

CLUSTERS IN CARBON MARTENSITE. PART I. MARTENSITE IN THERMODYNAMIC EQUILIBRIUM

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An original method of evaluation of the cluster population in carbon martensite has been developed [1]. Using this method it was shown that the Kurdiumov's model of a carbon redistribution within the different octahedral site sublattices can quantitatively account for both observed normal and abnormal tetragonality in carbon martensite. It was also shown that an existence of the internal strains in martensite constitutes a necessary and sufficient condition for the energetic preference of tetrahedral site over the cubic one. The presence of the residual tetragonal distortion in the quasi-cubic phase of κ- martensite is associated with the presence of the mixed clusters formed of the atoms belonging to O_c sublattice as well as to remaining ones.

The long - range ordering parameter η is associated with the mean carbon concentration on sublattice $C(O_a)$, $C(O_b)$ and $C(O_c)$ by the following relations :

$$C(O_a) = C(O_b) = C(1-\eta)/3$$
 (1)
 $C(O_c) = C(1+2\eta)/3$

On the other hand, the internal energy, may be expressed as:

$$U = N \sum V_n C_n \tag{2}$$

where: V_n - binding energy per atom,

C_n - probability that a given atom belongs to cluster n

N - total number of iron atoms.

Taking into account eq. (1) and the law of conservation of the total number of particles in the sublattices, we obtain:

$$\sum_{(\alpha)} C_{\alpha} + \sum_{(\gamma)} (a_{\gamma} / n_{\gamma}) C_{\gamma} = C(1 + 2\eta)/3$$

$$\sum_{(\beta)} C_{\beta} + \sum_{(\gamma)} (b_{\gamma} / n_{\gamma}) C_{\gamma} = C(1 + 2\eta)/3$$
(3)

where the subscript α labels the clusters composed of the atoms located in O_C sites, while the subscript β - labels the clusters composed of

atoms located in O_a or O_b sites. The γ subscript labels the mixing of the atoms in n-particle clusters with a_{γ} atoms in O_c and b_{γ} atoms in O_a or O_b sublattices.

The entropy S can be expressed as:

$$S = S_1 + S_{sh} \tag{4}$$

where S_L is the entropy connected with longrange ordering, S_{sh} the entropy connected with short-range ordering. Taking into account eq. (1), the entropy S_1 can be expressed as:

$$S_{L} = kN[C(1+2\eta)/3 \ln C(1+2\eta)/3 + (1-C/3 - 2C\eta) \\ \ln C(1+2\eta)/3 + 2C(1-\eta)/3 \ln C(1-\eta)/3 + 2(1-C/3+C\eta/3) \ln (1-C/3+C\eta/3)]$$
 (5)

The entropy S_{sh} , may be obtain in the form:

$$\begin{split} S_{sh} &= -kN[\Sigma_{(\alpha)}R_{\alpha}ln(1-C_{\alpha}/R_{\alpha}) + 2\Sigma_{(\beta)}R_{\beta} \\ & ln(1-C_{\beta}/R_{\beta}) + \Sigma_{(\gamma)}R_{\gamma}ln(1-C_{\gamma}/R_{\gamma}) + \Sigma_{(\alpha)}C_{\alpha} \\ & lnC_{\alpha}/(R_{\alpha}-C_{\alpha}) + 2\Sigma C_{(\beta)}lnC_{\beta}/(R_{\beta}-C_{\beta}) + \Sigma_{(\gamma)}C_{\gamma} \\ & ln_{\gamma}C_{\gamma}/(R_{\gamma}-C_{\gamma})] \end{split}$$
(6)

where $R_{\alpha,\beta,\gamma}$ are numbers of possibilities of rotations transforming a given cluster into another one. Concentrations C_{α} , C_{β} and C_{γ} must satisfy the equation (3). For equilibrium state, from the condition of minimum free energy, we have a system of equations from which all order parameters can be derived. Omitting technical details of calculations we obtain:

$$C_{\alpha}/(R_{\alpha} - C_{\alpha}) = N_1 \exp(-V_{\alpha}/kT)$$

$$C_{\beta}/(R_{\beta} - C_{\beta}) = N_2 \exp(-V_{\beta}/kT)$$

$$C_{\gamma}/(R_{\gamma} - C_{\gamma}) = (N_1)^{\alpha\gamma/n\gamma}(N_2)^{b\gamma/n\gamma} \exp(-V_{\gamma}/kT)$$
(7)

where normalization constants N_1 and N_2 can be found numerically using jointly eq.(7) and eq.(3).

REFERENCES:

[1]. L. Dąbrowski: Metall. Mater. Trans. -in press.