

VOC species and emission inventory from vehicles and their SOA formation potentials estimation in Shanghai, China

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Abstract: Volatile organic compound (VOC) species from vehicle exhaust and gas evaporation were investigated by chassis dynamometer and on-road measurements of 9 gasoline vehicles, 7 diesel vehicles, 5 motorcycles, and 4 gas evaporation samples. The secondary organic aerosol (SOA) mass yields of gasoline, diesel, motorcycle exhausts, and gas evaporation were estimated based on the mixing ratio of measured C2-C12 VOC species and inferred carbon number distributions. High aromatic contents were measured in gasoline exhaust and contributed more SOA yield comparatively. A vehicular emission inventory was compiled based on a local survey of on-road traffic in Shanghai and real-world measurements of vehicle emission factors from previous studies in the cities of China. The inventory-based vehicular organic aerosol (OA) productions to total CO emissions were compared with the observed OA to CO concentrations ($\Delta\text{OA}/\Delta\text{CO}$) in the urban atmosphere. The results indicate that vehicles dominate the primary organic aerosol (POA) emissions and OA productions, which contributed about 40% and 60% of OA mass in the urban atmosphere of Shanghai. Diesel vehicles, which accounted for less than 20% of vehicle kilometers of travel (VKT), contribute more than 90% of vehicular POA emissions and 80%-90% of OA mass derived by vehicles in urban Shanghai. Gasoline exhaust could be an important source of SOA formation. Tightening the limit of

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27 aromatic content in gasoline fuel will be helpful to reduce its SOA contribution.
28 Intermediate-volatile organic compounds (IVOCs) in vehicle exhausts have great
29 contributions to SOA formation in the urban atmosphere of China. However, more
30 experiments need to be conducted to determine the contributions of IVOCs to OA
31 pollution in China.

32 **Key words:** SOA; VOC species; vehicle emission; emission inventory; organic
33 aerosol

34 **1. Introduction**

35 SOA accounts for a significant fraction of ambient tropospheric aerosol
36 (Hallquist et al., 2009; Jimenez et al., 2009). De Gouw and Jimenez (2009) suggested
37 that SOA from urban sources may be the dominant source of organic aerosol globally
38 between 30 and 50 latitude.

39 Gas-phase oxidation of VOCs has traditionally been considered to be the major
40 source of urban SOA formation. VOCs are oxidized to low vapor pressure reaction
41 products by OH radical, ozone, and NO₃ radical, and eventually form OA in the
42 atmosphere. Odum et al. (1997) investigated the SOA formation from vaporized
43 reformulated gasoline and found single light aromatic hydrocarbons are responsible
44 for the majority of SOA formation. Kleindienst et al. (2002) verified that 75–85% of
45 the SOA was due to reaction products of C6-C9 light aromatic compounds from
46 automobile exhaust. Robinson et al. (2007) further recognized that IVOCs and
47 semi-volatile organic compounds (SVOCs) are also important sources for OA
48 productions based on the smog chamber studies of diesel exhaust and wood fire
49 (Weitkamp et al., 2007; Grieshop et al., 2009). Their subsequent study pointed IVOCs
50 such as long-chain and branched alkanes from vehicle exhaust play more important
51 roles in SOA production compared with other combustion emissions (Jathar et al.,
52 2013). A recent study from Zhao et al. (2014) concluded that primary IVOCs were
53 estimated to produce about 30% of newly formed SOA in the afternoon during
54 CalNex campaign in Pasadena, California.

55 Due to the abundance of reactive organic compounds, vehicle emission has been
56 recognized as a major source of urban SOA formation (Stone et al., 2009; Liu et al.,
57 2012; Borbon et al., 2013). Laboratory chamber studies also report significant SOA
58 production from diesel, gasoline, and motorcycle exhaust photo-oxidation (Hung et al.,
59 2006; Weitkamp et al., 2007; Chirico et al., 2010; Nordin et al., 2013; Platt et al.,
60 2013). Current research is now focusing on the relative importance of gasoline and
61 diesel vehicles to urban SOA formation. Bahreini et al. (2012) and Hayes et al. (2013)
62 suggested gasoline emissions dominate over diesel in urban SOA formation by field
63 studies. Gentner et al. (2012) argued diesel is responsible for 65% to 90% of
64 vehicular-derived SOA based on the estimation of SOA formation from gasoline and
65 diesel fuel compositions.

66 Shanghai is one of the most urbanized cities in the Yangtze River Delta (YRD)
67 region in China. The YRD region occupies 2% of land area and generates 8%-12% of
68 the primary PM_{2.5} and the emissions of its precursors in China (Huang et al., 2011).
69 Motor vehicles are the fastest growing source of pollution in the megacities of China.
70 The number of vehicles in Shanghai has doubled in the last decade and reached 2.6
71 million (about 107 units per 1000 capita) in 2012 (SCCTPI, 2013). Gasoline and
72 diesel vehicles increased by 2.8 and 1.3 times, respectively, while motorcycle
73 decreased by 36%. Vehicular emission has been recognized as the largest source of
74 VOCs in urban Shanghai, which contributes 25%~28% of the measured VOC
75 concentrations. Other VOC emission sources were solvent usage, chemical industry,
76 petrochemical industry, and coal burning, etc. (Cai et al., 2010; Wang et al., 2013).
77 Yuan et al. (2013) indicated that VOC emissions are large contributors to SOA
78 formation through field measurements at a receptor site in eastern China. Huang et al.
79 (2012, 2013) reported that 28.7%-32.1% of the fine particle mass is organic matter
80 (OM) and 30.2%-76% of OM is contributed by SOA in the atmosphere of Shanghai
81 and its surrounding areas. Based on the historical measurement data of organic (OC)
82 and element carbon (EC) in PM_{2.5} in the atmosphere in urban Shanghai, the OC/EC

83 ratio shows growing trend from 1999 to 2011, which implies that the secondary
84 fraction of organic matter is playing an increasing role in urban Shanghai (Ye et al.,
85 2003; Feng et al., 2005; Hou et al., 2011; Cao et al., 2013; Feng et al., 2013).
86 However, the contribution of VOC emissions to SOA formation and the relative
87 importance of vehicular emission remain unclear. At present, vehicle use is
88 experiencing a rapid growth episode in the cities of China. Understanding the
89 contribution of vehicular VOC emissions to SOA formation will be helpful to identify
90 the source of OA and PM_{2.5} pollution in China.

91 In this study, we first constructed a vehicular emission inventory of Shanghai for
92 the year of 2012. Then the SOA yields of VOCs emissions from different vehicle
93 types were discussed based on the new measurements of VOCs species from a fleet of
94 vehicles in Shanghai. Finally, we calculated the inventory-based vehicular OA
95 production with the ambient observation data to evaluate the OA contribution of
96 vehicle emission. The main purpose of this study is to discuss: (1) the contribution of
97 vehicle emission to OA in urban Shanghai; (2) the relative contributions of gasoline
98 and diesel vehicles to vehicle derived OA.

99 **2. Materials and methods**

100 2.1 Vehicular emission inventory establishment

101 2.1.1 Methodology of emission inventory compilation

102 We developed emission inventories for the pollutants including VOCs, CO, EC,
103 and OC with the IVE (International Vehicle Emission) model for Shanghai, China.
104 The methodology of the model has been introduced by Wang et al. (2008). VKT,
105 vehicle flow distribution, driving pattern, fleet composition and emission factor of
106 each vehicle type were 5 key parameters for the development of vehicle emission
107 inventory. Vehicle emissions can be calculated with Eq. (1).

$$108 \quad E = \sum_t \{ VKT \times f_{[t]} \times EF_{[t]} \times \sum_d [f_{[dt]} \times K_{[dt]}] \} \quad (1)$$

109 Where, E is emission amount of each vehicle type (g). VKT is Vehicle kilometers of

110 travel of each vehicle type (km). $f_{[t]}$ is the fleet composition of the specific technology
111 of each vehicle type (%), such as fuel type, engine size, and emission standard. $EF_{[t]}$
112 is the emission factor of each vehicle technology ($\text{g}\cdot\text{km}^{-1}$). $f_{[dt]}$ is the fraction of the
113 driving pattern (%). $K_{[dt]}$ is the correction factor of each driving pattern determined by
114 the model (unitless). Evaporative emissions are also calculated with Eq. (1). $EF_{[t]}$ will
115 be evaporative emission factor of each vehicle technology as the evaporative
116 emissions are calculated.

117 2.1.2 Road traffic data survey

118 VKTs and their weights on 3 road types (including highway, arterial road, and
119 residential road) were surveyed from transportation for the year of 2012. VKTs on
120 each road type were further separated into 7 vehicle types by the use of video camera
121 surveys. The distributions of each vehicle type were surveyed on various road types
122 with video cameras from March to May. About 4000 valid hours were obtained on 15
123 roads covering 3 road types. Survey days included weekdays and weekends and each
124 day covered 24 h. The results show that light-duty vehicles (including light-duty cars,
125 light-duty trucks, and taxis) are the major vehicle types on the road, accounting for
126 56% of the total flows. Heavy-duty vehicles (including heavy-duty bus, heavy-duty
127 truck, and city bus) comprise 19% of the whole VKTs in Shanghai. GPS data were
128 used to determine the driving patterns of various vehicle types. The driving patterns
129 were determined by the average speeds and VSP (Vehicle Specific Power)
130 distributions. We installed GPS units on light-duty cars, taxis, buses, and heavy-duty
131 trucks to record the driving speeds and altitudes second by second. About 150 hours
132 of valid GPS data were collected in this study. The data covered 2831 km of roads and
133 were composed of 3 road types and 4 vehicle types. VSP of each vehicle and road
134 type can be calculated with Eq. (2) introduced by Jimenez (1999).

$$135 \quad VSP(kW \cdot t^{-1}) = v \times [1.1a + 9.81 \times (a \cdot \tan(\sin(\text{grade}))) + 0.132] + 0.000302 \times v^3 \quad (2)$$

136 Where, v is vehicle speed ($\text{m}\cdot\text{s}^{-1}$). a is vehicle acceleration ($\text{m}\cdot\text{s}^{-2}$). grade is vertical
137 rise/slope length. Table 1 shows the daily VKT and average speeds of various vehicle

138 and road types in 2012.

139 2.1.3 Fleet composition data survey

140 Fleet composition data were used to separate the VKT of each vehicle type (as
141 shown in Table 1) into the fractions of specific technologies, such as fuel type, engine
142 size, and emission standard. The data were determined by the ratios of the populations
143 of specific technologies in the vehicle information database from the Vehicle
144 Management Department of Public Security Bureau of Shanghai. We call this “static”
145 fleet. Light-duty cars and taxis were mainly composed of gasoline vehicles, which
146 occupied 98% and 97%, respectively. Diesel vehicles dominated in light-duty truck,
147 heavy-duty bus, heavy-duty truck, and city bus, comprising 56%, 91%, 89%, and 98%,
148 respectively. Euro 2 vehicles were the majority of light-duty cars and light-duty trucks,
149 accounting for 51% and 68%, respectively. Heavy-duty buses and trucks were mainly
150 composed of Euro 2 and Euro 3 diesel vehicles, which comprised 40% and 45% of
151 each vehicle type. However, the fraction of each specific technology should be
152 changed with its occurrence frequency in the real-world. Generally, older vehicles
153 show less occurrence frequency than newer vehicles, which means the annual mileage
154 of older vehicle should be less than the newer one. For this reason, we considered to
155 adjust the fleet compositions according to their real-world annual average mileages.
156 About 30,000 vehicles were surveyed at 4 inspection stations in this study. Vehicle
157 age and odometer reading were recorded for each vehicle. The survey data showed
158 that the annual average mileages of light-duty truck, heavy-duty bus, and heavy-duty
159 truck tended to decrease with the increase of their vehicle ages. The “adjusted” fleet
160 compositions were determined by the multiplication of vehicle populations and their
161 surveyed annual mileages. Fig. 1 shows the static and adjusted fraction by each
162 vehicle type in Shanghai. It is indicated that the adjusted fractions of the older
163 vehicles with pre-Euro and Euro 1 emission standard for light-duty truck, heavy-duty
164 bus, heavy-duty truck, and city bus were much lower than those of the static ones.
165 Correspondingly, the adjusted fractions of the newer vehicles with Euro 3 emission

166 standard increased a lot compare with the static ones.

167 2.1.4 Vehicle emission factors

168 The emission factors of each vehicle technology were modeled with the IVE
169 model. However, most of the default emission factors in the model are based on the
170 measurements in the US. To localize the emission factors in this study, we collected
171 the published emission factors based on the real-world measurements in the previous
172 studies to adjust the modeled emission factors. The measurements were all conducted
173 with Portable Emission Measurement Systems (PEMS) under designed driving routes
174 in the cities of China. The cities included Shanghai, Beijing, Guangzhou, Xi'an,
175 Shenzhen, Jinan, and Yichang (Chen et al., 2007; Huo et al., 2012a; Huo et al., 2012b;
176 Wu et al., 2012; Huang et al., 2013). Fig. 2 shows the comparisons of the adjusted
177 emission factors with the measured ones. The evaporative emission factors were not
178 adjusted due to the lack of measurement data. Default factors in the model were used
179 to calculate evaporative emissions in this study. It is indicated that the adjusted
180 emission factors of each vehicle type generally fit well with the measured results. The
181 emission factors are reliable to be used to establish the emission inventory.

182 2.2 VOC species measurements and SOA yield estimation

183 2.2.1 VOC sampling

184 The exhaust from 4 light-duty gasoline vehicles (LDGVs), 5 taxis, 5 heavy-duty
185 diesel trucks (HDDTs), 2 city buses (buses), and 5 motorcycles (MTs) were measured
186 in June 2010. LDGVs, taxis, and MTs were fueled by gasoline. HDDTs and buses
187 were fueled by diesel. The emission standards of the tested vehicles covered Euro 1 to
188 Euro 3 and their model years covered 2001 to 2009. All MTs were 4-stroke with 125
189 cc displacement and without catalytic converter or any other pollution control device.
190 All gasoline vehicles were equipped with catalytic converters. Diesel vehicles didn't
191 install any aftertreatment device like DPF (Diesel Particle Filter). Table 2 lists the
192 detailed information of the tested vehicles. Commercially available fuels were used in
193 the test. The fuel quality met the requirements of the local standard in Shanghai. The

194 sulfur contents of both gasoline and diesel fuel were below 50 ppm.

195 All the automobiles were measured on chassis dynamometers. LDGVs and taxis
196 were measured utilizing a vehicle mass analysis system (VMAS), which was widely
197 used in in-use vehicle inspection stations in China. VOC sample was collected from a
198 1-bag test of the Economic Commission of Europe (ECE) urban cycle. The highest
199 speed reaches $50 \text{ km}\cdot\text{h}^{-1}$ and the average speed is about $18.8 \text{ km}\cdot\text{h}^{-1}$. HDDTs and
200 buses were measured on a loaded mode test cycle. The tested vehicles were operated
201 on idling and a test cycle which simulates high engine loads under 100%, 90%, and
202 80% of their maximum powers. The highest speed reaches $70 \text{ km}\cdot\text{h}^{-1}$. MT exhausts
203 were sampled while operating on the road. A GPS unit was installed on the tested
204 motorcycles to record the speeds second by second. The highest speed reached 50
205 $\text{km}\cdot\text{h}^{-1}$ and the average speed was about $20 \text{ km}\cdot\text{h}^{-1}$. Vehicle exhaust was sampled into
206 a Summa canister (Entech Inst., USA) during the whole driving cycle. We also
207 collected the samples of gasoline vapor at 4 gas stations in Shanghai to analyze the
208 VOC species of non-tailpipe gasoline.

209 2.2.2 VOC analysis

210 Concentration of C2-C12 VOCs in samples were determined by a GC-MS
211 system (Agilent 7890A/5975C) with standard gases prepared by Spectra Gas. The
212 samples collected in the Summa canister were pre-concentrated to an acceptable level
213 for the analytical devices using a 7100A pre-concentrator (Entech Inst., USA) with an
214 Entech 7016CA automatic sample injector. A 50 mL sample was extracted by the
215 pre-concentrator into a 1/4 inch liquid nitrogen cold trap to remove water and CO_2 ,
216 and then separated by GC and detected by MS. The carrier gas was helium.

217 2.2.3 SOA yield estimation

218 To investigate the SOA formation potentials of VOC emissions in vehicle
219 exhausts and gas evaporation, we calculated the SOA yields of the exhausts from
220 gasoline, diesel, and motorcycle vehicles and evaporative emissions with the
221 following equation.

222
$$Y_j = \frac{\sum (C_{i,j} \times Y_i)}{\sum VOC_j} \quad (3)$$

223 Where, Y_j is the SOA yield of source j (unitless). $C_{i,j}$ is the weight percent (by carbon)
224 of species i which can be identified by measurements or references from source j
225 (wtC%). VOC_j is the weight percent (by carbon) of total identified SOA precursors
226 and unidentified species. Identified non-SOA precursors were excluded from total
227 VOC emissions. The weight percentages of identified species were determined by the
228 measurements above. The unidentified species accounted for about 25%, 60%, and
229 50% in gasoline, diesel, and motorcycle exhausts, respectively. Considering IVOCs
230 which had high SOA formation potentials were not measured in this study, we
231 estimated the amounts of IVOCs including alkanes and aromatics larger than C12 and
232 polycyclic aromatics based on Gentner et al. (2012) and varied them by one order of
233 magnitude. Y_i is the yield of species i under high-NO_x condition considering the study
234 was focusing on urban area with high NO_x concentration (unitless). The yield for
235 each SOA precursor was referenced from Gentner et al. (2012), which listed the yields
236 of known and estimated compounds using a combination of measured SOA yields
237 derived from laboratory-chamber experiments and approximate SOA yields based on
238 box modeling. However, considering the average organic loading in Shanghai
239 ($15.5\mu\text{g}\cdot\text{m}^{-3}$) was relatively higher than that in the reference ($10\mu\text{g}\cdot\text{m}^{-3}$), we
240 recalculated the SOA yields of the compounds using a semi-empirical model based on
241 absorptive gas-particle partitioning of two semi-volatile products introduced by Odum
242 et al. (1997). The SOA yields of straight alkanes increased by an average of 16% in
243 the range of 12-17 carbon atoms, C6-8 aromatics increased by ~19%, and naphthalene
244 increased by ~12%. The yields of the compounds in similar chemical class were
245 corrected based on the increments above.

246

247 2.3 Air pollution observation and vehicular OA contribution determination

248 2.3.1 Air pollution observation

249 To estimate the vehicular OA production in the atmosphere, we calculated the
250 SOA formation potentials from vehicular VOC emissions based on observation data
251 with a photochemical- age-based parameterization method. The observation data were
252 obtained from a monitoring site on the roof of a 5-floor building (15 m high above the
253 ground) at Shanghai Academy of Environmental Science (31.17°N, 121.43°E), which
254 was located southwest of urban area of Shanghai. The site was mostly surrounded by
255 commercial properties and residential dwellings. Vehicle exhaust was a major source
256 of pollutants near this site. Fig. S1 shows the location of monitoring site in this study.
257 Carbon monoxide was continuously measured by an ECOTECH EC9820 CO analyzer.
258 PM_{2.5} concentration was measured by a Thermo Fisher commercial instrument β -ray
259 particulate monitor. Organic carbon (OC) and elemental carbon (EC) were measured
260 by a carbon analyzer (model RT-4, Sunset Laboratory Inc.). Water soluble ions were
261 measured by a commercial instrument for online monitoring of aerosols and gases
262 (MARGA, model ADI 2080, Applikon Analytical B.V.). Individual VOC species were
263 continuously measured every 30 minutes by two on-line gas chromatographs with
264 flame ionization detector (GC-FID) systems (Chromato-sud airmoVOC C2–C6
265 #5250308 and airmoVOC C6–C12 #2260308, France). Fig. S2 shows the time series
266 data of meteorological parameters and concentrations of major air pollutants observed
267 in urban Shanghai in summer (August in 2013) and winter (January in 2013). We
268 didn't measure the OA concentration due to the lack of observation equipment. The
269 OA concentrations were determined by OC concentrations multiplying the OM/OC
270 ratio. Turpin et al. (2001) suggested a ratio around 1.2-1.6 from fresh emission to aged
271 air mass in remote area. Considering our study was mainly focusing on urban area
272 where emissions were not fully aged, we used the ratio of 1.4 to convert OC
273 concentrations.

274 2.3.2 OA production estimation

275 The evolution of primary VOC emissions to SOA formation is determined by
 276 OH exposure in the atmosphere. The OH exposure can be calculated with Eq. (4)
 277 developed by de Gouw et al. (2005, 2008).

$$278 \quad \Delta t \cdot [OH] = \frac{1}{(k_X - k_E)} \times \left[\ln\left(\left[\frac{X}{E}\right]_{t=0}\right) - \ln\left(\left[\frac{X}{E}\right]\right) \right] \quad (4)$$

279 Here, Δt is photochemical age (h). $[OH]$ is the average OH radical concentrations
 280 ($\text{molecules} \cdot \text{cm}^{-3}$). The ratio of m,p-xylene to ethylbenzene (X/E) was considered as a
 281 photochemical clock. k_X and k_E are the OH rate constants of m,p-xylene (18.9×10^{-12}
 282 $\text{cm}^3 \cdot \text{molecule}^{-1} \cdot \text{s}^{-1}$) and ethylbenzene ($7.0 \times 10^{-12} \text{ cm}^3 \cdot \text{molecule}^{-1} \cdot \text{s}^{-1}$) (Yuan et al.,
 283 2013). $[X]/[E]_{t=0}$ and $[X]/[E]$ are initial emission ratio and the ratio after
 284 photochemical reaction of m,p-xylene to ethylbenzene. The concentrations of
 285 m,p-xylene and ethylbenzene showed good correlations during the observation (Fig.
 286 S3). The different diurnal variations of m,p-xylene and ethylbenzene indicated that
 287 they are oxidized at different rates from each other (Fig. S4). Fig. 3 illustrates the
 288 diurnal distributions of the ratios of observed m,p-xylene to ethylbenzene in summer
 289 and winter of 2013. The initial emission ratios of m,p-xylene to ethylbenzene were
 290 determined by the X/E ratio on 97.5 percentiles, which were 2.17 and 1.68 in summer
 291 and winter, respectively.

292 OA production is determined by the loss terms and formation rates of OA
 293 concentration after POA emissions exhaust into the atmosphere. de Gouw et al. (2008)
 294 introduced a method to explain the OA evolution during photochemical aging of
 295 urban plumes as shown by Eq. (5).

$$296 \quad \frac{\Delta OA}{\Delta CO} = \frac{\Delta POA}{\Delta CO} + \frac{\Delta SOA}{\Delta CO} \quad (5)$$

$$= ER_{POA} \times \exp(-L_{OA} \cdot \Delta t) + ER_{VOC_j} \times Y_j \times \frac{P_{OA}}{L_{OA} - P_{OA}} \times [\exp(-P_{OA} \cdot \Delta t) - \exp(-L_{OA} \cdot \Delta t)]$$

297 Here, $\Delta OA/\Delta CO$ is the ratio of OA formation versus CO emission after photochemical
 298 reaction ($\mu\text{g} \cdot \text{m}^{-3} \cdot \text{ppmv}^{-1}$). ER_{POA} is the primary emission ratio of OA to CO emission
 299 ($\mu\text{g} \cdot \text{m}^{-3} \cdot \text{ppmv}^{-1}$). ER_{VOC_j} is the primary emission ratios of VOC (including SOA

300 precursors and unidentified species) from source j to CO emission in the unit of
 301 $\text{ppbv}^{-1} \cdot \text{ppmv}^{-1}$. Y_j is the SOA yield of source j , which is determined by Eq. (3). L_{OA}
 302 and P_{OA} are the loss and formation rate of organic aerosol, respectively. We used the
 303 empirical parameters derived by de Gouw et al. (2008), which were 0.00677 h^{-1} and
 304 0.0384 h^{-1} , respectively. Δt is the photochemical age calculated by equation (4).
 305 Because there is no OH measurement in Shanghai, we reference the 24-h average OH
 306 concentration ($3 \times 10^6 \text{ molecules} \cdot \text{cm}^{-3}$) from de Gouw et al. (2008).

307 2.3.3 Determination of vehicular OA contribution

308 The observed $\Delta OA/\Delta CO$ in the atmosphere represents the ratio of total POA
 309 emissions and their SOA formation to total CO emissions from all sources, which can
 310 be explained by Eq. (6) as follows.

$$311 \quad \left(\frac{\Delta OA}{\Delta CO} \right)_{obs} = \frac{POA_{total} + SOA_{total}}{CO_{total}} = \frac{(POA_{veh} + SOA_{veh})}{CO_{total}} + \frac{(POA_{oth} + SOA_{oth})}{CO_{total}} = \frac{OA_{veh}}{CO_{total}} + \frac{OA_{oth}}{CO_{total}} \quad (6)$$

312 The contribution of vehicular OA to total OA production in the atmosphere can be
 313 determined by the ratio of OA_{veh}/CO_{total} to $(\Delta OA/\Delta CO)_{obs}$. Here, $(\Delta OA/\Delta CO)_{obs}$ is the
 314 ratio of observed ΔOA (Obs, $OA - OA_{background}$) to observed ΔCO (Obs,
 315 $CO - CO_{background}$) in the unit of $\mu\text{g} \cdot \text{m}^{-3} \cdot \text{ppmv}^{-1}$. OA_{veh}/CO_{total} is the ratio of vehicular
 316 POA emission and SOA formation to total CO emissions, which can be calculated by
 317 Eq. (5). ER_{POA} and ER_{VOCj} in Eq. (5) represented for the ratios of vehicular POA and
 318 VOC emissions to total CO emissions from vehicles and other sources. The total
 319 amount of CO emission was 1.2×10^6 tons according to the annually updated emission
 320 inventory in Shanghai for the year of 2012. Iron & steel manufacturing was the major
 321 source of CO emission, which accounted for 54% of the total. The sector produced
 322 19.7×10^6 and 18×10^6 tons of pig irons and crude steels, and consumed more than
 323 10×10^6 tons of coal in 2012. Vehicles were the second largest source, accounting for
 324 27.8% of the total. However, considering the observation site was located in the urban
 325 area and most of the industrial CO emission sources were located at the surrounding
 326 areas (about 30-50 km from city center), it would be more reasonable to simulate the
 327 contribution of CO concentration from vehicle exhausts and other sources to the

328 receptor by using numerical model. Here we used CMAQ model and brute force
329 method to simulate the CO concentrations during January and August in 2013 under
330 two scenarios of with or without vehicular CO emission. The meteorological data was
331 from the results of the Weather Research and Forecasting Model (WRF). Detailed
332 information is shown in the supplement materials. The results showed that vehicles
333 dominated CO emission in urban Shanghai, accounting for 66% and 70% of total CO
334 concentrations in summer and winter, respectively. On this account, the vehicular
335 fraction of total CO emissions used in this study was determined to 68%.

336 **3. Results and discussion**

337 3.1 Vehicle emission inventory

338 The emissions of CO, NO_x, VOCs, EVA (gas evaporation), EC, and POA
339 (OC*1.2) from vehicles were 343.9, 110.9, 39.4, 8.9, 4.0, and 4.3 k tons in Shanghai
340 for the year of 2012 (See Table 3). Gasoline vehicles (including LDGV, Taxi, HDGV,
341 and Motorcycle) were the major sources of CO, VOCs, and EVA emissions,
342 accounting for 91%, 69%, and 100%, respectively. Diesel vehicles (including LDDV,
343 HDDV, and Bus) were the major source of NO_x, EC, and POA emissions, comprising
344 82%, 99%, and 96%, respectively. CO and VOC (including EVA) emissions
345 decreased by 40% and 38% compared with the results for the year of 2004 from Wang
346 et al. (2008). NO_x emission increased by 21%. PM emission were low estimated in
347 that article since the PM emission factors were much lower than real-world
348 measurement data as shown in Fig. 2. Gasoline vehicle emissions have been well
349 controlled even though their VKTs were nearly doubled in the past few years. In
350 comparison, the control effect of diesel vehicle emission was relatively poor. It is
351 clear that diesel exhausts dominate the primary PM (including EC and OA) emissions
352 in Shanghai. However, since VOC emissions are mainly from gasoline vehicles, we
353 will further discuss the contributions of gasoline and diesel exhausts to SOA.

354 3.2 VOC species of vehicle emissions and gas evaporation

355 Fig. 4 compares the VOC compositions of the exhausts from different vehicle
356 types and gas evaporation in this study to the results from other countries or regions.
357 Since the VOC species measured in different studies are not the same, we normalized
358 the concentrations of the common species including C2-C12 alkanes, alkenes, alkynes,
359 and single-ring aromatics in each study as 100%. Other compounds and unidentified
360 VOCs were excluded in the comparison. The weighted percentages of individual VOC
361 for the exhausts from different vehicle types and gas evaporation were listed in Table
362 S1.

363 The exhausts from gasoline vehicles (including LDGVs and taxis) had similar
364 VOC compositions. Single-ring aromatics were the major species of the exhausts
365 from gasoline vehicles, accounting for 50% of the total VOCs approximately.
366 Straight-chain alkanes, branched alkanes, and cycloalkanes comprised 17.0%, 18.1%,
367 and 6.1% of the total VOCs, respectively. Toluene, m,p-xylene, o-xylene, and
368 ethylbenzene were the main compounds in LDGV and taxi exhausts, accounting for
369 7.54%, 6.71%, 5.20%, and 4.42% of the total VOCs, respectively. Motorcycle emitted
370 more branched alkanes and less single-ring aromatics than LDGVs and taxis.
371 2-methylhexane (23.43%) was the most abundant VOC in motorcycle exhausts,
372 followed by m,p-xylene (9.34%), ethylbenzene (5.53%) and o-xylene (4.37%). It was
373 indicated from Fig. 3 that the proportion of single-ring aromatics in LDGV exhausts
374 were higher and the proportion of alkene were lower in this and previous studies in
375 China (Liu et al., 2008; Wang et al., 2013) than those in Hong Kong (Guo et al., 2011)
376 and US (Schauer et al., 2002; Gentner et al., 2013; May et al., 2014). The differences
377 of aromatic content in gasoline fuel in different regions and countries could be the
378 main reason of the difference in the proportion of aromatic compounds in LDGV
379 exhausts. The limit of aromatic content in current gasoline standard in China was 40
380 vol%, which was much higher than the limits of gasoline standards of the US (22-25
381 vol%) and Europe (35 vol%).

382 High proportion of straight-chain alkanes were measured in the exhausts from
383 diesel vehicles, which accounted for 34.9% and 35.6% of the total VOCs from HDDT
384 and bus exhausts, respectively. N-dodecane, propene, n-undecane, acetone, and
385 n-decane were major species in diesel exhausts, accounting for 13.65%, 10.85%,
386 8.69%, 7.00%, and 6.86% of the total VOCs, respectively. The proportions of
387 straight-chain alkanes in diesel exhausts in this study were much higher than those in
388 the previous studies of the US (Schauer et al., 1999; May et al., 2014). Incomplete
389 combustion of diesel fuel caused by poor engine maintenance could be the main
390 reason for the high straight-chain alkane emissions.

391 High proportion of alkenes was measured in gas evaporation in this study, which
392 accounted for 40% of the total VOCs. Propane, isopentane, isobutene, 1-pentene, and
393 n-butane were major species in gas evaporation emissions, accounting for 15.99%,
394 11.87%, 9.69%, 8.87%, and 6.51% of the total VOCs, respectively. The proportions of
395 VOC species in gas evaporation in this study was close to the results in the other
396 study of China (Zhang et al., 2013), but different from the studies in the US (Harley et
397 al., 2000) and Korea (Na et al., 2004), which reported less alkenes and more branched
398 alkanes in gas evaporation.

399 3.3 SOA yields of different vehicle exhausts and gas evaporation

400 VOC species of vehicle emissions and gas evaporation were classified into 5
401 categories by their chemical classes, and their distributions of carbon numbers were
402 shown in Fig. 5(a). Previous studies have confirmed that IVOCs, which were not
403 detected in this study, played important roles to SOA production (Jathar et al., 2013;
404 Zhao et al., 2014). For this reason, we introduced the amounts of alkanes and aromatic
405 larger than C12 and polycyclic aromatics in unburned fuels from Gentner et al. (2012)
406 as the inferred S/IVOCs as shown in the light colored bars in Fig. 5(a). The carbon
407 numbers of VOCs in gasoline and motorcycle exhausts mainly concentrated in the
408 intervals between C6 to C9 whether the IVOCs were merged or not. Comparatively,
409 exhausts from diesel vehicles had a wider distribution of carbon number, ranging from

410 C2 to C25. More than half of the species were S/IVOCs and most of them were
411 alkanes. The carbon numbers of VOCs in gas evaporation were mainly distributed
412 within the range of C3-C7, which were much smaller than those in vehicle exhausts.

413 Fig. 5(b) shows the SOA mass yields of different chemical class. To compare the
414 differences of the SOA yields with or without S/IVOC species, we set two scenarios
415 as “Y1” and “Y2”. Y1 indicates the SOA yields of measured C2-C12 VOCs in this
416 study. Y2 includes the extra SOA yields of inferred S/IVOCs. S/IVOC species had
417 little effect on the SOA yields of gasoline exhausts and evaporative emissions.
418 Aromatics dominated the yields which accounted for almost 100% of the total.
419 However, the SOA yield of diesel exhaust was significantly affected by S/IVOCs. The
420 yield increased from 0.010 to 0.191 when the inferred S/IVOCs were considered. In
421 Y2 scenario, Aromatics were still the largest contributors (34.9%) to but not
422 dominating the yield. Next were branched alkanes, polycyclic aromatics, and
423 straight-chain alkanes, which accounted for 24.8%, 17.1%, and 12.7%.

424 The SOA yield of gasoline exhaust was larger than the yield of liquid gasoline
425 reported by Gentner et al. (2012). We found the aromatic contents in gasoline
426 exhausts of this study which dominated the SOA yield were much higher than those in
427 the reference. This may be due to the loose limit of aromatics for the gasoline fuel in
428 China. However, the estimated yield of gasoline exhaust was much lower than the
429 effective yields (3-30%) of LEV-1 (tier 1 of low emission vehicle standard in
430 California, US) gasoline vehicles (similar to Euro 1-3 LDGVs tested in this study)
431 investigated using a smog-chamber experiment by Gordon et al. (2014a). The reason
432 for the underestimate was still unclear. In contrast, the estimated SOA yield of diesel
433 exhaust which combined the inferred S/IVOCs (Y2) was higher than the average
434 effective yields ($9\pm 6\%$) for HDDVs without DPF based on smog-chamber
435 experiments by Gordon et al. (2014b). Since there were few experiments on
436 motorcycle exhaust, we compared the SOA yield of motorcycle exhaust with the
437 experiment results from the exhausts of 2- and 4-stroke gasoline off-road engines

438 (Gordon et al., 2013). Neither motorcycles nor off-road engines had catalytic
439 converter. The estimated yield was close to the experiment results of off-road engine
440 exhausts (2-4%).

441

442 3.4 Primary emission ratio and SOA formation potential based on observation

443 Fig. 6 is a scatterplot of OA versus CO concentrations measured in urban
444 Shanghai in the summer and the winter of 2013. The observation data were
445 color-coded by OH exposure ($\Delta t \cdot [\text{OH}]$) determined by equation (4). It was indicated
446 from the figure that the ratios of OA to CO concentrations generally showed growing
447 trends with the increase of OH exposure both in summer and winter. The results were
448 similar to the previous studies in the United States, Japan, and Mexico (Bahreini et al.,
449 2012; de Gouw et al., 2008; Takegawa et al., 2006; DeCarlo et al., 2008). The primary
450 emission ratios of POA to CO were determined by the minimum slopes of the
451 observed OA to CO concentrations, about $12 \mu\text{g}\cdot\text{m}^{-3}\cdot\text{ppmv}^{-1}$ in both summer and
452 winter (as shown by the dotted grey lines). The maximum slopes of OA to CO were
453 50 and $35 \mu\text{g}\cdot\text{m}^{-3}\cdot\text{ppmv}^{-1}$ (as shown by the dotted black lines) in summer and winter,
454 respectively. The SOA formation ratio in summer was much higher than in winter.
455 The inventory-based vehicular POA emission to total CO emission was 10.6
456 $\mu\text{g}\cdot\text{m}^{-3}\cdot\text{ppmv}^{-1}$ in the urban area of Shanghai (shown by the dotted yellow lines), close
457 to the primary emission ratio observed in the atmosphere, which indicated that
458 vehicles dominated the POA emissions in urban Shanghai. The dotted pink and orange
459 lines in Fig. 6 represent the maximum OA production ratios (assuming SOA
460 precursors were 100% reacted) calculated with the SOA yields in Y1 (only detected
461 VOCs in this study) and Y2 scenarios (detected VOCs plus the inferred S/IVOCs).
462 The maximum OA production ratios were 13.8 and $18.7 \mu\text{g}\cdot\text{m}^{-3}\cdot\text{ppmv}^{-1}$, respectively.
463 It was indicated that S/IVOCs played much more important roles to SOA productions
464 of vehicle exhaust. However, the max. OA production ratio for Y2 scenario was still
465 considerable underestimated compared with the observation data, which implied that

466 the SOA yields derived by known and estimated species were still far from explaining
467 the actual SOA formation rate in the atmosphere. For this reason, we introduced the
468 measured SOA yield of gasoline exhaust (~ 0.190) from Gordon et al. (2014a) to
469 substitute the yield for gasoline vehicles (0.039) in Y2 scenario and defined the new
470 group as Y3 scenario. The maximum OA production ratio (shown by the dotted red
471 lines) increased to $27.6 \mu\text{g}\cdot\text{m}^{-3}\cdot\text{ppmv}^{-1}$, but still failed to reach the max. observed
472 $\Delta\text{OA}/\Delta\text{CO}$. There must be other emission sources of SOA precursors in the
473 atmosphere of Shanghai. Previous studies have revealed that VOC emissions from
474 solvent usage, chemical and petrochemical industrial, and coal burning, etc.
475 comprised more than 70% of the observed VOCs in the atmosphere of urban Shanghai
476 (Cai et al., 2010; Wang et al., 2013). The SOA productions of VOC emissions from
477 these sources cannot be ignored.

478 3.5 Estimation of vehicular OA contribution in the urban atmosphere

479 Fig. 7 shows the diurnal variations of average observed $\Delta\text{OA}/\Delta\text{CO}$ and OH
480 exposure in summer and winter. There was a strong correlation between the observed
481 $\Delta\text{OA}/\Delta\text{CO}$ and OH exposure, which indicated the photooxidation dominated the SOA
482 formation in the atmosphere. In the role of photochemical reaction, the observed
483 $\Delta\text{OA}/\Delta\text{CO}$ showed rapid growth trend in the afternoon in summer and reached a peak
484 around 13:00~14:00. The average observed $\Delta\text{OA}/\Delta\text{CO}$ in urban atmosphere of
485 Shanghai were 33.2 and $21.1 \mu\text{g}\cdot\text{m}^{-3}\cdot\text{ppmv}^{-1}$ in summer and winter, respectively.

486 To evaluate the contribution of vehicle emission to OA production in urban
487 atmosphere, we estimated the vehicular OA formation ratio to total CO emissions with
488 Eq. (5) in three scenarios. Fig. 7(a) and (b) showed the results of vehicular OA
489 formation ratios to total CO emissions in Y1 scenario. The SOA yields of gasoline,
490 diesel, and motorcycle exhausts and gas evaporation were 0.046, 0.010, 0.024, and
491 0.0009, respectively. Fig. 7(c) and (d) showed the results in Y2 scenario where the
492 inferred S/IVOCs were merged. The SOA yields of gasoline, diesel, and motorcycle
493 exhausts and gas evaporation were 0.047, 0.191, 0.025, and 0.0009, respectively. Fig.

494 7(e) and (f) showed the results in Y3 scenario. The SOA yield of gasoline exhaust was
495 replaced to 0.190 based on the experiment by Gordon et al. (2014a). The
496 photochemical age (Δt) in each hour was calculated with Eq. (4). Due to the lack of
497 OH measurement in Shanghai, we referenced the 24-h average OH concentration
498 (3×10^6 molecules $\cdot\text{cm}^{-3}$) from de Gouw et al. (2008). The grey and yellow lines were
499 the ratio of vehicular POA and OA production to total CO emissions. The average
500 vehicular OA production ratios to total CO emission in the urban area were 10.6
501 $\mu\text{g}\cdot\text{m}^{-3}\cdot\text{ppmv}^{-1}$ and 10.8 $\mu\text{g}\cdot\text{m}^{-3}\cdot\text{ppmv}^{-1}$ in summer and winter in Y1 scenario, 11.8
502 $\mu\text{g}\cdot\text{m}^{-3}\cdot\text{ppmv}^{-1}$ and 11.4 $\mu\text{g}\cdot\text{m}^{-3}\cdot\text{ppmv}^{-1}$ in Y2 scenario, and 13.3 $\mu\text{g}\cdot\text{m}^{-3}\cdot\text{ppmv}^{-1}$ and
503 12.4 $\mu\text{g}\cdot\text{m}^{-3}\cdot\text{ppmv}^{-1}$ in Y3 scenario. The vehicular OA mass accounted for 34% and
504 52% of the average observed OA in summer and winter in the urban atmosphere of
505 Shanghai in Y1 scenario. The contributions would increase to 37% and 55% in Y2
506 scenario, and 41% and 59% in Y3 scenario. It was indicated that vehicle emission was
507 the major source of OA mass in the urban atmosphere of Shanghai. For Y2 scenario
508 where the inferred IVOC species were merged to SOA yield estimation, the vehicular
509 OA production ratios increased about 3%. For Y3 scenario where the SOA yield of
510 gasoline exhausts was enhanced to the smog chamber experiment result (0.190), the
511 vehicular OA production ratios further increased about 4%. Vehicular SOA formation
512 ratios accounted for 4% of the total vehicular OA in Y1 scenario, 9%-13% in Y2
513 scenario and 16%-23% in Y3 scenario. The SOA formation ratios in both scenarios
514 were lower than expected. There were two possible reasons for the underestimation.
515 One reason was that other emission sources with high SOA formation potentials in
516 addition to vehicles were not considered in this study. The non-fossil VOC emissions
517 from solvent use, chemical and petrochemical industrials, etc. reported by the
518 previous studies could be the rest of contributors (Cai et al., 2010; Wang et al., 2013).
519 Another possible reason was the SOA yields were still underestimated in this study.
520 There were about 30%, 50% and 15% of VOC species still unidentified in gasoline,
521 diesel, and motorcycle exhausts even after we combined the inferred S/IVOC species

522 reported in Gentner et al. (2012). The SOA formation potentials of the identified VOC
523 species may contribute more SOA than expected.

524 At present, few SOA observations in the cities of China can be referenced to
525 verify the results in this study. Huang et al. (2014) has reported the fossil OA
526 dominated the OA mass (~40%) in Shanghai based on the observation data in the first
527 quarter of 2013, which was slightly lower than our result. The possible reason for the
528 difference could be the location of observation site (close to urban or suburban).
529 However, the studies both indicated that vehicle emission was the major source of OA
530 mass in large cities of China.

531 3.6 SOA formation contributions of different vehicle types

532 Fig. 8(a) and (b) show the changes of OA formation ratios in different fuel and
533 vehicle types in Y1 scenario with the increase of the photochemical age. The OA
534 produced from evaporative emissions were combined in gasoline vehicles and
535 corresponding vehicle types. The inventory-based $\Delta\text{OA}/\Delta\text{CO}$ show downward trends
536 in Y1 scenario. Diesel exhausts dominated the OA productions, which accounted for
537 96%, 91% and 84% after 0, 6, and 24 hours of photochemical reaction. HDDV and
538 bus were major sources of OA productions. Fig. 9(c) and (d) show the changes of OA
539 formation ratios in Y2 scenario. The $\Delta\text{OA}/\Delta\text{CO}$ show upward trends in Y2 scenario.
540 The contributions of diesel exhausts in this scenario increased to 92% and 87% after 6
541 and 24 hours of photochemical reaction. Fig. 9(e) and (f) show the changes of OA
542 formation ratios in Y3 scenario. The contribution of gasoline vehicles in this scenario
543 increased a lot. Although gasoline vehicles only accounted for 4% of POA emission,
544 their contributions to vehicular OA formation increased to 18% and 34% after 6 and
545 24 hours of photochemical reaction, respectively. LDGV would be the second large
546 contributor after HDDV. It can be indicated that diesel vehicles were the largest
547 contributors to vehicle derived OA in both scenarios although they only accounted for
548 less than 20% of VKTs in Shanghai. Control of the POA emissions and SOA
549 precursors from diesel vehicles are equally important. Gasoline vehicle could be

550 another important contributor to vehicular OA formation. However, there still exist
551 some debates on the SOA yield of gasoline exhaust. It will be meaningful to find out
552 their actual SOA yield and key precursors for urban OA pollution control.

553 **4. Conclusions**

554 To evaluate the OA contribution of vehicle emissions in the urban atmosphere of
555 Shanghai, we developed a vehicular emission inventory and estimated the SOA yields
556 of gasoline, diesel, and motorcycle exhausts and gas evaporation based on measured
557 C₂-C₁₂ VOC species and inferred S/IVOC species based on Gentner et al. (2012).
558 Higher contents of aromatic were measured in this study and other studies in China
559 compared with the results from the US and European. Loose limit to aromatic
560 contents in the standard of gasoline fuel in China should be responsible for the high
561 aromatic contents, which resulted in larger SOA yield of gasoline exhaust than the
562 results reported by Gentner et al. (2012) based on the same method. However, the
563 estimated yield was still much lower than the results from smog-chamber experiments
564 (Gordon et al., 2014a), which implied the unidentified species were considerable to
565 SOA formation.

566 Vehicles dominated the POA emissions and OA productions in the urban
567 atmosphere of Shanghai. Their contributions to OA productions were about 40% and
568 60% in summer and winter, respectively. The rest of the contributors could be the
569 non-fossil VOC emissions from solvent use, chemical and petrochemical industrials,
570 etc. and the underestimated SOA productions from unidentified VOC or IVOC
571 species in the exhausts. At present, vehicles are experiencing rapid growth trends in
572 the cities of China. Primary emissions and secondary formation of OA derived from
573 vehicles will lead to further deterioration of fine particle pollution in the urban area.
574 Reduction of primary PM emissions and SOA precursors from vehicle exhausts will
575 be helpful to improve the air quality in the cities of China. The results also indicate
576 diesel exhausts dominate the POA emissions in the urban area. Therefore,
577 strengthening the primary PM emission control of diesel vehicles, especially for the

578 older diesel vehicles with loose emission standards as shown by Fig. 2, plays an
579 important role in OA pollution prevention. Now China is conducting the large-scale
580 elimination of “yellow-labeled” diesel vehicles whose emission standards were lower
581 than Euro 3. It can be expected to effectively reduce the OA pollution caused by
582 diesel vehicles. On the other hand, gasoline exhausts have high potential impacts on
583 SOA formation in the urban area. Tightening the limit of aromatic contents in gasoline
584 fuel will be meaningful to reduce the SOA contributions of gasoline vehicles.

585 There are still some uncertainties need to be improved in the future. First is the
586 SOA mass yield. More experiments on SOA yields of vehicle exhausts in China will
587 be helpful to the SOA formation potentials of different vehicle types. Especially for
588 gasoline exhausts, the estimated SOA yield was much lower than the experiment
589 results in the US. Vehicular OA contributions will increase about 4%-5% if we replace
590 the estimated SOA yield of gasoline exhaust to the experiment result. It will be
591 meaningful to find out their actual SOA yield and key precursors for urban OA
592 pollution control. Emission inventory is another important source of uncertainty in
593 this study. To reduce the uncertainty of vehicular emission inventory, we localized the
594 vehicle mileage and emission factor data based on the traffic surveys in Shanghai and
595 real-world measurements in some cities of China. However, the CO emission
596 inventories of other sources shown in Fig. S5 still have large uncertainties according
597 to the previous study (Huang et al., 2011). More accurate emission inventory will be
598 helpful to reduce the uncertainty of vehicular OA contribution in this study. However,
599 it can be concluded that vehicle emissions are the most important contributors to OA
600 pollution in the cities of China. Another implication is the potential roles of IVOCs in
601 vehicle exhausts are very important on the SOA formation in the urban area.
602 Therefore, further studies need to pay more attentions to determine the contributions
603 of IVOC emissions to OA pollution in China.

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857 **Table 1.** Daily VKT and average speeds of various vehicle and road types in Shanghai in 2012.

Road types	Daily vehicle kilometers traveled (million km)							Average speed (km·h ⁻¹)	
	Light-duty car	Light-duty truck	Taxi	Heavy-duty bus	Heavy-duty truck	City bus	Motor-cycle		Total
Highway	38.3	0.62	3.96	3.10	11.82	0.23	0.00	58.04	57.9
Arterial road	22.6	2.93	6.16	1.12	4.73	1.58	5.29	44.41	36.0
Residential road	18.3	3.11	8.92	0.89	1.90	1.38	4.15	38.64	28.5
Total	79.2	6.66	19.04	5.11	18.45	3.19	9.44	141.09	43.0

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Table 2. Test vehicle specifications.

ID	Vehicle type	Fuel type	Emission standard	Model year	Odometer reading (km)
LDGV-1	Light-duty car	Gasoline	Euro 1	2002	245306
LDGV-2	Light-duty car	Gasoline	Euro 2	2005	59790
LDGV-3	Light-duty car	Gasoline	Euro 3	2008	87662
LDGV-4	Light-duty car	Gasoline	Euro 3	2008	80856
Taxi-1	Light-duty taxi	Gasoline	Euro 1	2001	270000
Taxi-2	Light-duty taxi	Gasoline	Euro 1	2002	~100000
Taxi-3	Light-duty taxi	Gasoline	Euro 2	2003	99638
Taxi-4	Light-duty taxi	Gasoline	Euro 3	2007	281315
Taxi-5	Light-duty taxi	Gasoline	Euro 3	2008	361180
HDDT-1	Heavy-duty truck	Diesel	Euro 1	2003	331387
HDDT-2	Heavy-duty truck	Diesel	Euro 1	2003	271000
HDDT-3	Heavy-duty truck	Diesel	Euro 2	2004	271125
HDDT-4	Heavy-duty truck	Diesel	Euro 2	2004	204193
HDDT-5	Heavy-duty truck	Diesel	Euro 3	2009	70000
Bus-1	City bus	Diesel	Euro 2	2006	295236
Bus-2	City bus	Diesel	Euro 3	2006	175122
MT-1	Motorcycle	Gasoline	Euro 1	2003	15000
MT-2	Motorcycle	Gasoline	Euro 1	2003	11191
MT-3	Motorcycle	Gasoline	Euro 2	2004	96969
MT-4	Motorcycle	Gasoline	Euro 2	2003	13912
MT-5	Motorcycle	Gasoline	Euro 2	2003	5379

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Table 3. Vehicle emission inventory in Shanghai.

Vehicle type	Emission inventory (k ton)					POA (OC*1.2)
	CO	NOx	VOCs	EVA	EC	
in vehicle type						
LDGV	192.03	13.30	15.59	6.15	0.02	0.07
LDDV	1.89	5.72	0.32	0.00	0.17	0.11
Taxi	68.89	3.86	5.56	1.96	0.01	0.03
HDGV	36.79	2.20	2.29	0.29	0.00	0.01
HDDV	24.71	67.56	9.74	0.00	3.16	3.40
Bus	5.53	17.56	2.06	0.00	0.58	0.62
Motorcycle	14.01	0.67	3.85	0.49	0.02	0.06
in fuel type						
Gasoline	311.71	20.04	27.28	8.88	0.05	0.17
Diesel	32.14	90.84	12.12	0.00	3.91	4.13
Total	343.85	110.88	39.40	8.88	3.96	4.30

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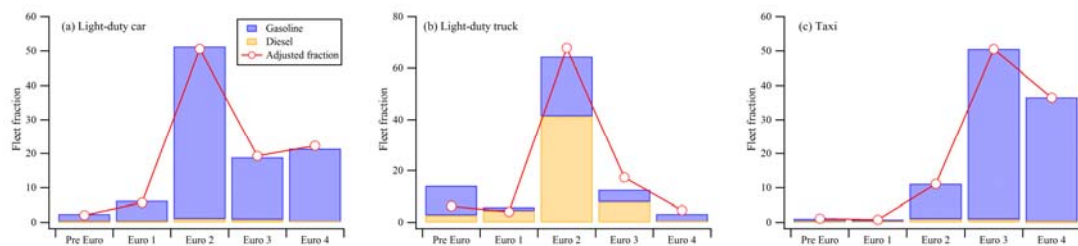
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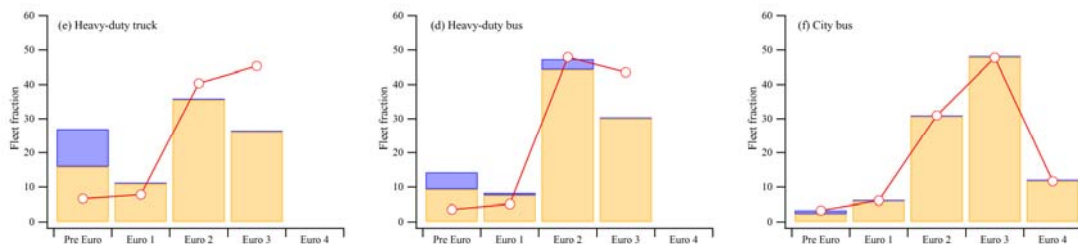
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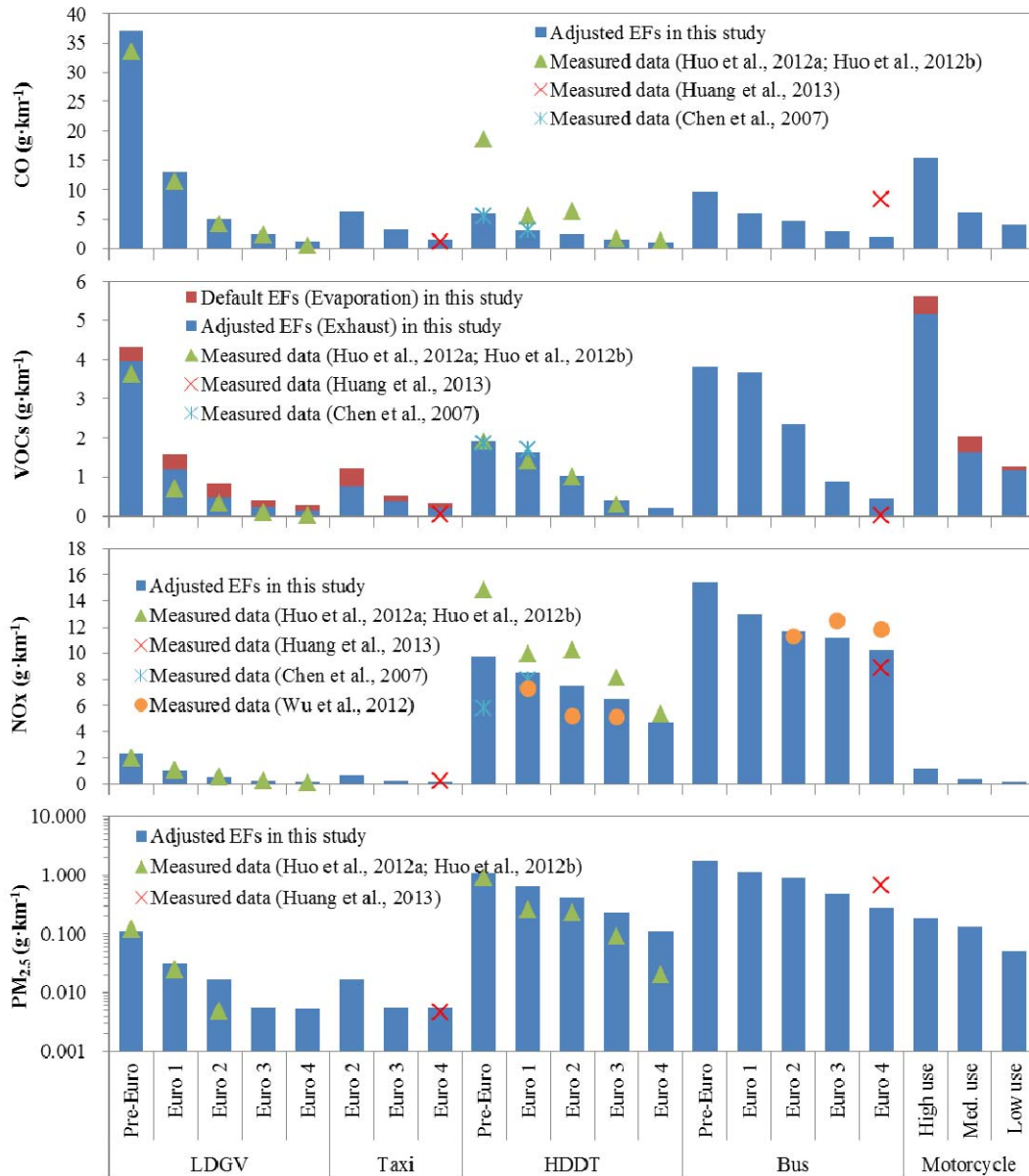
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Fig. 1. Static and adjusted fractions of each vehicle type in Shanghai.

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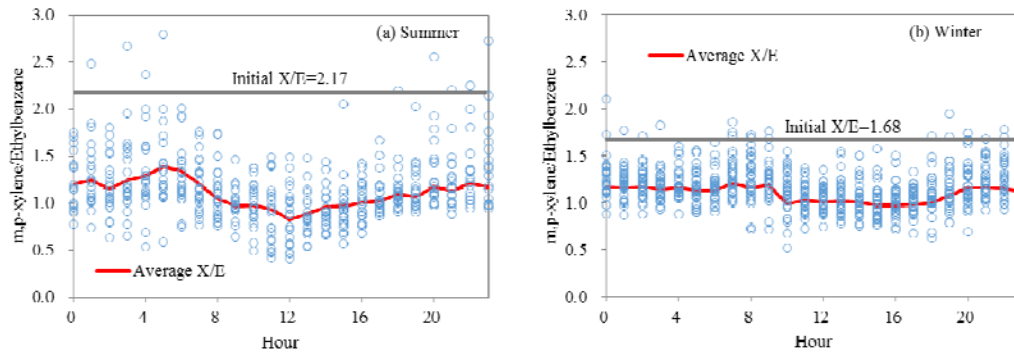
881 **Fig. 2.** Adjusted emission factors of various vehicle types (blue bars) and their comparisons with

882 measured emission factors in the previous studies (dots).

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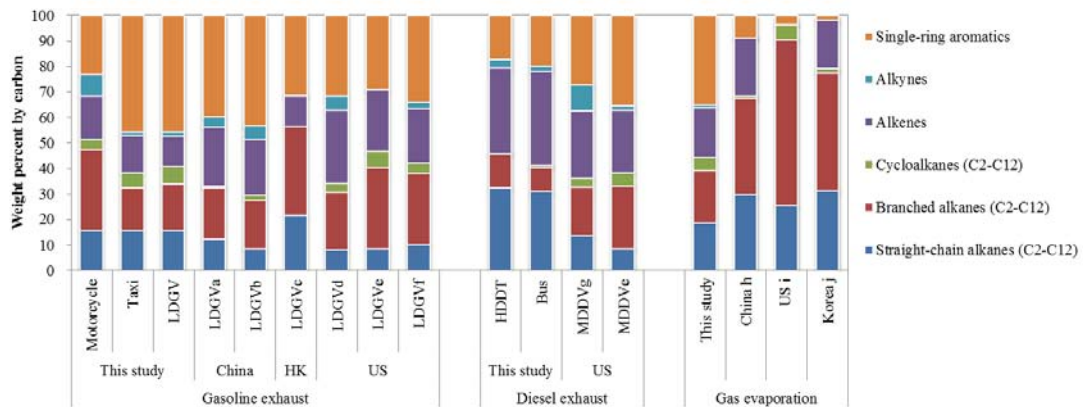
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887 **Fig. 3.** Diurnal distributions of the ratios of m,p-xylene to ethylbenzene concentrations in summer
 888 and winter in the urban atmosphere in 2013.

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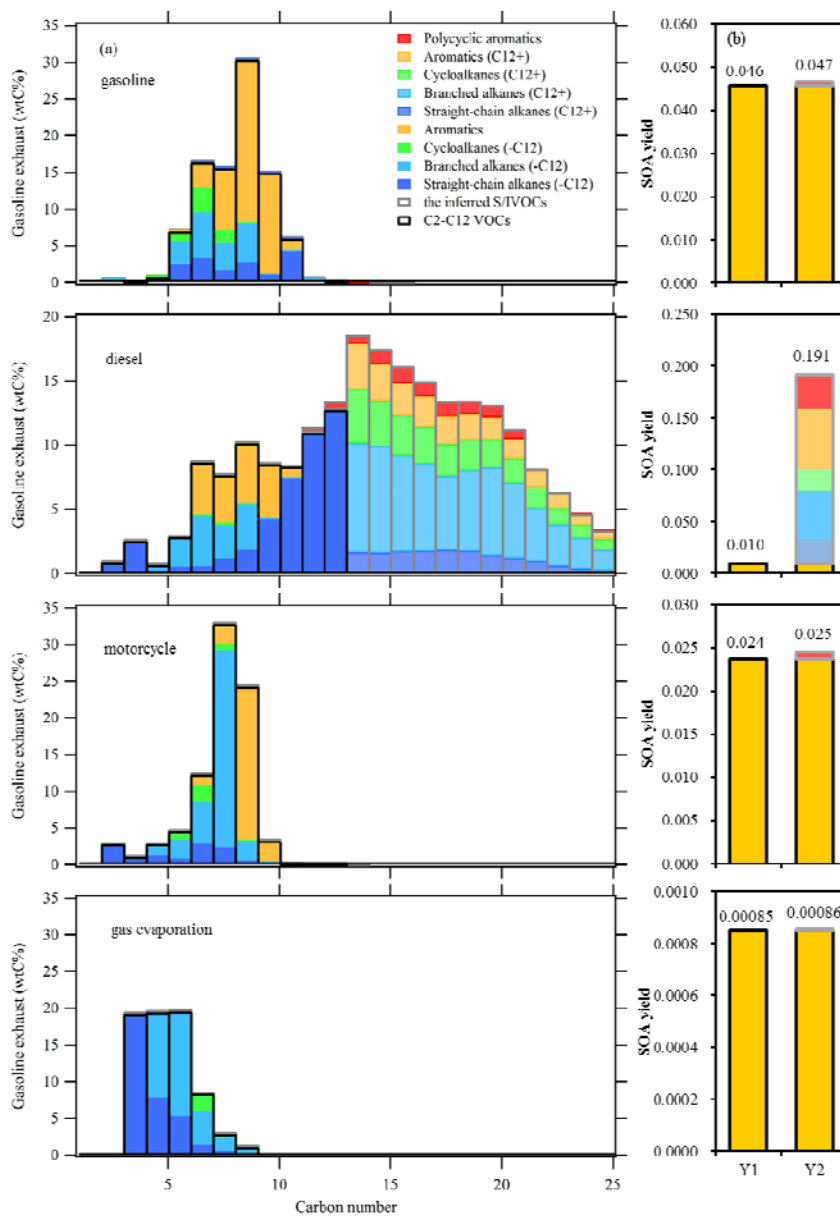
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893 **Fig. 4.** Comparisons of measured VOC compositions of the exhausts from different vehicle types
 894 and gas evaporation to the results in other studies (a. Liu et al., 2008; b. Wang et al., 2013; c. Guo
 895 et al., 2011; d. Schauer et al., 2002; e. May et al., 2014; f. Gentner et al., 2013; g. Schauer et al.,
 896 1999; h. Zhang et al., 2013; i. Harley et al., 2000; j. Na et al., 2004).

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901 **Fig. 5.** (a) Distributions of mass by chemical class in carbon number of different vehicle exhausts

902 and evaporative emissions; (b) Calculated SOA yields based on C2-C12 VOCs measured in this

903 study (Y1) and C2-C12 VOCs plus the inferred S/IVOC species (Y2). The inferred S/IVOC

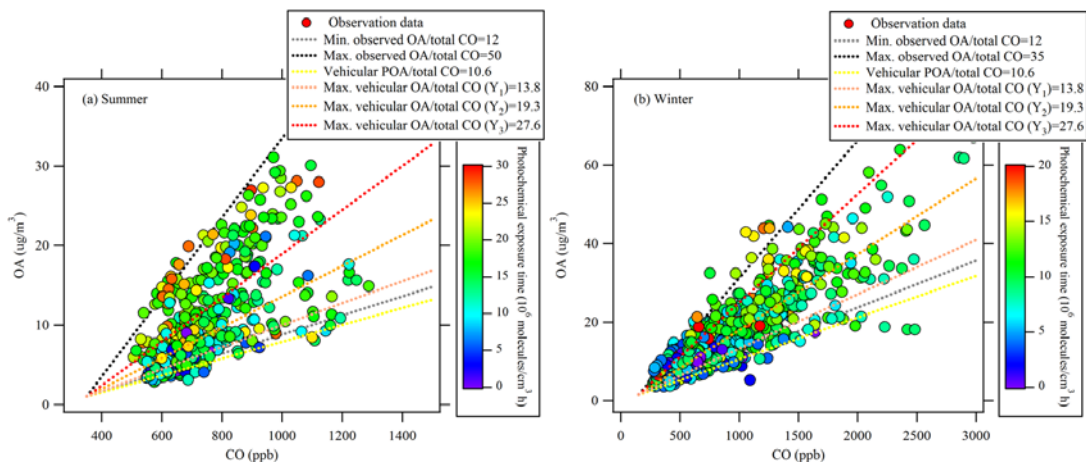
904 species are shown in light colored bars.

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910 **Fig. 6.** Relationship of measured OA and CO concentrations color-coded by the photochemical
 911 exposure in the summer (a) and winter (b) of 2013 in urban Shanghai according to equation (4).
 912 Minimum and maximum ratios of observed OA to CO concentrations are shown by dotted grey
 913 and black lines. Vehicular POA/Total CO is shown by dotted yellow line. The minimum and
 914 maximum OA formation ratios of vehicle emissions calculated with three different SOA yields of
 915 Y₁, Y₂ and Y₃ are shown by the dotted pink, orange and red lines, respectively.

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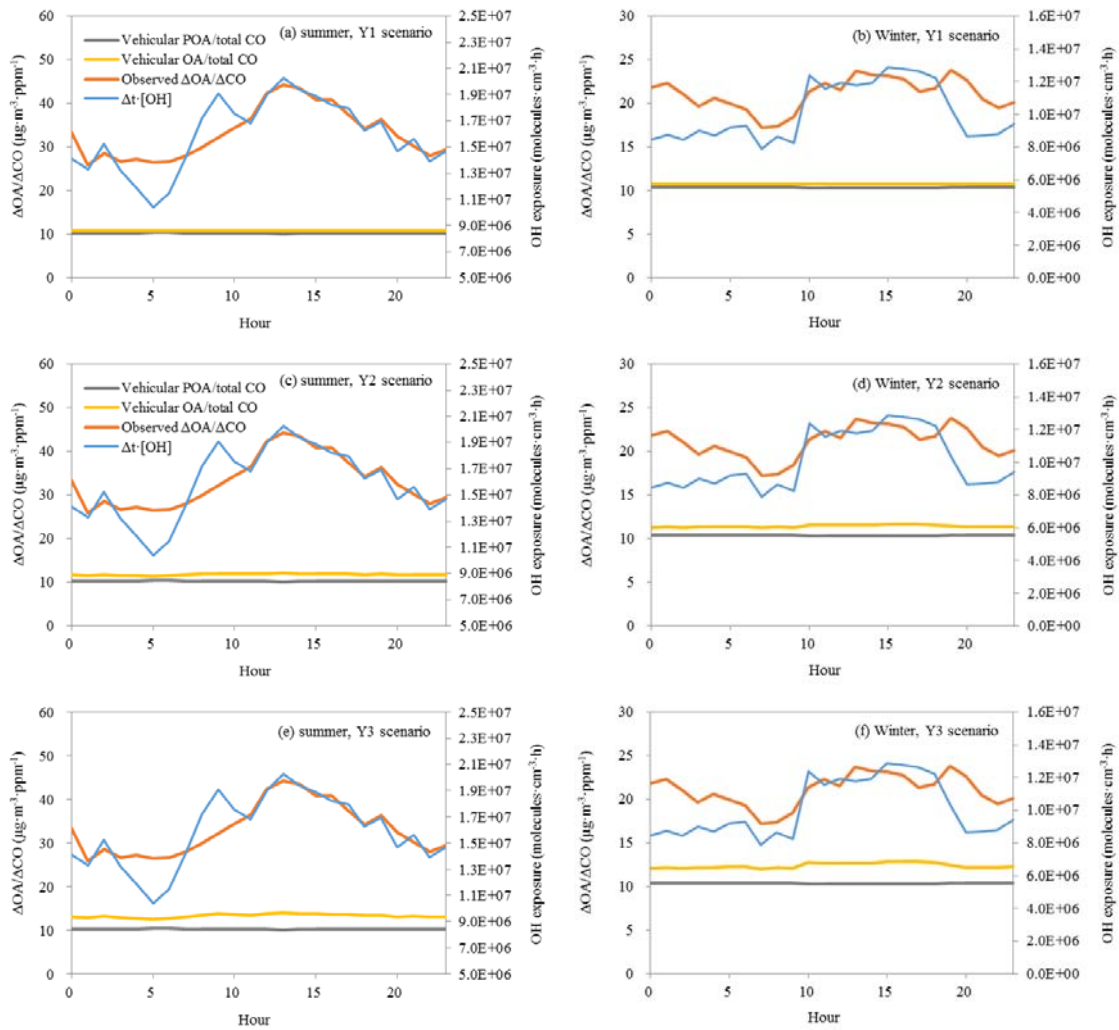
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930 **Fig. 7.** Diurnal variations of observed $\Delta\text{OA}/\Delta\text{CO}$ in the atmosphere (red line), OH exposures (blue
 931 line), and the ratios of vehicular POA emission (grey line) and OA formation (orange line) to total
 932 CO emissions with the SOA yields in three scenarios (Y1, Y2 and Y3) in summer and winter in the
 933 urban area of Shanghai for the year of 2013.

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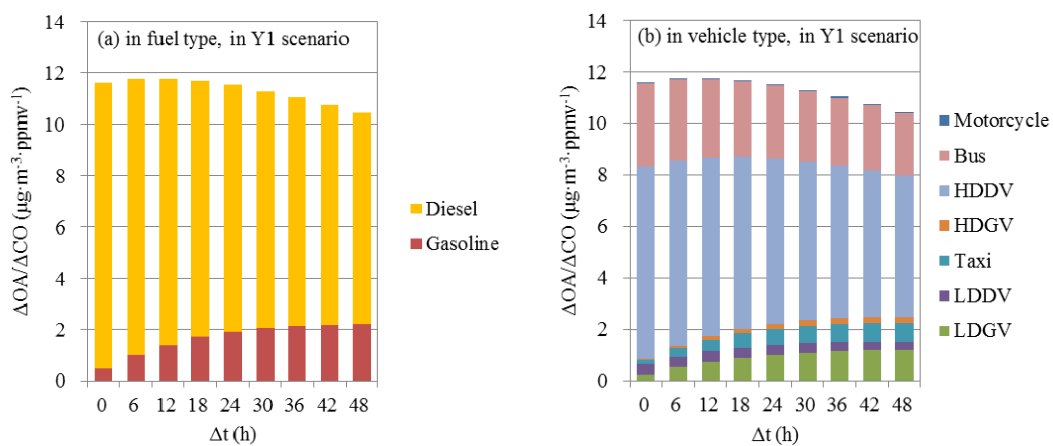
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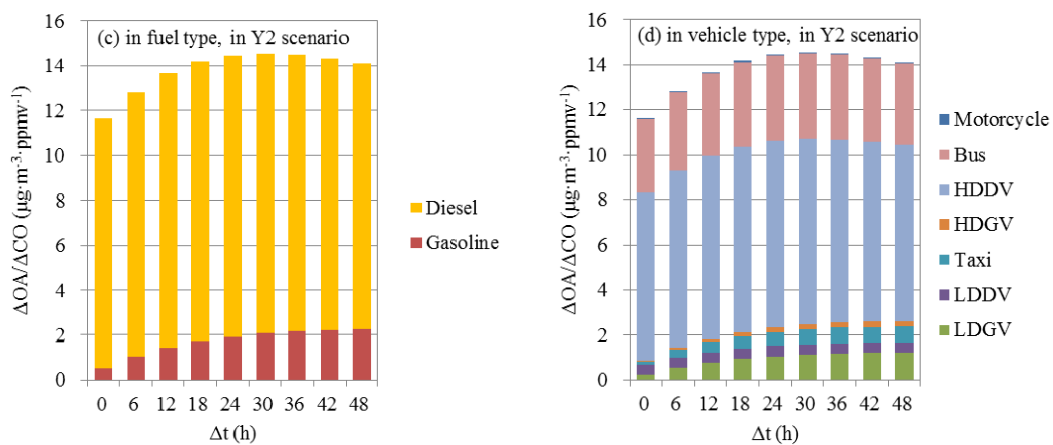
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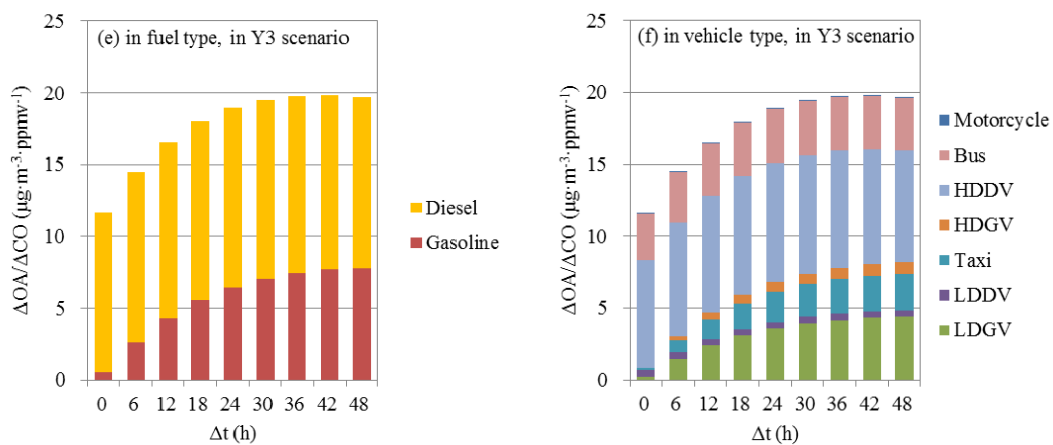
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947 **Fig. 8.** Contributions of vehicle emissions to OA formation ratios in different vehicle and fuel

948 types in Y1, Y2 and Y3 scenarios with the changes of photochemical ages.

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