



# Characteristics and sources of aerosol aminiums over the eastern coast of China: insights from the integrated observations in a coastal city, adjacent island and surrounding marginal seas

Shengqian Zhou<sup>1</sup>, Haowen Li<sup>1</sup>, Tianjiao Yang<sup>1</sup>, Ying Chen<sup>1,2</sup>, Congrui Deng<sup>1,2</sup>, Yahui Gao<sup>3,4</sup>, Changping Chen<sup>3,4</sup>, and Jian Xu<sup>1</sup>

<sup>1</sup>Shanghai Key Laboratory of Atmospheric Particle Pollution Prevention, Department of Environmental Science & Engineering, Fudan University, Jiangwan Campus, Shanghai 200438, China

<sup>2</sup>Institute of Eco-Chongming (IEC), 3663 N. Zhongshan Rd., Shanghai 200062, China

<sup>3</sup>Key Laboratory of the Ministry of Education for Coastal and Wetland Ecosystems, School of Life Sciences, Xiamen University, Xiamen 361005, China

<sup>4</sup>State Key Laboratory of Marine Environmental Science, Xiamen University, Xiamen 361005, China

**Correspondence:** Ying Chen (yingchen@fudan.edu.cn) and Congrui Deng (congruideng@fudan.edu.cn)

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**Abstract.** An integrated observation of aerosol aminiums was conducted in a coastal city (Shanghai) in eastern China, a nearby island (Huaniao Island), and over the Yellow Sea and East China Sea (YECS). Triethylaminium (TEAH<sup>+</sup>) was abundant over Shanghai but not detected over the island and the open seas, suggesting its predominantly terrestrial origin. By contrast, relatively high concentrations of dimethylaminium (DMAH<sup>+</sup>) and trimethylaminium + diethylaminium (TMDEAH<sup>+</sup>) were measured over the ocean sites, indicating the significant marine source contribution. Environmental factors, including boundary layer height (BLH), temperature, atmospheric oxidizing capacity and relative humidity, were found to be related to aminium concentrations. All the detected aminiums demonstrated the highest levels in winter in Shanghai, consistent with the lowest BLH and temperature in this season. Aminiums mainly existed in fine particles and showed a bimodal distribution, with two peaks at 0.18–0.32 μm and 0.56–1.0 μm, indicating that condensation and cloud processing were the main formation pathways for aminiums in analogy with NH<sub>4</sub><sup>+</sup> and non-sea-salt SO<sub>4</sub><sup>2-</sup> (nss-SO<sub>4</sub><sup>2-</sup>). Nonetheless, a unimodal distribution for aerosol aminiums was usually measured over the YECS or over Huaniao Island when influenced mainly by the marine air mass, which suggested that aminiums in marine aerosols may undergo different formation pathways from those on the land. Terrestrial

anthropogenic sources and marine biogenic sources were both important contributors for DMAH<sup>+</sup> and TMDEAH<sup>+</sup>, and the latter exhibited a significantly higher TMDEAH<sup>+</sup> to DMAH<sup>+</sup> ratio. By using the mass ratio of methanesulfonate (MSA) to nss-SO<sub>4</sub><sup>2-</sup> as an indicator of marine biogenic source, we estimated that marine biogenic source contributed to 26%–31% and 53%–78% of aerosol aminiums over Huaniao Island in the autumn of 2016 and summer of 2017, respectively. Due to the important role of atmospheric amines in new particle formation, the estimation results highlighted the importance of marine biogenic emission of amines on the eastern coast of China, especially in summer.

## 1 Introduction

Low molecular weight amines are commonly found in the atmosphere in both gas and particle phases (Ge et al., 2011a, b). Based on present theoretical calculations (Kurtén et al., 2008; Loukonen et al., 2010; Paasonen et al., 2012; Olenius et al., 2017), laboratory simulations (Wang et al., 2010a, b; Kurtén et al., 2014; Erupe et al., 2011; Almeida et al., 2013; Yu et al., 2012) and field observations (Smith et al., 2010; Kürten et al., 2016; Tao et al., 2016), amines in the atmosphere have been proved to play an important

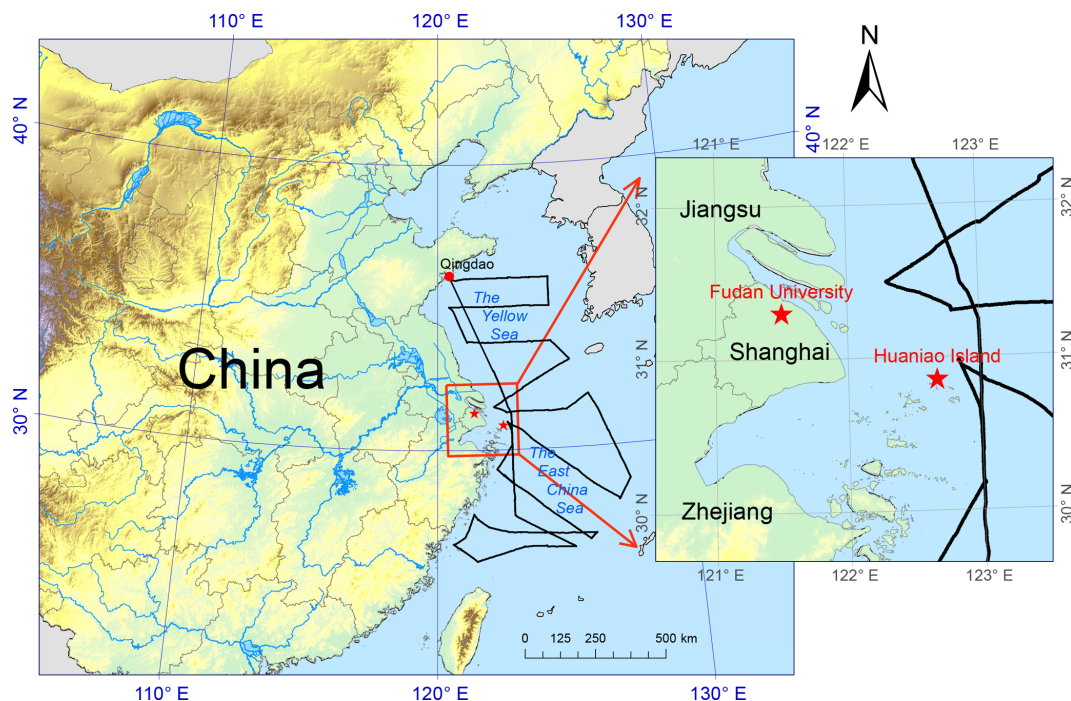
role in new particle formation and subsequent particle growth and thus affect both the number concentrations of aerosols and cloud condensation nuclei that are closely relevant to regional climate (Tang et al., 2014; Yao et al., 2018). For example, dimethylamine (DMA) was found to be a key species involved in new particle formation events in the urban area of Shanghai, and the nucleation mechanism was likely to be  $\text{H}_2\text{SO}_4$ -DMA- $\text{H}_2\text{O}$  ternary nucleation (Yao et al., 2018). Gaseous amines in the atmosphere can react with oxidants such as  $\cdot\text{OH}$  and  $\text{O}_3$  to form secondary organic aerosols (SOA) (Murphy et al., 2007) or gaseous oxidation products such as imines, formamides, nitrosamines and nitramines (Nielsen et al., 2012). In aerosols, amines are mainly in the form of protonated cations, namely aminiums (Ge et al., 2011a), and the formation of aminium salts from an acid-base reaction or heterogeneous reaction, such as replacing the  $\text{NH}_4^+$  in particles, is another important pathway for amines to form SOA in the atmosphere (Pankow, 2015; Kupiainen et al., 2012; Liu et al., 2012; Chan and Chan, 2013).

Amines originate from a wide range of sources, including anthropogenic sources such as animal husbandry and industrial emissions, as well as natural sources such as marine sources, vegetation emissions and soil processing (Ge et al., 2011b; Hemmilä et al., 2018). Dawson et al. (2014) measured concentrations of trimethylamine (TMA, 1.3–6.8 ppt) near a cattle farm, which were 2–3 orders of magnitude higher than those in ambient environments. Shen et al. (2017) demonstrated that coal combustion could emit abundant methylaminium ( $\text{MMAH}^+$ ), ethylaminium ( $\text{MEAH}^+$ ) and diethylaminium ( $\text{DEAH}^+$ ) through combustion experiments, and the corresponding emission factors were  $18.0 \pm 16.4$ ,  $30.1 \pm 25.6$  and  $14.6 \pm 10.1$  mg (kg coal) $^{-1}$ , respectively. In the marine boundary layer, marine source is an important contributor for amines and it was found to be closely related to the biological activities on the ocean surface. In the North Atlantic, the concentrations of dimethylaminium ( $\text{DMAH}^+$ ) and  $\text{DEAH}^+$  were significantly higher during the periods with high biological activity and clean air masses than those with low biological activity or polluted air masses advecting to the sampling site, and the contributions of these two aminiums to SOA and water-soluble organic nitrogen (WSO) reached 11% and 35%, respectively (Facchini et al., 2008). The observation in Cabo Verde also showed that the concentrations of aminiums were higher during the occurrence of algal blooms (Müller et al., 2009). In addition to gas-to-particle conversion, which has been generally considered to be the major formation pathway (Facchini et al., 2008; Rinaldi et al., 2010), aminiums in the marine boundary layer may also be generated with primary marine aerosols. For example, Fourier transform infrared (FTIR) spectroscopy measurements demonstrated that the submicron organic carbon was composed of 50% hydroxyl, 33% alkane, and 14% amine in nascent sea spray aerosols artificially generated off the California coast (Bates et al., 2012) and of 55% hydroxyl, 32% alkane, and 13% amine over

the open ocean (Frossard et al., 2014). Aerosol time-of-flight mass spectrometry (ATOFMS) analyses of ambient aerosols in the Antarctic sympagic environment also indicated that 11%–25% of aminiums were contributed by primary marine source (Dall'Osto et al., 2019).

Given the potentially important roles of amines in the atmosphere and the complexity of their sources, it is important to conduct a systematic analysis of their concentrations, affecting factors, formation pathways and source contributions. Eastern China is a densely populated region with strong human activity and large emissions of atmospheric pollutants. Under the influence of the summer monsoon, marine source components can be vital to the atmospheric composition of the coastal area. Although the lifetime of gaseous amines in the atmosphere is only a few hours, it can be prolonged after amines partition into the particle phase, and thus they may be transported over a long range (Nielsen et al., 2012). Many studies have been done on the gas and/or particle phases of amines over eastern China and adjacent seas (Huang et al., 2012, 2016; Hu et al., 2015; Zheng et al., 2015; Tao et al., 2016; Yu et al., 2016; Shen et al., 2017; Xie et al., 2018; Yao et al., 2016, 2018). For example, C1 to C6 amines over Shanghai were measured during the summer of 2015, of which C1, C2 and C4 amines were the dominant species with average concentrations of 15.7, 40.0 and 15.4 ppt, respectively (Yao et al., 2016). Zheng et al. (2015) measured an average concentration 7.2 ppt of total amines in a suburban site of Nanjing during the summer of 2012, derived mainly from industrial emissions in adjacent areas. The aminiums in fine particles over Shanghai in the summer of 2013 were found to exhibit a high concentration (mean  $86.4 \text{ ng m}^{-3}$ ) and play an important role in the new particle formation events (Tao et al., 2016). Previous studies on aminiums over the marginal seas off the coast of China indicated that  $\text{DMAH}^+$  and trimethylaminium ( $\text{TMAH}^+$ ) were overwhelmingly from marine sources (Hu et al., 2015; Yu et al., 2016; Xie et al., 2018). In May 2012, the concentrations of  $\text{DMAH}^+$  and  $\text{TMAH}^+$  over the Yellow Sea (YS) and Bohai Sea even reached 4.4 and  $7.2 \text{ nmol m}^{-3}$ , which was 1–3 orders of magnitude higher than those reported in other oceanic regions (Hu et al., 2015). These extremely high concentrations were thought to be associated with high biological activities. In spite of these field studies, long-term observations of aminiums over coastal seas and the quantitative estimates of the contribution of marine biogenic source to aerosol aminiums are still lacking.

In this study, the aminiums over a coastal megacity (Shanghai), a nearby island (Huaniao Island) and surrounding marginal seas (the Yellow Sea and East China Sea, YECS) were measured. The relationships between aminium concentrations and environmental factors were systematically analyzed. The size distributions of aminiums were investigated with speculation on the main formation pathways. Aside from this, the dominant sources determining the concentrations and ratios between aminium species were eluci-



**Figure 1.** Map of sampling sites and area. The red stars represent the locations of Shanghai (Fudan University) and Huaniao Island, and the black line in the surrounding marginal seas represents the cruise track in the spring of 2017.

dated, and the contributions of terrestrial anthropogenic and marine biogenic sources to aminiums were quantitatively estimated. Our results will be a great help for understanding the chemical properties, reaction pathways, and sources of aerosol aminiums over the coastal area and the ocean.

## 2 Sampling and analysis

### 2.1 Aerosol sampling

The sampling site in Shanghai was located on top of the No. 4 teaching building of Fudan University (31.30° N, 121.50° E) (Fig. 1). This site is affected by the school, residential, commercial and traffic activities and can be taken as representative of coastal cities. Particulate matter with an aerodynamic diameter less than 2.5  $\mu\text{m}$  ( $\text{PM}_{2.5}$ ) was simultaneously collected by two medium-flow samplers (100 L  $\text{min}^{-1}$ , HY-120B, Hengyuan) using a 90 mm pre-combusted quartz filter (Whatman) and a cellulose filter (Grade 41, Whatman), respectively. A total of 131 samples were collected within four seasons with a sampling duration of around 24 h (spring: 25 March–26 April 2013; summer: 16 July–17 August 2013; autumn: 30 October–30 November 2013; winter: 1 December 2013–23 January 2014) (Table 1).

Aerosols were also collected on Huaniao Island (HNI, 30.86° N, 121.67° E), which is about 80 km away from Shanghai in the East China Sea (ECS) (Fig. 1). The locally anthropogenic emissions were negligible, but the site was

affected by the terrestrial transport and the ship emissions from nearby container ports (Wang et al., 2016, 2018).  $\text{PM}_{2.5}$  samples were collected in the summer of 2016 (4–18 August) and size-segregated samples were obtained using a 10-stage Micro-Orifice Uniform Deposit Impactor (30 L  $\text{min}^{-1}$ , MOUDI, MSP Model 110-NR) and 47 mm PTFE filters (Zeflour, PALL) in the autumn of 2016 (12 November–3 December), the spring of 2017 (11–19 March) and early and late summer 2017 (22 June–9 July and 27 August–12 September, respectively) (Table 1). The 50 % cutoff diameters for the 10 stages were 18, 10, 5.6, 3.2, 1.8, 1.0, 0.56, 0.32, 0.18, 0.10 and 0.056  $\mu\text{m}$  and the sampling durations were 24–48 h.

The size-segregated samples were also collected over the YECS on board research vessel (R/V) *Dongfanghong II* in the spring of 2017. The cruise started from Qingdao on 27 March and returned on 15 April (Fig. 1), and a total of nine sets of samples were obtained.

### 2.2 Chemical analysis

One-fourth of  $\text{PM}_{2.5}$  and half of MOUDI sample filters were cut and placed into a polypropylene jar (Nelgene) with 15 and 20 mL of ultrapure water (18.25  $\text{M}\Omega\text{ cm}^{-1}$ ), respectively, for a 40 min ultrasonic extraction. The extract was filtered through a 0.45  $\mu\text{m}$  PTFE filter (Jinteng) and stored at 4° for ion measurement. An ion chromatograph (IC, DIONEX ICS-3000, Thermo Fisher) assembled with AG11-HC and AS11-HC was used to determine anions, including  $\text{Cl}^-$ ,  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{HCOO}^-$ , methane-

**Table 1.** Summary of sampling information from the different campaigns.

Sampling site	Sampler	Sampling period	Number of samples or sample sets
Fudan University, Shanghai	Medium-flow PM <sub>2.5</sub> sampler	25 March–26 April 2013 (spring)	29
		16 July–17 August 2013 (summer)	26
		30 October–30 November 2013 (autumn)	29
		1 December 2013–23 January 2014 (winter)	37
Huaniao Island	Medium-flow PM <sub>2.5</sub> sampler	4–18 August 2016 (summer)	14
Huaniao Island	MOUDI	12 November–3 December 2016 (autumn)	9
		11–19 March 2017 (spring)	4
		22 June–9 July 2017 (early summer)	8
		27 August–12 September 2017 (late summer)	7
Yellow Sea and East China Sea	MOUDI	27 March–14 April 2017 (spring)	9

sulfonate (MSA), malonate, succinate, glutarate, maleate and C<sub>2</sub>O<sub>4</sub><sup>2-</sup>. The columns CG17 and CS17 were used to measure inorganic cations including Na<sup>+</sup>, NH<sub>4</sub><sup>+</sup>, K<sup>+</sup>, Mg<sup>2+</sup> and Ca<sup>2+</sup> and aminiums. Six aminiums, including DMAH<sup>+</sup>, TMAH<sup>+</sup> + DEAH<sup>+</sup>, propylaminium (MPAH<sup>+</sup>), triethylaminium (TEAH<sup>+</sup>), ethanolaminium (MEOAH<sup>+</sup>) and triethanolaminium (TEOAH<sup>+</sup>), could be effectively separated and measured using the IC method. The MMAH<sup>+</sup> and MEAH<sup>+</sup> in the aerosols could not be quantified because their peaks were obscured by the wide and distorted peak of NH<sub>4</sub><sup>+</sup>. It should be noted that TMAH<sup>+</sup> and DEAH<sup>+</sup> could not be completely separated using the IC system (VandenBoer et al., 2011, 2012; Zhou et al., 2018; Huang et al., 2014). Therefore, the sum of TMAH<sup>+</sup> and DEAH<sup>+</sup> concentrations (TMDEAH<sup>+</sup>) were quantified using the calibration curve of TMAH<sup>+</sup> with errors less than 3% (Zhou et al., 2018). With the sampling volumes of 144 and 86 m<sup>3</sup> for PM<sub>2.5</sub> and MOUDI samples, respectively, the detection limits of DMAH<sup>+</sup>, TMDEAH<sup>+</sup>, TEAH<sup>+</sup>, MPAH<sup>+</sup>, MEOAH<sup>+</sup> and TEOAH<sup>+</sup> were 0.55, 0.78, 1.93, 2.59, 1.94 and 4.96 ng m<sup>-3</sup> for PM<sub>2.5</sub> samples and 0.20, 0.29, 0.71, 0.95, 0.56 and 1.82 ng m<sup>-3</sup> for samples collected in each MOUDI stage. MPAH<sup>+</sup>, MEOAH<sup>+</sup> and TEOAH<sup>+</sup> were rarely detected in the aerosol samples (< 10%) and thereby not reported in this study. Detailed information about analysis of aminiums is given in Zhou et al. (2018).

One-fourth of the PM<sub>2.5</sub> cellulose sample filter was cut and digested with 7 mL of HNO<sub>3</sub> and 1 mL of HF (both

acids were purified from GR using a sub-boiling system) at 185 °C for 30 min in a microwave digestion system (MARS5 Xpress, CEM). Inductively coupled plasma optical emission spectroscopy (ICP-OES, SPECTRO) was used for determining the following elements: Al, Ca, Fe, Na, P, S, Cu, K, Mg, Mn, Zn, As, Ba, Cd, Ce, Co, Cr, Mo, Ni, Pb, Ti and V. The detailed procedures refer to Wang et al. (2016).

### 2.3 Auxiliary data

The 3 h resolution meteorological data from Baoshan station in Shanghai (WMO index: 58362) were obtained from the National Climatic Data Center (NCDC, <https://www.ncdc.noaa.gov/isd>, last access: 10 October 2018). The 10 s resolution meteorological data were recorded by a shipborne meteorological station during the cruise. The planetary boundary layer height (BLH) and 6 h accumulated precipitation (TPP6) were extracted from NCEP's Global Data Assimilation System Data (GDAS). The daily concentrations of gaseous pollutants (SO<sub>2</sub>, CO, NO<sub>2</sub> and O<sub>3</sub>) averaged from nine real-time monitoring stations in Shanghai were obtained from the Shanghai Environmental Monitoring Center (<http://www.semc.gov.cn/aqi/home/DayData.aspx>, last access: 3 May 2017). The farthest station, Chuansha, is about 23.5 km away from Fudan site, and the daily concentrations of gaseous pollutants varied consistently in the nine stations.



The 3 d air mass backward trajectories were calculated using a Hybrid Single Particle Lagrangian Integrated Trajectory (HYSPLIT) model (<http://ready.arl.noaa.gov/HYSPLIT.php>, last access: 3 March 2018) with the starting height of 100 m.

### 3 Results and discussion

#### 3.1 Seasonal and spatial variations in aminium concentrations

The mean concentrations of  $\text{NH}_4^+$  and aminiums in each campaign of this study and reported in the literature were listed in Table 2. It should be noted that  $\text{TEAH}^+$  concentrations over Huaniao Island and the YECS were mostly below the detection limits ( $< \text{DL}$ ). For other aminiums and  $\text{TEAH}^+$  over Shanghai, the number of samples below detection limits was generally less than 30%. These undetectable concentrations were considered to be zero for the calculation of means and standard deviations. Three aminiums,  $\text{DMAH}^+$ ,  $\text{TMDEAH}^+$  and  $\text{TEAH}^+$ , were commonly detected in the aerosol samples collected from Shanghai. The most abundant aminiums were  $\text{DMAH}^+$  and  $\text{TEAH}^+$ , with annual means of 15.6 and 16.0  $\text{ng m}^{-3}$ , respectively. By comparison, the average  $\text{TMDEAH}^+$  concentration (4.4  $\text{ng m}^{-3}$ ) was significantly lower. All three aminiums showed the highest concentrations in winter and the lowest levels in spring ( $\text{DMAH}^+$ ) and summer ( $\text{TMDEAH}^+$  and  $\text{TEAH}^+$ ), which generally agreed with the seasonal trends of  $\text{PM}_{2.5}$  and  $\text{NH}_4^+$  concentrations in Shanghai (Fig. 2). Specifically, the average  $\text{TEAH}^+$  reached 35.2  $\text{ng m}^{-3}$  in winter in Shanghai, about 40 times as much as that in summer.  $\text{TEAH}^+$  was mostly below the detection limit in the aerosols collected over Huaniao Island and the YECS, suggesting its dominant land sources and negligible marine contribution. By contrast, the average  $\text{DMAH}^+$  and  $\text{TMDEAH}^+$  concentrations (14.0 and 13.2  $\text{ng m}^{-3}$ ) over Huaniao Island were close to and significantly higher than those over Shanghai, respectively. Similarly high concentrations of  $\text{DMAH}^+$  and  $\text{TMDEAH}^+$  (11.9 and 14.6  $\text{ng m}^{-3}$ ) were also observed over the YECS (Fig. 2 and Table 2), suggesting that the two aminiums might have notable marine sources. Accordingly, both species reached the highest levels during the summer campaigns in 2017 on Huaniao Island, consistent with the highest primary productivity in the coastal ECS and prevailing winds from the ocean in summer. As a major component of fine particles over eastern China with similar chemical properties to aminiums,  $\text{NH}_4^+$  was mainly from terrestrial sources and its concentrations over Huaniao Island and YECS were much lower than those over Shanghai (Fig. 2).

Our measurement of  $\text{DMAH}^+$  in Shanghai was comparable to those previously reported from the urban sites but generally higher than those measured in the forest areas of Toronto (VandenBoer et al., 2012), Hyytiälä (Hemmilä et al.,

2018) and Guangdong (Liu et al., 2018a). This implies that anthropogenic activities may be crucial sources of  $\text{DMAH}^+$  in the urban atmosphere. The  $\text{TMDEAH}^+$  concentrations in our study were much lower than those reported by Tao et al. (2016) in Shanghai. Their sampling location was close to the residential areas and could be influenced by local sources, such as human excreta emissions (Zhou et al., 2018). The aerosol  $\text{TEAH}^+$  concentrations in China were reported in our study for the first time and could not be compared to previous work. According to previous measurement results for gaseous amines in the same site from 25 July to 25 August in 2015, the average mass concentrations of C2, C3 plus C4 and C6 amines were 80.4, 53.1 and 15.8  $\text{ng m}^{-3}$ , respectively (Yao et al., 2016). The order of concentrations was consistent with that of the corresponding aerosol aminiums in the summer of 2013, which was  $\text{DMAH}^+ > \text{TMDEAH}^+ > \text{TEAH}^+$  (9.1, 1.7 and 0.9  $\text{ng m}^{-3}$ , respectively) in this study. Based on these concentrations, the ratios of each amine vs. aminium were roughly calculated and C2 amines /  $\text{DMAH}^+$ , (C3 plus C4 amines) /  $\text{TMDEAH}^+$  and C6 amines /  $\text{TEAH}^+$  were 8.8, 30.1 and 17.9, respectively. These values were comparable to the ratio of total amines to total aminiums (14.9) over a mountain site in southern China (Liu et al., 2018a). Except for the three aminiums commonly detected in this study,  $\text{MMAH}^+$  and  $\text{MEAH}^+$  (Liu et al., 2018a; Ho et al., 2015; Shen et al., 2017) were other abundant aminiums detected in the urban site.

Aerosols were collected using a MOUDI over Huaniao Island and the YECS. Aminiums in  $\text{PM}_{1.8}$  of the MOUDI samples were compared to those of  $\text{PM}_{2.5}$ , since MOUDI does not have the 50% cutoff diameter of 2.5  $\mu\text{m}$  and aminiums in  $\text{PM}_{1.8}$  accounted for over 60% of the concentrations over the whole size range of aerosols. Our measurements of aminiums over Huaniao Island and the YECS were comparable to those previously observed over the seas off the eastern coast of China (Hu et al., 2015; Yu et al., 2016; Xie et al., 2018), but they were apparently higher than many other oceanic regions, such as the Arabian Sea (Gibb et al., 1999) and Cabo Verde (Müller et al., 2009). The high aminiums over the YECS were probably associated with the severe air pollution in eastern China and the high ocean productivity in nearby marginal seas.

#### 3.2 Environmental factors affecting aminium concentrations

##### 3.2.1 Boundary layer height (BLH)

The concentrations of  $\text{PM}_{2.5}$ ,  $\text{NH}_4^+$  and three aminiums sampled in Shanghai in 2013 dropped significantly when the BLH increased from 200 to 500 m and then slowly decreased with the further increase in BLH (Figs. 3a and S1), due to the enhanced ventilation. Specifically, the concentrations of  $\text{DMAH}^+$ ,  $\text{TMDEAH}^+$  and  $\text{TEAH}^+$  (58.4, 13.9 and 80.5  $\text{ng m}^{-3}$ ) in Shanghai reached its maximum, along

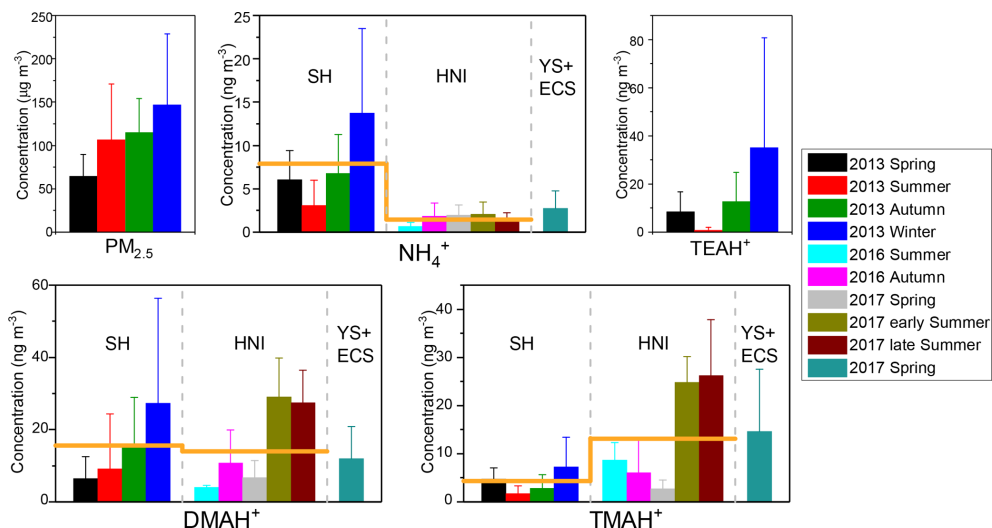
**Table 2.** The mass concentrations (mean values  $\pm 1$  standard deviation) of  $\text{NH}_4^+$  and aminiums over Shanghai, Huaniao Island and the YECs compared to other sites reported in the literature. The values below the detection limits are indicated by < DL.

No.	Site	Site type	Sampling period	Particle size	Aminium ( $\text{ng m}^{-3}$ )							Reference
					$\text{NH}_4^+$ ( $\mu\text{g m}^{-3}$ )	MMAH <sup>+</sup>	DMAH <sup>+</sup>	TMDEAH <sup>+</sup>	MEAH <sup>+</sup>	TEAH <sup>+</sup>		
1	Shanghai, China	urban	Spring (March–April 2013)	PM <sub>2.5</sub>	6.0 $\pm$ 3.4		6.4 $\pm$ 6.1	4.8 $\pm$ 2.3			8.4 $\pm$ 8.4	This study
2			Summer (July–August 2013)	PM <sub>2.5</sub>	3.1 $\pm$ 2.9		9.1 $\pm$ 15.2	1.7 $\pm$ 1.6			0.9 $\pm$ 1.0	
3			Autumn (November 2013)	PM <sub>2.5</sub>	6.8 $\pm$ 4.5		15.5 $\pm$ 13.4	2.8 $\pm$ 2.9			12.7 $\pm$ 12.2	
4			Winter (December 2013–January 2014)	PM <sub>2.5</sub>	13.7 $\pm$ 9.8		27.3 $\pm$ 29.0	7.3 $\pm$ 6.2			35.2 $\pm$ 45.6	
5	Shanghai, China	urban	July–August 2013	PM <sub>1.8</sub>	2.5 $\pm$ 1.3	8.9 $\pm$ 6.1	15.7 $\pm$ 7.9	38.8 $\pm$ 17.0	11.5 $\pm$ 17.4			Tao et al. (2016)
6				PM <sub>10</sub>	2.6 $\pm$ 1.3	9.9 $\pm$ 6.9	20.1 $\pm$ 10.7	47.0 $\pm$ 19.9	15.7 $\pm$ 26.4			
7	Shanghai, China	urban	January 2013	PM <sub>2.5</sub>			2.4			0.2		Huang et al. (2016)
8			July–August 2013	PM <sub>2.5</sub>			3.9			0.3		
9	Yangzhou, China	urban	November 2015–April 2016	PM <sub>2.5</sub>		4.9 $\pm$ 1.9	4.3 $\pm$ 2.4			15.4 $\pm$ 8.1		Shen et al. (2017)
10	Nanjing, China	urban	April–May 2016	PM <sub>2.5</sub>		7.6	4.2			21.7		
11			August 2014	PM <sub>1.8</sub>		7.2 $\pm$ 4.1	18.0 $\pm$ 11.7			36.4 $\pm$ 18.6		
12	Xi'an, China	urban	July 2008–August 2009	PM <sub>2.5</sub>		14.4 $\pm$ 9.6				3.3 $\pm$ 2.4		Ho et al. (2015)
13	Guangzhou, China	urban	September–October 2014	PM <sub>0.95</sub>	4.3 $\pm$ 1.1	41.8 $\pm$ 11.4	14.5 $\pm$ 3.2	3.7 $\pm$ 0.9	3.2 $\pm$ 0.4			Liu et al. (2017)
14				PM <sub>3</sub>	5.1 $\pm$ 1.4	50.4 $\pm$ 13.7	17.7 $\pm$ 3.6	4.8 $\pm$ 1.4	4.0 $\pm$ 0.5			
15				PM <sub>10</sub>	5.2 $\pm$ 1.4	51.8 $\pm$ 13.9	19.0 $\pm$ 3.8	5.4 $\pm$ 1.6	4.2 $\pm$ 0.6			
16	Tampa Bay, Florida, USA	urban	July–September 2005	PM <sub>2.5</sub>	1.4 $\pm$ 1.2		31.6 $\pm$ 28.3					Calderón et al. (2007)
17	A traffic site, Milan, Italy	urban	October 2013	TSP	4.2 $\pm$ 2.9		90 $\pm$ 20				360 $\pm$ 20	Perrone et al. (2016)
18	A limited traffic site, Milan, Italy	urban	October 2013	TSP	4.0 $\pm$ 3.0		100 $\pm$ 10				420 $\pm$ 100	
19	Qingdao, China	semi-urban	May 2013, November–December 2013, November–December 2015	PM <sub>0.056–10</sub>			6.3					Xie et al. (2018)
20	Resort beach site in Qingdao, China	coastal, rural	August 2016	PM <sub>0.056–10</sub>			28.5 $\pm$ 23.0	9.0 $\pm$ 6.6				
21	Egbert, Toronto, Canada	agricultural and semi-forested	October 2010	PM <sub>2.5</sub>			0.1 $\pm$ 0.2	1 $\pm$ 0.6				VandenBoer et al. (2012)
22	Hyytiälä, southern Finland	boreal forest	March 2015	PM <sub>10</sub>	0.4 $\pm$ 0.1		6.8	1.5			1.1	Hemmiä et al. (2018)
23			April 2015	PM <sub>10</sub>	0.1 $\pm$ 0.1		2.9	3.1			0.7	
24			July 2015	PM <sub>10</sub>	0.1 $\pm$ 0.1		3.0	8.4 $\pm$ 4.9			1.8 $\pm$ 1.4	0.4
25	Nanling, Guangdong, China	forest	October 2016	PM <sub>2.5</sub>	0.9 $\pm$ 0.6	8.8 $\pm$ 7.8	2.4 $\pm$ 3.2	1.1 $\pm$ 1.8				Liu et al. (2018a)
26			May–June 2017		1.8 $\pm$ 1.6	11.9 $\pm$ 9.8	5.0 $\pm$ 2.2	1.7 $\pm$ 1.7				

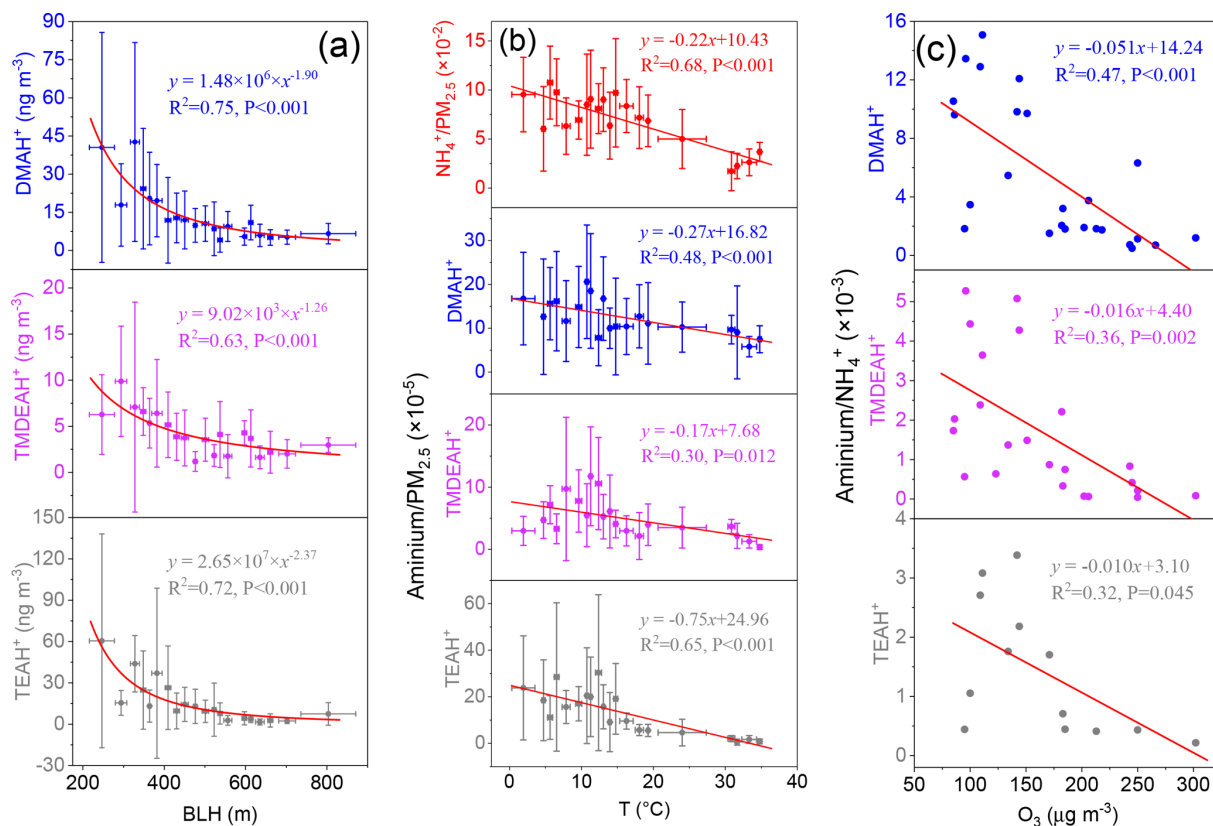
Table 2. Continued.

No.	Site	Site type	Sampling period	Particle size	NH <sub>4</sub> <sup>+</sup> (μg m <sup>-3</sup> )	Aminium (ng m <sup>-3</sup> )					Reference
						MMAH <sup>+</sup>	DMAH <sup>+</sup>	TMDEAH <sup>+</sup>	MEA <sup>+</sup>	TEAH <sup>+</sup>	
27	Brent, Alabama, USA	forest	1 June–15 July 2013	submicron	0.52	148*					You et al. (2014)
28	Huaniao Island, China	marine	August 2016	PM <sub>2.5</sub>	0.7 ± 0.4		4.0 ± 0.6	8.7 ± 3.7		< DL	This study
29			November–December 2016	PM <sub>1.8</sub>	1.9 ± 1.5		10.7 ± 9.3	6.0 ± 6.8		< DL	
30				PM <sub>10</sub>	2.1 ± 1.8		15.1 ± 12.4	8.4 ± 8.8		< DL	
31			March 2017	PM <sub>1.8</sub>	2.0 ± 1.2		6.8 ± 4.6	2.7 ± 1.8		< DL	
32				PM <sub>10</sub>	2.3 ± 1.4		11.4 ± 11.6	3.1 ± 2.2		< DL	
33			June–July 2017	PM <sub>1.8</sub>	2.1 ± 1.4		29.0 ± 10.8	24.8 ± 5.4		< DL	
34				PM <sub>10</sub>	2.2 ± 1.6		32.2 ± 11.0	27.5 ± 5.7		< DL	
35			August–September 2017	PM <sub>1.8</sub>	1.4 ± 0.7		25.8 ± 8.7	25.0 ± 11.0		< DL	
36	Yellow Sea and East China Sea	marine	March–April 2017	PM <sub>10</sub>	1.5 ± 0.8		27.4 ± 9.1	26.3 ± 11.6		< DL	
37				PM <sub>1.8</sub>	2.8 ± 2.0		11.9 ± 9.0	14.6 ± 12.9		< DL	
38				PM <sub>10</sub>	3.0 ± 2.2		13.5 ± 10.1	16.6 ± 14.5		< DL	
39	Yellow Sea and the northwestern Pacific	marine	April 2015	PM <sub>0.056–10</sub>			12.9 ± 10.6	13.2 ± 13.8			Xie et al. (2018)
40	East China Sea	marine	January 2016	PM <sub>0.056–10</sub>			30.8 ± 9.7	12.0 ± 6.6			
41	Yellow Sea and Bohai Sea	marine	August 2015, June–July 2016	PM <sub>0.056–10</sub>			33.3	19.4			
42	Southern Yellow Sea	marine	November 2013	PM <sub>0.056–10</sub>			18.9 ± 16.6	31.8 ± 19.2			
43	Yellow Sea and Bohai Sea	marine	May 2012	PM <sub>11</sub>			202 ± 170	432 ± 426			Htu et al. (2015)
44	Southern Yellow Sea	marine	November 2012	PM <sub>10</sub>			13.3 ± 4.6	30.0 ± 12.6			Yu et al. (2016)
45	Northern Yellow Sea and Bohai Sea	marine	November 2012	PM <sub>10</sub>			–	15.0 ± 6.6			
46	Arabian Sea	marine	August–October 1994	PM <sub>0.9</sub>	0.04	3.2	2.1	0.3			Gibb et al. (1999)
47			November–December 1994	PM <sub>0.9</sub>	0.1	3.7	11.1	0.5			
48	Mace Head, Ireland	marine	January–December 2006	PM <sub>1</sub>			4.7 ± 6.0	7.6 ± 9.4			Facchini et al. (2008)
49	Irish West Coast	marine	June–July 2006	PM <sub>1</sub>			14.7 ± 14.3	14.3 ± 8.7			
50	São Vicente, Cabo Verde	marine	May–June, December 2007	PM <sub>0.14–0.42</sub>	0.1	0.1	0.4	0.2			Müller et al. (2009)
51	Off the Central Coast of California, USA	marine	July 2007	PM <sub>1</sub>			22				Sorooshian et al. (2009)
52	Eastern Mediterranean, Greece	marine	2005–2006	PM <sub>1</sub>			9.2 ± 36.8	< DL			Violaki and Mihalopoulos (2010)

\* Fourier transform infrared spectroscopy (FTIR) measured total primary aminiums (R-NH<sub>3</sub><sup>+</sup>).



**Figure 2.** The mass concentrations of  $\text{PM}_{2.5}$ , fine-particle  $\text{NH}_4^+$ , and three aminiums ( $\text{TEAH}^+$ ,  $\text{DMAH}^+$  and  $\text{TMDEAH}^+$ ) in different campaigns in Shanghai (SH), Huaniao Island (HNI) and the Yellow and East China seas (YECS). The columns and error bars represent average concentrations and standard deviations, respectively. The orange horizontal lines represent the annual average concentrations of aminiums in SH and HNI.



**Figure 3.** (a) Relationships between concentrations of aminiums and boundary layer height (BLH) over Shanghai in 2013. (b) Relationships between mass ratios of aminiums and  $\text{NH}_4^+$  to  $\text{PM}_{2.5}$  and temperature over Shanghai in 2013. (c) Relationships between mass ratios of aminiums to  $\text{NH}_4^+$  and  $\text{O}_3$  concentrations over Shanghai during the summer of 2013.

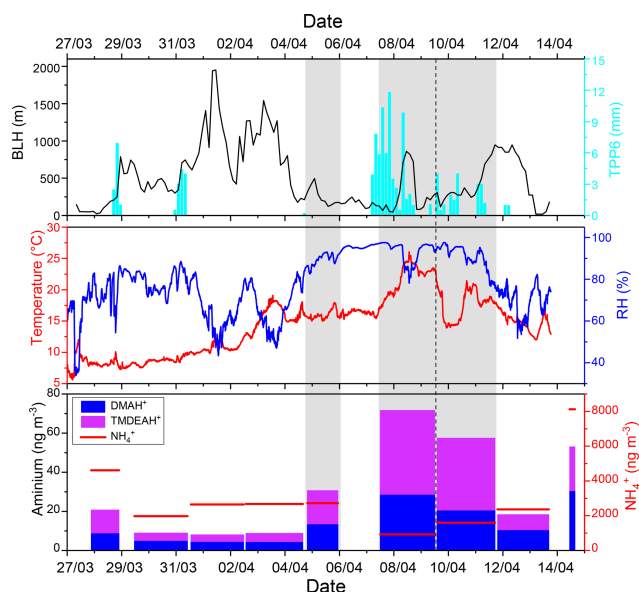
with  $\text{PM}_{2.5}$  ( $447 \mu\text{g m}^{-3}$ ), during the severe haze event between 30 November and 8 December 2013, when the average BLH and wind speed were  $298 \text{ m}$  and  $1.35 \text{ m s}^{-1}$ , respectively (Fig. S2 in the Supplement). By comparison, the average concentrations of  $\text{DMAH}^+$ ,  $\text{TMDEAH}^+$  and  $\text{TEAH}^+$  ( $8.9$ ,  $4.0$  and  $10.1 \text{ ng m}^{-3}$ ) were much lower prior to the haze event (on 26–29 November 2018) associated with the higher BLH ( $636.4 \text{ m}$ ) and wind speed ( $2.73 \text{ m s}^{-1}$ ). Thus, the generally stagnant meteorology in winter (Liu et al., 2013) could cause a substantial accumulation of aerosol aminiums and lead to the seasonal variation in aminiums in Shanghai.

### 3.2.2 Temperature

To eliminate the synchronous change of aminiums and  $\text{NH}_4^+$  with  $\text{PM}_{2.5}$ , the mass ratios of aminiums to  $\text{PM}_{2.5}$  (aminiums/ $\text{PM}_{2.5}$ ) and  $\text{NH}_4^+$  to  $\text{PM}_{2.5}$  ( $\text{NH}_4^+/\text{PM}_{2.5}$ ) were applied for analysis. These ratios were found to be negatively correlated with air temperature in Shanghai (Fig. 3b). Similar to  $\text{NH}_4^+$ , aminiums combined with  $\text{NO}_3^-$ ,  $\text{Cl}^-$  and organic acids are semi-volatile and can dissociate in the atmosphere (Tao and Murphy, 2018). Thus, the negative correlations may be explained by the movement of gas-particle partitioning equilibrium to the gas phase at higher temperatures (Ge et al., 2011a). This is consistent with the previous observation that the proportion of particles containing aminiums in the urban area of Shanghai was much higher in winter (23.4 %) than in summer (4.4 %) (Huang et al., 2012). The seasonal variation in temperature may also lead to the change of concentrations of aerosol aminiums. It should be pointed out that environmental variables like BLH and temperature are constantly changing with time and their impacts on aminium concentrations may vary within the sampling duration (24 or 48 h). However, these variables must be averaged over the same time interval as aminium concentrations. This analysis may eliminate the instant discordance and improve the correlations between environmental variables and aminiums or aminiums/ $\text{PM}_{2.5}$ , and the results could explain the seasonal variation in aminiums well.

### 3.2.3 Oxidizing capacity

As gaseous amines can be oxidized by oxidants such as  $\cdot\text{OH}$ ,  $\text{O}_3$  and  $\text{NO}_3\cdot$  in the atmosphere before partitioning into the particulate phase (Ge et al., 2011b; Nielsen et al., 2012; Yu and Luo, 2014), aminium concentrations in aerosols may decrease with enhanced atmospheric oxidizing capacity. Ozone concentration can represent oxidizing capacity of the lower atmosphere (Thompson, 1992). Here the relationship between aminium/ $\text{NH}_4^+$  ratios and  $\text{O}_3$  was examined because the formation of particulate aminiums and  $\text{NH}_4^+$  were both temperature-dependent and using their ratios could avoid the temperature effect to some extent. Besides, the residence time of  $\text{NH}_3$  in the atmosphere due to the oxidation reaction is about 72.3 d (Ge et al., 2011b), and therefore  $\text{NH}_4^+$  con-



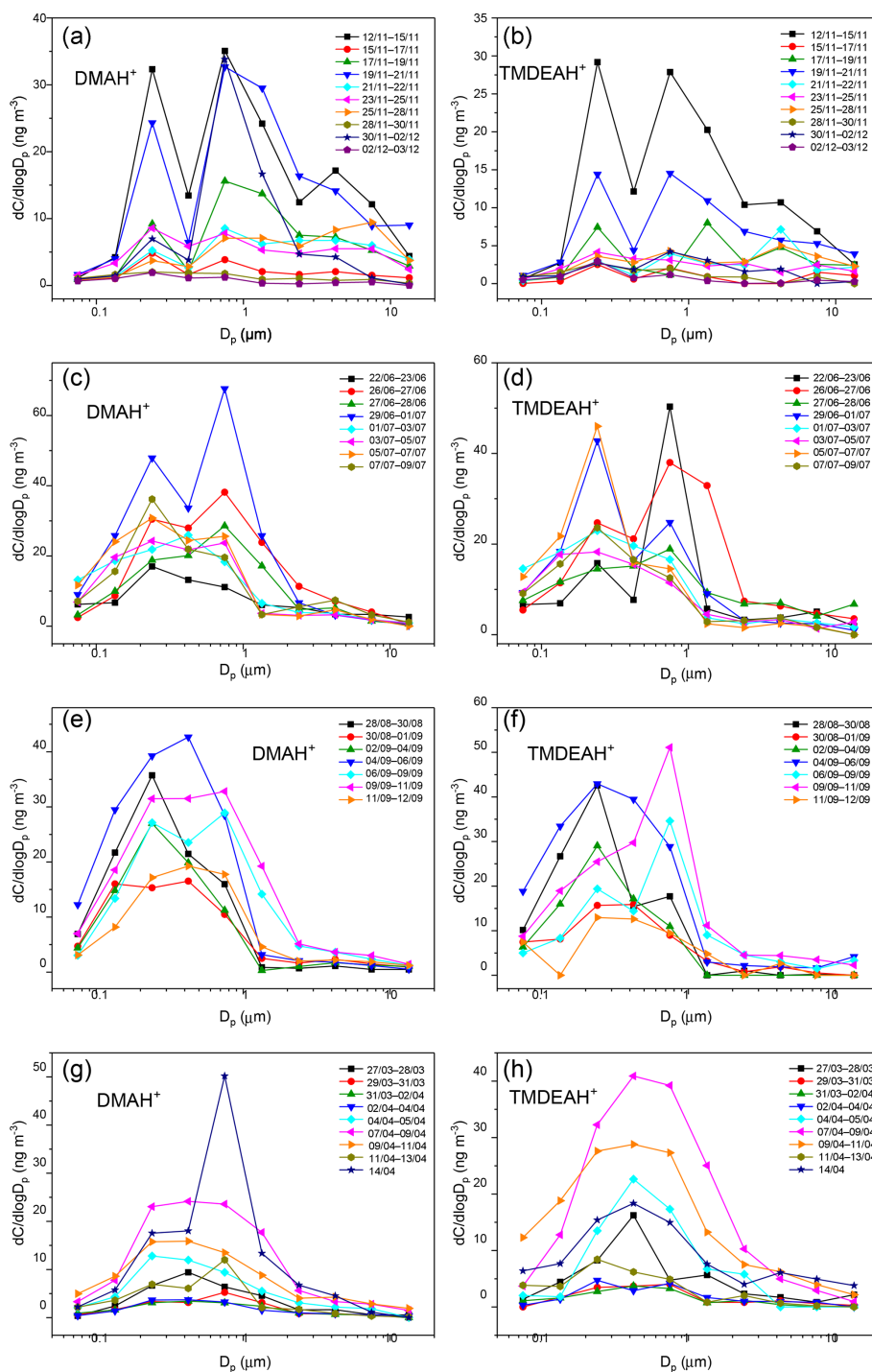
**Figure 4.** Time series of meteorological parameters and the concentrations of aminiums and  $\text{NH}_4^+$  during the 2017 spring cruise. The time range spanned by the column of each aminium concentration corresponds to the sampling time.

centrations in aerosols should not be affected by  $\text{O}_3$ . Negative correlations were found between aminiums/ $\text{NH}_4^+$  and  $\text{O}_3$  concentrations in Shanghai during the summer of 2013 (Fig. 3c). In other seasons, the correlations were not obvious, especially in winter when  $\text{O}_3$  concentrations were the lowest and neither of the aminiums/ $\text{NH}_4^+$  were correlated with  $\text{O}_3$  (Fig. S3). In general, atmospheric oxidizing capacity is the strongest in summer (Logan, 1985; Liu et al., 2010), and the results verified that high oxidizing capacity in summer may reduce the formation of particulate aminiums by oxidizing gaseous amines. It was consistent with the diurnal pattern of gaseous amines with the lowest values at noon and the negative correlations between the concentrations of amines and  $\text{O}_3$  observed in Shanghai during the summer of 2015 (Yao et al., 2016). It should be noted that there was no significant variation in temperature and little rainfall during the sampling periods in the summer of 2013. In other seasons, due to the relatively weak photochemistry and more complex sources and meteorology, other factors except oxidizing capacity played more important roles in affecting aerosol aminiums.

### 3.2.4 Relative humidity and fog processing

In the spring of 2017 over the YECS, although the sample of 4–5 April was influenced by high Chl *a* concentrations and low BLH, the concentrations of  $\text{DMAH}^+$  and  $\text{TMDEAH}^+$  ( $13.3$  and  $17.4 \text{ ng m}^{-3}$ ) were about half of those on 7–9 April (Fig. 4). This was probably due to the intense fog event that occurred on 7–9 April with relative humidity  $> 90 \%$ , which





**Figure 5.** Size distributions of aminiums during different campaigns: (a–b) in the autumn of 2016 on Huaniao Island, (c–d) in early summer 2017 on Huaniao Island, (e–f) in late summer 2017 on Huaniao Island, and (g–h) during the 2017 spring cruise over the Yellow and East China seas.

could enhance the gas-to-particle partitioning of amines. The enhancement of TMA gas to particles by cloud and fog processing has been observed in both field and laboratory simulations (Rehbein et al., 2011). It was also found that the number fraction of TMA-containing particles dramatically

increased from  $\sim 7\%$  on clear days to  $\sim 35\%$  on foggy days and that number-based size distribution of TMA-containing particles shifted towards larger mode, peaking at the droplet mode ( $0.5\text{--}1.2\ \mu\text{m}$ ) in Guangzhou (Zhang et al., 2012). The investigation over the Yellow and Bohai seas in the summer

of 2015 found significantly positive correlations between the concentrations of DMAH<sup>+</sup> and TMAH<sup>+</sup> and relative humidity (Yu et al., 2016). Therefore, fog and high relative humidity (RH) are also favorable conditions for gas-to-particle conversion of amines.

### 3.3 Size distributions and formation pathways of aerosol aminiums

The aminiums were mainly distributed in fine aerosols with diameter less than 1.8 μm, and the mass percentages of DMAH<sup>+</sup> and TMDEAH<sup>+</sup> in the coarse mode were around 36 % in the autumn of 2016 on Huaniao Island and less than 15 % in all other campaigns on Huaniao Island and over the YECS (Fig. 5a–d). This is consistent with the previous reports that > 70 % of aminiums were distributed in fine particles over Shanghai during the summer of 2013 (Tao et al., 2016) and over the western North Pacific and its marginal seas (Xie et al., 2018). The aminiums mostly demonstrated a bimodal distribution in the autumn and early summer campaigns on Huaniao Island with peaks at 0.18–0.32 μm (condensation mode) and 0.56–1.0 μm (droplet mode). This is similar to the size distributions of DMAH<sup>+</sup> and TMDEAH<sup>+</sup> observed in Shanghai (Tao et al., 2016) and to NH<sub>4</sub><sup>+</sup> and non-sea salt (nss-SO<sub>4</sub><sup>2-</sup>) in all campaigns over Huaniao Island and the YECS (Figs. S4–S5). The size distribution suggests that the gas-to-particle condensation (condensation mode) and cloud processing (droplet mode) seem to be major mechanisms for the formation of aminiums and other secondary species NH<sub>4</sub><sup>+</sup> and nss-SO<sub>4</sub><sup>2-</sup>.

In order to compare the contributions between condensation and cloud processing to the formation of specific species, the ratio of its concentrations in droplet mode (0.56–1.0 μm) to those in condensation mode (0.18–0.32 μm) was calculated (denoted as  $\alpha$ ). It could be seen that the  $\alpha$  values of NH<sub>4</sub><sup>+</sup> and nss-SO<sub>4</sub><sup>2-</sup> were significantly greater than 1, especially in the case of high concentrations, indicating that the cloud processing probably determined the concentrations of these species (Fig. 6). In contrast, aminiums had  $\alpha$  values around 1, suggesting that condensation and cloud processing might be equally important to the formation of aminiums.

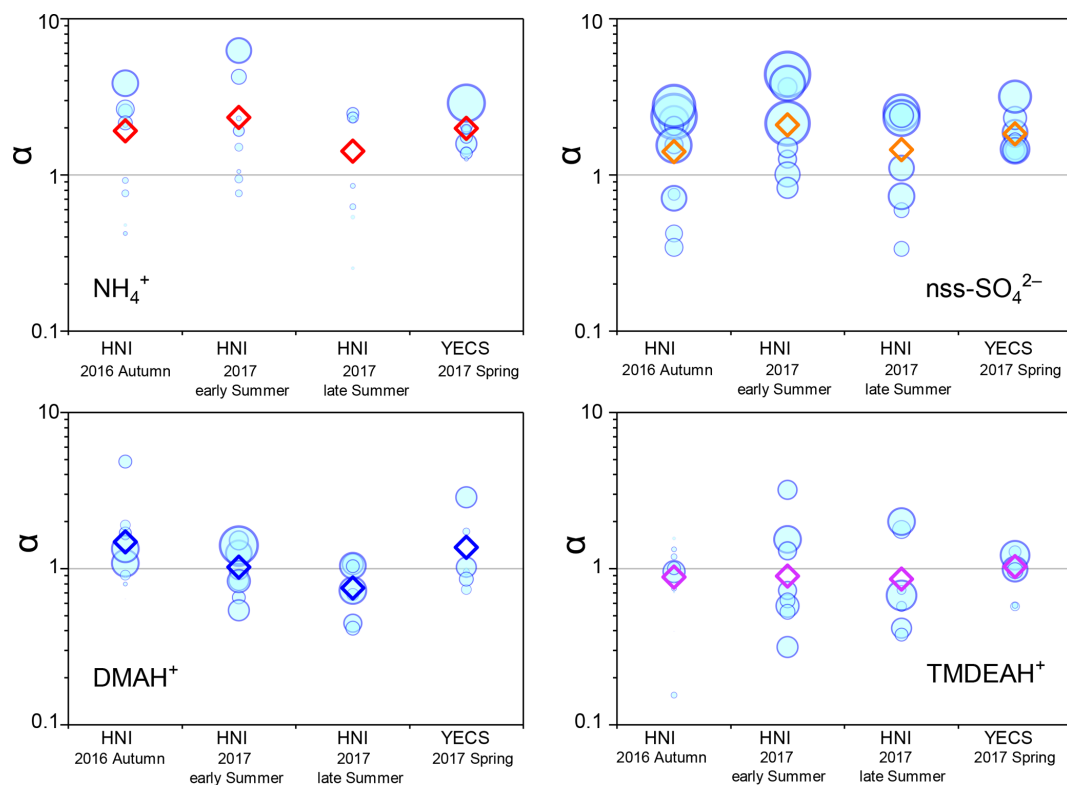
In late summer on Huaniao Island and the spring cruise over the YECS, when air masses were mainly from oceanic regions (see Sect. 3.4.3), the aminiums generally exhibited a unimodal distribution with one wide peak at 0.18–1.0 μm due to the increased concentrations at 0.32–0.56 μm (Fig. 5e–h). The concentrations of NH<sub>4</sub><sup>+</sup> and nss-SO<sub>4</sub><sup>2-</sup> also showed a significant elevation in the size range of 0.32–0.56 μm during these periods. The deviation of MOUDI cutoff diameters during the sampling could be ruled out because the concentrations of particulate matter always presented a trimodal distribution with peaks at 0.18–0.32, 0.56–1.8 and 3.2–10 μm. The unimodal distributions of aminiums with the peak at 0.18–1.0 μm have been widely reported over the seas off the eastern coast of China (Hu et al., 2015; Yu et al., 2016;

Xie et al., 2018). This suggests that the formation mechanisms of aerosol aminiums over the ocean may be different from those over the land. It was indicated that the high concentration and unique size distribution of TMAH<sup>+</sup> observed over the oligotrophic western North Pacific were mainly attributed to the primary TMAH<sup>+</sup> in sea-spray aerosols (Hu et al., 2018). In addition, some studies have demonstrated that artificially generated sea spray aerosols and actual primary marine aerosol both contained amines or aminiums (Bates et al., 2012; Frossard et al., 2014; Dall'Osto et al., 2019). Thus, we speculate that the elevated concentrations of aminiums at 0.32–0.56 μm over the seas off the eastern coast China may be also associated with the increased concentration of sea-spray aerosols that contain substantial primary aminiums. On the other hand, the heterogeneous formation of secondary aminiums on the surface of sea spray aerosols cannot be ruled out (Yu et al., 2016).

### 3.4 Sources of aerosol aminiums

#### 3.4.1 Anthropogenic sources on the land

Correlation analysis was carried out between aminiums, other PM<sub>2.5</sub> components and gaseous pollutants measured in Shanghai (Fig. 7). It can be seen that the secondary inorganic components SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup> (SNA), PM<sub>2.5</sub>, and DMAH<sup>+</sup> were significantly correlated with each other with the correlation coefficients above 0.6. This suggests that anthropogenic sources may have a great contribution to the atmospheric DMA in Shanghai, which is consistent with previous findings in Nanjing (Zheng et al., 2015). Considering the unique role of DMA in new particle formation (Almeida et al., 2013), our results reinforce that the frequent new particle formation events observed in extremely polluted Chinese cities are indeed, at least in part, due to amines (Yao et al., 2018). The correlations between TEAH<sup>+</sup> and SNA were relatively weak, but TEAH<sup>+</sup> was found to be significantly correlated with the components mainly from industrial sources (represented by the high concentrations of K, Mn, Cd, Pb, Zn and Cl<sup>-</sup>) (Tian et al., 2015; Liu et al., 2018b), indicating that industrial emissions could be an important source of triethylamine. It was consistent with the observation result in a suburban site, where gaseous C4 to C6 amines had some abrupt and frequent increases during the night, and may be caused by some local emissions (You et al., 2014). Compared to the DMAH<sup>+</sup> and TEAH<sup>+</sup>, TMDEAH<sup>+</sup> showed much weaker correlation with the anthropogenically derived components. Weak correlations were also found between all the aminiums and V, Ni, Al, Mg, Ca and Fe, suggesting that ship emissions (traced by V and Ni) and soil dust (represented by Al, Ca and Fe) were not the main sources of aminiums in PM<sub>2.5</sub> over Shanghai.



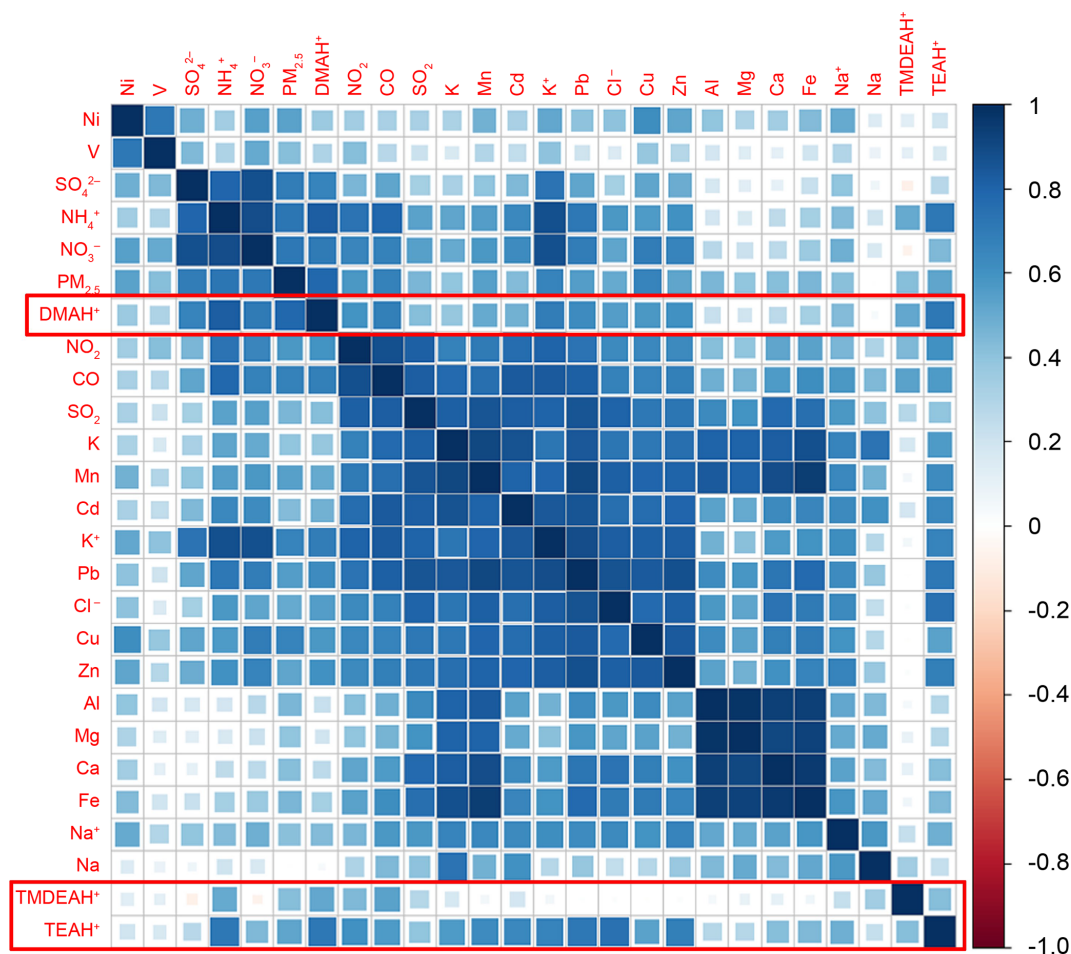
**Figure 6.** The  $\alpha$  values of  $\text{NH}_4^+$ ,  $\text{nss-SO}_4^{2-}$  and aminiums in different campaigns. The diameter of the circle is proportional to the concentration and the diamond-shaped symbol represents the average value of  $\alpha$  for each campaign. It should be noted that the bottom of column is the line of  $\alpha = 1$ .

### 3.4.2 Marine biogenic source

As discussed in Sect. 3.1, the relatively high concentrations of  $\text{DMAH}^+$  and  $\text{TMDEAH}^+$  over Huaniao Island and the YECS implied that the marine sources contributed substantially to these two aminiums. Accordingly, a spatial variation in aminium concentrations was observed over the YECS during the spring cruise. The concentrations of  $\text{DMAH}^+$  and  $\text{TMDEAH}^+$  increased by a factor of 3–5 in the southern ECS (average  $24.4$  and  $40.3 \text{ ng m}^{-3}$  for the samples of 7–11 April, respectively) compared to the YS and northern ECS (average  $7.0$  and  $8.4 \text{ ng m}^{-3}$  for the samples of 27 March–5 April, respectively) (Fig. 8). This is consistent with the noticeable difference of *Chl a* concentrations between the southern and northern YECS (2.3 times higher in the southern YECS than that in northern YECS, unpublished data). Furthermore, the highest  $\text{TMDEAH}^+$  and lowest  $\text{NH}_4^+$  concentrations observed on 7–11 April corresponded to the air mass back trajectories originating from the ocean, suggesting that the metabolic activities of surface plankton in highly productive seas could be a strong source of amines, as previously reported (Facchini et al., 2008; Müller et al., 2009; Sorooshian et al., 2009; Hu et al., 2015). In contrast, the high concentrations of aminiums observed on 14 April near Qingdao were affected by the air masses transported from eastern

China (Fig. 8) and thereby contributed to mainly by terrestrial sources.

Fine-mode  $\text{NH}_4\text{NO}_3$  could decompose during its transport from the land to the ocean, and the released  $\text{HNO}_3$  gas would react with dust and sea salt aerosols to form coarse-mode  $\text{NO}_3^-$ . Therefore, negative correlations were observed between the concentrations of fine-mode  $\text{NO}_3^-$  and alkaline species ( $\text{Na}^+ + \text{Ca}^{2+}$ ) over East Asia (Bian et al., 2014; Uno et al., 2017). Since only one dust event was encountered on 12–13 April during the cruise (unpublished data), the coarse-mode  $\text{NO}_3^-$  in this study should be mostly formed by heterogeneous reaction with sea salts. Therefore, the importance of terrestrial transport to marine aerosols could be roughly estimated by the percentage of  $\text{NO}_3^-$  in the fine mode. For aerosols collected on 29–31 March and 4–5, 7–9, and 9–11 April, over two-thirds of concentrations of  $\text{NO}_3^-$  were in the coarse mode ( $> 1.8 \mu\text{m}$ , Fig. 9a). These samples should be less affected by the terrestrial air masses (referred to hereafter as Category 1) compared to other samples (referred to hereafter as Category 2), and the analysis was consistent with the forward directions of air mass trajectories (Fig. S6). Aminiums were negatively correlated with  $\text{NH}_4^+$  for Category 1 samples, suggesting that aminiums were probably dominated by marine biogenic sources, whereas  $\text{NH}_4^+$  was influenced by terrestrial transport (Fig. 9b). For Category 2 samples, a pos-



**Figure 7.** Correlation coefficient matrix among the concentrations of PM<sub>2.5</sub> components and gaseous pollutants over Shanghai in 2013.

itive correlation was found between aminiums and  $\text{NH}_4^+$ , indicating that terrestrial sources could contribute significantly to aminiums over the YECS in these cases (Fig. 9c).

### 3.4.3 Source contributions to aminiums over the coastal sea

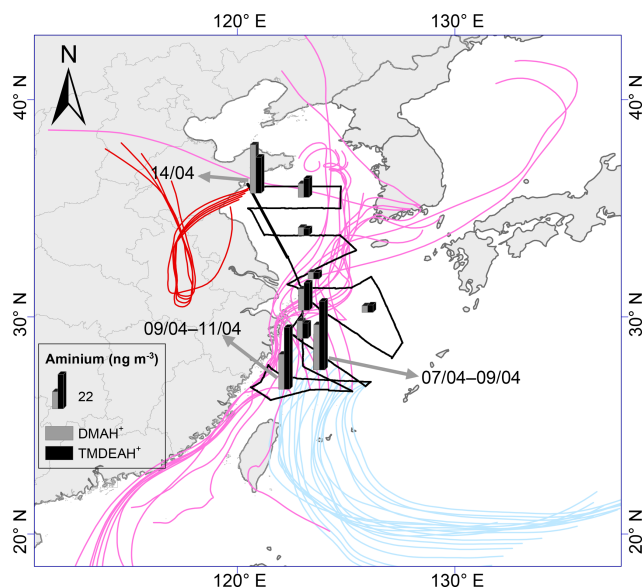
Huaniao Island is located on the front line of terrestrial transport to the ECS and influenced by the air masses from the land or ocean, depending on the seasonal variation in prevailing winds. Significantly positive correlations were found between the concentrations of aminiums and  $\text{NH}_4^+$  in the autumn but not in the summer of 2016 or in late summer 2017 (Fig. 10). Accordingly, the majority of backward trajectories pointed towards northern China in autumn, whereas air masses predominantly originate from the ECS in summer (Fig. 11). Meanwhile,  $\text{NO}_3^-$  demonstrated a trimodal distribution with three peaks at 0.18–0.32  $\mu\text{m}$  (condensation mode), 0.56–1.0  $\mu\text{m}$  (droplet mode) and 3.2–5.6  $\mu\text{m}$  (coarse mode) in autumn but only one peak at 3.2–5.6  $\mu\text{m}$  in late summer 2017 (Fig. S7). This implies that terrestrial transport could be a dominant source for aminiums over the coastal

ECS in autumn, whereas marine sources were dominant in late summer. In early summer 2017, the mass ratios of aminiums to  $\text{NH}_4^+$  were significantly lower on 26–28 June than those on other days (Fig. S8), corresponding to different origins and properties of the air masses. Removing the data measured on 26–28 June, we found a significantly positive correlation between the concentrations of DMAH<sup>+</sup> and  $\text{NH}_4^+$  but not between TMDEAH<sup>+</sup> and  $\text{NH}_4^+$ . This suggests that DMAH<sup>+</sup> and TMDEAH<sup>+</sup> may be predominantly derived from terrestrial and marine sources, respectively.

Good positive correlations were generally found between the concentrations of TMDEAH<sup>+</sup> and DMAH<sup>+</sup> over Huaniao Island and the YECS, and the slope for autumn samples dominated by terrestrial sources was significantly lower than those influenced primarily by marine air masses (e.g., late summer on Huaniao Island and spring over the YECS, Fig. 12). The highest slope of TMDEAH<sup>+</sup> vs. DMAH<sup>+</sup> (1.98) occurred in the summer of 2016, which was also mainly affected by marine sources. Therefore, it is speculated that aminiums derived from marine biogenic source might have significantly higher TMDEAH<sup>+</sup> to DMAH<sup>+</sup> ratios than

**Table 3.** Calculated terrestrial and marine source contributions to aerosol aminiums over Huaniao Island (mean; minimum–maximum).

Campaign	DMAH <sup>+</sup>		TMDEAH <sup>+</sup>	
	Terrestrial contribution (%)	Marine contribution (%)	Terrestrial contribution (%)	Marine contribution (%)
2016 - Autumn	74.1 (42.5–100)	25.9 (0–57.5)	69.1 (34.3–100)	30.9 (0–65.7)
2017 - Early summer	46.7 (20.3–98.8)	53.3 (1.2–79.7)	25.8 (11.0–48.6)	74.2 (51.4–89.0)
2017 - Late summer	37.0 (19.2–57.4)	63.0 (42.6–80.8)	21.7 (9.0–42.1)	78.3 (57.9–91.0)

**Figure 8.** The spatial distribution of aminiums over the YECS in the spring of 2017. The light blue, pink and red lines represent 72 h backward trajectories corresponding to sample sets collected on 7–9, 9–11 and 14 April, respectively.

those from terrestrial sources. Similarly, Hu et al. (2015) observed a significant correlation between the TMDEAH<sup>+</sup> and DMAH<sup>+</sup> concentrations over the Yellow Sea with a slope of 1.27–2.49. In early summer 2017, the weak correlation between the DMAH<sup>+</sup> and TMDEAH<sup>+</sup> and very low slope (0.29) suggested the mixing of terrestrial and marine influence on aminiums over Huaniao Island during that period as discussed above.

Dimethylsulfide (DMS) produced in seawater by the metabolism of plankton will be released into the atmosphere, and SO<sub>2</sub>, MSA, SO<sub>4</sub><sup>2-</sup> and other products can be formed through a series of oxidation reactions (Saltzman et al., 1985; Charlson et al., 1987; Faloona, 2009; Barnes et al., 2006). MSA is often used as a tracer of marine biogenic source to calculate the marine biogenic contribution to nss-SO<sub>4</sub><sup>2-</sup> (Yang et al., 2009, 2015). Therefore, the mass ratio of MSA to nss-SO<sub>4</sub><sup>2-</sup> (MSA/nss-SO<sub>4</sub><sup>2-</sup>) can be used to indicate the contribution of marine sources to relevant aerosol compo-

nents. A significant linear relationship was found between aminium/NH<sub>4</sub><sup>+</sup> and MSA/nss-SO<sub>4</sub><sup>2-</sup> for the samples collected in the autumn of 2016 and summer of 2017 over Huaniao Island (Fig. 13). The value of aminium/NH<sub>4</sub><sup>+</sup> increased with the increasing contribution of marine sources to the aerosol aminium. When the marine biogenic source contribution is 0, the corresponding aminium/NH<sub>4</sub><sup>+</sup> values (*b* in Eq. 1) represent the average ratios that are completely contributed by terrestrial sources. By multiplying the ratios by NH<sub>4</sub><sup>+</sup> concentrations, the aerosol aminiums contributed by terrestrial sources can be calculated (Eq. 2). Therefore, the contributions of terrestrial and marine sources to aerosol aminiums can be quantitatively estimated.

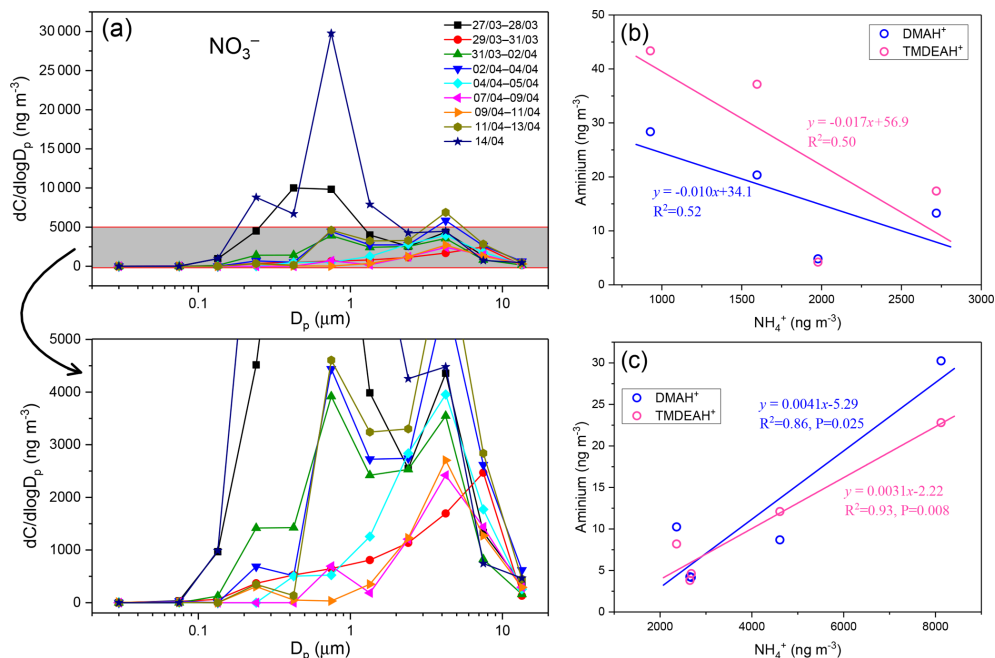
$$([\text{aminium}]/[\text{NH}_4^+])_{\text{terrestrial}} = k \times \left( [\text{MSA}]/[\text{nss-SO}_4^{2-}] \right)_{\text{terrestrial}} + b, \quad (1)$$

$$[\text{aminium}] = ([\text{aminium}]/[\text{NH}_4^+])_{\text{terrestrial}} \times [\text{NH}_4^+] + [\text{aminium}]_{\text{marine}}, \quad (2)$$

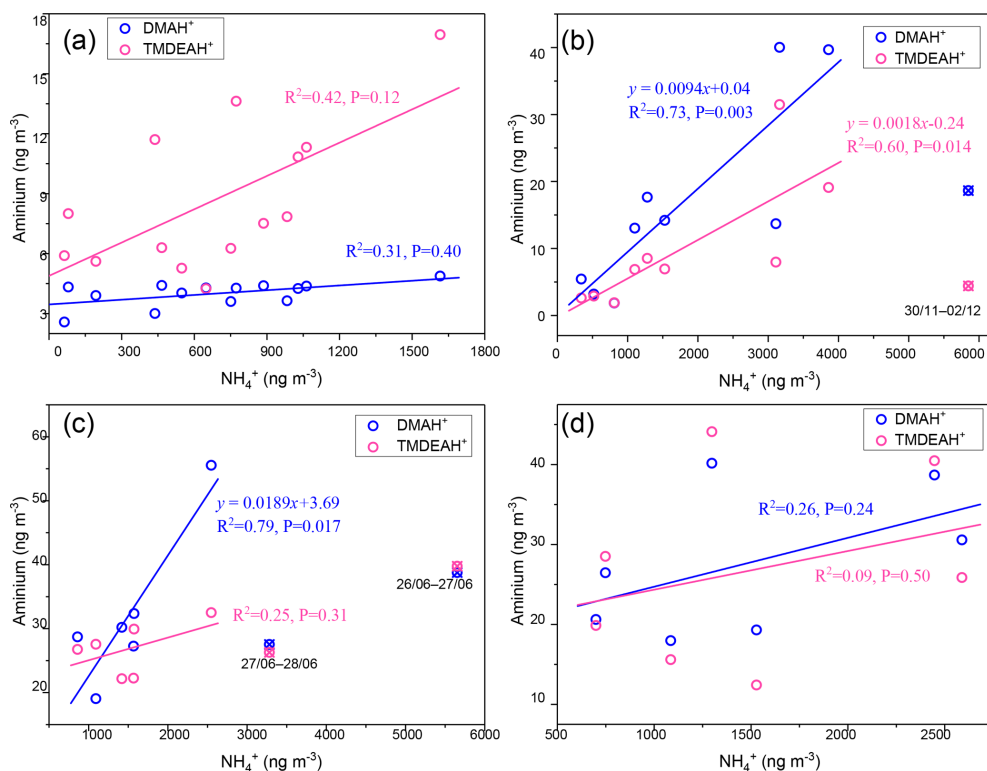
where *k* and *b* are the slope and intercept of the linear fitting equation of [aminium]/[NH<sub>4</sub><sup>+</sup>] and [MSA]/[nss-SO<sub>4</sub><sup>2-</sup>], respectively (Fig. 13).

Although most of MSA comes from marine sources, the terrestrial sources may also make up a certain contribution (Yuan et al., 2004). Therefore, MSA/nss-SO<sub>4</sub><sup>2-</sup> = 0 was not used as the end member value for calculating the terrestrial contribution. We have simultaneously measured MSA and nss-SO<sub>4</sub><sup>2-</sup> in a total of 64 total suspended particle (TSP) samples collected in the autumn of 2016 and the summer of 2017. The retention percentage of air mass over the land (*R<sub>L</sub>*) was calculated for each sample based on 3 d backward trajectories (see Figure S9 and supplementary text for more information). Samples with the largest 10% *R<sub>L</sub>* values (*n* = 7, *R<sub>L</sub>* > 74%) were considered to be terrestrially dominant with an average MSA/nss-SO<sub>4</sub><sup>2-</sup> (±1 standard deviation) of 0.0021 ± 0.0013. Therefore, this value was regarded as the end member value of terrestrial MSA/nss-SO<sub>4</sub><sup>2-</sup> in these seasons. Substituting it into the previous fitting equation, the values of  $([\text{DMAH}^+]/[\text{NH}_4^+])_{\text{terrestrial}}$  and  $([\text{TMDEAH}^+]/[\text{NH}_4^+])_{\text{terrestrial}}$  were 0.0068 (0.0038–0.0105) and 0.0034 (0.00005–0.0076), respectively. Then the average contributions of terrestrial and marine sources to

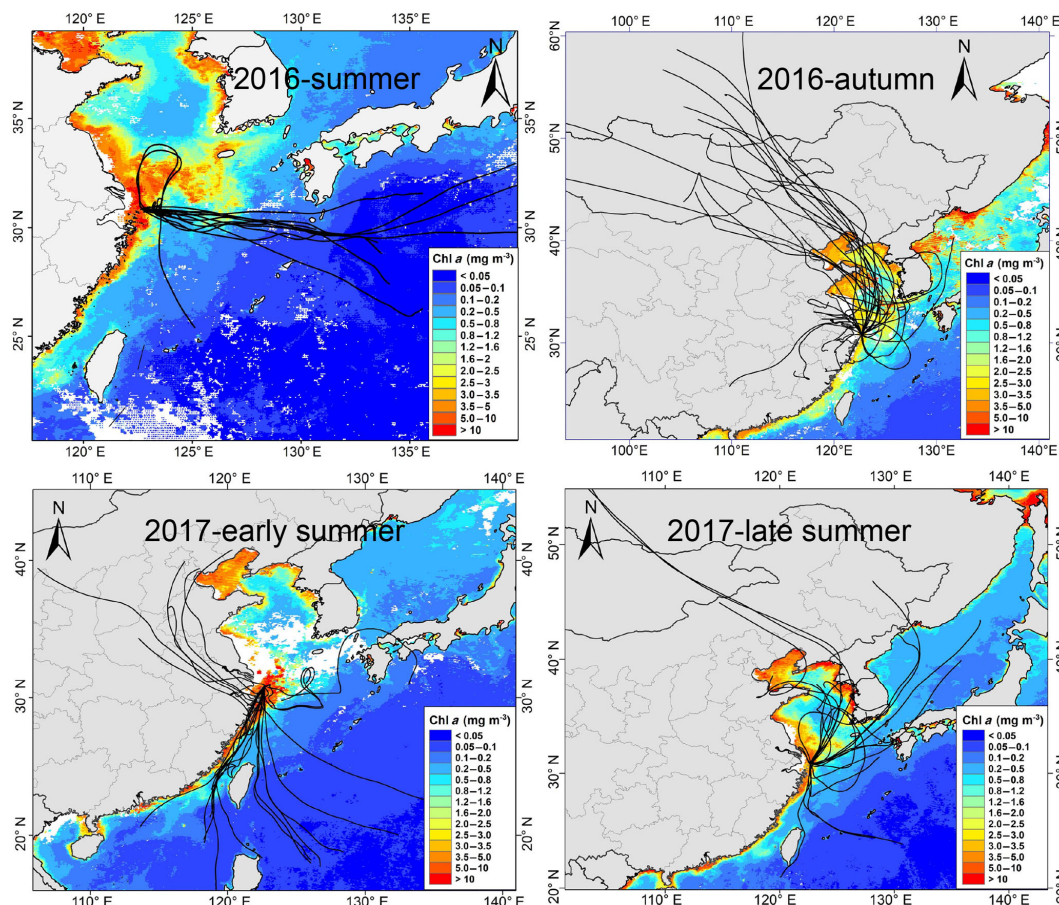




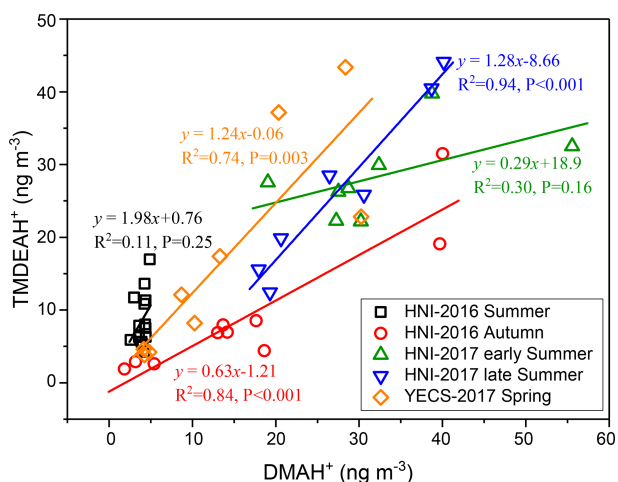
**Figure 9.** (a) Size distributions of  $\text{NO}_3^-$  over the YECS in the spring of 2017. (b) Correlations between concentrations of aminiums and  $\text{NH}_4^+$  for the samples mainly influenced by marine air masses. (c) Correlations between concentrations of aminiums and  $\text{NH}_4^+$  for the samples predominantly influenced by terrestrial transport.



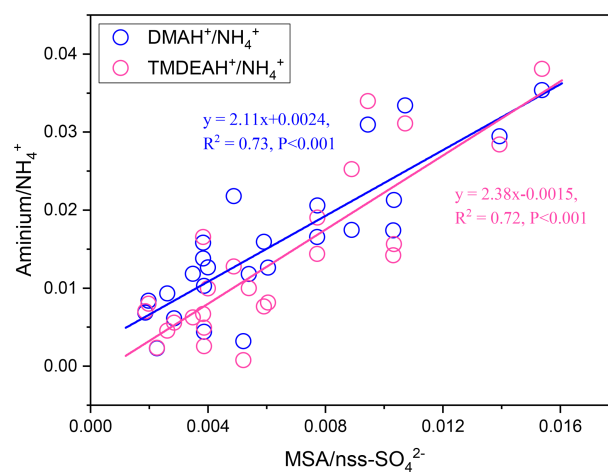
**Figure 10.** Correlations between aminiums and  $\text{NH}_4^+$  concentrations over Huaniao Island for each campaign: (a) in the summer of 2016, (b) in the autumn of 2016, (c) in early summer 2017, and (d) in late summer 2017.



**Figure 11.** The 72 h backward trajectories starting from Huaniao Island and the average Chl *a* concentration retrieved and combined from Aqua-MODIS and Terra-MODIS during the sampling period. Each sample during the summer of 2016 corresponds to one trajectory with a starting time in the middle of sampling period. Each sample set during the autumn of 2016 and the summer of 2017 corresponds to three trajectories and the starting times are taken at equal intervals in the sampling period.



**Figure 12.** Correlations between DMAH<sup>+</sup> and TMDEAH<sup>+</sup> for each campaign over Huaniao Island and the YECS.



**Figure 13.** Correlations between aminium / NH<sub>4</sub><sup>+</sup> and MSA / nss-SO<sub>4</sub><sup>2-</sup> over Huaniao Island during autumn 2016 and summer 2017.

the two aminiums in each campaign were calculated and shown in Table 3. It can be seen that the average terrestrial contributions to DMAH<sup>+</sup> and TMDEAH<sup>+</sup> in aerosols were both more than 60 % in autumn, higher than those in summer. The contributions of marine sources during late summer 2017 (63.0 % for DMAH<sup>+</sup> and 78.3 % for TMDEAH<sup>+</sup>) were higher than those in early summer (53.3 % for DMAH<sup>+</sup> and 74.2 % for TMDEAH<sup>+</sup>), which was consistent with previous hypothesis. Furthermore, the contribution of marine sources was greater to TMDEAH<sup>+</sup> than to DMAH<sup>+</sup> in all campaigns, which corresponded to the higher ratio of TMDEAH<sup>+</sup> / DMAH<sup>+</sup> in the samples influenced primarily by marine air masses (Fig. 12). It should be pointed out that aminium / NH<sub>4</sub><sup>+</sup> ratios could vary with the chemistry of aerosols due to slightly different gas-to-particle partitioning of the amines and NH<sub>3</sub> (Chan and Chan, 2013; Pankow, 2015; Xie et al., 2018) and marine aminiums may also partially originate from primary sources, as discussed above. Therefore, our discussion is constrained in the source analysis of aerosol aminiums but not gaseous or total amines (gaseous amines + aerosol aminiums). Although NH<sub>4</sub><sup>+</sup> was mainly derived from the land, marine sources may also have made a certain contribution (Altieri et al., 2014; Paulot et al., 2015). This was neglected in our calculation and might lead to the overestimate of terrestrial contributions to aminiums. Besides, the relatively small number of data points used in the fitting (25 points) and the treatment of  $([\text{aminium}]/[\text{NH}_4^+])_{\text{terrestrial}}$  as a fixed value ignoring its variation could cause uncertainty in the results. Nonetheless, our method is the first attempt to calculate the contributions of marine biogenic and terrestrial sources to aerosol aminiums over the coastal sea, which will provide an insight of sources and roles of amines in the atmosphere.

#### 4 Conclusions

Amines in the atmosphere play an important role in new particle formation and subsequent particle growth, and studying aerosol aminiums can provide insight into the sources, reaction pathways and environmental effects of amines. An integrated observation was conducted on aerosol aminiums (DMAH<sup>+</sup>, TMDEAH<sup>+</sup> and TEAH<sup>+</sup>) in a coastal city (Shanghai), a nearby island (Huaniao) and the surrounding marginal seas (the YECS). All three aminiums exhibited significant seasonal variation in Shanghai, with their highest concentrations in winter, which was consistent with relatively severe air pollution associated with the winter monsoon (continental winds) and the lowest BLH and temperature in this season. Atmospheric oxidizing capacity and relative humidity may also influence the concentrations of aerosol aminiums to some extent by oxidizing gaseous amines and enhancing gas-particle partitioning, respectively. By comparing the ocean sites to Shanghai, similar DMAH<sup>+</sup> concentrations and 3-fold higher TMDEAH<sup>+</sup> concentrations

were observed, suggesting that these two aminiums may have significant marine sources. In contrast, TEAH<sup>+</sup> was abundant in Shanghai but was below the detection limit over Huaniao Island and the YECS, implying its terrestrial origin.

Aminiums influenced substantially by terrestrial transport showed a bimodal distribution with two peaks at 0.18–0.32 μm (condensation mode) and 0.56–1.0 μm (droplet mode), suggesting that the gas-to-particle condensation and cloud processing were main formation pathways for aerosol aminiums. Nonetheless, aminiums demonstrated a unimodal distribution with a wide peak at 0.18–1.0 μm over the YECS and in late summer on Huaniao Island, and the elevated concentration at 0.32–0.56 μm might be related to sea-spray aerosols that either contain primary aminiums or provide a surface for heterogeneous reactions to form secondary aminiums. This indicates that aminiums in marine aerosols may undergo different formation pathways from those on the land.

We distinguished the contributions of terrestrial and marine sources to aerosol aminiums for the first time by taking the mass ratio of MSA to nss-SO<sub>4</sub><sup>2-</sup> as an indicator of marine biogenic sources. In the autumn of 2016, the contributions of terrestrial sources to aminiums over Huaniao Island were estimated to be more than 60 %. By contrast, marine biogenic sources dominated aminium concentrations especially for TMDEAH<sup>+</sup> (~ 80 %) in the summer of 2017. Our results indicated that marine biogenic emission of amines could not be ignored on the eastern coast of China, especially in summer. Therefore, it is necessary to add this source into the emission inventory of amines and recent modeling of amines over eastern China without a marine source (Mao et al., 2018) may result in significant deviations. Besides, the role of amines in new particle formation over the open ocean is likely to be more important, due to much lower pollution compared to the coastal area, which should be further studied.

*Data availability.* Data are available from the corresponding author on request (yingchen@fudan.edu.cn).

*Supplement.* The supplement related to this article is available online at: <https://doi.org/10.5194/acp-19-10447-2019-supplement>.

*Author contributions.* SZ, YC and CD conceived the study. SZ, YC and CD wrote the paper. SZ, HL and JX collected the samples. SZ, TY and JX performed the measurements. All authors contributed to the review of the manuscript.

*Competing interests.* The authors declare that they have no conflict of interest.

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