Parameter	Description	Value
Condensed phase		
$D_{p,S}$	Surface-weighted middle diameter of initial dry particles	125 nm (initial)
$ ho_{ m p}$	Density of SOA	$1.3\mathrm{gcm^{-3}}$
$MW_p$	Molecular weight of SOA	$175 \mathrm{g} \mathrm{mol}^{-1}$
GF <sup>1</sup>	Hygroscopic growth factor	1.1 (high RH) or 1.0 (low RH)
γ	Uptake coefficient of OH and HO <sub>2</sub>	0–1
$p_{\rm frag}$	A combined probability for fragmentation	0–1
$k_{\rm RO_2+RO_2}$	Reaction rate coefficient of $RO_2 \cdot + RO_2 \cdot$	$1 \times 10^{-21}$ -1 × $10^{-15}$ cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup>
$d_{ m SL}$	Depth of surface layer	0.76 nm
$D_{ m org}$	Bulk diffusivity of SOA compound	$1 \times 10^{-16} - 1 \times 10^{-11} \mathrm{cm}^2 \mathrm{s}^{-1}$
Gas-phase		
[OH] <sub>g</sub>	OH concentration along the flow tube	$0-1.84 \times 10^{11}$ molecule cm <sup>-3</sup> -air
[HO <sub>2</sub> ] <sub>g</sub>	HO <sub>2</sub> concentration along the flow tube	Same as [OH]
$\overline{c}_{\mathrm{OH}}$	Mean speed of OH radicals	$610\mathrm{ms^{-1}}$
$\overline{c}_{\mathrm{HO}_2}$	Mean speed of HO <sub>2</sub> radicals	$440\mathrm{ms^{-1}}$
$D_{ m g,OH}$	Diffusion coefficient of OH radicals	$0.21\mathrm{cm}^2\mathrm{s}^{-1}$
$D_{\mathrm{g,HO}_2}$	Diffusion coefficient of HO <sub>2</sub> radicals	$0.25\mathrm{cm}^2\mathrm{s}^{-1}$