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CONFORMATIONAL CHANGE OF SPIN LABELLED MYOGLOBIN - ELECTRON PARAMAGNETIC RESSONANCE IN SOLUTION AND POWDERED SAMPLES

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ABSTRACT. A conformational change of spin labelled myoglobin have been followed by measuring the spin label's (isothiocyanate) correlation time for temperatures between 18°C and 44°C. The correlation time was calculated from Electron Paramagnetic Ressonance Spectra using the components of the espectroscopic and hiperfine tensors obtained by fitting the powder spectra using Lefebvre and Maruani's program.

RESUNO. A mudança conformacional da molécula de mioglobina de cachalote marcada do terminal amina foi acompanhada através da medida do tempo de correlação do marcador isotiocianato para temperaturas entre 18°C e 44°C. O tempo de correlação foi determinado a partir dos espectros de Ressonância Paramagnética Eletrônica, utilizando-se as componentes dos tensores espectroscópico e hiperfino obtidos do ajuste dos espectros de po com o programa de Lefebvre e Maruani.

I. INTRODUCTION

Several parameters have been observed to present an anomalous change with temperature in a range below that at which normal denaturation process takes place in some globular proteins (1,2), as myoglobin. (3,4,5) This change occurs in a temperature range close to the homeotherm regulation temperature and has been associated with a conformational change of the protein.

In the present work EPR spectra of isothiocyanate spin labelled myoglobin (4-isothiocyanate 2,2,6,6 tetramethyl piperdinooxyl) were observed in powdered samples and in solution. This spin label binds to the terminal amino group, since no SH groups are present. The binding is sensitive to pH and to the molecular derivatives. (6) These features make isothiocyanate a convenient spin label to probe the helix A region of myoglobin.

The paramagnetic parameters of the isotyocyanate spin label have been obtained from the powder spectrum.

The conformational change of myoglobin were followed by means of measurements of the label's correlation time.

II. EXPERIMENTAL

Lyophilized sperm-whale myoglobin (Sigma Chemical Co) and the 103 spin label of Synvar Associates were utilized. A 10% myoglobin solution at pH 7.9 has been prepared by

addition of 0.05 M phosphate buffer. The spin label has been used at a ratio of two molecules to one of myoglobin. This solution has been mixed gently at a temperature around 0°C. In order to eliminate free spin label the solution was dialyzed against the buffer at a temperature around that used in the mixing. The dialyzed solution was dried to a powder form with nitrogen gas.

The EPR spectra have been obtained with Varian V-4502 spectrometer and V-3400 magnet at X-band, and with a mixed electronic spectrometer and a V-3900 magnet at Q-band. At X-band the frequency has been measured with the Hewlett-Packard hp 5545M counter and the magnetic field has been calibrated with the MgO:Cr g-marker (isotropic spectroscopic factor g = 1.9797). At Q-band the frequency and field have been calibrated using simultaneously two g-markers; MgO:Cr and DPPH (g = 2.0036).

The Varian V-4540 temperature controller has been used and the temperature has been measured with a chromel-constantant thermocouple immersed in the solution.

III. RESULTS

III.1. Powder Samples

Figures 1 and 2 show the powder experimental spectra at X and Q Lands. These spectra have been simulated with a slightly modified Lefebvre and Maruani program⁽⁷⁾ using the Simpson method of integration with 100 intervals

of integration for each angle. Simulated spectra of Lorentzian and Gaussian shapes are also shown in Figures 1 and 2. The parameters below give the best fit at each band:

where g_i and A_i are the principal values of the spectroscopic and hiperfine tensors respectively.

III.2. Solution

The spin label spectra in myoglobin solution are of the weakly immobilized kind. This allows one to obtain the isotropic parameters, g_0 and A_0 , by measuring the position of the central line and the distance between two neighbouring lines in the spectra. The values obtained are: $g_0 = 2.0063 \pm 0.0003$ and $A_0 = 17 \pm 0.5$ G.

Kivelson's theory $^{(8)}$ may be used to obtain the correlation time, τ_{c} . The linewidths are given by

$$\Delta H(M) = A + BM + CM^2$$

where M is the nuclear spin quantum number of each line in the spectrum. For Lorentzian lines one has:

$$A = \Delta H(0)$$
 $B = \frac{\Delta H(0)}{2} \left[\sqrt{\frac{I(0)}{I(-1)}} - \sqrt{\frac{I(0)}{I(1)}} \right]$

$$C = \frac{3H(0)}{2} \left[\sqrt{\frac{I(0)}{I(-1)}} + \sqrt{\frac{I(0)}{I(1)}} - 2 \right]$$

where I(M) is the experimentally measured amplitude of the line. The theoretical development for the relaxation time T_2 shows that the parameters B and C are proportional to the correlation time and depend on the x, y and z components of the \bar{g} and \bar{A} tensors. B and C allow for two independent ways of measuring τ_c . As the value of B depends on the microwave power (9) it is more convenient to use the values obtained from C.

In Figure 3 we show the experimentally obtained values of B and C as a function of temperature. The parameter C curve show that the correlation time takes maximum and minimum almost constant values for temperatures below 18°C and above 44°C respectively. The mean value occurs at 32°C and this temperature was taken as the transition temperature.

IV. DISCUSSION AND CONCLUSION

IV.1. Powder Samples

figures 1 and 2 make it clear that the theoretical fit of the spectra cannot be made taking into account only one of the line shapes. The experimental line shape can probably be obtained using a mixture of Lorentzian and

Gaussian lines. The fit is better using an angular dependence of the form $X = a + B \cos^2\theta$ for the components linewidths (a and a + B are the minimum and maximum linewidths).

Fitting the data at X and Q band we have obtained the magnetic parameters of the 4-isothiocyanate 2,2,6,6 tetramethyl piperdinoxyl bounded to myoglobin. They are:

$$g_x = 2.0093 \pm 0.0006$$
 $A_x = 9.0 \pm 1.3 G$
 $g_y = 2.0066 \pm 0.0006$ $A_y = 7.1 \pm 0.9 G$
 $g_z = 2.0030 \pm 0.0005$ $A_z = 33.7 \pm 0.5 G$

which correspond to g_0 and A_0 values of 2.0063 \pm 0.0006 and 16.7 \pm 0.9 gauss in agreement with those obtained from solution spectra.

IV.2. Solution

Using the parameters B and C and the parameters obtained in the preceeding section, the $\tau_{\rm C}$ values at the transition temperature are $0.78\times10^{-9}\,{\rm s}$ and $0.96\times10^{-9}\,{\rm s}$ respectively.

The temperature changes of the parameter C can be taken as corresponding to the temperature dependence of the correlation time $\tau_{\rm C}$. This kind of curve cannot be associated with the temperature change of the viscosity, since the latter is monotonic. The change in $\tau_{\rm C}$ must be interpreted as being a consequence of a change in the conformation of myoglobin. At temperatures below $18^{\rm O}{\rm C}$ the structure of the molecule restricts the label movement while above $44^{\rm O}{\rm C}$ it allows for

a less restricted movement. This conformational transition of myoglobin is probably related with a structural change near the region corresponding to helix A (where the isothiocyanate is bound) as well as to the CD, DF and GR regions suggested by Kendrew⁽¹⁰⁾. Figure & shows the regions envolved in the discussion.

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Louro for introducing Lefebvre and Maruani's program to our
computer program library and the electronic and mechanical
staff members for help in designing the apparatus.

LEGENOS

Fig. 1 - Experimental and simulated rigid spectra of isothiocyanate spin label at X hand experimental, simulated with lorentzian components of 4.6 width and ---- simulated with Gaussian components of 6.8 G width. Simulated spectra with $g_{\rm x} = 2.0092$, $g_{\rm y} = 2.0065$, $g_{\rm z} = 2.0029$, $A_{\rm x} = 8.0$ C, $A_{\rm y} = 8.0$ C, $A_{\rm z} = 34.0$ G.

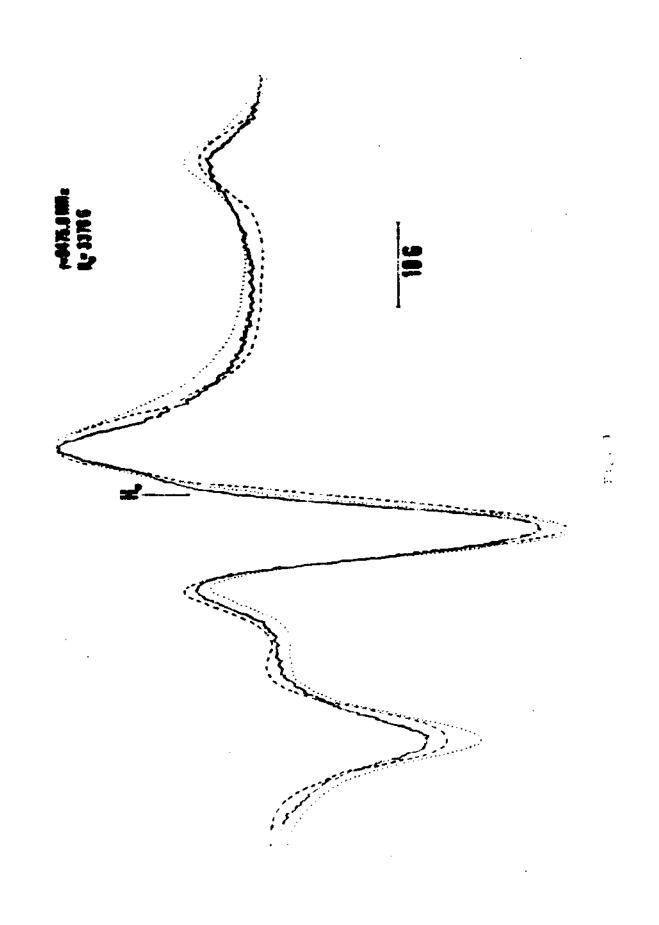
Fig. 2 - Experimental and simulated rigid spectra of isothiocyanate spin label at C-band components, simulated with lorentzian components with 4.6 G width and ---- simulated with Gaussian components with 7.1 G linewidth. Simulated spectra with $g_{\rm X} = 2.0093$, $g_{\rm Y} = 2.0067$, $g_{\rm Z} = 2.0036$, $h_{\rm X} = 10.5$ G, $h_{\rm Y} = 6.2$ G, $h_{\rm Z} = 33.4$ G.

Fig. 3 - Temperature dependence of the parameters B(0) and C(0).

Fig. 4 - The relative positions of helix A, CD, DF and GE. (11)

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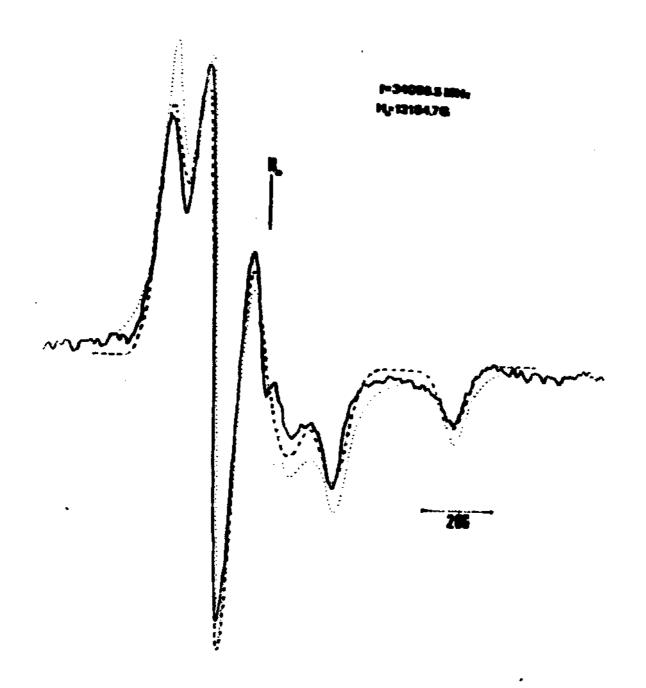


Fig. 2

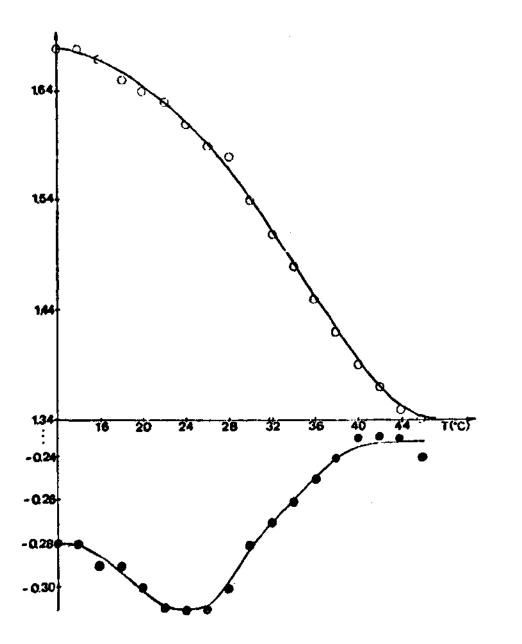


Fig. 3

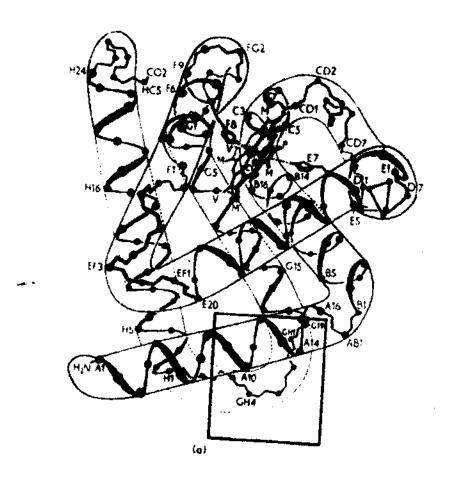


Fig. 4