

## **DIFFUSION OF TRACE ELEMENTS AND RADIONUCLIDES IN SOILS AND CLAYS**

A. **Gosman,** J. L. **Martinez Nieto**

CTU, Fac. of Nucl. Sci. *k* Phys. Eng., Dept. of Nuclear Chemistry Bfehova7, 115 19 Praha 1

**Key words:** trace elements, radionuclides, diffusion, distribution coefficient, moisture, soils, clays, prediffusion time

The previous communications [1-2] provided some results radionuclide diffusion in soils at different moistures, and the influence of pH of soils on diffusion. Capillary modification of the thin layer method was verified. The prepared soil of a given moisture is inserted into a glass capillary, which is hermetically closed. The 6-7 days represent the standard prediffusion rest time  $t_k$  used for establishing the physico-chemical equilibrium in the system. After  $t_k$  the capillary is opened at one end, paper disc with spread radionuclide is inserted and contacted with the soil, and the capillary is immediately closed: diffusion can begin. After the chosen time *t* the capillary is opened, very thin layers of a thickness of approx. 0.05 cm are pressed out, and their specific activities are measured and used for calculating *D.*

This communication discusses the next data on  ${}^{65}Zn, {}^{85}Sr, {}^{137}Cs$  diffusion in soil (LP 26 at Lesni potok, Central Bohemia, locality of Jevany). Further experiments were extended to clays in Northern Bohemia 99 (overbed, Sokolov). It was important for the data and the methods analysis used to find out if and how the diffusion parameters *Def* depend on the "prediffusion rest time" - i.e. on establishing the physico-chemical equilibrium, distribution of the liquid phase, etc. Besides  $t_k$ , shaking was also used in the closed cell, and after the chosen time  $t_p$  the material was placed into the capillary (start of the diffusion time). Cheung  $[3]$  also supposed certain dependence of diffusivity on time - in this case on the diffusion time proper. However, Cho, Oscarson, Hahn [4] have proved that even tens of days (e.g. 24 to 188 days) of diffusion excerted no influence on the diffusivity.

From the results obtained it follows that for soil LP 26 with particle diameter  $< 0.2$  cm, <sup>65</sup>Zn and near moistures (20 and 25.5 w.%) the  $D_{ef}$  values equal  $2.34 \cdot 10^{-7} \text{ cm}^2 \text{s}^{-1}$  and  $3.02 \cdot 10^{-7}$  cm<sup>2</sup>s<sup>-1</sup> corresponding to  $t_k = 1$  week; however, for the same case but  $t_k = 2$ weeks, the  $D_{ef}$  values are  $4.17 \cdot 10^{-7}$  cm<sup>2</sup>s<sup>-1</sup> and  $5.25 \cdot 10^{-7}$  cm<sup>2</sup>s<sup>-1</sup>. For the same soil, particle dimensions less than 0.2 cm and greater than 0.047 cm, and near moistures for  $t_k = 0$  day it follows that  $D_{ef}(^{137}\text{Cs}) < D_{ef}(^{65}\text{Zn}) < D_{ef}(^{85}\text{Sr})$ . It corresponds to changes of distribution coefficients  $K_d$ :  $K_d(^{137}Cs)$  being  $\approx 10$  times greater than  $K_d(^{65}Zn)$  and  $K_d(^{85}Sr)$ , which are of the same order. On the other hand, for the given radionuclide and change of  $t_k$  in the same soil it was found that for <sup>85</sup>Sr and near moistures  $D_{ef}(^{85}Sr)$  increases with  $t_k = 0$ ,  $t_k = 1$ ,  $t_k = 2$  days (there is the maximum of  $D_{ef}$ ), and with the next change of  $t_k = 6$ ,  $t_k = 29$  days the  $D_{ef}(^{85}Sr)$  decreases. Similarly, the maximum was found for  $D_{ef}(^{65}Zn)$  if  $t_k$  changed (0-1-2-6 days).

Of importance may be the comparison of diffusivities at near moistures and  $t_k = \text{const}$ in soil (LP 26) and in clay, where *Def* values are by one order of magnitude lower. In case of clay (at near moistures and  $t_k = \text{const}$ ) the following has been found: at  $t_k = 0$ day is  $D_{ef}(^{85}Sr) > D_{ef}(^{137}Cs) > D_{ef}(^{65}Zn)$ . This corresponds with the  $K_d$  change in clay:  $K_d(^{65} \text{Zn}) > K_d(^{137} \text{Cs}) > K_d(^{85} \text{Sr})$ . The result is the same for  $t_k = 1$  day. The explanation consists in that the contribution of the diffusion flow in the liquid phase is greatest for minimum  $K_d(^{85}Sr)$ . As regards  $t_p$  and  $t_k$  values, it has been found that in case of <sup>85</sup>Sr the influence of  $t_p$  is greater than the influence of  $t_k$  in clay compared with soil. Further, for all experiments with <sup>65</sup>Zn, <sup>85</sup>Sr and <sup>137</sup>Cs it follows that  $D_{ef}$  values at all moistures are less than the corresponding limiting Nernst *Do* values.

Conclusions and results:

- a) in soil medium diffusivity data have been completed
- b) diffusivity of <sup>65</sup>Zn, <sup>85</sup>Sr, and <sup>137</sup>Cs in clay has been determined by using the checked capillary modification of the thin layer method
- c) it has been found that *Dej* values in the clay medium are by one order of magnitude lower compared with the diffusion data for soil medium
- d) the influence of prediffusion time of establishing physico chemical equilibrium in moistadsorbing medium on diffusion of trace elements and radionuclides has been verified
- e) the data analysis shows the possible use of the checked and carried out experimental methods for determining the diffusivity under the given and changing conditions of diffusion transport in soils, clays and other materials.

## **References:**

- [1] GOSMAN, A. BLAZfCEK, J.: *Study of Diffusion of Trace Elements and Radionuclides in Soils. Capillary Modification of the Thin Layer Method. Diffusion of l37Cs.* J. Radioanl. Nucl. Chem., Articles, Vol. 182, No. 2 (1994), pp. 179-191.
- [2] GOSMAN, A. BILA, N. LEBEDA, 0. SUSTROVA, L.: *Diffusion of Trace Elements and Radionuclides in Soils.* Workshop 96 - Part III, Czech Technical University, Prague 1996, pp. 949-950.
- [3] CHEUNG, S. C: *Methods to Measure Apparent Diffusion Coefficient in Compacted Bentonite Clays and Data Interpretation.* Can.J.Civ.Eng., Vol. 16 (1989), pp. 434-443.
- [4] CHO, W. J. OSCARSON, D. W. HAHN, P. S.: *The Measurement of Apparent Diffusion Coefficient in Compacted Clays: an Assessment of Methods.* Applied Clay Science, Vol. 8 (1993), pp. 283-294.

*This reseaixh has been conducted at the Department of Nuclear Chemistry as part of the research project "Modelling of the Migration of Toxic ana 1 Radiotoxic Substances in Soils, Rocks and Sediments" and has been supported by CACR grant No. 104/94/1511.*