



Supplement of

Interactions between aerosol organic components and liquid water content during haze episodes in Beijing

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Figure S1. The comparison of "metastable" water and "stable" water predicted from ISORROPIA-II in different seasons in Beijing.



Figure S2. κ_{org} in four seasons in Beijing calculating from different methods. The box plots represent the 10th, 25th, 50th, 75th, and 90th percentiles of the corresponding data. The squares represent the mean value of the corresponding data. The methods used here were described in detail in Table S2.



Figure S3. Diurnal variations of RH, temperature, inorganic aerosol fraction, organic O/C ratio, and κ_{total} in different seasons in Beijing.



Figure S4. Diurnal variation of mass concentrations for different NR-PM₁ chemical compositions and ALW contents in different seasons in Beijing. ALW during night time is larger than those during day time.



Figure S5. Variations of the relative humidity (RH) and particle chemical compositions (a) the typical high ALW episode in Figure 3a and (b) the typical low ALW episode in Figure 3b. For the high-ALW episode, ALW_{inorg} predicted from ISORROPIA "stable" mode was almost the same as from "metastable" mode. However, for low-ALW episode, "stable" mode resulted in much less ALW_{inorg} than from "metastable" mode.

 $Table \ S1. \ Summary \ of \ the \ data \ periods \ used \ in \ this \ research, \ the \ averaged \ NR-PM_1 \ concentration \ mass$

concentrations, component mass concentrations, <i>f</i> ₄₄ , <i>T</i> , and RH in the four seasons in Beijing.											
Season	Time Period	Days	$NR-PM_1(\mu g m^{-3})$	Org	SO ₄	NO ₃	NH4	Cl	f_{44}	Т	RH
2013-spring	2013.3.9-5.28	81	81.1	26.2	18.0	21.0	13.6	2.3	0.172	15	48
2017-summer	2017.6.1-8.7	68	54.2	22.1	10.3	10.1	6.6	0.4	0.205	28	61
2013-Autumn	2013.10.11-12.2	53	63.9	28.1	11.0	12.3	9.7	2.9	0.144	10	47
2017-winter	2016.12.1-2017.2.28	90	63.2	29.0	10.2	13.3	8.2	2.3	0.109	2	51

Table S2. Calculation methods of κ_{org} from the second sec	om AMS/ACSM measured <i>f44</i> or O/C.

Calculation Method	Application range	Tested Species	Calculated κ_{org} in this study	Reference
$\kappa_{org} = (0.18 \pm 0.04) \times O/C + 0.03$	0.05 < O/C < 1.42	Alkanes, biogenic terpenoids, and aromatics-derived SOA; POA	0.16±0.04	Lambe et al., 2011 ¹
κ _{org} = (0.29±0.05)×O/C	0.3 < O/C < 0.6	Ambient aerosol, Egbert, a rural site in Ontario, Canada during the spring of 2007	0.22±0.07	Chang et al., 2010 ²
κ _{org} = (0.26±0.03)×O/C	0.38 < O/C < 0.98	α -pinene and m-xylene derived SOA	0.19±0.06	Massoli et al., 2010 ³

$= 2.2 \times f_{\odot} = 0.12$			Dupplicay et al	
$K_{org} = 2.2 \times J44 = 0.13$	$0.04 < f_{44} < 0.17$	Jungfraujoch, Switzerland and	0.21 ± 0.08	20114-5
$\kappa_{org} = 0.3 / \times 0 / C - 0.09$		Mexico city		2011

Table S3. Summary of the haze episodes in different seasons in Beijing.							
	Haze episode	High ALW	Low ALW	Nontype			
Spring	7	4	2	1			
Summer	1	1	0	0			
Autumn	6	3	3	0			
winter	8	4	3	1			
total	22	12	8	2			
Average NR-PM ₁		100.8	76.2				
Average fracorg		0.51	0.63				
Average O/C		0.75	0.68				
Ktotal		0.38	0.32				

Reference

1. Lambe, A. T.; Onasch, T. B.; Massoli, P.; Croasdale, D. R.; Wright, J. P.; Ahern, A. T.; Williams, L. R.; Worsnop, D. R.; Brune, W. H.; Davidovits, P., Laboratory studies of the chemical composition and cloud condensation nuclei (CCN) activity of secondary organic aerosol (SOA) and oxidized primary organic aerosol (OPOA). *Atmos Chem Phys* **2011**, *11* (17), 8913-8928.

2. Chang, R.-W.; Slowik, J.; Shantz, N.; Vlasenko, A.; Liggio, J.; Sjostedt, S.; Leaitch, W.; Abbatt, J., The hygroscopicity parameter (κ) of ambient organic aerosol at a field site subject to biogenic and anthropogenic influences: relationship to degree of aerosol oxidation. *Atmos Chem Phys* **2010**, *10* (11), 5047-5064.

3. Massoli, P.; Lambe, A.; Ahern, A.; Williams, L.; Ehn, M.; Mikkilä, J.; Canagaratna, M.; Brune, W.; Onasch, T.; Jayne, J., Relationship between aerosol oxidation level and hygroscopic properties of laboratory generated secondary organic aerosol (SOA) particles. *Geophys Res Lett* **2010**, *37* (24).

4. Duplissy, J.; DeCarlo, P. F.; Dommen, J.; Alfarra, M. R.; Metzger, A.; Barmpadimos, I.; Prevot, A. S. H.; Weingartner, E.; Tritscher, T.; Gysel, M.; Aiken, A. C.; Jimenez, J. L.; Canagaratna, M. R.; Worsnop, D. R.; Collins, D. R.; Tomlinson, J.; Baltensperger, U., Relating hygroscopicity and composition of organic aerosol particulate matter. *Atmos Chem Phys* **2011**, *11* (3), 1155-1165.

 Jimenez, J. L.; Canagaratna, M. R.; Donahue, N. M.; Prevot, A. S. H.; Zhang, Q.; Kroll, J. H.; DeCarlo, P. F.; Allan, J. D.; Coe, H.; Ng, N. L.; Aiken, A. C.; Docherty, K. S.; Ulbrich, I. M.; Grieshop, A. P.; Robinson, A. L.; Duplissy, J.; Smith, J. D.; Wilson, K. R.; Lanz, V. A.; Hueglin, C.; Sun, Y. L.; Tian, J.; Laaksonen, A.; Raatikainen, T.; Rautiainen, J.; Vaattovaara, P.; Ehn, M.; Kulmala, M.; Tomlinson, J. M.; Collins, D. R.; Cubison, M. J.; Dunlea, E. J.; Huffman, J. A.; Onasch, T. B.; Alfarra, M. R.; Williams, P. I.; Bower, K.; Kondo, Y.; Schneider, J.; Drewnick, F.; Borrmann, S.; Weimer, S.; Demerjian, K.; Salcedo, D.; Cottrell, L.; Griffin, R.; Takami, A.; Miyoshi, T.; Hatakeyama, S.; Shimono, A.; Sun, J. Y.; Zhang, Y. M.; Dzepina, K.; Kimmel, J. R.; Sueper, D.; Jayne, J. T.; Herndon, S. C.; Trimborn, A. M.; Williams, L. R.; Wood, E. C.; Middlebrook, A. M.; Kolb, C. E.; Baltensperger, U.; Worsnop, D. R., Evolution of Organic Aerosols in the Atmosphere. *Science* **2009**, *326* (5959), 1525-1529.