign: background (green), fresh smoke plume (red) and aged smoke plume (blue).

90

91

92

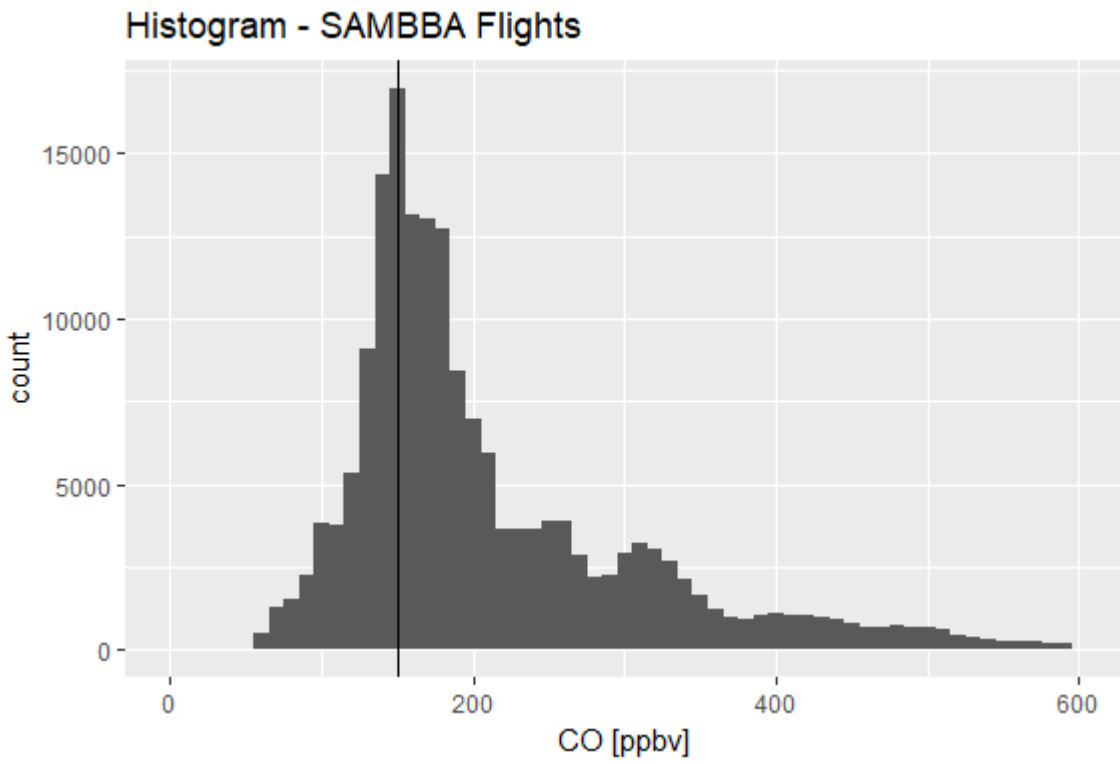
93

94

95

96

97



99

100 Figure 5. Histogram that present the frequency distribution of CO [ppbv] for all SAMBBA flights in
101 Amazon rainforest.

102

103

104

105

106

107

108

109

110

111

112 **References**

113

114 Apel, E. C. (2002). Measurement and interpretation of isoprene fluxes and isoprene, methacrolein, and methyl vinyl ketone
115 mixing ratios at the PROPHET site during the 1998 Intensive. *Journal of Geophysical Research*, 107(D3), 1–15.

116 <https://doi.org/10.1029/2000JD000225>

117 St Clair, J. M., Rivera-Rios, J. C., Crouse, J. D., Knap, H. C., Bates, K. H., Teng, A. P., ... Wennberg, P. O. (2015).

118 Kinetics and Products of the Reaction of the First-Generation Isoprene Hydroxy Hydroperoxide (ISOPOOH) with OH.

119 *Journal of Physical Chemistry A*, 150915075517008. <https://doi.org/10.1021/acs.jpca.5b06532>

120