

MASTER

Fission Product Release

from

Defected LWR Fuel Rods*

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FISSION PRODUCT RELEASE FROM DEFECTED LWR FUEL RODS*

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Experiments conducted at Oak Ridge National Laboratory both with fission product simulants and with irradiated commercial fuel have been utilized to develop a semi-empirical model of fission product release from defected Light Water Reactor (LWR) fuel rods. At fuel temperatures less than 1200°C, releases occur from fission products previously accumulated in the pellet-to-cladding gap region. In this temperature range, the release of species of moderate volatility is postulated to result from two processes. The first of these, which occurs during the period of fuel clad rupture, is due to the transport of the fission and fission product gases as they are vented through the cladding defect. This mode of release is virtually instantaneous. The second mechanism for release, which is time-dependent, involves the diffusional transport of the semi-volatile species to the point of clad rupture through the interconnected voids (the pellet-to-cladding gap and cracks in fuel pellets) within the fuel rod.

The burst release component in the temperature range 900 to 1200°C can be described by the mathematical relation

$$M_B = \alpha V_B (M_0/A)^a \exp (-C/T), \quad (1)$$

in which M_B denotes the mass (in g) of fission product released during the period of cladding rupture, V_B is the volume (in cm^3) of plenum gas vented, as determined at 0°C and system pressure, M_0 is the inventory (in g) of the fission product that is associated with interconnected voids, A is the internal area (in cm^2) of cladding, T represents the temperature (in K) at the point of rupture, and the parameters α , a , and C are adjustable constants.

The diffusional release component has the form

$$M_D = M_0 [1 - \exp(-R_0 t/M_0)], \quad (2)$$

in which M_D is the mass (in g) of fission product released due to diffusional transport during the time period t (in h) following rupture, and R_0 is the initial rate of diffusional release (in g/h). Note that Eq. (2), as written, contains no provision for radioactive decay.

The initial rate of release by diffusional transport in the temperature range 500 to 1200°C is given by the expression

$$R_0 = (\delta W/P)(M_0/A)^a \exp(-\gamma/T), \quad (3)$$

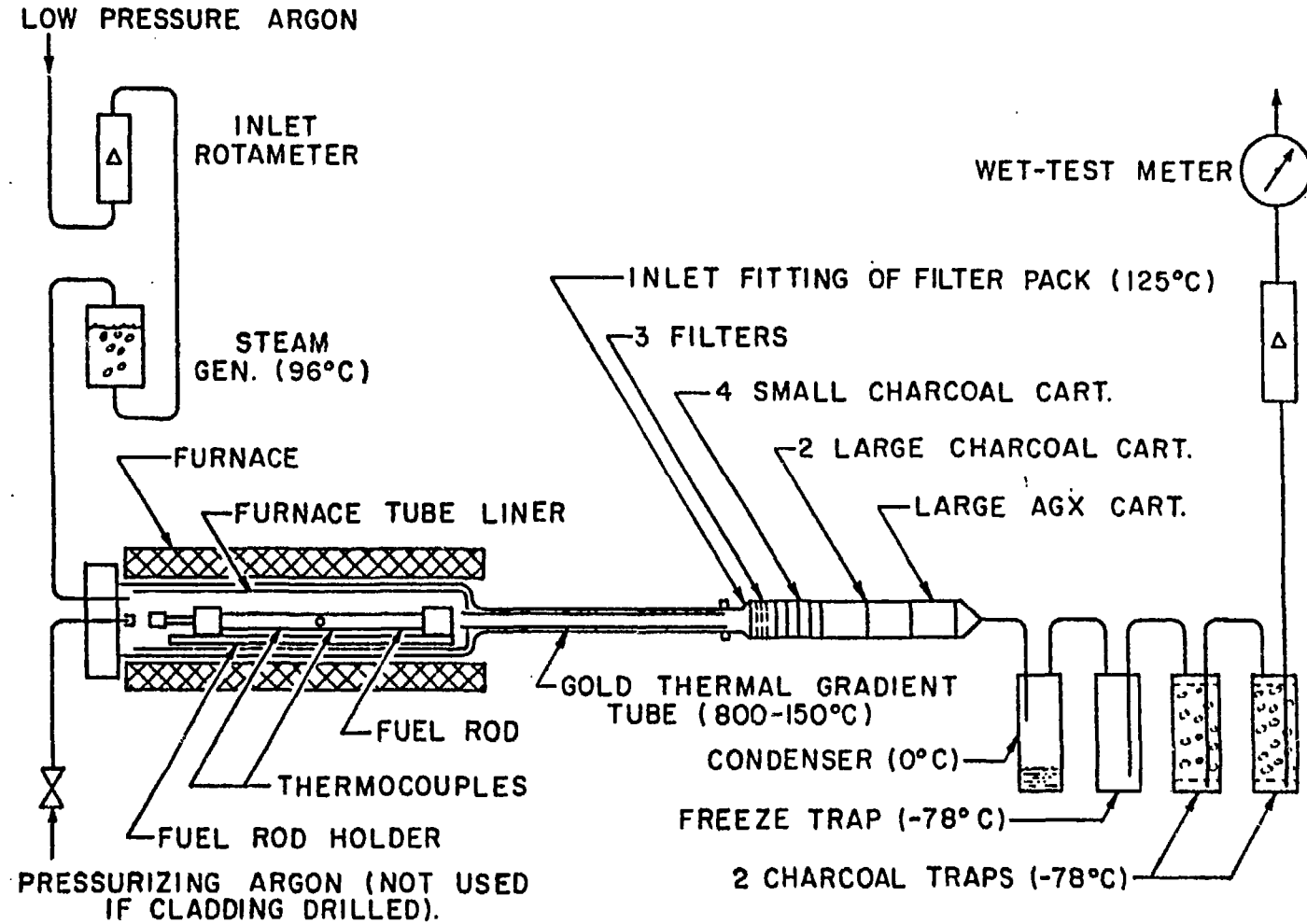
in which W is the width (in μm) of the radial gap in the defected rod, P is the system pressure (in MPa), and δ and γ are adjustable constants. (The amount of cladding expansion and the size of the rupture opening that occur in our tests result in an equivalent gap width, W , of $200 \mu\text{m}$.)

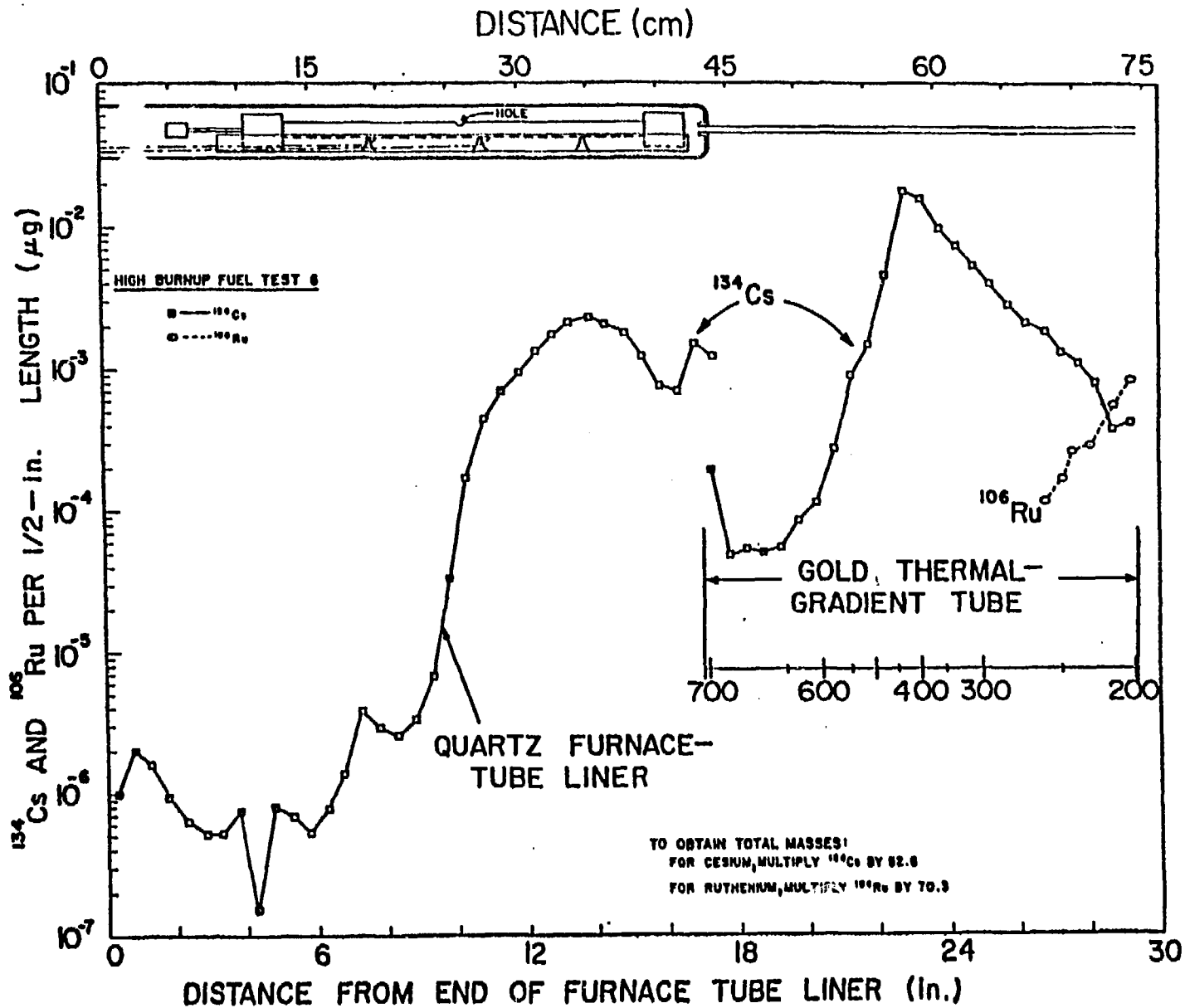
Values of the various constants which were derived from the experimentally determined release data for cesium and iodine species are presented in the following table:

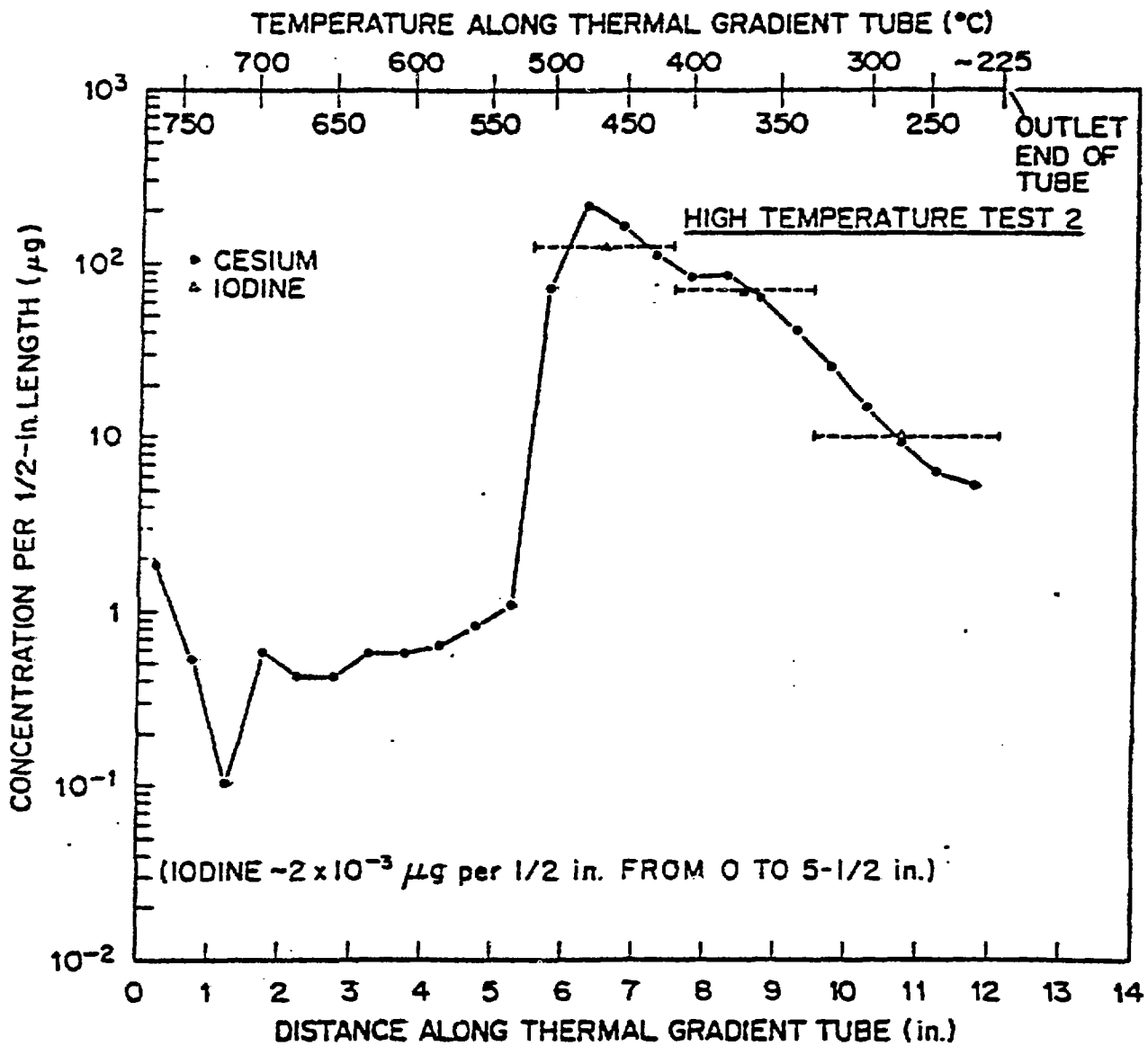
Parameter	Cesium species	Iodine species
$\alpha, (\text{g/cm}^3) \cdot (\text{g/cm}^2)^{-a}$	3.49	0.163
a	0.8	0.8
C, K^{-1}	7.42×10^3	3.77×10^3
$\delta, (\text{g} \cdot \text{MPa} / \mu\text{m} \cdot \text{h}) \cdot (\text{g/cm}^2)^{-a}$	1.90×10^3	1.22×10^2
γ, K^{-1}	1.98×10^4	1.48×10^4

Two series of tests were conducted to define the limitations of the model in describing the releases of cesium and iodine species. The first of these involved an extension of the temperature range of the experiments from 1200°C to 1600°C. These tests indicate that release from high-burnup fuel proceeds by a third process over the approximate temperature range 1200 to 1450°C. This process is probably due to linkage of fission gas bubbles which are formed at grain boundaries. Beyond 1450°C, releases appear to result from diffusion of the fission products through the UO_2 matrix.

The second series of experiments involved an extension of the studies to moderate-burnup fuel that had been irradiated at a higher temperature and thus contained larger quantities of fission products associated with interconnected voids (the "gap inventory"). The results indicate that the model correctly predicts releases from fuel rods which have gap inventories up to 12% of total inventory.







FISSION PRODUCT RELEASE MODEL

$$M_t = M_B + M_D$$

- M_t TOTAL QUANTITY OF FISSION PRODUCT RELEASED
- M_B QUANTITY OF FISSION PRODUCT RELEASED DURING
RUPTURE
- M_D QUANTITY OF FISSION PRODUCT RELEASED BY
DIFFUSIONAL TRANSPORT

BURST RELEASE COMPONENT

$$M_B = \alpha V_B (M_0/A)^a \exp [- C/T]$$

M_B - MASS OF FISSION PRODUCT RELEASED DURING RUPTURE

V_B - VOLUME OF PLENUM GAS VENTED

M_0 - INVENTORY OF FISSION PRODUCT ASSOCIATED WITH INTERCONNECTED VOIDS (THE "GAP INVENTORY")

A - INTERNAL AREA OF THE CLADDING

T - TEMPERATURE AT BURST LOCATION

α, a, C - ADJUSTABLE CONSTANTS

DIFFUSIONAL RELEASE COMPONENT

$$M_D = M_0 [1 - \exp(-R_0 t/M_0)]$$

t - TIME DURING WHICH DIFFUSIONAL
RELEASE OCCURS

INITIAL RELEASE RATE

$$R_0 = (\delta W/P)(M_0/A)^a \exp [-\gamma/T]$$

W - WIDTH OF THE RADIAL GAP ALONG WHICH
DIFFUSION OCCURS

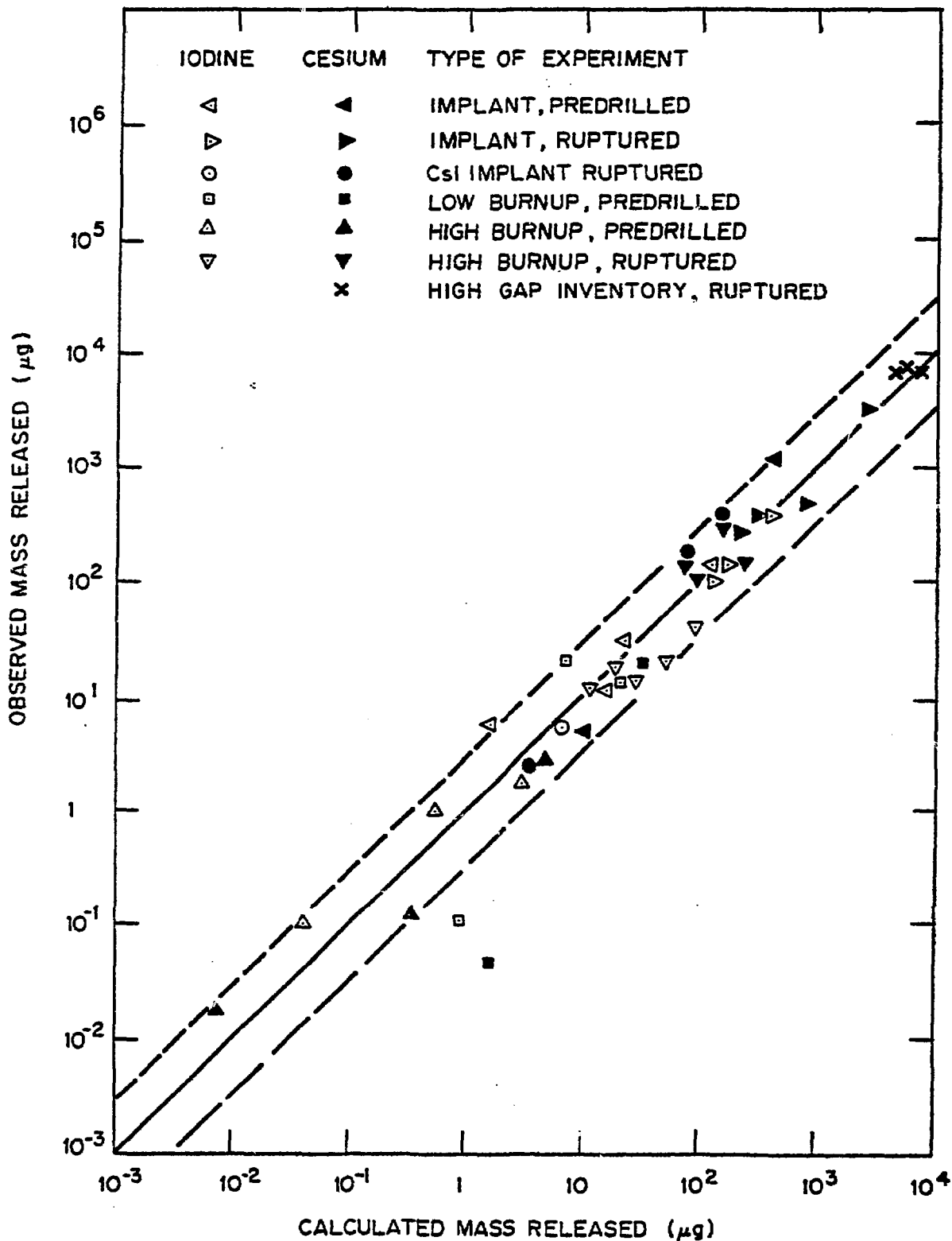
P - SYSTEM PRESSURE

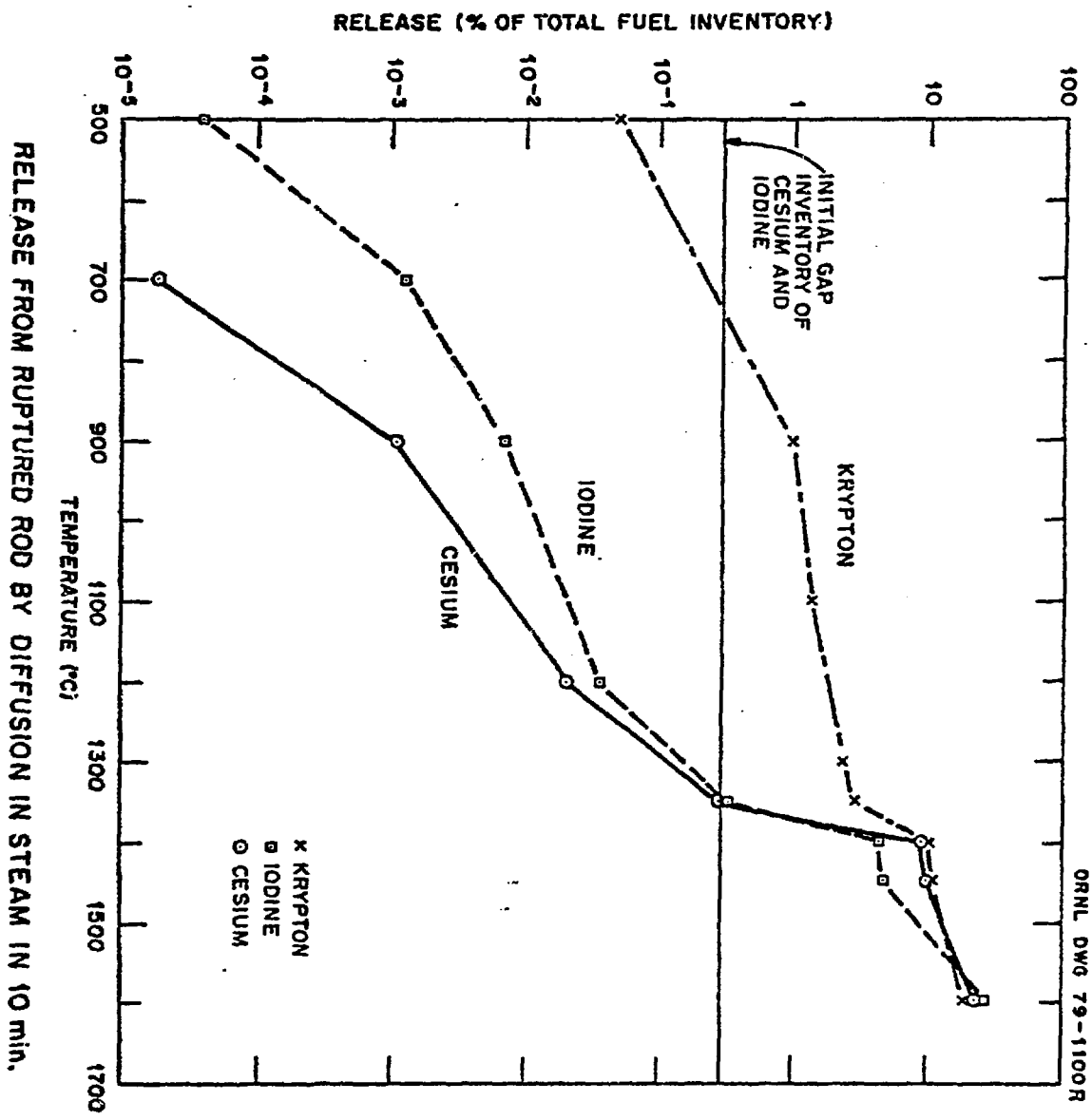
δ, γ - ADJUSTABLE CONSTANTS

COMPARISON OF BURST AND DIFFUSIONAL
RELEASE COMPONENTS

T (°C)	M_B/M_D (THR)	
	CESIUM	IODINE
700	220	41
800	68	14
900	25	5.9
1000	6.8*	2.2*
1100	2.2*	0.9*
1200	0.8*	0.5*

* Assumes rod rupture at 900°C





DISTRIBUTION OF CESIUM AND IODINE IN IRRADIATED FUEL TESTS

TYPE OF TEST	TEST TEMPERATURE (°C)	TEST TIME (min)	FURNACE TUBE		THERMAL GRADIENT TUBE AND FILTER PAPERS		CHARCOAL	
			CESIUM (μg)	IODINE (μg)	CESIUM (μg)	IODINE (μg)	CESIUM (μg)	IODINE (μg)
GAP PURGE, He	700-1200	417	100	0.1	1900	190	0.0	0.3
BURST, STEAM	900	1	110	4.0	17	7.0	0.0	0.4
DIFFUSIONAL, STEAM	1200	27	40	0.1	110	19.0	0.0	0.9
DIFFUSIONAL, STEAM	1445	7	21,000	34.0	1200	960.0	0.0	0.6
DIFFUSIONAL, DRY AIR	700	300	2	0.03	5	0.5	0.0	18.4

SUMMARY OF CHEMICAL BEHAVIOR

1. IN THE GAP-PURGE TESTS WITH PURIFIED HELIUM, IODINE APPEARED TO BE DEPOSITED IN THE APPARATUS AS CsI.
2. IN TESTS CONDUCTED IN STEAM, CsOH AND CsI APPEARED TO BE THE MAJOR CHEMICAL SPECIES OF CESIUM AND IODINE RELEASED.
3. IN TESTS CONDUCTED IN DRY AIR, THE INDICATED MAJOR CHEMICAL SPECIES OF CESIUM AND IODINE RELEASED WERE Cs₂O AND I₂.
4. ANTIMONY APPEARED TO BE RELEASED IN THE STEAM TESTS AS ELEMENTAL ANTIMONY. NO ANTIMONY WAS RELEASED IN THE DRY AIR TESTS.
5. IN DRY AIR TESTS, SIGNIFICANT AMOUNTS OF VERY VOLATILE RUTHENIUM WERE RELEASED, PROBABLY AS RuO₄. ONLY TRACE AMOUNTS WERE RELEASED IN STEAM.