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Preliminary Tank Characterization Report for Single-Shell Tank 241-TX-117: Best-Basis Inventory

D. E. Place

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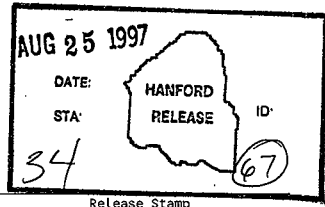
Abstract: An effort is underway to provide waste inventory estimates that will serve as standard characterization source terms for the various waste management activities. As part of this effort, an evaluation of available information for single-shell tank 241-TX-117 was performed, and a best-basis inventory was established. This work follows the methodology that was established by the standard inventory task.

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Revision 0

**PRELIMINARY TANK
CHARACTERIZATION REPORT
FOR SINGLE-SHELL TANK
241-TX-117:
BEST-BASIS INVENTORY**

June 1997

D. E. Place
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Prepared for
U.S. Department of Energy
Richland, Washington

**PRELIMINARY TANK CHARACTERIZATION REPORT
FOR SINGLE-SHELL TANK 241-TX-117:
BEST-BASIS INVENTORY**

This document is a preliminary Tank Characterization Report (TCR). It only contains the current best-basis inventory (Appendix D) for single-shell tank 241-TX-117. No TCRs have been previously issued for this tank, and current core sample analyses are not available. The best-basis inventory, therefore, is based on an engineering assessment of waste type, process flowsheet data, early sample data, and other available information.

The *Standard Inventories of Chemicals and Radionuclides in Hanford Site Tank Wastes* (Kupfer et al. 1997) describes standard methodology used to derive the tank-by-tank best-basis inventories. This preliminary TCR will be updated using this same methodology when additional data on tank contents become available.

REFERENCE

Kupfer, M. J., A. L. Boldt, B. A. Higley, K. M. Hodgson, L. W. Shelton, B. C. Simpson, and R. A. Watrous (LMHC), S. L. Lambert, and D. E. Place (SESC), R. M. Orme (NHC), G. L. Borsheim (Borsheim Associates), N. G. Colton (PNNL), M. D. LeClair (SAIC), R. T. Winward (Meier Associates), and W. W. Schulz (W²S Corporation), 1997, *Standard Inventories of Chemicals and Radionuclides in Hanford Site Tank Wastes*, HNF-SD-WM-TI-740, Rev. 0, Lockheed Martin Hanford Corporation, Richland, Washington.

APPENDIX D

**EVALUATION TO ESTABLISH
BEST-BASIS INVENTORY FOR
SINGLE-SHELL TANK 241-TX-117**

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APPENDIX D**EVALUATION TO ESTABLISH BEST-BASIS INVENTORY FOR
SINGLE-SHELL TANK 241-TX-117**

An effort is underway to provide waste inventory estimates that will serve as standard characterization source terms for the various waste management activities (Hodgson and LeClair 1996). As part of this effort, an evaluation of available information for single-shell tank 241-TX-117 was performed and a best-basis inventory was established. This work, detailed in the following sections, follows the methodology that was established by the standard inventory task.

D1.0 CHEMICAL INFORMATION SOURCES

Available chemical and radiological inventory estimates for tank 241-TX-117 consist only of the inventory estimate generated by the Hanford Defined Waste (HDW) model (Agnew et al. 1997a). No Tank Characterization Report (TCR) has been previously issued for this tank, and core sample analyses are not available. The best-basis inventory, therefore, is based on the waste types contained in tank 241-TX-117 and composition data from other Hanford Site tanks containing similar waste types.

D2.0 COMPARISON OF COMPONENT INVENTORY VALUES

The tank 241-TX-117 chemical and radionuclide inventory predicted by the HDW model (Agnew et al. 1997a) is provided in Table D2-1. The chemical species are reported without charge designation per the best-basis inventory convention. The HDW model inventory is based on 1,904 kL (503 kgal) of salt cake produced by the 242-T Evaporator and 110 kL (29 kgal) of diatomaceous earth.

Table D2-1. Hanford Defined Waste Model Prediction of
Tank 241-TX-117 Inventory. (2 sheets)

Analyte	Hanford Defined Waste model inventory* (kg)
Al	37,400
Bi	2,840
Ca	4,500
Cl	8,960
CO ₃	30,900
Cr	5,280
F	3,440
Fe	6,310
Hg	3.75
K	2,520
La	1.39 E-04
Mn	139
Na	534,000
Ni	908
NO ₂	96,200
NO ₃	725,000
OH	139,000
Pb	305
PO ₄	103,000
Si	19,700
SO ₄	31,300
Sr	0
TOC	4,060
U	15,200
Zr	69.7

Table D2-1. Hanford Defined Waste Model Prediction of
Tank 241-TX-117 Inventory. (2 sheets)

Analyte	Hanford Defined Waste model inventory ^a (kg)
Radionuclide ^b (Ci)	
¹³⁷ Cs	315,000
⁹⁰ Sr	86,000

^a Agnew et al. (1997a)

^b Revision 4 of the HDW model (Agnew et al. 1997a) contains estimates for 46 radionuclides. Only the two most prevalent, ⁹⁰Sr and ¹³⁷Cs, are listed in this table. The HDW model radionuclide predictions are baselined to January 1, 1994.

D3.0 COMPONENT INVENTORY EVALUATION

D3.1 CONTRIBUTING WASTE TYPES

The HDW model (Agnew et al. 1997a), the Sort on Radioactive Waste Type (SORWT) model (Hill et al. 1995) and the waste tank summary report (Hanlon 1997) are not entirely consistent as to their reporting of the waste types present in tank 241-TX-117 and the total tank waste volume. The HDW model includes a diatomaceous earth layer that is not reported by the other two documents. The HDW model tank inventory is based on total waste volume of 2,014 kL (532 kgal), which is 356 kL (94 kgal) less than that reported in the other documents.

The HDW model (Agnew et al. 1997a) predicts that the tank contains 746 kL (197 kgal) of T1 SltCk, 1,158 kL (306 kgal) of SMMT2 salt cake predicted from the Supernatant Mixing Model (SMM), and 110 kL (29 kgal) of diatomaceous earth. The HDW model (Agnew et al. 1997a) refers to 242-T Evaporator salt cakes formed during the time period of 1951-1955 as T1 SltCk. The HDW model refers to 242-T Evaporator salt cakes formed during the period of 1965-1976 as T2 SltCk on a global basis, or SMMT2 when calculated for an individual tank by the SMM.

The Sort on Radioactive Waste Type (SORWT) model (Hill et al. 1995) lists EB (evaporator bottoms) and 1C (first decontamination cycle BiPO₄ waste) as the primary and secondary waste types, respectively, but credits the entire tank 241-TX-117 volume (2,369 kL [626 kgal]) to salt cake with 30 kL (8 kgal) of interstitial liquid. Hanlon (1997) also indicates that the entire tank inventory is salt cake.

D3.2 EVALUATION OF TECHNICAL FLOWSHEET INFORMATION

Waste transaction records (Agnew et al. 1997b) show that tank 241-TX-117 was initially filled with 1C waste supernatant transferred from tanks 241-T-106 and 241-T-105 in the first and second quarters of 1951. Tank 241-T-105 contained a significant volume of well-settled 1C and 2C sludge at the time. Some solid material may have been included in the transfer, depending on the transfer method used. The 1C supernatant was transferred to tank 241-TX-118 in the third quarter of 1951, leaving the tank essentially empty.

The waste transaction records (Agnew et al. 1997b) show that tank 241-TX-117 began receiving 242-T Evaporator bottoms in the fourth quarter of 1951. A total of 21,631 kL (5,715 kgal) of concentrated salt solutions were routed through tank 241-TX-117 between the fourth quarter of 1951 and third quarter of 1955. The evaporator bottoms received in tank 241-TX-117 were often transferred to other tanks, including 241-T-108, 241-T-109, 241-TX-112, 241-TX-114, 241-TY-101, and 241-TY-102. The supernatants from salt cakes formed in tank 241-TX-117 were recycled to the evaporator feed tank (241-TX-118). The deposition of a large salt cake layer in tank 241-TX-117 would normally be expected.

Tank 241-TX-117 again received concentrated evaporator bottoms from the 242-T Evaporator and recycled supernatants to the evaporator beginning in the first quarter of 1966 and continuing intermittently until the fourth quarter of 1971 (Agnew et al. 1997b). Anderson (1990) indicates that the supernatant had been removed from the tank by the first quarter of 1971, and that an initial removal of interstitial liquid occurred in the second quarter of 1971. Regardless of the exact ending date, salt would have precipitated as the concentrated salt solutions cooled, and a large T2 salt cake layer would have been deposited on top of previously existing salt cake.

Three truck loads of diatomaceous earth (approximately 40.8 MT) were added to the tank in November 1970 (Buckingham and Metz 1974). The diatomaceous earth, consisting primarily of silicon dioxide, was added to demonstrate liquid immobilization using a absorbent solid.

Agnew et al. (1997b) indicates that salt well pumping of the interstitial liquid occurred in fourth quarter of 1981, in the first and second quarters of 1982, and in the fourth quarter of 1983. Anderson (1990) does not report events beyond 1980.

D3.3 DETERMINATION OF WASTE VOLUMES

There is a possibility that some 1C and 2C sludges were transferred to tank 241-TX-117 with the 241-T-105 1C waste supernatant transfers in 1951. The sludges in tank 241-T-105 would have been well settled at the time, and no significant solids would have been transferred if the supernatant was decanted. Since there were no solids level measurements in tank 241-TX-117 to indicate a sludge layer, it will be assumed that the volume of solids transferred, if any, is negligible.

The HDW model estimates the T1 salt cake volume based on a solids volume measurement made in the first quarter of 1965 (746 kL [197 kgal] of T1 salt cake). A total of 21,631 kL (5,715 kgal) of T1 salt cake supernatant was transferred to tank 241-TX-117, which is considered sufficient to have produced the 746 kL (197 kgal) of T1 SlcCk.

Waste transaction records (Agnew et al. 1997b) show that tank 241-TX-117 received 33,164 kL (8,762 kgal) of concentrated 242-T Evaporator waste between 1966 and 1971. One transfer of evaporator bottoms was made to tank 241-TX-108 in the third quarter of 1966 (1,779 kL [470 kgal]). With the exception of the one transfer, the evaporator bottoms were cooled in tank 241-TX-117, salt precipitated, and the salt cake supernatants were recycled to the evaporator feed tank (241-TX-118). This volume is considered sufficient to account for T2 salt cake accumulated in tank 241-TX-117.

The HDW model estimates that the T2 salt cake volume is 1,158 kL (306 kgal). This volume of T2 salt cake is calculated by subtracting the T1 salt cake volume (746 kL [197 kgal]), the diatomaceous earth volume (110 kL [29 kgal]), and the estimated interstitial liquid removed (356 kL [94 kgal]) from the waste volume indicated by a December 31, 1971 solids volume measurement (2,369 kL [626 kgal]).

Later solids level measurements in tank 241-TX-117 are unreliable because the manual tape is located above a 2-foot diameter hole in the salt cake created by the removal of a pump (Swaney 1993). The 356 kL (94 kgal) of interstitial liquid removed reported by Agnew et al. (1997b) is based on low-quality data, and does not agree with the 206 kL (54.3 kgal) volume reported by Hanlon (1997). The HDW model assumption that removal of interstitial liquid causes an equal decrease in the salt cake volume is unsubstantiated. Anderson 1990 reports that removal of 102 kL (27 kgal) of interstitial liquid from tank 241-TX-117 in the second quarter of 1971 did not cause a corresponding drop in the level of the solids. The total waste volume of 2,369 kL (626 kgal) reported by Hanlon (1997) will be used for estimating the tank inventory.

The HDW model assumes a density of 0.39 g/mL for diatomaceous earth. This is roughly the equivalent of the 24 lbs/cubic foot as-delivered packed density reported by the vendor (Buckingham and Metz 1974). The HDW model makes no provision for absorption of liquid waste by the diatomaceous earth. However, this appears to be an appropriate assumption for the tank 241-TX-117 inventory determination.

An April 11, 1983 composite photograph of the tank 241-TX-117 interior (Brevick et al. 1997) shows the mounding of bright white solids on the waste surface. The diatomaceous earth does not appear to be dispersed into the salt cake. The composite photograph for the interior tank 241-TX-116, which also received diatomaceous earth, does not show a separate layer of diatomaceous earth on the waste surface, and the tank 241-TX-116 analytical results which showed high silicon concentrations extending 203 cm (80 in.) down into the salt cake (Horton 1977). As the result of these observations, it will be assumed that the diatomaceous earth has the density corresponding to the as-delivered packed density and occupies the

110 kL (29 kgal) layer predicted by the HDW model. It will also be assumed that the diatomaceous earth layer in tank 241-TX-117 includes a negligible quantity of absorbed waste components:

D3.4 COMPOSITION OF TANK 241-TX-117 WASTE

D3.4.1 Composition of T1 Salt Cake

Operation of the 242-T Evaporator between 1951 and 1955 resulted in 2,903 kL (767 kgal) of salt cake, which is contained in 10 underground storage tanks in the T, TX, and TY Tank Farms (Agnew et al. 1997a). The evaporator feeds during this time period consisted largely of 1C and tributyl phosphate (TBP) waste supernatants. The HDW model refers to this salt cake as T1 SlcCk on a global basis. The HDW model uses this average T1 SlcCk composition to calculate the T1 salt cake inventories for individual tanks rather than its SMM because of the lack of detailed evaporator feed composition data. The salt cake produced by the 242-T Evaporator from 1951 through 1955 will be referred to as T1 salt cake hereafter in this report. Seventy-nine percent of the T1 salt cake is contained in the TX Tank Farm. With the exception of tank 241-T-109, all tanks containing T1 salt cake also contain other waste types. Five of the tanks containing T1 salt cake have been core or auger sampled (tanks 241-T-108, 241-T-109, 241-TX-116, 241-TY-101, and 241-TY-102).

The auger samples for tanks 241-T-108 and 241-T-109 are recent (1995) and laboratory analyses should meet all *Hanford Federal Facility Agreement and Consent Order* (Tri-Party Agreement) (Ecology et al. 1994) requirements. Tank 241-T-108 is expected to contain 1C/CW sludge as well as T1 salt cake (Agnew et al. 1997a), however the analytical results indicate that the tank 241-T-108 sample retrieved was primarily salt cake as evidenced by the high sodium concentration (223,000 $\mu\text{g/g}$) reported for the composite (Baldwin 1996). Tank 241-T-109 contains only T1 salt cake generated from the 242-T Evaporator concentration of TBP and 1C/CW supernatants. The composition of the tank 241-T-109 salt cake is somewhat unusual in that it is primarily sodium phosphate rather than sodium nitrate. The composition reported by the TCRs for tank 241-T-108 (Baldwin et al. 1996) and for tank 241-T-109 (Brown et al. 1996) are included in Table D3-1.

T1 salt cake was deposited in tank 241-TX-116 between 1951 and 1955. The tank 241-TX-116 core sample was taken with the initial prototype of a rotary core sampler in 1976 to 1977 (Allen 1977). Sample recoveries were relatively poor and no material was recovered from several segments. Additionally, analytical methods and quality assurance requirements differed significantly from current practices. The analytical data are provided in a letter report (Horton 1977). Core segments 6, 7, 9, and 10 are expected to be T1 salt cake based on the HDW model layer volumes, and this is confirmed by differences in the core sample results as compared to segments 1 through 4 (T2 salt cake). No material was recovered in segments 5 and 8. The analytical results were corrected to a silicon-free basis

since diatomaceous earth (92 percent SiO_2) was added to tank 241-TX-116 in November of 1970 (Buckingham and Metz 1974). The analytical results are included in Table D3-1.

Tanks 241-TY-101 and 241-TY-102 were core sampled in 1985. As with the tank 241-TX-116 core sample, the analytical methods and quality assurance differed from current practices. Tank 241-TY-101 contains ferrocyanide scavenging wastes as well as salt cake. The relatively low sodium concentration reported for the composite (121,000 $\mu\text{g/g}$, Weiss and Mauss 1987a) indicates that the sample was primarily sludge and that the data are not appropriate examples of T1 salt cake. Tank 241-TY-102 contains both T1 and T2 salt cakes (about 45 percent T1 salt cake). Since only composite analyses were performed, the results can not be used as an example of T1 salt cake.

However, it should be noted that the phosphate concentration for tank 241-TY-102 is relatively low (29,000 $\mu\text{g/g}$, Weiss and Mauss 1987b), indicating that the phosphate concentration of the T1 salt cake added to tank 241-TY-102 could not have been comparable to concentrations measured for tanks 241-T-108 and 241-T-109 (125,000 and 246,000 $\mu\text{g/g}$, respectively, Table D3-1). Phosphate concentrations exceeding 100,000 $\mu\text{g/g}$ are not necessarily typical of T1 salt cakes based on the analytical results for tanks 241-TY-102 and 241-TX-116. The reason for this wide variation in phosphate concentration is not known, but supernatants recycled from salt receiving tanks to the 242-T Evaporator might have been depleted in phosphate, and consequently the salt cakes formed from recycled supernatants would have a lower phosphate concentration.

The analytical data for tanks 241-T-108, 241-T-109, and 241-TX-116 are tabulated in Table D3-1. The relative standard deviation of the mean for all components except sodium and zirconium are extremely high, indicating that the composition of the waste type is extremely variable. Any model which assumes that T1 salt cake has a relatively consistent composition, including the prediction in Table D3-1 or the HDW model (Agnew et al. 1997a), will have very limited usefulness in predicting the inventory of a tank containing T1 salt cake. The composition predicted by the HDW model for the global composition of T1 salt cake is included in Table D3-1 for comparison. With the exception of sodium and nitrate, the predicted T1 salt cake composition differs significantly from the HDW model T1 SlcK concentrations for most chemical analytes. The predicted T1 salt cake concentrations used to calculate the tank 241-TX-117 inventory for this evaluation will be based on the average of the concentrations in tanks 241-T-108, 241-T-109, and 241-TX-116.

Table D3-1. Composition of T1 Salt Cakes (2 Sheets).

Analyte	Tank 241-T-108 ^a ($\mu\text{g/g}$)	Tank 241-T-109 ^b ($\mu\text{g/g}$)	Tank 241-TX-116 ($\mu\text{g/g}$) ^{c,d}	Relative std dev of mean (%)	Average - predicted T1 salt cake ($\mu\text{g/g}$)	HDW model T1 SlcCk ^e ($\mu\text{g/g}$)
Ag	<7.96	18.6	NR	NA	<13.3	NR
Al	2,290	1,250	1,720	17.2%	1,750	140.1128
Bi	605	170	NR	56.1%	388	1,806.784
Ca	177	324	NR	29.3%	251	2,116.939
Cd	<7.96	<5	NR	NA	<5 ^f	NR
Cl	<905	341	NR	NA	341 ^f	1,376.542
CO ₃	NR	10,400	32,800	67.2%	21,600	6,832.004
Cr	19.2	40	150	58.3%	69.9	128.6514
F	10,700	13,000	3,140	33.3%	8,950	948.0084
Fe	6,110	5,490	16,000	37.0%	9,200	4,040.613
Hg	NR	NR	NR	NA	NR	0.601441
K	<239	<500	NR	NA	<239 ^f	270.302
La	<39.8	<50	NR	NA	<39.8 ^f	0
Mn	182	1,030	NR	70.0%	606	0
Na	223,000	181,000	246,600	8.85%	216,900	185,809.8
Ni	<15.9	<20	NR	NA	<18	396.1703
NO ₂	6,210	492	210	84.8%	2,300	5,525.867
NO ₃	392,000	20,800	574,700	49.5%	329,200	333,726.3
OH	NR	NR	NR	NA	NR	8,933.119
Pb	533	303	NR	27.5%	418	0
P as PO ₄	125,000	246,000	13,500	52.4%	128,200	70,614.37
Si	1,500	889	NA	25.6%	1,200	287.0366
S as SO ₄	1,110	516	34,200	93.2%	11,900	5,974.895
Sr	21.6	<10	NR	NA	<15.8	0
TOC	NR	NR	NR	NA	NR	1.34 E-06 (wt%)
U	1,130	<500	0.0052	NA	<543	9,724.072
Zr	10.9	12.2	NR	5.63%	11.6	19.18255

Table D3-1. Composition of T1 Salt Cakes (2 Sheets).

Analyte	Tank 241-T-108 ^a ($\mu\text{g/g}$)	Tank 241-T-109 ^b ($\mu\text{g/g}$)	Tank 241-TX-116 ($\mu\text{g/g}$) ^{c,d}	Relative std dev of mean (%)	Average - predicted T1 salt cake ($\mu\text{g/g}$)	HDW model T1 SltCk ^e ($\mu\text{g/g}$)
Radio- nuclide	$\mu\text{Ci/g}$	$\mu\text{Ci/g}$	$\mu\text{Ci/g}$	%	$\mu\text{Ci/g}$	$\mu\text{Ci/g}$
²⁴¹ Am	<0.123	NR	NR	NA	<0.123 ^g	4.67 E-04
⁶⁰ Co	<0.0133	NR	NR	NA	<0.0162 ^g	5.77 E-05
¹³⁴ Cs	NR	NR	2.44	NA	0.0080 ^g	2.43 E-06
¹³⁷ Cs	2.00	NR	4.74	40.7%	2.63 ^g	34.44064
¹⁵⁴ Eu	<0.0455	NR	NR	NA	<0.0514 ^g	0.001026
¹⁵⁵ Eu	<0.0407	NR	NR	NA	<0.0503 ^g	0.004959
Density (g/mL)	2.35	1.55 ^h	NR	NA	1.95	1.742038
% H ₂ O	19.5%	47.70%	NR	NA	33.6%	37.7268%

HDW = Hanford Defined Waste

NA = Not applicable

NR = Not reported

^a Baldwin (1996)

^b Brown et al. (1996)

^c Horton (1977)

^d Silica-free basis due to the addition of diatomaceous earth to this tank

^e Agnew et al. (1997a)

^f Since these analytes were not expected in this waste, the lower value was used instead of an average

^g Predicted T1 salt cake radionuclides are decayed to January 1, 1994. The radionuclides for tanks 241-T-108, 241-T-109, and 241-TX-116 are reported as of the date analyzed, therefore, the average predicted values may not match the reported values

^h The density reported by the 241-T-109 TCR (Brown et al. 1996) was not actually measured, but based on a HDW model Rev. 3 estimate (Agnew et al. 1996).

The density reported for tank 241-T-108 is relatively high and reflects a particle density measurement, rather than a bulk density. The density of the predicted T1 salt cake was somewhat arbitrarily set to 1.7 g/ml to avoid over-reporting of the waste components. T1 salt cake constitutes at roughly 31 vol% of the tank inventory, so the impact of the tank 241-T-108 high waste density could not be ignored. The selected value is in good agreement with the HDW model prediction for T1 salt cake.

D3.4.2 Composition of T2 Salt Cake

Post-1965 operation of the 242-T Evaporator resulted in 22,672 kL (5,990 kgal) of salt cake that is contained in 26 underground storage tanks in the S, SX, U, T, TX, and TY Tank Farms (Agnew et al. 1997a). The HDW model refers to this salt cake as T2 SlcK on a global basis or as SMMT2 when calculated by the SMM for an individual tank. The salt cake produced by the 242-T Evaporator from 1965 to 1976 will be referred to as T2 salt cake, hereafter, in this report. Ninety-one percent of the T2 salt cake is contained in the TX Tank Farm. All tanks containing T2 salt cake also contain other waste types.

Eight tanks containing T2 salt cake have been core sampled, 241-S-107, 241-U-102, 241-U-105, 241-U-107, 241-TX-107, 241-TX-116, 241-TY-102, and 241-TY-103. Only three of these tanks (241-U-102, 241-U-105 and 241-TX-116) have T2 salt cake layers large enough to differentiate it from other waste types in core sample data at the segment level.

T2 salt cake was formed in tanks 241-U-102 and 241-U-105 from 1975 through 1976 (Agnew et al. 1997a). Core sampling of tanks 241-U-102 and 241-U-105 was performed in early 1996. Based on the HDW model, segments 4, 5, and 6 for the two cores from tank 241-U-102 and segment 8 of two cores from tank 241-U-105 are expected to be representative of the T2 salt cake waste type. An independent determination of these levels is not possible because of a lack of solids volume measurements in this time period. Furthermore, a significant composition change between the expected S2 salt cake and T2 salt cake layers can not be seen in the core sample data. The recent analytical data should meet all Tri-Party Agreement requirements. Descriptions of the core sampling events and analytical data are available in the respective TCRs (Hu et al. 1997 and Brown and Franklin 1996).

T2 salt cake was deposited in tank 241-TX-116 between 1966 and 1971. The tank 241-TX-116 core sample was taken with the initial prototype of a rotary core sampler from 1976 to 1977 (Allen 1977). Sample recoveries were relatively poor. Additionally, analytical methods and quality assurance differed significantly from current practices. However, this sample event provides the only composition data for early production of the T2 salt cake waste type. Inclusion of an early T2 salt cake type is important since 242-T Evaporator feeds and operating practices changed over time.

The analytical data for tank 241-TX-116 are provided in a letter report (Horton 1977). Core segments 1 through 4 are expected to be representative T2 salt cake from the HDW

model, and this is confirmed by vertical differences in the core sample results. It was necessary to correct the analytical results to a silicon-free basis since diatomaceous earth (92 percent SiO₂) was added to tank 241-TX-116 in November of 1970 (Buckingham and Metz 1974). The silica from the diatomaceous earth had migrated into the top four core segments (approximately 203 cm [80 in.]) of the salt cake.

The composition data for tanks 241-U-102, 241-U-105, and 241-TX-116 are summarized in Table D3-2. The analytical results for tanks 241-U-102 and 241-U-105 are mass-weighted averages based on the mass of the partial core segment corresponding to each analytical result. Mass-weighted averages, rather than simple arithmetic averages, were calculated because the core segments were not of equal length and the mass of the partial core segments analyzed varied from approximately 30 g to 250 g. Similarly, a mass-weighted average was created for the combination of the T2 salt cake in the two U Farm tanks (81.5% tank 241-U-102 and 18.5% tank 241-U-105). The analytical results for tank 241-TX-116 core segments were simply averaged since the core segments were of equal length. The T2 salt cake prediction is the arithmetic average of the U Tank Farm and tank 241-TX-116 concentrations. The data for tank 241-TX-116 were intentionally given more emphasis (50 percent of the predicted concentration) in the generalized T2 salt cake prediction as it represents an operating period that is more applicable to the TX Tank Farm. The global HDW model composition for T2 salt cake (T2 SlkCk) is included in the Table D3-2 for comparison.

The use of the composition data from tanks 241-U-102, 241-U-105, and 241-TX-116 to represent the composition of other T2 salt cakes should be viewed only as an approximation. None of these three tanks had undergone salt well pumping at the time of the respective core samples. In the case of tank 241-TX-117, these data are being applied to a salt cake which has been salt well pumped and has collapsed to a reduced volume as the result of the removal of interstitial liquid. Additionally, the T2 salt cake projected by the HDW model in tanks 241-U-102 and 241-U-105 could be erroneous if the transfers were TX Tank Farm supernatants (i.e., saturated salt solutions that had already cooled and would not form additional salt cake) rather than actual evaporator bottoms.

Table D3-2. Composition of T2 Salt Cakes (2 Sheets).

Analyte	241-U-102 T2 salt cake wt. avg. ^{a,b} ($\mu\text{g/g}$)	241-U-105 T2 salt cake wt. avg. ^{a,c} ($\mu\text{g/g}$)	U Tank Farm T2 salt cake wt. avg. ^a ($\mu\text{g/g}$)	241-TX-116 T2 salt cake mean ^{d,e} ($\mu\text{g/g}$)	T2 salt cake prediction ^f ($\mu\text{g/g}$)	HDW T2 SlitCk ^g ($\mu\text{g/g}$)
Ag	11.6	19.7	13.1	NR	13.1	NR
Al	18,000	12,900	17,100	38,000	27,500	17,912
Bi	<70.5	<47.2	<66.2	NR	<66.2	220.81
Ca	308	253	298	NR	298	1,462
Cd	<5.94	12.8	<7.21	NR	<7.21	NR
Cl	5,100	5,790	5,230	NR	5,230	3,327.8
CO ₃	53,500	36,500	50,300	58,000	54,200	17,093
Cr	2,310	2,100	2,270	353	1,310	4259.6
F	<125	1,110	<307	3,540	<1,920	930.79
Fe	391	2,270	737	23,900	12,300	620.58
Hg	NR	NR	NA	NR	NA	1.1338
K	1750	1,470	1,700	NR	1,700	1060.7
La	<35.2	29.7	<34.2	NR	<34.2	1.0 E-04
Mn	123	743	237	NR	237	160.31
Na	262,600	220,500	254,800	166,700	210,800	192,764
Ni	91.5	89.5	91.1	NR	91.1	405.82
NO ₂	56,700	40,100	53,600	7,840	30,700	46,096
NO ₃	284,700	395,700	305,200	308,700	306,946	268,197
OH	NR	NR	NA	NA	NA	68,079
Pb	<119	214	<136	NR	<136	109.91
P as PO ₄	5,050	14,100	6,720	8,620	7,670	7,707.9
Si	152	232	167	NR	167	1,817.7
S as SO ₄	17,900	8,350	16,200	16,400	16,300	13,823
Sr	<7.04	<4.72	<6.61	NR	<6.61	0
TOC	8,810	11,000	9,210	NR	9,210	5,191
U	<353	545	<388	NR	<388	2,174.3
Zr	10.8	45.4	17.2	NR	17.2	14.707

Table D3-2. Composition of T2 Salt Cakes (2 Sheets).

Analyte	241-U-102 T2 salt cake wt. avg. ^{a,b} ($\mu\text{g/g}$)	241-U-105 T2 salt cake wt. avg. ^{a,c} ($\mu\text{g/g}$)	U Tank Farm T2 salt cake wt. avg. ^a ($\mu\text{g/g}$)	241-TX-116 T2 salt cake mean ^{d,e} ($\mu\text{g/g}$)	T2 salt cake prediction ^f ($\mu\text{g/g}$)	HDW T2 SlCk ^g ($\mu\text{g/g}$)
Radionuclide ^h ($\mu\text{Ci/g}$)						
²⁴¹ Am	<37.0	<0.95	<30.3	NR	<30.3	0.0285
⁶⁰ Co	<0.155	0.086	<0.142	NR	<0.142	0.027
¹³⁴ Cs	NR	NR	NA	9.64 E-04	9.64 E-04	0.0016
¹³⁷ Cs	197	145	188	34.8	111	163.24
¹⁵⁴ Eu	<0.475	0.61	<0.499	NR	<0.499	0.431
¹⁵⁵ Eu	<1.10	0.82	<1.05	NR	<1.05	0.1849
Density (g/mL)	1.66	1.73	1.70 ⁱ	NR	1.70	1.634

HDW = Hanford Defined Waste

NA = Not applicable

NR = Not reported

^a Weighted average based on the weight of each partial core segment analyzed

^b Hu et al. (1997)

^c Brown and Franklin (1996)

^d Silica-free basis due to the addition of diatomaceous earth to this tank

^e Horton (1977)

^f Average of U Tank Farm and tank 241-TX-116 data

^g Agnew et al. (1997)

^h Decayed to January 1, 1994

ⁱ A simple average is used for the density.

D3.4.3 Composition of Diatomaceous Earth

Approximately 40.8 MT of diatomaceous earth was added to tank 241-TX-117 in January 1971 to demonstrate absorption of free liquids in waste tanks (Buckingham and Metz 1974). Diatomaceous earth is the siliceous skeletal remains of single-cell algae. The composition of the diatomaceous earth added to tank 241-TX-117 is provided in Table D3-3.

Table D3-3. Diatomaceous Earth Addition to 241-TX-117.

Chemical compound	Weight percent	Element	Element added (kg)
SiO ₂	92.3	Si	17,600
Al ₂ O ₃	1.1	Al	238
Fe ₂ O ₃	2.0	Fe	571
CaO	0.9	Ca	263
MgO	0.4	Mg	98
Loss on ignition	3.1	Unknown	1,265

D3.5 PREDICTED INVENTORY FOR TANK 241-TX-117

The chemical and radionuclide inventory of tank 241-TX-117 can be estimated from the T1 salt cake, and T2 salt cake volumes (746 kL [197 kgal], and 1,514 kL [400 kgal], respectively), densities (1.7 g/mL) and the average of chemical/radionuclide concentrations calculated for the 242-T salt cake wastes that have been analyzed. The chemicals included in the diatomaceous earth (110 kL [29 kgal]) were previously calculated in Table D3-3. The resulting inventories are provided in Table D3-4. The inventories estimated by the HDW model (Agnew et al. 1997a) are included in the table for comparison.

Table D3-4. Estimated Chemical and Radionuclide Inventory for Tank 241-TX-117. (3 sheets)

Analyte	T1 salt cake layer inventory ^a (kg)	T2 salt cake layer inventory ^b (kg)	Diatomaceous Earth ^c (kg)	Predicted 241-TX-117 inventory (kg)	HDW model inventory (kg)
Ag	< 16.8	33.8	NR	< 50.6	NR
Al	2,220	70,900	238	73,400	37,400
Bi	491	< 170	NR	< 661	2,840
Ca	318	766	263	1,350	4,500
Cd	< 6.34	18.5	NR	< 24.9	NR
Cl	432	13,500	NR	13,900	8,960
CO ₃	27,300	139,400	NR	166,800	30,900
Cr	88.6	3,380	NR	3,470	5,280
F	11,300	< 4,950	NR	< 16,300	3,440
Fe	11,700	31,700	571	44,000	6,310
Hg	NR	NR	NR	NA	3.75

Table D3-4. Estimated Chemical and Radionuclide Inventory for Tank 241-TX-117. (3 sheets)

Analyte	T1 salt cake layer inventory ^a (kg)	T2 salt cake layer inventory ^b (kg)	Diatomaceous Earth ^c (kg)	Predicted 241-TX-117 inventory (kg)	HDW model inventory (kg)
K	<303	4,370	NR	<4,670	2,520
La	<50.5	<88.1	NR	<139	1.39 E-04
Mn	768	611	NR	1,380	139
Na	274,900	542,500	NR	817,400	534,000
Ni	<22.8	235	NR	<257	908
NO ₂	2,920	79,100	NR	82,000	96,200
NO ₃	417,300	790,000	NR	1.207 E+06	725,000
OH	8,120	84,600	NR	92,800	139,000
Pb	530	<351	NR	<881	305
PO ₄	162,400	19,700	NR	182,200	103,000
Si	1,510	430	17,600	19,600	19,700
SO ₄	15,100	41,900	NR	57,100	31,300
Sr	<20.0	<17.0	NR	<37.0	0
TOC	NR	23,700	NR	23,700	4,060
U	<689	<999	NR	<1,690	15,200
Zr	14.6	44.3	NR	58.9	69.7
Radio-nuclide ^d	T1 Salt cake layer ^a (Ci)	T2 Salt cake layer ^b (Ci)	Diatomaceous Earth ^c	Predicted 241-TX-117 inventory (Ci)	HDW model inventory (Ci)
²⁴¹ Am	<156	<78,000	NA	<78,200	39.7
⁶⁰ Co	<20.5	<365	NA	<386	26.2
¹³⁴ Cs	10.2	2.48	NA	12.7	2.02
¹³⁷ Cs	3,340	286,300	NA	289,600	315,000
¹⁵⁴ Eu	<65.2	<1,280	NA	<1,350	413

Table D3-4. Estimated Chemical and Radionuclide Inventory for Tank 241-TX-117. (3 sheets)

Radio-nuclide ^d	T1 salt cake layer ^a (Ci)	T2 salt cake layer ^b (Ci)	Diatomaceous Earth ^c	Predicted 241-TX-117 inventory (Ci)	HDW model inventory (Ci)
¹⁵⁵ Eu	<63.8	<2,690	NA	<2,760	162

HDW = Hanford Defined Waste

NA = Not applicable.

NR = Not reported

^a Based on the T1 salt cake prediction in Table D3-1.

^b Based on the T2 salt cake prediction in Table D3-2.

^c Based on the diatomaceous earth composition provided in Table D3-3.

^d Radionuclides decayed to January 1, 1994.

D3.6 COMPARISON OF TANK 241-TX-117 INVENTORY ESTIMATES

The lack of sample-based inventory data adds considerable uncertainty to estimation of chemical and radionuclide inventories for tank 241-TX-117. The use of waste composition data from tanks 241-T-108, 241-T-109, 241-TX-116, 241-U-102, and 241-U-105 to represent the wastes in tank 241-TX-117 is a reasonable approach in the absence of analytical data. However, it should be noted that the operating history of tank 241-TX-117 is different from any other Hanford Site tank containing similar waste types. Estimation based on compositions measured in other tanks should be regarded as only an approximation.

The tank 241-TX-117 inventories predicted by the HDW model and the estimate based on waste analyses in other tanks are generally of the same order of magnitude, although the HDW model inventories are generally somewhat lower. Part of the explanation for this difference may be that the HDW model calculated density for the tank 241-TX-117 T2 salt cake is 1.44 g/cc based on the sodium, aluminum, and hydroxide concentrations. This HDW calculated density is often much lower than is generally found when salt cakes are analyzed. The calculated density is used in determining the HDW model inventory for all analytes. Additionally, the HDW model reduced the T2 salt cake inventory by 356 kL (94 kgal) based on the quantity of interstitial liquid removed.

Aluminum. The estimated aluminum inventory is 1.96 times higher than that predicted by the HDW model. Part of this difference is attributable to the low T2 salt cake density calculated by the HDW model (1.44 g/cc), as compared to the 1.7 g/mL estimated from analytical data for T2 salt cakes, and the lower T2 salt cake volume used by the HDW model. The tank 241-TX-116 analytical results show a much higher aluminum concentration and Anderson (1990) indicates processing of a substantial volume of aluminum coating

wastes in the 1967-1968 time period. The estimated aluminum inventory will be used for the best-basis inventory.

Carbonate and Hydroxide. The estimated tank 241-TX-117 carbonate inventory is 5.4 times the HDW model inventory. The hydroxide ion in Hanford waste tanks is converted to carbonate by the absorption of carbon dioxide from the ambient air. One mole of absorbed carbon dioxide will react with two moles of hydroxide ion to form one mole of carbonate ion. The rate is difficult to model at best, and is accelerated by the use of airlift circulators installed in many Hanford underground storage tanks. Conversion of 77,000 kg of hydroxide to carbonate would account for the difference. The HDW model does not account for the absorption of carbon dioxide from the atmosphere.

Fluoride. The estimated fluoride inventory is 4.7 times that predicted by the HDW model. This is possibly the result of the HDW model assumptions that sodium fluoride is the only chemical compound containing fluoride and that it does not precipitate. The formation of insoluble fluoride compounds (such as sodium fluorophosphate) may be causing some fluoride to precipitate and remain in the tank.

Iron. The estimated iron inventory is skewed by the high iron concentration (2.4 wt% on silicon-free basis) reported for tank 241-TX-116 T2 salt cake. A later analysis of the tank 241-TX-116 salt cake (Schulz 1980) indicated very little insoluble material. The high iron concentration is not likely for a salt cake since iron is insoluble in alkaline solutions and significant iron concentration would not be expected in the evaporator feed solutions. Therefore, the HDW model iron inventory will be used for the best-basis.

Nitrate. The estimated nitrate inventory is 1.7 times that predicted by the HDW model. Much of the HDW model salt cake inventory is predicted by the SMM, and it is, therefore, difficult to determine the specific causes of this discrepancy. The global HDW model T1 and T2 salt cake concentrations (see Tables D3-1 and D3-2) are very reasonable, indicating that either the problem lies within the SMM model or that some feed inputs have been missed.

Phosphate. The estimated phosphate inventory is 1.8 times that predicted by the HDW model. Most of the phosphate (roughly 90 percent) is associated with the T1 salt cake layer. The discrepancy is primarily due to the different phosphate concentrations predicted for T1 salt cake.

Sodium. The predicted HDW sodium inventory is about 65 percent that predicted from analytical data for other T1 and T2 salt cake tanks. The waste density predicted by the HDW model for the T2 salt slurry is 1.49 g/cc, which is 13 percent below that normally expected for a salt cake. The total volume of salt cake predicted by the HDW model is 356 kL (94 kgal) less, which could explain another 16 percent of the difference. The remainder of the difference is the result of the low sodium concentration predicted for the T2 salt cake layer by the SMM. The global HDW model T2 salt cake sodium concentration is

very reasonable (see Table D3-2). Either there is an internal problem in the SMM model calculations or some feed inputs have been missed.

Total Hydroxide. Once the best-basis inventories were determined, the hydroxide inventory was calculated by performing a charge balance with the valences of the other analytes. In some cases, this approach requires that other analyte (e.g., sodium or nitrate) inventories be adjusted to achieve the charge balance. During such adjustments, the number of significant figures is not increased. The charge balance approach is consistent with that used by Agnew et al. (1997a). The revised total hydroxide inventory based on the best-basis inventory is 163,000 kg, which is about 17 percent higher than the HDW model estimate. The difference results from the generally higher chemical inventories indicated by analytical data for T1 and T2 salt cake tanks.

Cesium-137 and Strontium-90. The heat load for tank 241-TX-117 was not estimated (Kummerer 1995) due to a lack of waste and vapor space temperature data. The lack of a heat load estimate makes evaluation of validity of ^{137}Cs and ^{90}Sr inventories difficult. The HDW ^{137}Cs inventory is only 9 percent higher than predicted from analytical data from other waste tanks. To be conservative, the higher HDW model inventory will be used as the best-basis. The HDW model inventory for ^{90}Sr will also be used for the best-basis as insufficient ^{90}Sr analytical data are available for the salt cakes. Tank 241-TX-117 did not store complexed wastes (waste transactions designated as EVT, Agnew et al. 1997b) containing relatively high ^{90}Sr concentrations during 1974-1976, so the 86,000 Ci ^{90}Sr HDW inventory is not unreasonably low.

D4.0 DEFINE THE BEST-BASIS AND ESTABLISH COMPONENT INVENTORIES

Information about chemical, radiological, and/or physical properties is used to perform safety analyses, engineering evaluations, and risk assessment associated with waste management activities, as well as regulatory issues. These activities include overseeing tank farm operations and identifying, monitoring, and resolving safety issues associated with these operations and with the tank wastes. Disposal activities involve designing equipment, processes and facilities for retrieving wastes and processing them into a form that is suitable for long-term storage.

Chemical and radiological inventory information are generally derived using three approaches: (1) component inventories are estimated using the results of sample analyses, (2) component inventories are predicted using the HDW model based on process knowledge and historical information, or (3) a tank-specific process estimate is made based on process flowsheets, reactor fuel data, essential material usage, and other operating data.

An effort is underway to provide waste inventory estimates that will serve as the standard characterization for the various waste management activities (Hodgson and LeClair 1996). As part of this effort, an evaluation of available information for tank 241-TX-117 was performed including the following:

- Waste transactions and operating data to confirm that only 242-T Evaporator salt cakes and diatomaceous earth were expected in this tank
- Composition data from three waste tanks (241-T-108 [Brown et al. 1996], 241-T-109 [Brown et al. 1996], and 241-TX-116 [Horton 1977]) that are expected to have similar T1 salt cake compositions, and three waste tanks (241-U-102 [Hu et al. 1997], 241-U-105 [Brown and Franklin 1996], and 241-TX-116 [Horton 1977]) that are expected to have similar T2 salt cake compositions
- The quantity and composition of diatomaceous earth added to the tank
- An inventory estimate generated by the HDW model (Agnew et al. 1997a).

Based on this evaluation, a best-basis inventory was developed. No analytical data are available for the salt cake remaining in tank 241-TX-117 because no samples have been taken. The estimated inventory was, therefore, based on the composition of: the T1 salt cakes in tanks 241-T-108, 241-T-109, and 241-TX-116, and the T2 salt cakes in tanks 241-U-102, 241-U-105, and 241-TX-116 since the wastes in these tanks have actually been analyzed. The HDW model inventories were used when no other data were available or when analytical data were suspect.

The waste in tank 241-TX-117 consists of 2,260 kL (597 kgal) of T1 and T2 salt cake produced by the 242-T Evaporator and 110 kL (29 kgal) of diatomaceous earth. The

best-basis inventory for tank 241-TX-117 is presented in Tables D4-1 and D4-2. The inventory values reported in Tables D4-1 and D4-2 are subject to change. Refer to the Tank Characterization Database (TCD) for the most current inventory values.

Best-basis tank inventory values are derived for 46 key radionuclides (as defined in Section 3.1 of Kupfer et al. 1997), all decayed to a common report date of January 1, 1994. Often, waste sample analyses have only reported ⁹⁰Sr, ¹³⁷Cs, ^{239/240}Pu, and total uranium, or (total beta and total alpha) while other key radionuclides such as ⁶⁰Co, ⁹⁹Tc, ¹²⁹I, ¹⁵⁴Eu, ¹⁵⁵Eu, and ²⁴¹Am, etc., have been infrequently reported. For this reason it has been necessary to derive most of the 46 key radionuclides by computer models. These models estimate radionuclide activity in batches of reactor fuel, account for the split of radionuclides to various separations plant waste streams, and track their movement with tank waste transactions. (These computer models are described in Kupfer et al. 1997, Section 6.1 and in Watrous and Wootan 1997.) Model generated values for radionuclides in any of 177 tanks are reported in the HDW Rev. 4 model results (Agnew et al. 1997a). The best-basis value for any one analyte may be either a model result or a sample or engineering assessment-based result if available. (No attempt has been made to ratio or normalize model results for all 46 radionuclides when values for measured radionuclides disagree with the model.) For a discussion of typical error between model derived values and sample derived values, see Kupfer et al. 1997, Section 6.1.10.

Table D4-1. Best-Basis Inventory Estimates for Nonradioactive Components in Tank 241-TX-117 (Effective May 31, 1997). (2 Sheets)

Analyte	Total inventory (kg)	Basis (S, M, E, or C) ¹	Comment
Al	73,400	E	
Bi	<661	E	
Ca	1,350	E	
Cl	13,900	E	
TIC as CO ₃	166,800	E	
Cr	3,470	E	
F	<16,300	E	
Fe	6,310	M	
Hg	3.75	M	
K	<4,670	E	
La	1.39 E-04	M/E	
Mn	1,380	E	
Na	817,000	E	
Ni	<257	E	

Table D4-1. Best-Basis Inventory Estimates for Nonradioactive Components in Tank 241-TX-117 (Effective May 31, 1997). (2 Sheets)

Analyte	Total inventory (kg)	Basis (S, M, E, or C) ¹	Comment
NO ₂	82,000	E	
NO ₃	1.21 E+06	E	Concentration varies significantly between T1 salt cakes.
OH _{TOTAL}	157,000	C	Estimated by ion charge balance.
Pb	< 881	E	
P as PO ₄	128,000	E	Concentration varies significantly between T1 salt cakes.
Si	19,600	E	
S as SO ₄	57,100	E	
Sr	< 37	E	
TOC	23,700	E	
U _{TOTAL}	< 1,690	E	
Zr	59	E	

¹S = Sample-based

M = Hanford Defined Waste model-based, Agnew et al. (1997a)

E = Engineering assessment-based

C = Calculated by charge balance; includes oxides as hydroxides, not including CO₃, NO₂, NO₃, PO₄, SO₄, and SiO₃.

Table D4-2. Best-Basis Inventory Estimates for Radioactive Components in Tank 241-TX-117, Decayed to January 1, 1994 (Effective May 31, 1997). (2 sheets)

Analyte	Total inventory (Ci)	Basis (S, M, or E) [†]	Comment
³ H	168	M	
¹⁴ C	24.4	M	
⁵⁹ Ni	2.66	M	
⁶⁰ Co	26.2	E/M	
⁶³ Ni	255	M	
⁷⁹ Se	2.53	M	
⁹⁰ Sr	86,000	M	
⁹⁰ Y	86,100	M	
⁹³ Zr	12.4	M	
^{93m} Nb	9.04	M	
⁹⁹ Tc	174	M	
¹⁰⁶ Ru	0.00425	M	
^{113m} Cd	63.3	M	
¹²⁵ Sb	110	M	
¹²⁶ Sn	3.81	M	
¹²⁹ I	0.336	M	
¹³⁴ Cs	12.7	E	Based on analyses of 242-T Evaporator salt cakes in other tanks.
¹³⁷ Cs	315,000	M	
^{137m} Ba	298,000	M	
¹⁵¹ Sm	8,890	M	
¹⁵² Eu	2.68	M	
¹⁵⁴ Eu	413	M	
¹⁵⁵ Eu	162	M	
²²⁶ Ra	1.41 E-04	M	
²²⁷ Ac	9.48 E-04	M	
²²⁸ Ra	0.316	M	
²²⁹ Th	0.00729	M	
²³¹ Pa	0.00402	M	
²³² Th	0.0192	M	
²³² U	1.54	M	

Table D4-2. Best-Basis Inventory Estimates for Radioactive Components in Tank 241-TX-117, Decayed to January 1, 1994 (Effective May 31, 1997). (2 sheets)

Analyte	Total inventory (Ci)	Basis (S, M, or E) ¹	Comment
²³³ U	5.92	M	
²³⁴ U	5.12	M	
²³⁵ U	0.223	M	
²³⁶ U	0.0704	M	
²³⁷ Np	0.636	M	
²³⁸ Pu	1.25	M	
²³⁸ U	5.52	M	
²³⁹ Pu	81.7	M	
²⁴⁰ Pu	9.43	M	
²⁴¹ Am	39.7	M	
²⁴¹ Pu	74.0	M	
²⁴² Cm	0.0997	M	
²⁴² Pu	3.81 E-04	M	
²⁴³ Am	0.00135	M	
²⁴³ Cm	0.00909	M	
²⁴⁴ Cm	0.0898	M	

¹S = Sample-based

M = Hanford Defined Waste model-based, Agnew et al. (1997a)

E = Engineering assessment-based.

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D5.0 APPENDIX D REFERENCES

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