

20 **Abstract**

21 Subtropical forests in South China have received chronically large amounts of
22 atmogetic nitrogen (N) causing N saturation. Recent studies suggest that a significant
23 proportion of the N input is returned to the atmosphere, in part as nitrous oxide (N₂O).
24 We measured N₂O emission fluxes by closed chamber technique throughout two
25 years in a Masson pine-dominated headwater catchment with acrisols (pH~4) at
26 TieShanPing (Chongqing, SW China) and assessed the spatial and temporal
27 variability in two landscape elements typical for this region: a mesic forested hill
28 slope (HS) and a hydrologically connected, terraced groundwater discharge zone
29 (GDZ) in the valley bottom. High emission rates of up to 1800 μg N₂O-N m⁻² hr⁻¹
30 were recorded on the HS shortly after rain storms during monsoonal summer, whereas
31 emission fluxes during the dry winter season were generally low. Overall, N₂O
32 emission was lower in GDZ than in HS, rendering the mesic HS the dominant source
33 of N₂O in this landscape. Temporal variability of N₂O emissions on HS was largely
34 explained by soil temperature and moisture, pointing at denitrification as a major
35 process for N removal and N₂O production. The concentration of nitrate (NO₃⁻) in
36 pore water on HS was high even in the rainy season, apparently never limiting
37 denitrification and N₂O production. The concentration of NO₃⁻ decreased along the
38 terraced GDZ, indicating efficient N removal, but with moderate N₂O-N loss. The
39 extrapolated annual N₂O fluxes from soils on HS (0.54 and 0.43 g N₂O-N m⁻² yr⁻¹ for
40 a year with a wet and a dry summer, respectively) are among the highest N₂O fluxes

41 reported from subtropical forests so far. Annual N_2O -N emissions amounted to 8-10%
42 of the annual atmospheric N deposition, suggesting that forests on acid soils in South
43 China are an important, hitherto overlooked component of the anthropogenic N_2O
44 budget.

45 **1. Introduction**

46 The global atmospheric concentration of nitrous oxide (N_2O), an important
47 greenhouse gas and decomposer of stratospheric ozone, has increased from a
48 pre-industrial level of 270 ppbv to 322 ppbv in 2008 (WMO, 2009). The global source
49 strength is estimated to be $17.7 \text{ Tg N yr}^{-1}$, with agriculture contributing 2.8 (1.7-4.8)
50 Tg N yr^{-1} and soils under natural vegetation 6.6 (3.3-9.0) Tg N yr^{-1} (IPCC, 2007;
51 Hirsch et al., 2006). Both, estimates based on bottom-up approaches (Stehfest and
52 Bouwman, 2006) and on observations of the N_2O atmospheric column (Kort et al.,
53 2011; D'Amelio et al., 2009; Hirsch et al., 2006) suggest that 50-64% of the
54 atmospheric N_2O derive from the (sub)tropical zone 0° to 30°N . Much of the N_2O
55 attributable to this zone is emitted from forest soils (Melillo et al., 2001; Werner et al.,
56 2007a; Rowlings et al., 2012). So far, N_2O flux data from subtropical forests are
57 scarce making this biome an under-investigated component of the global N_2O budget.

58 Large parts of southern China are situated in the humid subtropics and the dominant
59 forest types are evergreen broadleaf and coniferous forest, many of which are found
60 as patches in densely populated areas with intensive agriculture. Due to strong
61 increases in the emission of nitrogen oxides (NO_x) and ammonia (NH_x), caused by
62 combustion of fossil fuels and massive fertilizer use, respectively (Xiong et al., 2008;
63 Liu et al., 2011b), these forests receive high amounts of reactive nitrogen (N_r) by
64 atmogenic deposition, mostly as ammonium (NH_4^+). A recent study of five forested

65 watersheds in South China found various degrees of N saturation (Chen and Mulder,
66 2007a), i.e. rates of atmospheric N_r input exceeding uptake by vegetation (Aber et al.,
67 2003) and causing elevated concentrations of nitrate (NO_3^-) in the root zone. Larsen
68 et al. (2011) reported moderate NO_3^- export with stream water in these watersheds,
69 despite low forest productivity, suggesting major unaccounted N sinks in these
70 ecosystems.

71 One possible fate of excess N may be gaseous emission as nitric oxide (NO), N_2O or
72 dinitrogen (N_2) produced during nitrification and denitrification. Known factors
73 controlling N_2O emissions are availability of inorganic N and degradable organic
74 carbon (C), soil temperature, soil moisture and soil pH (Parton et al., 1996; Flessa et
75 al., 1995; Smith et al., 2003; Weier et al., 1993; Simek and Cooper, 2002). The
76 monsoonal climate in South China provides favorable conditions for both nitrification
77 and denitrification as much of the annual N input, dominated by NH_4^+ , occurs during
78 rain storms in summer when soils are warm and N turnover rates high. Moreover,
79 forests in South China are often found on acidic soils. Acidity has been reported to
80 support high N_2O/N_2 product ratios in denitrification (Liu et al., 2010b; Bergaust et al.,
81 2010) which could result in high N_2O emissions. Acidity also inhibits autotrophic
82 nitrification by reducing the availability of ammonia (NH_3), the actual substrate for
83 membrane-bound ammonia mono-oxygenase in ammonia oxidizing bacteria (De Boer
84 and Kowalchuk 2001); on the other hand, low soil pH may enhance the apparent N_2O
85 yield of nitrification, possibly involving chemodenitrification of nitrite (NO_2^-) from

86 either nitrification or dissimilatory NO_2^- reduction (Mørkved et al. 2007). Therefore
87 the net effect of acidity on the production of nitrification-related N_2O is still uncertain.

88 Few studies have addressed N_2O emissions in Chinese subtropical forests. Tang et al.
89 (2006) studied the temporal variation of N_2O fluxes in a pine, a mixed and an
90 evergreen broadleaf forest on acid soils in the Dinghushan catchment, South China
91 throughout one year and found highest emissions during the wet-hot season. Based on
92 weekly to biweekly measurements, they estimated an annual emission of 3.2 kg
93 $\text{N}_2\text{O-N ha}^{-1}$ which is above the reported average N_2O emission of 1.2–1.4 kg N ha^{-1}
94 yr^{-1} for tropical forests (Stehfest and Bouwman, 2006; Werner et al., 2007b). Fang et
95 al. (2009) reported 2.0 to 2.4 kg $\text{N}_2\text{O-N loss ha}^{-1} \text{ yr}^{-1}$ based on monthly measurements
96 along a hillslope in an evergreen broadleaf forest in the same catchment. Lin et al.
97 (2010; 2012) estimated smaller average annual N_2O emissions of 0.13 and 0.71 kg N
98 ha^{-1} for pine forests in Hubei province, probably due to the higher pH of these forest
99 soils.

100 Estimates of annual N_2O emissions in forested catchments are fraught by large
101 temporal and spatial variability commonly reported for N_2O fluxes. Forests in
102 subtropical China are typically found on sloping terrain and mountain ridges, resulting
103 in hydrological gradients known to affect N_2O emissions. For example, Fang et al.
104 (2009) found that N_2O emissions increased with soil moisture from hilltop to the
105 bottom of a hill slope. Besides soil moisture, other soil factors, such as pH, texture,

106 vegetation type and productivity may vary along hydrological flow paths, resulting in
107 distinct spatial patterns of N₂O emissions (Osaka et al., 2006; Lark et al., 2004).
108 Ultimately, soil factors related to topography may shape distinct nitrifier and
109 denitrifier communities with respect to N₂O turnover (Philippot et al., 2009; Dörsch et
110 al., 2012; Wessen et al., 2011; Banerjee and Siciliano, 2012). Thus, understanding of
111 spatial and temporal variability of N₂O emission fluxes and their drivers on a
112 landscape level are indispensable for improving regional estimates of N₂O emissions.

113 The objective of the present study was to estimate N₂O fluxes in an N-saturated
114 forested catchment in Southwest China and to explore their main drivers in space and
115 time. More specifically, we investigated the seasonal distribution of emission fluxes
116 (dry versus wet season), the role of storm flow conditions on peak N₂O emissions and
117 the spatial distribution of N₂O emissions along a hydrological flow path on a hill
118 slope (HS) and a groundwater discharge zone (GDZ). We hypothesized that the
119 emission strength is higher during the wet season than the dry season and that a
120 substantial part of the annual N₂O emission occurs during transient N₂O emission
121 peaks triggered by rain episodes. In addition, based on previous observations that the
122 water-saturated GDZ has large N retention (Larssen et al. 2011), we hypothesized the
123 GDZ to act as a zone of increased denitrification and hence high N₂O emission
124 relative to the well-drained HS. Emissions of N₂O were measured together with soil
125 parameters from summer 2009 to autumn 2010, along two transects in each of the two
126 landscape elements, hillslope (HS) and ground water discharge zone (GDZ). To

127 improve flux estimates in time, a regression model is proposed interpolating
128 temporally discrete N₂O measurements on the HS based on continuously recorded soil
129 moisture and temperature.
130

131 **2. Materials and methods**

132 2.1 Site description

133 The Tieshanping (TSP) catchment is a 16.2 ha headwater catchment, 450 m asl,
134 located on a forested ridge about 25 km Northeast of Chongqing city, SW China
135 (29°38'N 104°41'E, Fig. 1a). Details on climate, vegetation and soil characteristics as
136 well as atmospheric N deposition can be found elsewhere (Chen and Mulder, 2007a).

137 The area has a typical subtropical monsoonal climate with a mean annual precipitation
138 of 1028 mm and a mean annual temperature of 18.2°C (three year average,
139 2001-2003). 75% of the rainfall occurs during summer (April to September). Mean
140 annual inorganic N deposition (from 2001-2003) was 4 g m⁻², 61% of which occurred
141 in the form of NH₄⁺-N (Chen and Mulder, 2007b). Data for more recent years show
142 increasing N deposition rates to > 5 g m⁻² yr⁻¹ and confirm the relative importance of
143 NH₄⁺ (Lei Duan, *pers. comm.*). The vegetation is a coniferous-broadleaf mixed forest
144 dominated by Masson pine (*Pinus massoniana*) with a well-developed understory of
145 evergreen shrubs (Chen and Mulder, 2007a). Land use is a naturally regenerated,
146 non-managed secondary forest after the original forest was cut during 1958-1962.

147 For the present study, a representative 4.6 ha sub-catchment was selected, consisting
148 of a NE facing hill slope (HS) and a hydrologically connected SE - NW oriented,
149 terraced valley bottom acting as a ground water discharge zone (GDZ) (Fig. 1b). The
150 HS is dominated by acidic, loamy yellow mountain soils (Haplic Acrisols; WRB, 2006),

151 developed from sandstone, the predominant bedrock in the area. The soils have only a
152 thin organic layer (O horizon; 0-2 cm) (Table 1), likely reflecting high turnover rates
153 of soil organic matter (SOM) in the warm and wet climate (Raich and Schlesinger,
154 1992; Zhou et al., 2008). Soils in the GDZ have developed from colluvium (Cambisols)
155 and were terraced during the 1960s for vegetable cultivation but abandoned shortly
156 after. The soils in the GDZ have a low hydraulic conductivity (Sørbotten, 2011), lack
157 horizontal differentiation (no distinct O horizon) and have high ground water levels.
158 The vegetation in the GDZ consists of sparse shrubs and grasses while trees taller than
159 2 m are absent. Clay mineralogy on HS and in GDZ is dominated by kaolinite. A
160 recent study (Sørbotten, 2011) showed that considerable interflow occurs along HS
161 during rainfall episodes. Excess water moves laterally down slope in surface horizons
162 overlaying the argic B horizon, which has a low hydrological conductivity, thus
163 draining the O/A and AB horizons on the HS relatively quickly. Discharge of
164 interflow into GDZ results in episodically high ground water levels and periodic
165 water-logging during summer (Sørbotten, 2011). Major soil characteristics for both
166 locations are summarized in Table 1.

167 2.2 N₂O flux measurements

168 N₂O emissions were measured manually by closed chamber technique along two
169 transects in the sub-catchment. Transect (T) was established perpendicular to the
170 contour lines on the HS stretching from the hill top close to the watershed divide to

171 the hill bottom close to the GDZ. Transect (B) was established in the GDZ following
172 an elevation gradient of six hydrologically connected terraces (Fig. 1b) towards a
173 flume, which was used to measure runoff from the sub-catchment. Six plots for N₂O
174 emission measurements (n=3) and sampling of soil and pore water were established
175 along the T transect on HS (T1 to T5 and B1) and five plots along the B transect in
176 GDZ (B2 to B6; Fig. 1b) with plot B1 representing the transition between HS and
177 GDZ. Two sampling strategies were applied to explore N₂O fluxes. A first
178 measurement campaign during summer 2009 (July to September) with frequent flux
179 measurements explored N₂O emission dynamics in response to sudden increases in
180 soil moisture after rainfall episodes. A more regular sampling scheme with biweekly
181 measurements was applied throughout the dry-cool season (Nov. 2009 to Mar. 2010)
182 and during the subsequent summer (from Apr. to Aug. 2010), interrupted by more
183 intensive measurement campaigns after individual rain storms (> 20 mm). For logistic
184 reasons, only a selection of plots (T1, T3, T5, B1, B2, B4 and B6) was monitored
185 during the dry season. In total 38 triplicate flux measurements were conducted on
186 plots T1, T3, T5, B1, B2, B4 and B6 and 20 on plots T2, T4, B3 and B5.

187 For flux sampling, closed, vented, zinc-coated iron chambers (30 cm diameter and 30
188 cm in height) (Hutchinson and Mosier 1981) were carefully pushed app. 2 cm into the
189 soil. Given the loamy texture and the limited number of roots in the mostly moist
190 surface soil, this deployment method was preferred over installing permanent
191 chamber bases which might impair surface run off and interflow during rain storms.

192 The chamber deployment was quality assured by checking for deviations of initial (1
193 min after deployment) chamber N₂O concentration from ambient (~0.34 ppmv) and
194 using CO₂ accumulation over time (not shown) to check for leakage, both of which
195 were regular in most of the cases. Gas samples were taken from a sampling port on
196 top of the chambers at 1, 5, 15 and 30 minutes after deployment by a 20-ml plastic
197 syringe. Samples were transferred immediately to pre-evacuated 12 ml-vials crimp
198 sealed with butyl septa (Chromacol, UK), resulting in an overpressure to avoid
199 contamination due to pressure changes during shipment. Chamber temperature was
200 recorded by inserting a handheld digital thermometer into the sampling port at
201 beginning and end of chamber deployment. Gas samples were shipped to the
202 Norwegian University of Life Sciences and analyzed within half a year after sampling.
203 The vials kept overpressure during shipment and storage. The samples were analyzed
204 by ECD-gas chromatography (Model 7890A, Agilent, Santa Clara, CA, US). Carbon
205 dioxide (CO₂) and N₂O were separated on a 20 m wide-bore (0.53 mm diameter)
206 Poraplot Q column run at 38°C after passing a packed Heysept column used for
207 back-flushing water. Helium (He 5.0) was used as carrier gas. The ECD was run at
208 375°C with 17 ml min⁻¹ Ar/CH₄ (90/10 vol%) as make-up gas. N₂O emission rates
209 (µg N m⁻² hr⁻¹) were calculated based on the rate of change in N₂O concentration in
210 the chamber, the internal volume of the chamber, the covered soil surface area and the
211 average chamber temperature during deployment. For determining the rate of change
212 in N₂O concentration in the chamber, which is the slope of a linear or a second order

213 polynomial fit of the concentration data against time, the R-square of both regressions
214 were compared. Cumulative N₂O emissions for observation periods in summer 2009
215 (27 days, Jul. 12 to Aug. 8) and summer 2010 (106 days, May 10 to Aug. 24) were
216 calculated assuming a constant flux rate between sampling dates. For the calculation
217 of cumulative N₂O emissions, only the periods with sampling frequency high enough
218 to obtain the cumulative N₂O emission rates of the whole summer based on
219 measurements were used.

220 2.3 Soil sampling and analysis

221 Composite soil samples were taken on Oct. 15, 2009 from the eleven plots used for
222 N₂O flux measurements, and analyzed for pH, total organic carbon (TOC) and total
223 nitrogen (TN). Samples at plots T1 to T5 and B1 were taken from the O- (0-2cm), A-
224 (2-8cm), AB- (8-20cm) and B- (20-40cm) horizons. In GDZ (plots B2 to B6), where
225 an organic horizon was absent, the homogeneous 0-20 cm and 20-40 layers were
226 sampled. Denitrification characteristics of the soils, investigated in a laboratory study,
227 are reported elsewhere (Zhu et al. 2013). Briefly, the instantaneous denitrification rate
228 (expressed in nmol N₂O-N g⁻¹ dw soil hr⁻¹) in soils from the different plots was
229 estimated from the N₂O accumulation rate in acetylene-treated, anoxic soil slurries at
230 20°C in the presence of ample NO₃⁻-N (2 mM). The soils' *ex situ* potential for N₂O
231 loss (in nmol NO₃⁻-N g⁻¹ dw soil) was estimated as the amount of NO₃⁻-N respired to
232 N₂O before N₂O reductase activity was fully expressed and rapid N₂O reduction to N₂

233 occurred. Since instantaneous denitrification rates were low in the deeper horizons,
234 only data from the O and A horizons of HS and the 0-20 cm layer of GDZ are
235 presented here.

236 Soil pH (H₂O) was measured in a suspension of 10 g dry weight (dw) soil in 50 ml
237 de-ionized water with an Orion SA720 electrode pH-meter connected to an Orion
238 ROSS Ultra pH Electrode. The contents of TOC and TN in the soil were measured in
239 the fine earth fraction (< 2mm), obtained after drying at 105°C and sieving, using an
240 element analyser (Vario EL III, Elementar Analysensysteme GmbH, Germany) at the
241 Research Center of Eco-environmental Science, Chinese Academy of Science
242 (RCEES, CAS). In addition, soil samples from the horizons and layers of all plots
243 were collected three times (Oct. 15, 2009, Jan. 22, 2010 and Apr. 29, 2010) to
244 measure KCl-extractable nitrate (NO₃⁻_{ex}) and ammonium (NH₄⁺_{ex}). After being
245 sampled, soils were stored in a freezer and extracted within one month, except the soil
246 taken on Jan. 22, 2010, which was extracted together with the samples taken on Apr.
247 29, 2010. Soil samples were extracted by shaking 10 g dw equivalents of fresh soil
248 with 50 ml of 2M KCl for 1 hr. The concentrations of NO₃⁻ and NH₄⁺ in the filtered
249 extract were determined using Flow Injection Analysis (San⁺⁺, Skalar, the
250 Netherlands) at RCEES, CAS.

251 Ceramic suction cup lysimeters (P80; Staatliche Porzellanmanufaktur, Berlin) and
252 Macrorhizon soil moisture sampler (Rhizosphere Research Products, The Netherlands)

253 were installed at each plot on HS and GDZ, respectively. On the HS, lysimeters were
254 installed in triplicate in the A- (4-5 cm), the top of the AB- (10 cm), the bottom of the
255 AB- (20 cm) and in the B horizon (40 cm). In the GDZ, one Macrorhizon was
256 installed at each of three depths (30, 60 and 100 cm). Soil water from HS was
257 sampled from Nov. 11, 2009, while groundwater in the GDZ was sampled from Aug.
258 5, 2009; both soil water and groundwater were sampled at weekly intervals until
259 spring 2010 and on each N₂O flux measurement date in summer 2010. Soil water
260 from triplicate lysimeters per soil depth on HS was pooled prior to analysis.
261 Concentrations of NO₃⁻ and NH₄⁺ in lysimeter water were measured in each sample
262 during wet-hot seasons and in pooled samples (four weeks) during the dry-cool season,
263 using ion chromatography (DX-120 for NH₄⁺ and DX-500 for NO₃⁻, DIONEX, USA)
264 at the Chongqing Academy of Environmental Sciences and Monitoring, China.

265 Soil bulk density (BD) of surface horizons/layers was measured from intact soil cores
266 (100cc steel cylinders) sampled in triplicate at two depths (3.5–7.2 cm and 10-13.7
267 cm), corresponding to the A and AB horizons, respectively (Sørbotten, 2011).

268 Soil temperature (ST) and volumetric moisture (VM, cm³/cm³) at 10 cm depth were
269 recorded every 10 minutes by TDR probes (Hydra Probe II) installed at plots T3 and
270 B1 on Oct. 11, 2009 and stored by a data logger (Campbell CR200). Data were
271 obtained until May 4, 2011, with a malfunctioning period from Dec. 25, 2010 to Feb.
272 8, 2011. Soil water filled pore space (WFPS) was calculated using bulk density (BD)

273 at depth of 10.0 - 13.7 cm, assuming a soil particle density (PD) of 2.65 g cm⁻³ (Linn
274 and Doran 1984) as:

$$275 \quad WFPS(\%) = \frac{VM}{1 - \left(\frac{BD}{PD}\right)} \times 100 \quad (\text{Eq. 1})$$

276 Groundwater level was monitored in PVC tubes, with a 30-cm nylon filter at the
277 lower end, and installed at each plot along the B-transect in GDZ (B1 to B6). The
278 PVC tubes were covered with a perforated lid to prevent pressure differences with the
279 atmosphere. Groundwater level was measured daily from Jul. 2009 to the beginning
280 of Aug. 2009 during a period of heavy rainstorms and on dates of N₂O measurements
281 thereafter.

282 Meteorological data (air temperature (AT), precipitation (precip) and relative
283 humidity (RH)) were obtained every 5 minutes from a weather station (WeatherHawk
284 232, USA), at the roof of the TSP Forest Bureau situated 1 km south of the catchment.

285 Vapor pressure deficit (VPD, kPa) was calculated from equation 2 (ASCE, 2005) as:

$$286 \quad VPD = \left(\frac{RH}{100} - 1\right) * 0.6108 * \exp(17.27 * AT / (AT + 237.3)) \quad (\text{Eq. 2})$$

287 2.4 Statistical analysis

288 Cumulative N₂O emissions for each of the eleven plots for summer 2010 (106 days)
289 were analyzed together with soil parameters in the top soil (depth-weighted for O and
290 A horizons on HS and from 0-20 cm in GDZ; Table 1) by Principal Component
291 Analysis (PCA). Bulk density, cumulative N₂O flux, NH₄⁺_{sw} and NO₃⁻_{sw} were

292 normalized using minus reciprocal, natural logarithmic, natural logarithmic and
293 sinusoidal transformation, respectively.

294 To explore the seasonal drivers of the N₂O emission flux, log-transformed average
295 N₂O emission rates on HS (average of 4 plots, T1, T3, T5 and B1, which had full N₂O
296 flux datasets including the dry season) were subjected to stepwise multiple linear
297 regression with soil temperature (ST), water filled pore space (WFPS) (average of
298 daily ST and WFPS obtained from TDR probes at both T3 and B1 from Oct. 12, 2009
299 onwards), as well as pore water NO₃⁻_{sw} and NH₄⁺_{sw} concentration in these four plots
300 as independent variables. The resulting empirical relation was used to interpolate the
301 N₂O flux on HS between measurement dates on the basis of continuously measured
302 soil temperature and WFPS and to estimate the annual N₂O emission of two complete
303 annual periods: one from May 5, 2009 to May 4, 2010 and the other one from May 5,
304 2010 to May 4, 2011.

305 Missing values for soil temperature (ST) and water filled pore space (WFPS) (from
306 May 1 to Oct. 11, 2009 and from Dec. 25, 2010 to Feb. 8, 2011) were estimated on a
307 daily basis, using a statistical model based on multiple linear regression, using
308 available meteorological data (average air temperature, rainfall and average VPD).
309 General linear model (GLM) with the mixed-effects (sampling dates and location)
310 was used for the comparison of different plots, the two landscape elements and
311 different seasons.

312 All statistical analyses were conducted with Minitab 16.1.1 (Minitab Inc.).

313

314 **3. Results**

315 3.1 Weather, soil moisture and soil temperature

316 Rainfall distribution during the wet season (Apr. to Oct.) differed between the two
317 years (Fig. 2). Summer 2009 (1054 mm) was wetter than summer 2010 (850 mm) and
318 precipitation occurred mostly as intensive rainstorms (up to 385 mm during 41 hours
319 between 3rd and 5th of Aug. 2009). In contrast, precipitation was more evenly
320 distributed in summer 2010. Only 117 mm rain fell in the dry-cool season from
321 November 2009 to March 2010. Soil water filled pore space (WFPS) at 10 cm depth
322 on the mid slope (T3, mean WFPS 55.8%) was significantly smaller than at the foot
323 slope of HS at the transition to GDZ (B1, mean WFPS 70.0%) (Fig. 3a), most likely
324 reflecting convergence of interflow from HS at B1. In winter, water filled pore space
325 (WFPS) was relatively stable with values around 56.7% for T3 and 68.5% for B1. In
326 contrast, large variations in water filled pore space (WFPS) were observed in spring
327 and summer, driven by precipitation, and subsequent drainage and evapotranspiration.
328 In plot T3 (HS), but not in plot B1 (transition to GDZ), water filled pore space (WFPS)
329 decreased significantly in late summer 2010 (July through September), indicating
330 drier conditions on the upper part of HS as described earlier for this site (Mulder et al.,
331 2005). Soil temperature at 10 cm depth varied seasonally between 10°C and 25°C with
332 no significant difference between T3 and B1, although the drier T3 plot tended to
333 have somewhat higher soil temperatures during the wet season (Fig. 3b).

334 The groundwater level in GDZ varied between +15 cm (above surface) and < -100
335 cm (below surface), with highest values at B5 and lowest at B1 (Fig. 4). The heavy
336 rainstorm between Aug. 3rd and 5th, 2009 resulted in a rapid increase of the
337 groundwater level. The groundwater level reached its lowest values from late summer
338 onwards, but increased in early spring, due to increased precipitation in Mar. and Apr.,
339 2010. In general, the groundwater level was deepest at B1 and increased to
340 near-surface in the order B1<B2=B3=B4<B5<B6 and was most stable close to the
341 outlet (plots B5 and B6). The groundwater level was only occasionally detectable
342 (>-100 cm depth) at plots B1, B2 and B3 (Fig. 4).

343 3.2 N₂O fluxes

344 Emission fluxes of N₂O showed a pronounced seasonal pattern with highest values
345 during summer (wet-hot season) and significantly lower emission rates during winter
346 (dry-cool season) (p=0.000, Fig. 5). During summer, N₂O emission rates varied 1-2
347 orders of magnitude, with highest values immediately following precipitation events
348 (compare Figs. 2 and 5) when water filled pore space (WFPS) was high (Fig. 3a). In
349 summer 2010, with its more even distribution of rainfall, peak N₂O emissions were
350 smaller (up to 450 μg m⁻² hr⁻¹) than in summer 2009 (up to 1730 μg N m⁻² hr⁻¹) (Fig.
351 5). The analysis with a general linear model (GLM) indicated that N₂O emission rates
352 were significantly higher on HS than in GDZ (p=0.011). The daily average N₂O
353 emission on HS was higher in summer 2009 (6.4 mg N m⁻² d⁻¹) than in summer 2010

354 (3.2 mg N m⁻² d⁻¹; Fig. 6). Furthermore, the average N₂O emission was smaller in
355 GDZ than on HS, emitting in average 3.2 mg N₂O-N m⁻² d⁻¹ and 1.8 mg N₂O-N m⁻²
356 d⁻¹ in 2009 and 2010, respectively. The spatial variability within the plots was high
357 particularly on HS (Fig. 6) and no significant differences in N₂O emissions were
358 found between plots within the same landscape element. The spatial pattern of the
359 cumulative N₂O emissions for the plots within one transect was not consistent for the
360 two summer periods (Fig. 6).

361 3.3 Spatial variability in soil factors controlling N₂O emissions

362 Surface soil had a greater bulk density in GDZ than on HS (Table 1) even though
363 there were exceptions (plots B4 and B5). The pH of the surface soil was significantly
364 higher in GDZ than on HS (Table 1). For TOC, TN and C/N the opposite was the case,
365 reflecting the lack of an organic surface layer in GDZ. Values for the concentration of
366 NO₃⁻ in soil extracts (NO₃⁻_{ex}) and the concentration of NO₃⁻ in soil water (NO₃⁻_{sw})
367 were significantly higher on HS than in GDZ. In the O-horizon, TOC and TN were
368 higher at T2 and T4 than at other plots on HS. In the GDZ, Plot B6 was highest with
369 respect to these variables. The NO₃⁻_{sw} was high in winter and early spring and
370 decreased sharply at the start of the growing season in early April, probably due to
371 biological uptake of N and dilution by ample rainfall at the end of March 2010 (Fig.
372 7). During winter, NO₃⁻_{sw} showed consistent differences among plots on HS with
373 highest values at plot T1 and lowest at plot T2, whereas concentrations were lower

374 and more uniform throughout summer, fluctuating with precipitation events. There
375 were consistent differences in $\text{NO}_3^-_{\text{sw}}$ in GDZ, with concentration levels decreasing in
376 the order $\text{B2} = \text{B4} > \text{B3} > \text{B5} \geq \text{B6}$. At B3, B5 and B6, $\text{NO}_3^-_{\text{sw}}$ concentrations were
377 below 5 mg N L^{-1} during most of the year with the greatest values at the end of the dry
378 season (Feb. and Mar., 2010). The concentration of NH_4^+ in soil water ($\text{NH}_4^+_{\text{sw}}$) was
379 small, with values amounting to less than 10% of total inorganic N in solution, except
380 for plots B2 and B6, and there was no seasonal pattern in any of the plots (data not
381 shown). No correlation was found between $\text{NH}_4^+_{\text{sw}}$ and $\text{NO}_3^-_{\text{sw}}$ in soil water.

382 We used PCA to assess the relationship between soil parameters and the cumulative
383 N_2O flux in summer 2010 (Fig. 6) at each plot (Fig. 8). The PCA divided the plots
384 into two groups (T1 to T5 and B1 as one group (HS); the others as the second group
385 (GDZ)). The first two components of PCA explained 63.9% of the total variation,
386 with most of the physicochemical parameters (pH, bulk density, TOC, TN, $\text{NO}_3^-_{\text{ex}}$,
387 $\text{NO}_3^-_{\text{sw}}$ and $\text{NH}_4^+_{\text{ex}}$) and the *ex situ* potential for N_2O loss contributing to the first
388 component (50.3%). The second component (13.6%) included $\text{NH}_4^+_{\text{sw}}$. The
389 instantaneous denitrification rate contributed to both components, but had relatively
390 small effect on the N_2O flux compared to other parameters. Total organic carbon
391 (TOC) and most of the variables associated with the availability of N (TN, $\text{NO}_3^-_{\text{ex}}$,
392 $\text{NO}_3^-_{\text{sw}}$ and $\text{NH}_4^+_{\text{ex}}$) were positively correlated with soil N_2O flux, whereas soil pH and
393 bulk density (BD) were negatively correlated with N_2O flux. Also, the instantaneous
394 denitrification rate and the *ex situ* potential for N_2O loss were positively correlated

395 with N₂O flux, whereas NH₄⁺_{sw} concentration in soil water did not correlate with N₂O
396 flux.

397 3.4 Temporal variability of seasonal drivers for N₂O emission

398 In summer, peak N₂O emission was associated with rainfall resulting in high water
399 filled pore space (WFPS) values. Emission of N₂O on HS was highest at water filled
400 pore space (WFPS) values between 63% and 72% (averages at 10 cm soil depth for
401 T3 and B1, respectively) and appeared somewhat lower at values larger than 75%
402 (Figs. 3a and 5). Figure 5 illustrates the large inter-annual variability of N₂O flux
403 between a wet year with intensive rainfall episodes (2009) and a relatively dry year
404 (2010) with only one episode with precipitation > 50 mm day⁻¹ (82 mm in 9 hrs on
405 July 5, 2010). The N₂O flux on HS was weakly, but negatively correlated with NO₃⁻_{sw}
406 (p=0.064) whereas no correlation was found with NH₄⁺_{sw} (p=0.659). At individual
407 sites in GDZ, the N₂O flux was weakly, but negatively correlated with the
408 groundwater level, i.e. higher N₂O flux occurred when the groundwater level was low.
409 Low soil temperature (ST) together with low water filled pore space (WFPS) seemed
410 to be the main constraints for N₂O emission activity in winter (Figs. 3 and 5).
411 Stepwise multiple linear regression indicated that soil temperature (ST) and water
412 filled pore space (WFPS) were the only significant variables explaining temporal
413 variability of N₂O emissions (Eq. 3) on HS; NO₃⁻_{sw} and NH₄⁺_{sw} concentrations did not
414 add significantly and were excluded by the regression. The residuals of the simulated

415 values were normally distributed (not shown), suggesting unbiased fitting of the
416 equation (standard errors of the coefficients in parenthesis):

$$417 \ln(N_2O_{HS}) = -4.80 (\pm 1.24) + 0.088 (\pm 0.018) WFPS + 0.178 (\pm 0.024) ST$$

418 (Eq. 3)

419 Adjusted coefficient of determination (R-Sq(adj)) = 68.9%

420 Since no data for soil temperature (ST) and water filled pore space (WFPS) was
421 available before Oct. 11, 2009 and from Dec. 25, 2010 to Feb. 8, 2011, we used
422 multiple linear regression to derive an equation that links soil temperature (ST) and
423 water filled pore space (WFPS) to meteorological data. The resulting regression
424 models, presented in equations 4 and 5 were then used to estimate the missing values
425 (standard errors of the coefficients in parenthesis):

$$426 ST = 7.84 (\pm 0.41) + 0.579 (\pm 0.029) AT + 1.20 (\pm 0.187) VPD \quad (\text{Eq. 4})$$

427 Adjusted coefficient of determination (R-Sq(adj)) = 67.4%

$$428 WFPS = 63.2 (\pm 0.27) + 0.209 (\pm 0.014) precip_5 + 1.84 (\pm 0.17) VPD_5 \quad (\text{Eq. 5})$$

429 Adjusted coefficient of determination (R-Sq(adj)) = 58.8%

430 where $precip_5$ (mm) is the day-weighted 5-day average precipitation prior to the date
431 for which the computation was done (calculated as $1.0 * precip_{day(0)} + 0.8 * precip_{day(-1)} +$
432 $0.6 * precip_{day(-2)} + 0.4 * precip_{day(-3)} + 0.2 * precip_{day(-4)}$) and VPD_5 (kPa) the
433 day-weighted 5-day average vapor pressure deficit calculated analogously. The

434 selected weights fitted the observations best. The residuals of the model simulations
435 were normally distributed (not shown). Equation 3 was used to estimate N₂O
436 emissions using measured and simulated values for soil temperature (ST; Eq. 4) and
437 water filled pore space (WFPS; Eq. 5). Measured and simulated values are shown in
438 figure 9.

439 3.5 Annual N₂O flux

440 Annual N₂O emissions on the HS were calculated at a daily time step from equation 3
441 for two 1-year periods (May 5 2009 – May 4 2010 and May 5 2010 to May 4, 2011)
442 (Fig. 9). The first annual period included a relatively wet summer, whereas the second
443 period had a relatively dry summer. Annual N₂O fluxes for HS were 0.54 and 0.43 g
444 N m⁻² for the two years, respectively, about 24% higher in the first year than in the
445 second. Higher annual N₂O emission in the first year may be attributed to the
446 contribution of the frequent rain episodes in the wet summer of 2009. The N₂O
447 emission occurring during the first 5 days after the largest rainfall episode (Aug. 4 to
448 5, 2009) contributed 15.8% to the annual N₂O emission. Emissions during the
449 dry-cool season in the first year (November 2009 to March 2010) contributed 12.0%
450 to the annual flux of N₂O. Also the emissions during the dry-cool season in the second
451 year (November 2010 to March 2011), solely based on model estimates, was found to
452 be low, contributing 16.9% to the annual flux of N₂O.

453

454 **4. Discussion**

455 N₂O emission rates recorded after the heavy rainstorm in 2009 (up to 1730 μg N₂O-N
456 m⁻² hr⁻¹; Fig. 5) are the highest found for forest ecosystems so far and exceed rates
457 reported for (sub)tropical forests which typically vary between 11 to 600 μg N₂O-N
458 m⁻² hr⁻¹ (Liu et al., 2011a; Zhang et al., 2008; Silver et al., 2005; Ishizuka et al., 2005;
459 Koehler et al., 2009; Rowlings et al., 2012; Kiese and Butterbach-Bahl, 2002).

460 Temporal variability (Fig. 5) exceeded spatial variability in both landscape elements
461 (HS and GDZ). However, small-scale (within-plot) variability of the cumulative N₂O
462 flux over the active season was high as seen from the high standard errors in figure 6
463 (particularly for HS). Probably this small scale variability was blurring existing trends
464 in flux strength between plots along the HS. Large small-scale spatial variability of
465 N₂O emissions is often reported for forest soils (Robertson and Klemmedtsson, 1996;
466 Bowden et al., 1992; Werner et al., 2007b) and is commonly attributed to small-scale
467 variation in mineral N availability, litter quality and soil moisture. We arranged the
468 plots along transects perpendicular to the contour lines in the two landscape units
469 because gradients in water filled pore space on HS and in groundwater level in GDZ
470 might affect N₂O emissions by controlling aeration, substrate level and ultimately
471 microbial activity (Garten et al., 1994). However, we did not find any consistent
472 relationship between cumulative N₂O emission and plot position in either of the two
473 landscape elements. Concentrations of NO₃⁻_{sw} on HS were similar at all plots during

474 the wet-hot season (Fig. 7a) and did not correlate with N₂O emissions. The
475 availability of NO₃⁻ was higher than in other studies (e.g., Koehler et al., 2009; Zhang
476 et al., 2008) and appeared to be non-limiting for N₂O production on the HS, probably
477 explaining the lack of a directional gradient in cumulative N₂O emissions as reported
478 by others (Nishina et al., 2009; Fang et al., 2009). Surprisingly, the clear difference in
479 water filled pore space between the mid slope position T3 and the foot slope position
480 B1 (Fig. 3a) in spring and early summer did not result in significantly different
481 cumulative N₂O emissions (Fig. 6). This was probably due to the fact that soil
482 moisture was measured at 10 cm depth (in the upper AB-horizon), rather than in the
483 zone of maximum denitrification activity, located in the uppermost soil layers (O and
484 A horizons; Zhu et al., 2013). Soil moisture data records for 5 cm depth were
485 incomplete due to sensor problems, but comparing existing data and bulk densities at
486 5 cm depth suggests that water filled pore space was more similar in the uppermost
487 soil layer (data not shown). The pronounced decrease in water filled pore space at T3
488 (but not B1), later in summer, markedly reduced N₂O emissions as compared with B1
489 (insert Fig. 5).

490 In GDZ, decreasing NO₃⁻ concentrations along the hydrological flow path (Fig. 7b)
491 suggested strong N retention. Both NO₃⁻ and the dissolved N₂O (data not shown)
492 concentrations in the groundwater were found to be very low at the outlet of GDZ (B5
493 and B6), suggesting more complete N₂O reduction along the flow path as NO₃⁻
494 became depleted. There was a weak trend of decreasing cumulative N₂O emissions

495 with increasing average groundwater level along the hydrological flow path in the
496 GDZ (with the exception of B3) during summer 2009, when the groundwater level
497 was very dynamic (Fig. 6a). No such pattern was seen in the hydrologically more
498 stable summer of 2010 (Fig. 6b). In general, groundwater level fluctuations are likely
499 to interact with NO_3^- in controlling N_2O emissions, as the rise of groundwater level
500 can transport NO_3^- into C-richer upper soil layers and create anoxia favorable for
501 denitrification. At the same time, the residence time of N_2O in the soil profile
502 increases, due to decreasing diffusion, thus increasing the chance of N_2O being
503 reduced to N_2 . Conversely, a drop in groundwater level may relieve gaseous diffusion
504 constraints leading to release of N_2O from the soil. In summer 2010 (Fig. 5), for
505 instance, N_2O emission rates were lowest at B6 until groundwater level dropped about
506 10 cm in the middle of July (Fig. 4), resulting in a single N_2O emission peak while
507 returning to low levels quickly thereafter. Similar N_2O dynamics have been observed
508 in rice paddies (e.g. Qin et al., 2010).

509 We found clear differences in N_2O source strength and environmental controls when
510 comparing HS and GDZ; average cumulative N_2O emission in GDZ was roughly half
511 of this at HS in both summers (Figs. 5 and 6), which was surprising, as we expected
512 GDZ to behave as a riparian zone with elevated denitrification activity. Riparian
513 zones are commonly considered as “hot spots” for N_2O emissions, because they
514 receive dissolved N (DON, NO_3^-) from surrounding slopes while providing optimal
515 conditions for denitrification (DOC, anoxia) (Groffman et al., 2000; Hefting et al.,

516 2003). However, unlike common riparian zones which typically develop soils rich in
517 organic matter, the GDZ studied here is formed by formerly cultivated terraces on
518 colluvial deposits. Soils are dense and have low hydrological conductivities, resulting
519 in low productivity and low TOC contents (Table 1). Laboratory incubations revealed
520 that the GDZ soils had instantaneous denitrification rates similar to HS soils, despite
521 their markedly lower TOC contents (Zhu et al. 2013). This indicated a similar N
522 removal potential in GDZ as compared with HS. However in their study, Zhu et al.
523 (2013) also found that denitrifying communities on HS and in GDZ differed
524 functionally with respect to their product stoichiometry, with the GDZ communities
525 being more efficient in reducing NO_3^- all the way to N_2 . Thus, the lower $\text{N}_2\text{O}/\text{N}_2$ ratio,
526 and consequently the reduced N_2O emission rate, in GDZ may be attributed to a
527 number of site-specific factors, directly or indirectly affecting denitrifier functioning
528 and stoichiometry: 1) pore water at GDZ had lower NO_3^- concentrations (Figs. 7b and
529 8) which may decrease the $\text{N}_2\text{O}/\text{N}_2$ ratio by forcing denitrifiers to utilize N_2O as an
530 additional electron acceptor; 2) diffusion of N_2O in soil is slow in dense GDZ soils,
531 resulting in higher dissolved N_2O concentrations in pore water (data not shown)
532 which supports denitrifier communities to express N_2O reductase and 3) soil pH at
533 GDZ was 0.5-0.6 units higher than at HS, which directly (Bergaust et al., 2010) or
534 indirectly (Liu et al., 2010a; Simek and Cooper, 2002; Dörsch et al., 2012) lowers
535 $\text{N}_2\text{O}/\text{N}_2$ ratios.

536 Even though this study was not designed to discriminate sources of N₂O emission,
537 several observations indicate that N₂O emission rates during the active periods were
538 more sensitive to denitrification controls than nitrification controls: 1) on the HS, high
539 N₂O emission occurred when soil water filled pore space was high; 2) in the GDZ,
540 dissolved N₂O decreased with NO₃⁻ concentration in groundwater; 3) the difference in
541 magnitude in N₂O flux on HS and in GDZ was congruent to differences in
542 denitrification stoichiometry observed in soils from the two landscape elements in
543 laboratory incubations (Zhu et al., 2013); 4) N₂O concentrations measured in soil air
544 were negatively correlated with *p*O₂ (R²=0.40, p=0.000, data not shown). Low
545 nitrification rates have been reported for acidic forest soils in SW China (Cai and
546 Zhao, 2009). However, given the prevalence of NH₄⁺ in deposited N and its rapid
547 disappearance in the soil solution, NH₄⁺ oxidizing processes could be a potentially
548 important source for N₂O emission. We conducted an *in situ* ¹⁵NO₃⁻ labeling
549 experiment in summer 2010 at two positions along the HS reported elsewhere (Zhu et
550 al., *submitted*) and found that 70-100% of the emitted N₂O derived from the labeled
551 soil NO₃⁻ pool. It seemed that even though the fate of the deposited NH₄⁺ remains
552 unclear in this catchment, nitrification is not the main source for the high N₂O
553 emission observed.

554 We found large temporal variability in N₂O fluxes. Multiple linear regression
555 identified soil temperature and water filled pore space as the main drivers, explaining
556 68.9% of the temporal variability of N₂O emission fluxes on HS. Such a high

557 coefficient of determination with only two variables is remarkable as most studies
558 conducted in forest ecosystems report much lower degree of explanation by ancillary
559 variables (Morishita et al., 2011; Gu et al., 2011). As shown in figure 9, the resulting
560 regression model captured reasonably well both the seasonal distribution of emission
561 fluxes and the transient emission peaks after rainfall events on HS except for the first
562 measurement point. Peak N₂O emissions (Fig. 5) were triggered by spikes in water
563 filled pore space (Fig. 3a) occurring after big rain episodes, resulting in a highly
564 skewed distribution of N₂O fluxes as found in many other studies (Khalil et al., 2007;
565 Venterea et al., 2009). Low soil temperature, together with relatively low water filled
566 pore space throughout winter resulted in low N₂O fluxes. This implies that N₂O
567 emission fluxes can be estimated fairly well from discontinuous flux measurements
568 when continuous data on soil temperature and water filled pore space are available.
569 However, it is important to sample fluxes closely around major rain fall events in
570 summer as they contribute disproportionally to overall emissions. Too long interval
571 between samplings may result in over- or under-estimation of the cumulative flux
572 (Parkin, 2008).

573 Annual N₂O emissions for the TSP catchment were estimated for the HS only, since
574 96.4% of the area consists of hill slopes, rendering GDZ emissions a minor
575 component of the overall N₂O budget. Annual fluxes showed considerable variation
576 between the wet year (May. 2009 to Apr. 2010) with several intensive rain episodes
577 (0.54 g N₂O-N m⁻² yr⁻¹) and the dryer year (May. 2010 to Apr. 2011) with more

578 evenly distributed precipitation ($0.43 \text{ g N}_2\text{O-N m}^{-2} \text{ yr}^{-1}$). The average annual N_2O
579 emission flux in TSP ($0.48 \text{ g N m}^{-2} \text{ yr}^{-1}$) is at the high end of annual N_2O emission
580 fluxes ($0.005\text{-}0.47 \text{ g N m}^{-2} \text{ yr}^{-1}$) reported for unfertilized forestland in China (Cai,
581 2012). The fluxes are even comparable to the high end of reported N_2O fluxes in
582 tropical rainforests where temperatures are high throughout the year (Dalal and Allen,
583 2008) and to a tropical lowland forest in Panama which was experimentally
584 N-enriched for 9-10 years (Koehler et al., 2009). Expressed as a fraction of the annual
585 atmogenic N-deposition, $\text{N}_2\text{O-N}$ emission amounted to between 8 and 10% of the
586 incoming inorganic N. This “emission factor” was much higher than the Tier 1 IPCC
587 default factor (1%) used for accounting both direct N_2O emissions arising from
588 mineral N fertilizer application to managed soils and indirect N_2O emissions arising
589 from the re-deposited N to uncultivated soils (IPCC, 2006). The emission factor
590 derived from our study is at the high end of the reported N_2O emission factors for
591 deciduous forests and coniferous forests, which were on average 6.5% and 2.4%,
592 respectively (van der Gon and Bleeker 2005). This pinpoints the prominent role of
593 nitrogen saturated, subtropical forests on acid soils as a secondary source of
594 anthropogenic N_2O which should be taken into account for regional N_2O budgets.

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Table 1 Characteristics of top soils at individual plots on the hill slope (HS) and in the groundwater discharge zone (GDZ)

plot	Horizon/layer	pH (H ₂ O)	TOC	TN	C/N	BD ^a	NH ₄ ⁺ _{ex}	NO ₃ ⁻ _{ex}	NH ₄ ⁺ _{sw}	NO ₃ ⁻ _{sw}	IDR ^b	<i>Ex situ</i> potential of N ₂ O loss ^b	
			mg g ⁻¹	g cm ⁻³	μg N g ⁻¹ dw soil	mgN L ⁻¹	nmol N ₂ O-N g ⁻¹ dw soil hr ⁻¹	nmol NO ₃ ⁻ g ⁻¹ dw soil					
HS	T1	O	3.96	226.1	8.7	26.1	-	49.6 (20.2)	37.9 (14.7)	-	-	8.4	523
		A	3.78	10.7	0.5	20.4	0.78 (0.05)	21.5 (8.6)	59.6 (29.2)	0.11 (0.05)	15.7 (3.1)	12.2	552
	T2	O	3.74	354.5	15.7	22.6	-	81.9 (11.7)	23.7 (39.4)	-	-	16.7	2336
		A	3.84	15.0	0.8	19.4	0.72 (0.03)	34.4 (11.9)	12.7 (19.2)	0.75 (0.50)	14.9 (4.6)	13.3	1056
	T3	O	4.07	220.2	9.9	22.2	-	85.9 (19.6)	35.7 (25.1)	-	-	40.4	1924
		A	4.13	12.1	0.6	19.1	0.73 (0.15)	59.6 (30.1)	20.1 (11.2)	0.15 (0.04)	13.3 (2.8)	31.4	1624
	T4	O	3.74	326.0	13.4	24.3	-	89.2 (16.6)	29.1 (14.2)	-	-	6.4	460
		A	3.76	14.3	0.7	20.6	0.47 (0.03)	32.1 (11.6)	13.0 (4.7)	0.07 (0.02)	13.5 (4.1)	16.3	1556
	T5	O	3.67	218.1	10.1	21.6	-	55.3 (19.4)	33.6 (43.6)	-	-	26.5	1706
		A	3.75	6.5	0.4	17.2	0.76 (0.16)	25.8 (6.7)	17.4 (6.5)	0.07 (0.02)	13.4 (1.7)	14.7	1394
B1	O	3.99	234.9	11.1	21.1	-	83.0 (16.6)	25.6 (2.9)	-	-	40.8	1236	
	A	4.02	6.05	0.3	17.6	0.42 (0.07)	23.3 (5.8)	6.6 (2.3)	0.17 (0.07)	13.8 (1.5)	11.3	1014	
GDZ ^a	0-20 cm	B2	4.46	14.5	0.9	15.6	1.57 (0.09)	3.6 (1.3)	6.3 (1.1)	1.25 (1.31)	6.2 (0.5)	1.8	318
		B3	4.31	12.8	0.7	18.6	1.67 (0.13)	2.8 (0.8)	2.6 (1.5)	0.01 (0.01)	1.5 (0.7)	12.3	494
		B4	4.38	23.1	1.6	14.5	0.69 (0.02)	7.9 (1.8)	1.6 (0.5)	0.02 (0.01)	5.4 (0.5)	21.0	53
		B5	4.42	11.8	0.8	14.4	0.63 (0.08)	2.9 (1.0)	2.4 (0.5)	0.02 (0.01)	0.7 (0.4)	20.1	201
		B6	4.80	40.5	2.8	14.7	1.01 (0.13)	4.3 (0.4)	0.6 (0.5)	1.12 (0.16)	0.2 (0.0)	28.0	357

TOC: total organic carbon; TN: total nitrogen; C/N: ratio of TOC and TN; BD: bulk density; NH₄⁺_{ex} and NO₃⁻_{ex}: average 2M KCl-extractable NH₄⁺ and NO₃⁻; NH₄⁺_{sw} and NO₃⁻_{sw}: dissolved NH₄⁺ and NO₃⁻ in soil water and groundwater in summer 2010 (May 11 – Sep. 26, 2010); IDR: instantaneous denitrification rate in anoxic soil slurries with ample NO₃⁻; *Ex situ* potential of N₂O loss: nmol NO₃⁻ g⁻¹ dw soil denitrified before maximum N₂O reduction to N₂ was observed in anoxic soil slurries with ample supply of NO₃⁻. Data in parentheses are the standard error.

^a data from Sørbotten (2011) obtained using 100 cm³ steel cylinders in three replicates for A horizon for each sampling plot.

^b data from Zhu et al. (2013)

707 Figure 1 Location of the Tieshanping (TSP) forest catchment, Chongqing, China (panel a) and
708 plot layout in the TSP catchment (panel b). Plots T1 to T5 constitute transect T on the hill
709 slope (HS); Plots B1 to B6 constitute transect B in the ground water discharge zone (GDZ).
710 Increasing plot number denote decreasing elevation along the transects. Plots T5 and B1 are at
711 the intersection of HS and GDZ.

712

713 Figure 2 Daily average air temperature and daily precipitation at TSP. The
714 temperature data from Sep. 5 to Sep. 18, 2009, Nov. 29, 2009 to Jan. 16, 2010 and Jan.
715 29 to Feb. 8, 2010 were missing. Shaded area indicates the dry-cool season.

716

717 Figure 3 Soil water filled pore space (WFPS, panel a) and soil temperature (ST, panel
718 b) at 10 cm depth at plots T3 and B1. Shaded area indicates the dry-cool season.

719

720 Figure 4 Groundwater level (GWL) in GDZ. The lower panel indicates plots and dates
721 at which the groundwater level was below the bottom of the monitoring well. Shaded
722 area indicates the dry-cool season.

723

724 Figure 5 Mean N_2O flux ($n=3$) at the plots on the hill slope (HS, panel a) and in the
725 ground water discharge zone (GDZ, panel b). Standard errors are not shown to
726 maintain readability of the figure. Shaded area indicates the dry-cool season.

727

728 Figure 6 Cumulative N₂O fluxes (g N₂O-N m⁻²) during summer 2009 (panel a, from
729 Jul. 12 to Aug. 8, 27 days) and 2010 (panel b, from May 10 to Aug. 24, 106 days) at
730 individual plots (n=3). Error bars represent standard error.

731

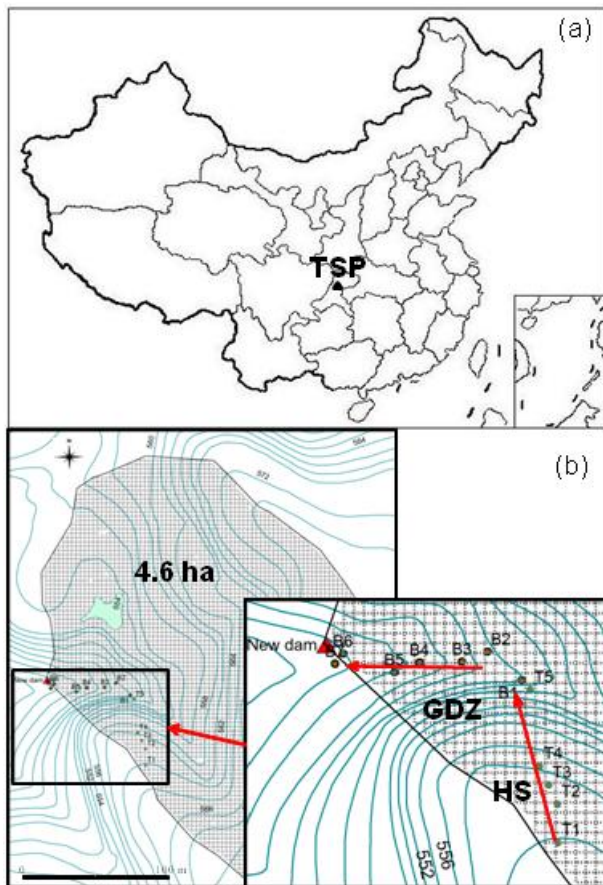
732 Figure 7 NO₃⁻ concentration (mg N L⁻¹) in soil pore water at 5 cm depth at each plot
733 of HS (panel a) and of groundwater at 30 cm depth in GDZ (panel b), respectively.
734 NO₃⁻ values from some of the HS plots during summer 2009 are missing because
735 samplers were damaged by worms. Shaded area indicates the dry-cool season.

736

737 Figure 8 Principal component analysis of cumulative N₂O fluxes and soil parameters
738 (Table 1) at TSP. Before analysis, the N₂O flux and several of the parameters were
739 transformed, whereas others (TOC: total organic carbon; TN: total nitrogen; C/N: the
740 ratio of TOC and TN; IDR: instantaneous denitrification rate; *ex situ* potential of N₂O
741 loss (see text for explanation); NH₄⁺_{ex} and NO₃⁻_{ex}: 2M KCl -extractable NH₄⁺ and
742 NO₃⁻) were not. The transformations were as follows: N₂O flux: ln(cumulative N₂O
743 flux from each plot in summer 2010; Fig. 6); bulk density (BD): -1/(bulk density);
744 NH₄⁺_{sw}: ln(dissolved NH₄⁺ in soil water); NO₃⁻_{sw}: sin(dissolved NO₃⁻ in soil water).
745 Red points indicate the individual plots.

746

747 Figure 9 Observed and simulated average N₂O flux at hill slope (HS). The shaded area
748 denotes the dry-cool season.

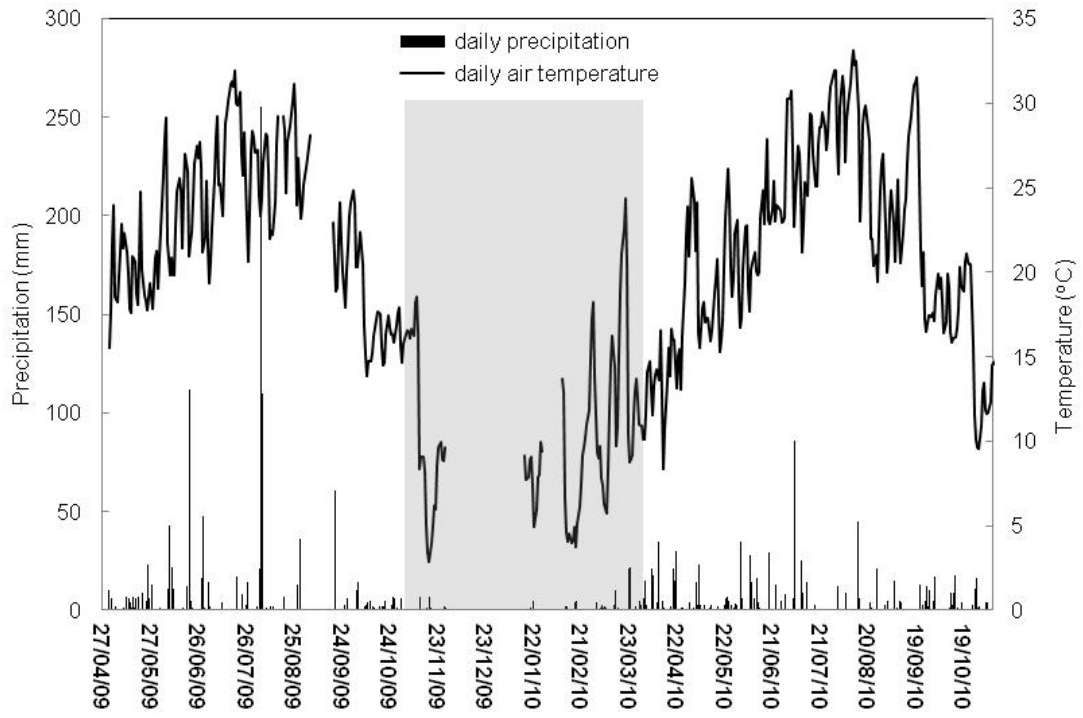


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750

751 Fig. 1

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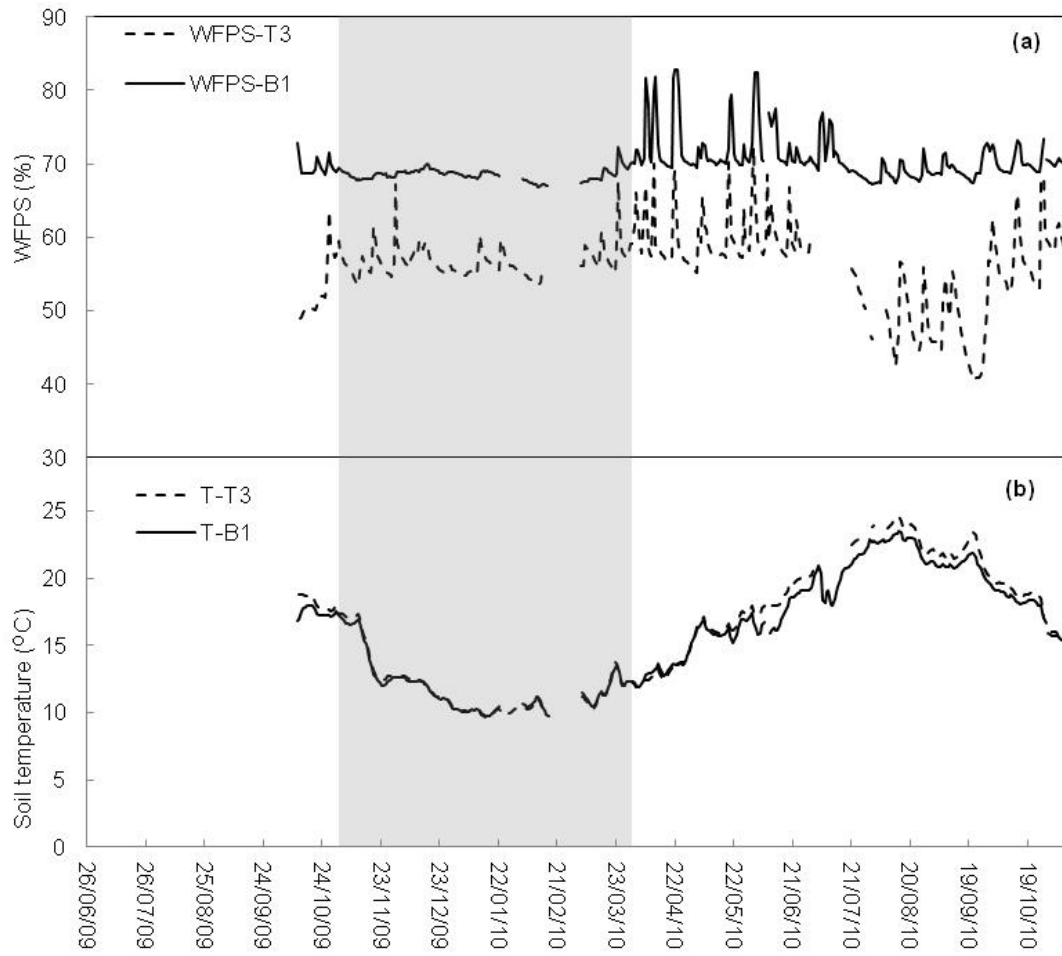


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755 Fig. 2

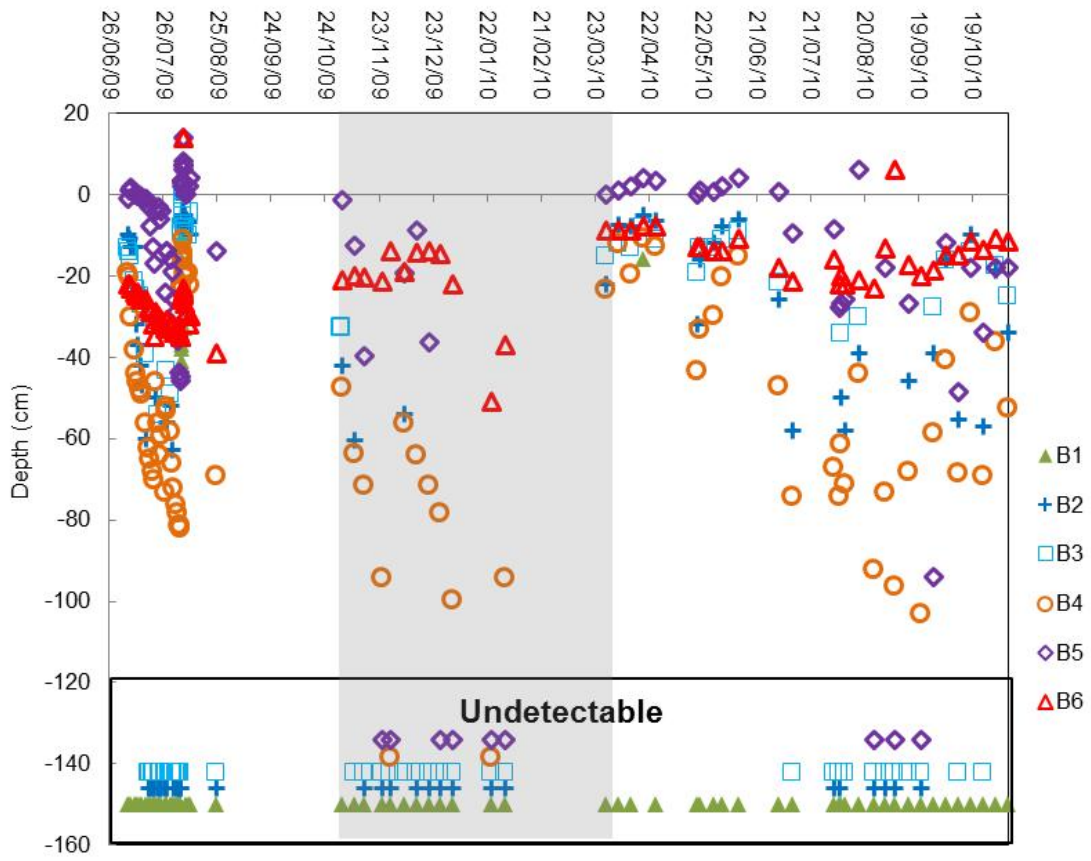
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759 Fig. 3

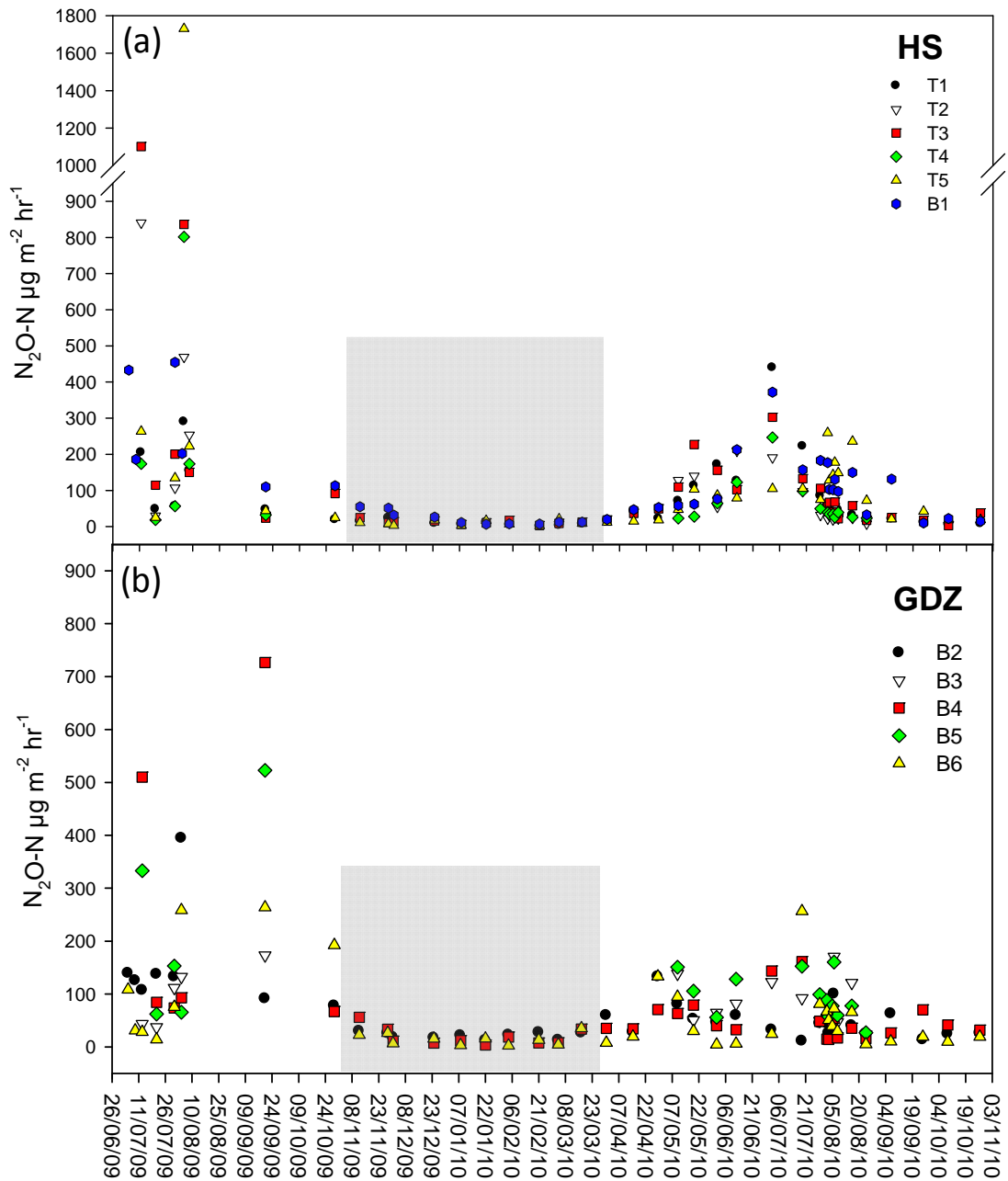


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762 Fig. 4

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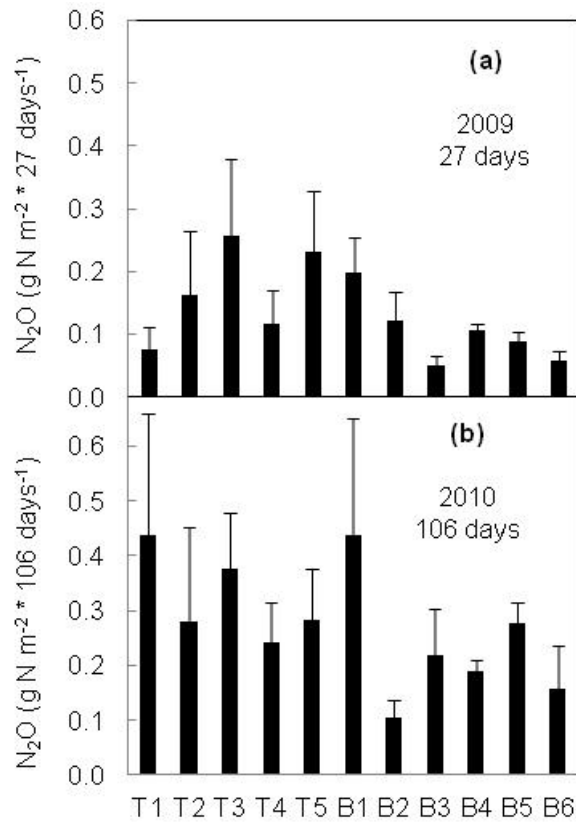


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766 Fig. 5

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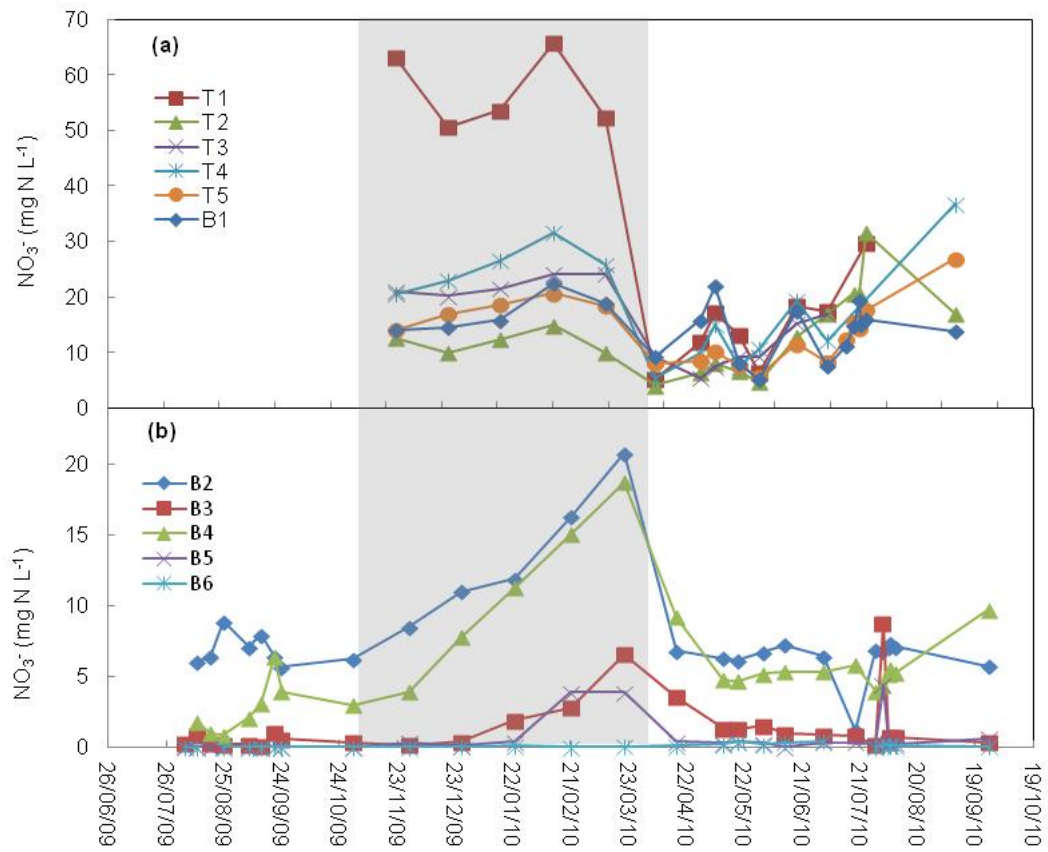


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770 Fig. 6

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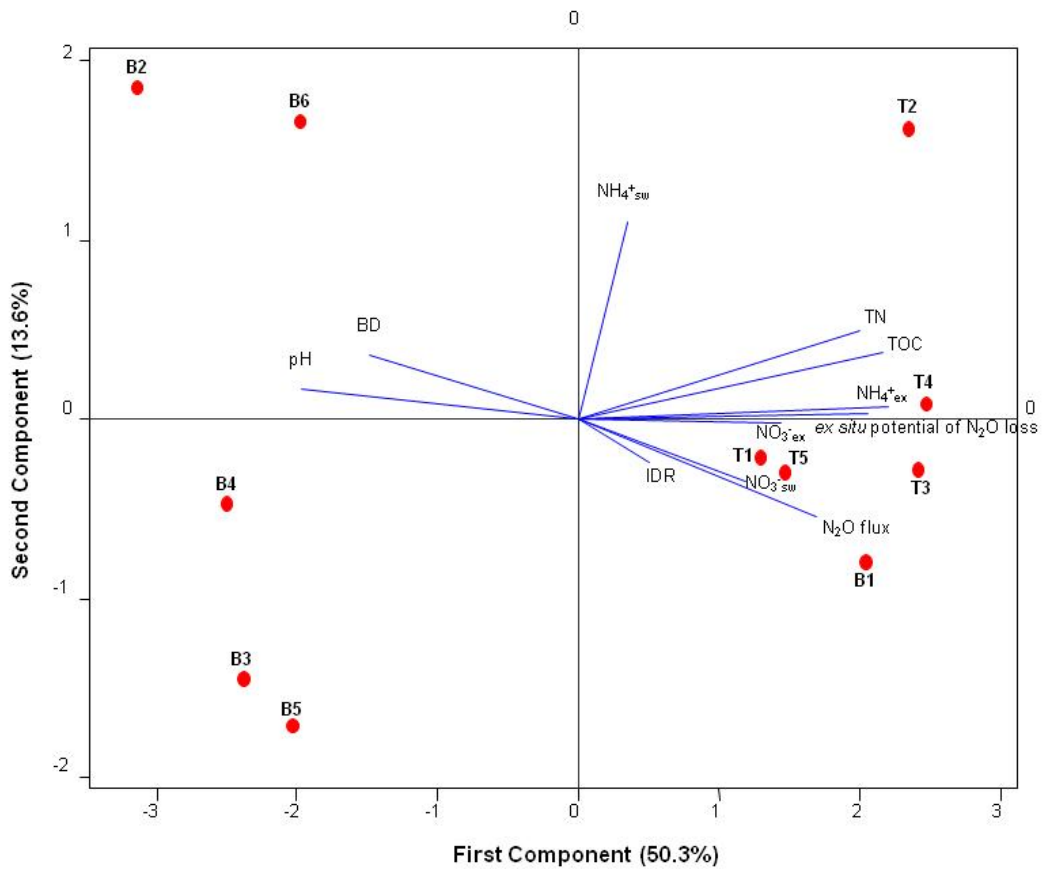


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774 Fig. 7

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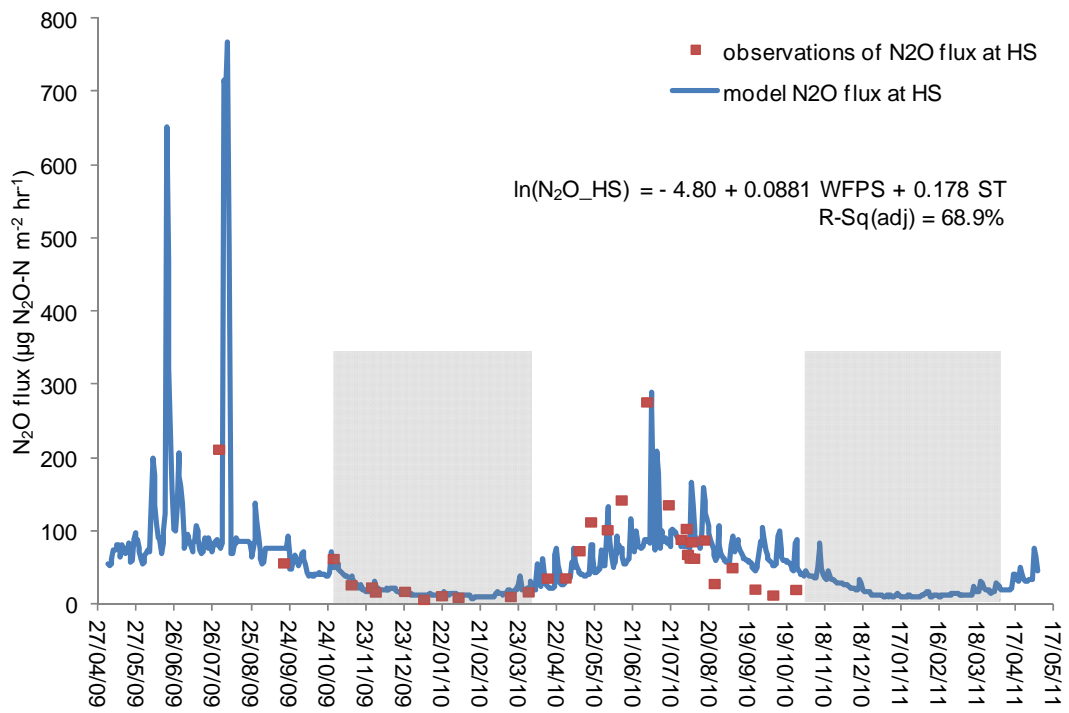
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778 Fig. 8

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780



781

782 Fig. 9

783