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DOUBLE PHOTON ELECTRIC DIPOLE AND QUADRUPOLE ABSORPTION
OF ATOMIC HYDROGEN *

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ABSTRACT

The two photon processes with one electric dipole photon and another electric quadrupole photon are investigated. These processes with selection rules different from two-electric dipole photon process may be tested by experiments with high power lasers.

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1. Introduction

With the high power laser techniques, it is possible to study the excited states of atoms and molecules by two photon spectroscopy. The two photon processes of electric dipole photon were calculated by many authors ¹⁾. In this paper we restricted our investigation to one dipole and one quadrupole photon absorption, which is slightly smaller than the two electric dipole photon process.

2. Theory of one electric dipole and one electric quadrupole photon absorption

The incoming beams are two plane polarized laser beams with the vector potential

$$\vec{A} = \frac{1}{\sqrt{2\omega\Omega}} \vec{e} e^{i\vec{k}\cdot\vec{r}} \quad (1)$$

$$\vec{A}' = \frac{1}{\sqrt{2\omega'\Omega'}} \vec{e}' e^{i\vec{k}'\cdot\vec{r}}$$

We now study the two photon absorption of electric dipole and quadrupole, these two photons are from two beams respectively.

The incoming plane polarized laser wave may be expressed as some spherical waves with different parities.

$$\vec{e} e^{i\vec{k}\cdot\vec{r}} = \sum_{L,M;\lambda=0,1} \vec{e} \cdot \vec{Y}_{LM}^{(\lambda)*} \left(\frac{\vec{r}}{\omega}\right) \vec{a}_{LM}^{(\lambda)} \quad (2)$$

$$\vec{a}_{LM}^{(1)} = \sqrt{\frac{L}{2L+1}} g_{L+1}(\omega r) \vec{Y}_{L,L+1,M}^{(1)} \left(\frac{\vec{r}}{r}\right) + \sqrt{\frac{L+1}{2L+1}} g_{L-1}(\omega r) \vec{Y}_{L,L-1,M}^{(1)} \left(\frac{\vec{r}}{r}\right)$$

$$\vec{a}_{LM}^{(0)} = g_L(\omega r) \vec{Y}_{LM}^{(0)} \left(\frac{\vec{r}}{r}\right) \quad (3)$$

$$g_L(\omega r) = (2\pi)^{3/2} \frac{1}{i} \frac{j_{L+1/2}(\omega r)}{\sqrt{\omega r}}$$

$j_{L+1/2}(\omega r)$ is the Bessel function, $\vec{Y}_{LM}^{(\lambda)}$, $\vec{Y}_{J,L,M}$ are the vector spherical harmonic functions, as shown in reference (2).

With the expression (2), the transition rate of the hydrogen atom from the initial state $|n_i, l_i\rangle$ to the final state $\langle n_f, l_f|$ through the absorption of one electric dipole photon and one quadrupole

photon may be derived by a straightforward application of Feynman diagram for perturbation theory of QED⁽²⁾. The result is

$$M_{n_1 l_1}^{n_2 l_2}(\omega) = \frac{e^4}{4(2l_1+1)\omega\omega'} \sum_{m_1, m_2} \left| \sum_{l, m} \frac{\langle n_2 l_2 m_2 | \vec{a}_{20}^{(\omega')} | n l m \rangle \langle n l m | \vec{a}_{10}^{(\omega)} | n_1 l_1 m_1 \rangle}{E_n - E_l - \omega} + \frac{\langle n_2 l_2 m_2 | \vec{a}_{10}^{(\omega)} | n l m \rangle \langle n l m | \vec{a}_{20}^{(\omega')} | n_1 l_1 m_1 \rangle}{E_n - E_l - \omega'} \right|^2 \quad (4)$$

where N, N' are the number of photons cross unit area per unit time in both laser beams which may be in connection to the Poynting vector of laser wave in classical approximation.

With the approximation $\omega r \ll 1$, one obtains

$$\langle n_2 l_2 m_2 | \vec{a}_{LM}^{(\omega')} | n l m \rangle = \sqrt{\frac{L+1}{L}} \frac{4\pi(\omega')^L}{(2L+1)!!} \langle n_2 l_2 | r^L | n l \rangle \langle l_2 m_2 | Y_{LM}^{(-)} | l m \rangle \quad (5)$$

$M_{n_1 l_1}^{n_2 l_2}(\omega)$, as shown in (5), may be reexpressed as

$$M_{n_1 l_1}^{n_2 l_2}(\omega) = \frac{(4\pi e)^4 \omega \omega'^3}{2700(2l_1+1)} \sum_{m_1, m_2} \left| \sum_{l, m} P_{n_1 l_1 l}^{n_2 l_2}(\omega, 1, 2) C_{l, m, 10}^{l, m} C_{l_2 m_2}^{l, m} + P_{n_1 l_1 l}^{n_2 l_2}(\omega', 2, 1) C_{l, m, 20}^{l, m} C_{l_2 m_2}^{l, m} \right|^2 \quad (7)$$

where $P_{n_1 l_1 l}^{n_2 l_2}(\omega, L_1, L_2)$ is the sum of the radial part of integral for the intermediate states.

$$P_{n_1 l_1 l}^{n_2 l_2}(\omega, L_1, L_2) = \sum_n \frac{\langle n_2 l_2 | r^{L_2} | n l \rangle \langle n l | r^{L_1} | n_1 l_1 \rangle}{E_n - E_l - \omega} \quad (8)$$

$$C_{l, m, LM}^{l, m} = \langle m | Y_{LM}^{(-)} | l, m \rangle$$

3. Selection rules

The electric dipole and quadrupole selection rules limit that

$$\begin{aligned} l_2 &= l_1 + 1; \quad l_2 = l_1 + 3; \\ l_2 &= l_1 - 1, \text{ for } l_1 > 0; \\ l_2 &= l_1 - 3, \text{ for } l_1 > 2. \end{aligned} \quad (9)$$

$M_{n_1 l_1}^{n_2 l_2}(\omega)$ in Eq.(5) for different cases are given as follows

$$\begin{aligned} M_{n_1 l_1}^{n_2 l_2+1}(\omega) &= \frac{(4\pi e)^4 \omega \omega'^3}{2700(2l_1+1)} \sum_{m_1, m_2} \left| \sum_m P_{n_1 l_1 l_1+1}^{n_2 l_2+1}(\omega, 1, 2) C_{l_1+1, m, 20}^{l_1+1, m} C_{l_2, m, 10}^{l_1+1, m} \right. \\ &+ P_{n_1 l_1 l_1}^{n_2 l_2+1}(\omega', 2, 1) C_{l_1, m, 20}^{l_1, m} C_{l_2, m, 10}^{l_1+1, m} \\ &+ P_{n_1 l_1 l_1-1}^{n_2 l_2+1}(\omega, 1, 2) C_{l_1, m, 10}^{l_1-1, m} C_{l_2, m, 20}^{l_1+1, m} \\ &\left. + P_{n_1 l_1 l_1+2}^{n_2 l_2+1}(\omega', 2, 1) C_{l_1, m, 20}^{l_1+2, m} C_{l_2, m, 10}^{l_1+1, m} \right|^2 \quad (10) \end{aligned}$$

here $l_2 = l_1 + 1$. For $l_1 > 0$, we have $l_2 = l_1 - 1$

$$\begin{aligned} M_{n_1 l_1}^{n_2 l_2-1}(\omega) &= \frac{(4\pi e)^4 \omega \omega'^3}{2700(2l_1+1)} \sum_{m_1, m_2} \left| \sum_m P_{n_1 l_1 l_1+1}^{n_2 l_2-1}(\omega, 1, 2) C_{l_1+1, m, 10}^{l_1+1, m} C_{l_2, m, 20}^{l_1-1, m} \right. \\ &+ P_{n_1 l_1 l_1}^{n_2 l_2-1}(\omega', 2, 1) C_{l_1, m, 20}^{l_1, m} C_{l_2, m, 10}^{l_1-1, m} \\ &+ P_{n_1 l_1 l_1-1}^{n_2 l_2-1}(\omega, 1, 2) C_{l_1, m, 10}^{l_1-1, m} C_{l_2, m, 20}^{l_1-1, m} \\ &\left. + P_{n_1 l_1 l_1-2}^{n_2 l_2-1}(\omega', 2, 1) C_{l_1, m, 20}^{l_1-2, m} C_{l_2, m, 10}^{l_1-1, m} \right|^2 \quad (11) \end{aligned}$$

For $l_2 = l_1 + 3$, Eq.(5) becomes

$$\begin{aligned} M_{n_1 l_1}^{n_2 l_2+3}(\omega) &= \frac{(4\pi e)^4 \omega \omega'^3}{2700(2l_1+1)} \sum_{m_1, m_2} \left| \sum_m P_{n_1 l_1 l_1+1}^{n_2 l_2+3}(\omega, 1, 2) C_{l_1+1, m, 10}^{l_1+1, m} C_{l_2, m, 20}^{l_1+3, m} \right. \\ &\left. + P_{n_1 l_1 l_1+2}^{n_2 l_2+3}(\omega', 2, 1) C_{l_1, m, 20}^{l_1+2, m} C_{l_2, m, 10}^{l_1+3, m} \right|^2 \quad (12) \end{aligned}$$

For $l_1 \geq 3$, $l_2 = l_1 - 3$,

$$\begin{aligned} M_{n_1 l_1}^{n_2 l_2-3}(\omega) &= \frac{(4\pi e)^4 \omega \omega'^3}{2700(2l_1+1)} \sum_{m_1, m_2} \left| \sum_m P_{n_1 l_1 l_1-1}^{n_2 l_2-3}(\omega, 1, 2) C_{l_1, m, 10}^{l_1-1, m} C_{l_2, m, 20}^{l_1-3, m} \right. \\ &\left. + P_{n_1 l_1 l_1-2}^{n_2 l_2-3}(\omega', 2, 1) C_{l_1, m, 20}^{l_1-2, m} C_{l_2, m, 10}^{l_1-3, m} \right|^2 \quad (13) \end{aligned}$$

4. Results for ground state of hydrogen atoms

Now we give the $M_{n,l}^{n_2,l_2}(\omega)$ of the ground state for atomic hydrogen, i.e. $n_1 = 1, l_1 = 0$.

$$M_{1,0}^{n_2,1}(\omega) = \frac{(4\pi e)^4 \omega \omega^3}{2700} \sum_{m_1, m_2} \left| \sum_m P_{m_1,0,1}^{n_2,1}(\omega, 1, 2) C_{0,0,10}^{1,m} C_{1,m,20}^{1,m_2} \right|^2 \quad (14)$$

For $n_2 > 1$, $C_{0,0,10}^{1,m} C_{1,m,20}^{1,m_2} = \frac{\sqrt{5}}{10\pi} \delta_{m_0} \cdot \delta_{m_1,0}$.

$$M_{1,0}^{n_2,3}(\omega) = \frac{(4\pi e)^4 \omega \omega^3}{2700} \sum_{m_1, m_2} \left| \sum_m P_{m_1,0,1}^{n_2,3}(\omega, 1, 2) C_{0,0,10}^{1,m} C_{3,m_2,20}^{1,m} + P_{1,0,2}^{n_2,3}(\omega', 2, 1) C_{2,m,10}^{3,m_2} C_{0,0,20}^{2,m} \right|^2 \quad (15)$$

where $C_{0,0,20}^{2,m} C_{2,m,10}^{3,m_2} = C_{0,0,10}^{1,m} C_{3,m_2,20}^{1,m} = \frac{3}{4\pi} \sqrt{\frac{3}{35}} \delta_{m_0} \delta_{m_1,0}$.

With the nonrelativistic approximation, $P_{1,0,l}^{n_2,l_2}(\omega, L_1, L_2)$ can be accurately evaluated by using the Coulomb Green's function method.

$$g_l(\nu, r, r') = \sum_n \frac{R_{n,l}(r) R_{n,l}^*(r')}{E_n + \frac{e^2}{2\nu r a}} = -\frac{2m}{\sqrt{rr'}} \int_0^\infty dx \exp\left(-\frac{r+r'}{a\nu} \text{Cosh}x\right) \left(\text{Coth}\frac{x}{2}\right)^{2l+1} \frac{2\sqrt{rr'}}{a\nu} \text{Sinh}x \quad (16)$$

where $\nu = \frac{1}{(1 - 2a\omega/e^2)^2}$, a is the Bohr radius.

A useful integral representation involving $g_l(\nu, r, r')$ has been introduced and derived by Rapoport and Zon (3)

$$J_l(\beta, \beta') = -\frac{1}{2m_0} \int dr dr' (rr')^{l+1} \exp\left(-\frac{r+\beta'r'}{a\nu}\right) g_l(\nu, r, r') = \frac{2^{l+1} (2l+1)! (a\nu)^{2l+3}}{(l+1-\nu) [(1+\beta)(1+\beta')]^{2l+1}} F(2l+2, l+1-\nu, l+2-\nu, \frac{(1-\beta)(1-\beta')}{(1+\beta)(1+\beta')}) \quad (17)$$

where ${}_2F_1$ is the usual hypergeometric function. Substituting wave functions $R_{n,l}(r)$ and $R_{n_2,l_2}^*(r)$ we obtain the radial second order matrix element in terms of $J_l(\beta, \beta')$

$$P_{n,l,l}^{n_2,l_2}(\omega, L_1, L_2) = [D_{n,l,l}^{n_2,l_2}(\beta, \beta')] J_l(\beta, \beta') \quad \beta = \frac{\nu}{n_2}, \beta' = \frac{\nu}{n_1} \quad (18)$$

A few explicit expressions of $D_{1,0,l,L_2}^{n_2,l_2}(\beta, \beta')$ are given as examples

$$D_{1,0,1,1,2}^{2,1}(\beta, \beta') = \frac{2m}{\sqrt{6}} \frac{\nu^4}{\partial\beta^2 \partial\beta'} \quad (19)$$

$$D_{1,0,1,1,2}^{4,3}(\beta, \beta') = \frac{2m\nu^6}{384\sqrt{35}} \frac{\partial^6}{\partial\beta^3 \partial\beta'}$$

$$D_{1,0,2,2,1}^{4,3}(\beta, \beta') = \frac{2m\nu^4}{384\sqrt{35} a^2} \frac{\partial^4}{\partial\beta^2 \partial\beta'}$$

where the units, in which $\hbar = c = 1$, are used.

In order to simplify the calculation we put

$$M_{n,l,l}^{n_2,l_2}(\omega) = \frac{(4\pi e)^4 a^3 m^2 \omega \omega^3}{2700 (2l_1+1)} M_{n,l,l}^{n_2,l_2}(\omega) \quad (20)$$

The quantity $M_{n,l,l}^{n_2,l_2}(\omega)$ has zero dimension. Using the expressions (16)---(20), the numerical calculation of $M_{1,0}^{2,1}(\omega)$, $M_{1,0}^{4,3}(\omega)$ has been done. The results are shown in Fig.1, Fig.2. It may be compared to the future experiments with high laser power.

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FIGURE CAPTIONS

- Fig.1 1s + 2p Two photon relative absorption probability and dipole photon frequency.
- Fig.2 1s + 4d Two photon relative absorption probability and dipole photon frequency.

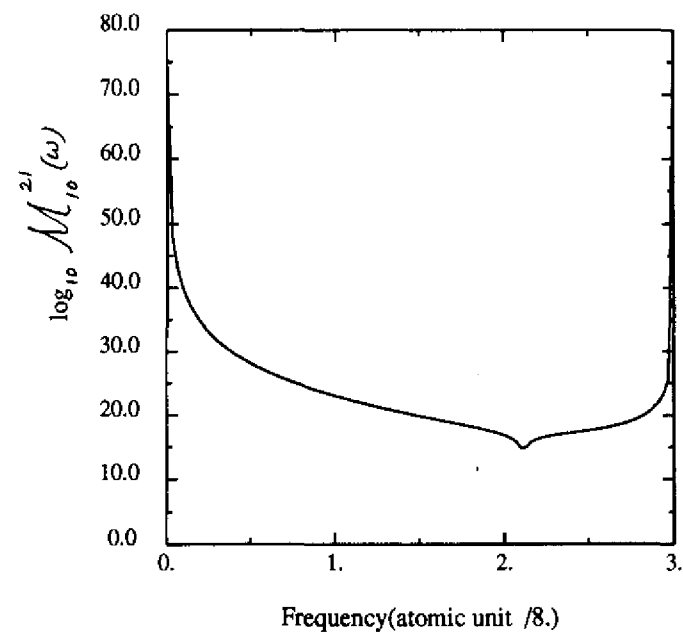


Fig.1

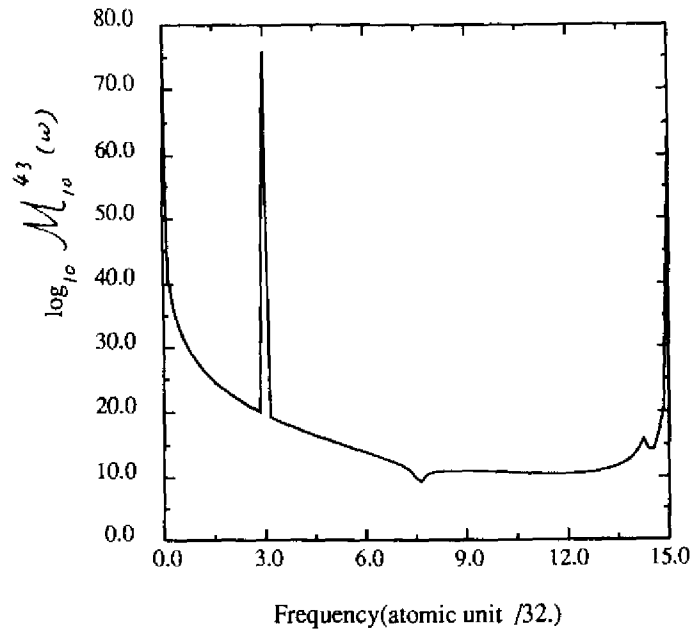


Fig.2

