

## NO<sub>X</sub> AND PAHS REMOVAL FROM INDUSTRIAL FLUE GAS BY USING ELECTRON BEAM TECHNOLOGY IN THE ALCOHOL ADDITION

## Y.-X. Sun<sup>1</sup>, A. G. Chmielewski<sup>1</sup>, J. Licki<sup>2</sup>, S. Bułka<sup>1</sup>, K. Kubica<sup>3</sup>, Z. Zimek<sup>1</sup>

<sup>1</sup>Institute of Nuclear Chemistry and Technology, 16 Dorodna str., 03-195 Warsaw, Poland <sup>2</sup>Institute of Atomic Energy, 05-400, Otwock-Świerk, Poland <sup>3</sup>Institute for Chemical Processing of Coal, 1 Zamkowa str., 44-803 Zabrze, Poland

The preliminary test of NO<sub>x</sub> and Polycyclic Aromatic Hydrocarbons (PAHs) removal from flue gas were investigated in the alcohol addition by using electron beam irradiation in EPS Kawęczyn. Experimental conditions were as follows: flue gas flow rate 5000 nM<sup>3</sup>/hr; humidity 4-5%; inlet concentrations of SO<sub>2</sub> and NO<sub>x</sub>, which were emitted from power station, were 192 ppm and 106 ppm, respectively; ammonia addition is 2.75 m<sup>3</sup>/hr; alcohol addition is 600 l/hr. It was found that NO<sub>x</sub> removal efficiency in the presence of alcohol was increased by 10% than without alcohol addition when the absorbed dose was below 6 kGy. The NO<sub>x</sub> removal efficiency was decreased when the absorbed dose was higher than 10 kGy.

In order to understand PAHs' behavior under EB irradiation, inlet PAHs (emitted from coal combustion process) sample and outlet PAHs (after irradiation) sample were collected by using a condensed bottle connected with XAD-2 adsorbent and active carbon adsorbent and were analyzed by a GC-MS. It is found that: at the 8 kGy absorbed dose, concentrations of PAHs with small aromatic rings ( $\leq$ 3, except Acenaphthylene) are reduced and concentrations of PAHs with large aromatic rings ( $\leq$ 4) are increased. A possible mechanism is proposed.

Acknowledgement

This work was supported by Polish Committee for Scientific Research, in the frame of grant No. 3T09B 061.