



WASHOUT BEHAVIOUR OF CHROMIA-DOPED UO_2 AND GADOLINIA FUELS IN LWR ENVIRONMENTS

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AREVA NP



▶ Introduction

- ◆ General background
- ◆ AREVA Cr₂O₃-doped fuel development
- ◆ Focus on washout fuel consequences

▶ Assessment of AREVA Cr₂O₃-doped fuel behaviour in a defective fuel rod

- ◆ Fuel specimens
- ◆ Oxidation behaviour
 - Experimental means
 - Influence of oxygen and pellets characteristics
- ◆ Washout behaviour
 - Experimental means
 - PWR and BWR conditions

▶ Conclusions



Introduction

General Background

▶ Action lines continuously reviewed to maintain nuclear power generation more economical and competitive

◆ **Long term** ▶ **Upgraded operating practices:**

- Power uprate / High burn-up / extended fuel cycles
- Improvement in nuclear plant availability and flexibility
 - Load-follow
 - Extended low power operations
 - Fast return to power

◆ **Short term** ▶ **Fuel reliability and robustness**

- Substantial lever for improving plant availability



Double challenge in the development of advanced fuel materials

AREVA Cr₂O₃-Doped Fuel Development

▶ Fuel development initiated by AREVA in 1990 with **the aim to provide a fuel product for:**

- ◆ **Better uranium utilization**
- ◆ **Higher reliability and robustness**

▶ AREVA optimized Cr₂O₃-doped UO₂ fuel **key advantages:**

- ◆ **Better Fuel Utilization**

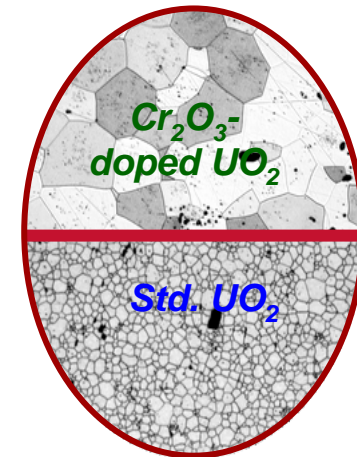
- High density fuel giving an **increase of the U²³⁵ mass** per fuel assembly

- ◆ **Improved PCI Performance**

- Very high adaptability in terms of power level variation (ramp rate)
- Bring **additional margins** in **fuel maneuverability**

- ◆ **Increased Operational Margins**

- **Enhanced fission gas retention** inducing **lower internal pressure** in fuel rods at end of life
 - Enable to **operate** the fuel assemblies up to high burnup levels, for reactor **power uprate, fuel stack length optimization**
 - **Improved back-end fuel cycle conditions**

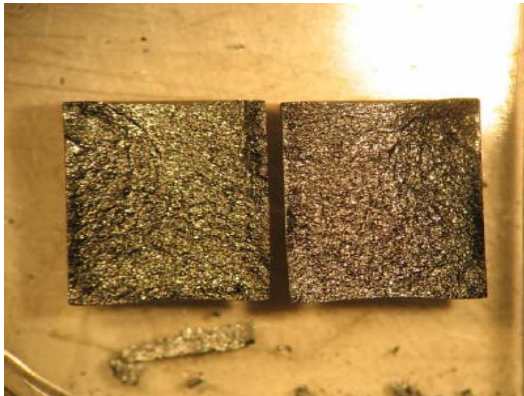


AREVA Cr₂O₃-Doped Fuel Development (cont.)

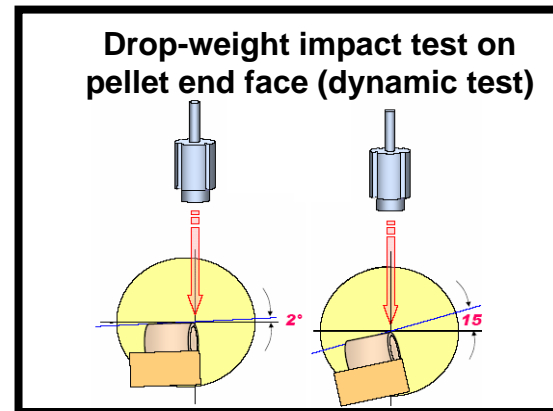
▶ AREVA optimized Cr₂O₃-doped UO₂ fuel **key advantages:**

◆ **Improved Reliability**

- Higher **resistance to chipping** (missing pellet surface)



Fracture of Cr₂O₃-doped pellet generates few particles



Chipping mass loss ratio :

$$\frac{\text{Cr}_2\text{O}_3 \text{ - doped fuel}}{\text{Std. UO}_2 \text{ fuel}} = 0.60$$

- **Enhanced wash-out behavior** for **lower activity release** in case of defective rods

Focus on Washout Fuel Consequences

- ▶ Failed fuel degradation after the occurrence of primary defects can affect nuclear reactor operation:
 - ◆ High **release of fission products to primary coolant**
 - ◆ **Fuel washout** due to direct contact of fuel pellet with the coolant
 - Tramp uranium results in increasing background activity level
- ▶ Previous experiments and observations reveal:
 - ◆ **UO₂ fuel readily oxidized** (high oxygen diffusion rate)
 - ◆ **Oxidation process controlled by surface reaction** (grain boundaries)



Fuel pellet characteristics may have a crucial impact on its dissolution tendency



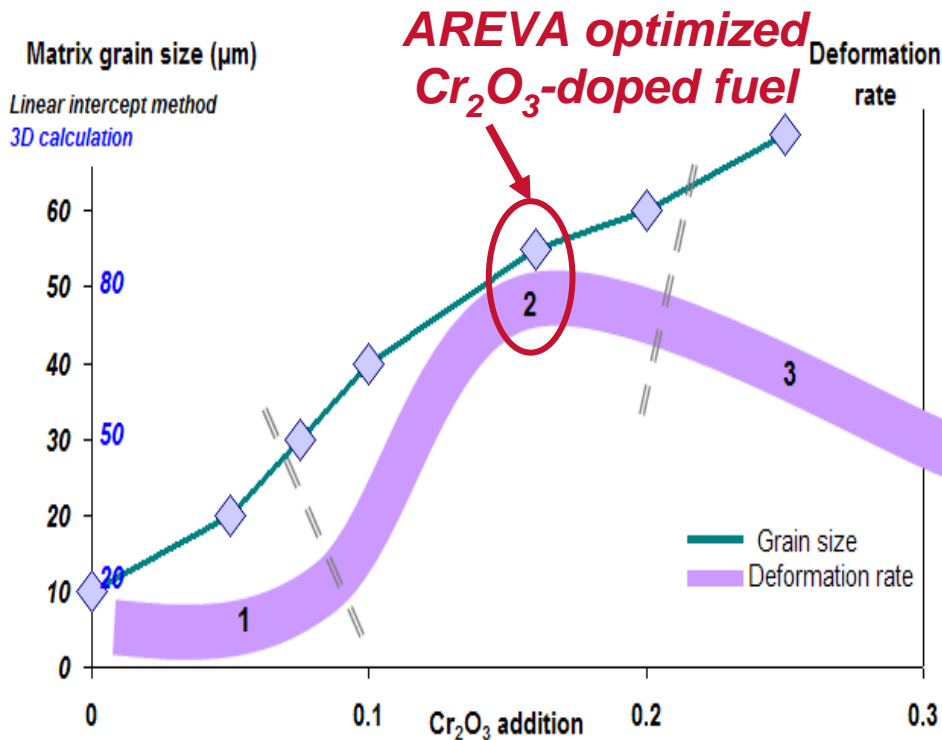
Assessment of AREVA Cr₂O₃-Doped Fuel Behaviour in a Defective Fuel Rod

Fuel Specimens

Fuel Specimens

AREVA Optimized Chromia-Doped UO₂

- ▶ **General features:** large grain (FGR target), viscoplastic (PCI target) microstructure with the right amount of Cr₂O₃



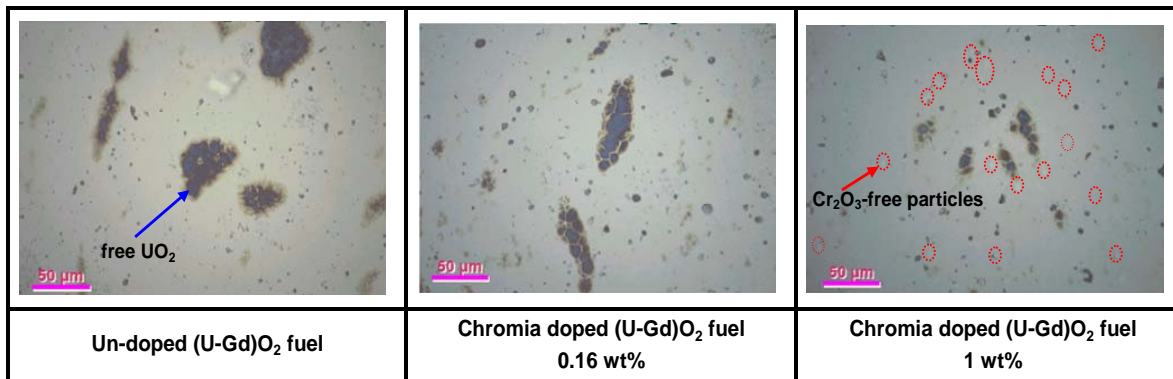
Maximum of improvement with a nominal Cr₂O₃ addition of 0.16 wt.%

- ◆ Homogeneous crystalline structure
 - Large grain size of 50-60 µm
 - No large Cr₂O₃-free particles in fuel matrix
- ◆ Enhanced densification of fuel matrix
 - High density 96 % TD easily achievable

Fuel Specimens

AREVA Optimized Chromia-Doped (U-Gd)O₂

- ▶ Development process launched on the model of Cr₂O₃-doped UO₂ fuel
- ▶ Fundamental and basic studies:
 - ◆ To assess the influence of Cr₂O₃ doping on (U-Gd)O₂ properties
 - ◆ To optimize the Cr₂O₃ doping level for maximum of improvements
- ▶ (U-Gd)O₂ **densification behaviour** can be significantly **enhanced**
 - Low doping amount: < 0.075 wt% → hardly no effect
 - High doping amount: >0.3 wt% → solarization effect
- ▶ (U-Gd)O₂ **phase structure** only modified when Cr₂O₃ in excess to its solubility limit into the fuel matrix:



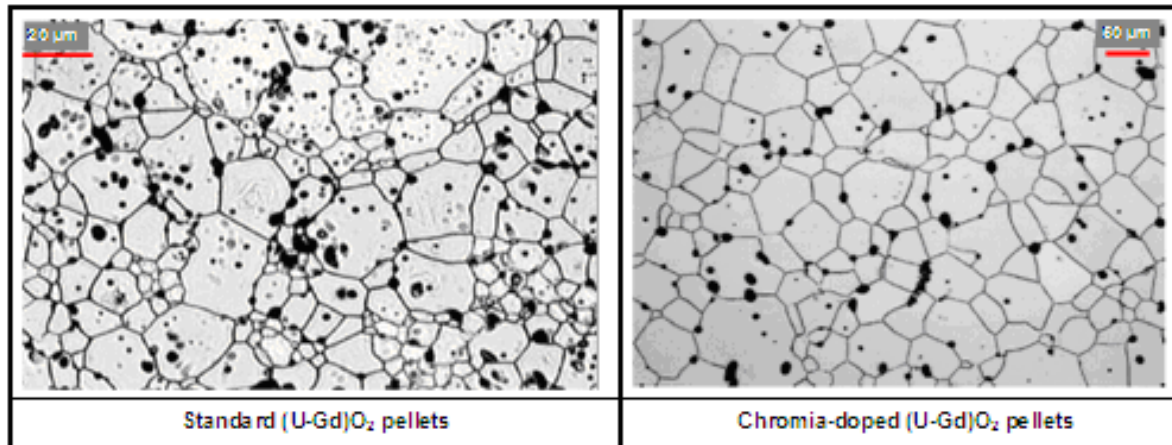
FUEL

Fuel Specimens

AREVA Optimized Chromia-Doped (U-Gd)O₂ (Cont.)

▶ (U-Gd)O₂ **polycrystalline structure:**

- ◆ Large grains > 40 μm from 0.16 wt% doping amounts



▶ (U-Gd)O₂ fuel **thermal conductivity:**

- ◆ Laser flash method up to 1500°C
- ◆ **No impact of Cr₂O₃-doping** in comparison to non-doped (U-Gd)O₂ fuel

Fuel Specimens

AREVA Optimized Chromia-Doped (U-Gd)O₂ (Cont.)

▶ (U-Gd)O₂ **fuel viscoplasticity:**

- ◆ Creep testing at high temperatures (1500°C) and compression stresses [30-60 MPa]
- ◆ In comparison to non-doped (U-Gd)O₂ fuel:
 - **Cr₂O₃ doping enhances distinctly the fuel plastic behaviour: improvement factor up to 10**
 - Positive effect attributed to the grain size enlargement allowing compensation of the fuel matrix hardening due to solid solution formation



Optimum Cr₂O₃ level of 0.16 wt% for (U-Gd)O₂ doping as for UO₂ resulting in a large grain and viscoplastic fuel matrix without detrimental effects on thermal characteristics



Assessment of AREVA Cr₂O₃-Doped Fuel Behaviour in a Defective Fuel Rod

Oxidation Behaviour

Fuel Oxidation Behaviour

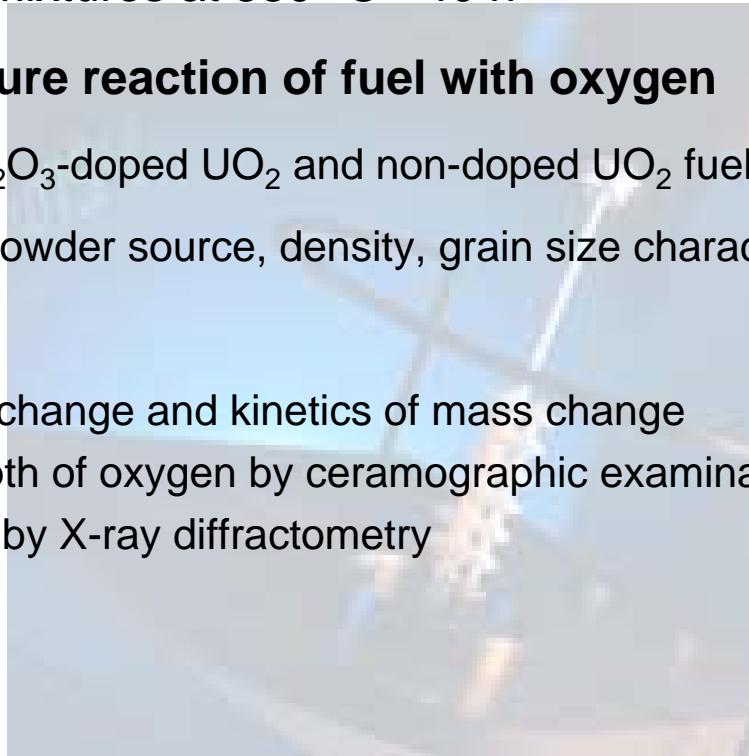
Experimental Means

► Test program to characterize:

- ◆ **The oxidation behaviour** of fuel pellets by thermogravimetry tests under Ar/O₂ gas mixtures at 380 °C - 40 h

↳ **Pure reaction of fuel with oxygen**

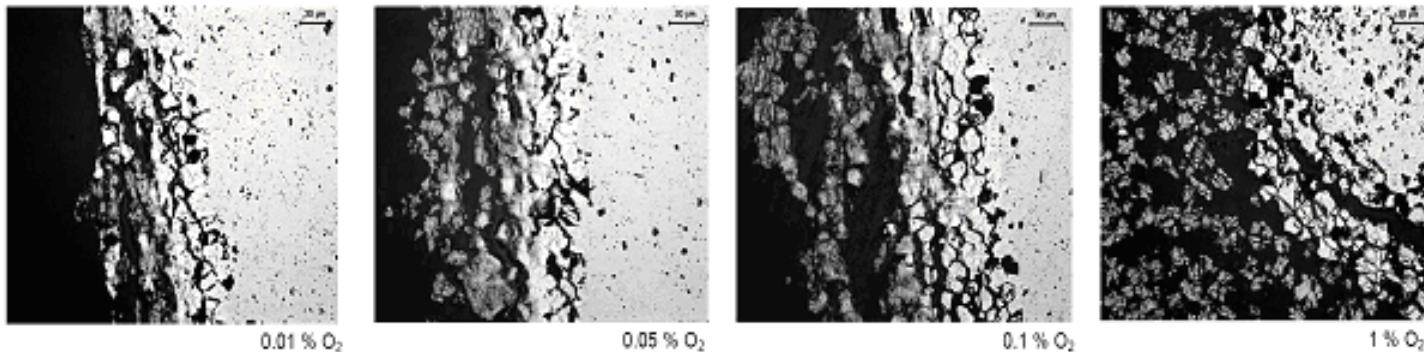
- Comparison Cr₂O₃-doped UO₂ and non-doped UO₂ fuel pellets
 - Variants: UO₂ powder source, density, grain size characteristics
- ◆ **Follow-up:**
 - Absolute mass change and kinetics of mass change
 - Penetration depth of oxygen by ceramographic examinations
 - Phase analysis by X-ray diffractometry



Fuel Oxidation Behaviour

Investigation of Pure Reaction with Oxygen

- ▶ Non-doped UO_2 samples - Ar+ O_2 mixtures – $0.01 < \text{O}_2\% < 1$



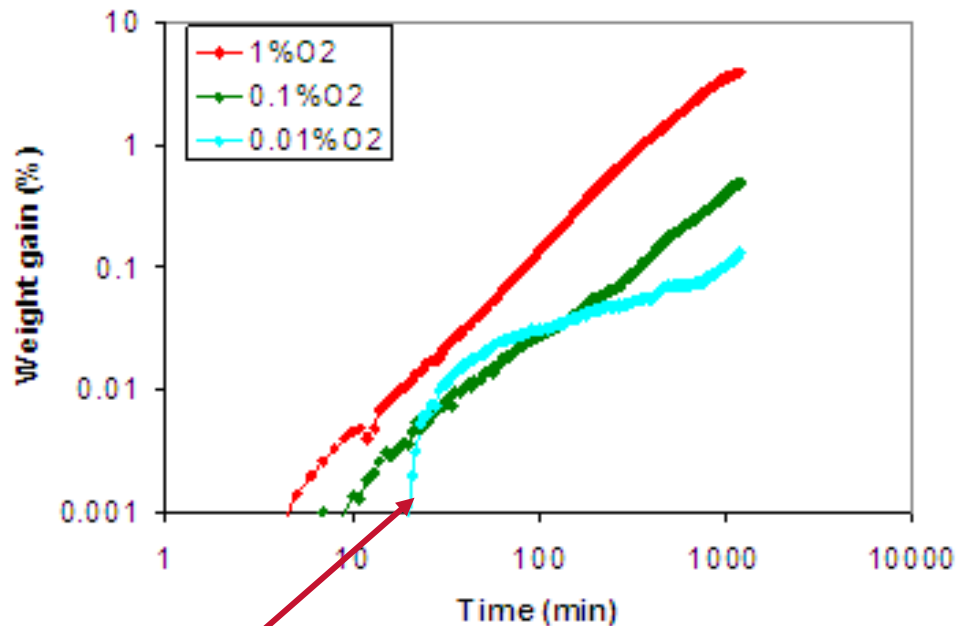
◆ Intergranular oxidation mode

- Consequent matrix volume increase: oxidizing-induced stresses leading to crack propagation and to strip deeper grain layers
- Significant attack from 0.1% O_2
- Pellet pulverization after 20h testing under Ar+1% O_2

Fuel Oxidation Behaviour

Investigation of Pure Reaction with Oxygen (Cont.)

- Oxidation kinetics varies according to $[O_2]$ in Ar

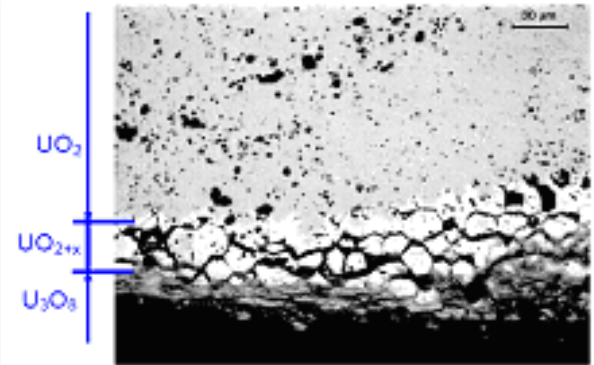
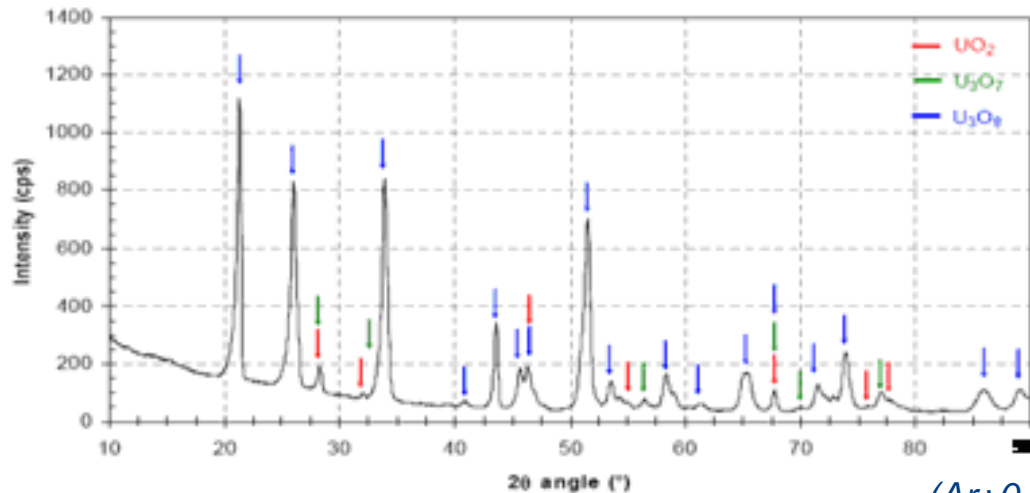


Induction period: oxidation governed by surface processes mainly

Fuel Oxidation Behaviour

Investigation of Pure Reaction with Oxygen (Cont.)

► Oxidation phase identification



(Ar+0.01% O₂ – 380 °C)



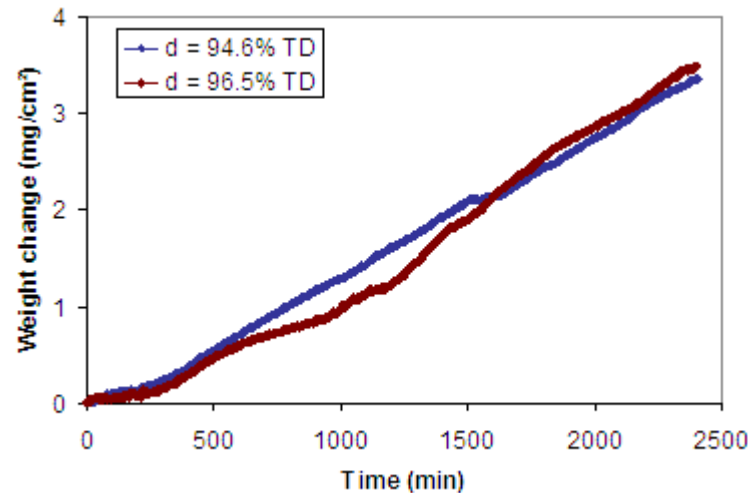
Two-step reaction process: $\text{UO}_2 \rightarrow \text{U}_4\text{O}_9/\text{U}_3\text{O}_7 \rightarrow \text{U}_3\text{O}_8$

Formation of intermediate $\text{U}_4\text{O}_9/\text{U}_3\text{O}_7$ compound corresponds to initial non-linear oxidation before intergranular cracks formation

Fuel Oxidation Behaviour

Influence of Pellet Characteristics

- ▶ UO_2 -source powder and pellet density (94.5 to 96.5 % TD) play no fundamental role in the UO_2 fuel oxidation:



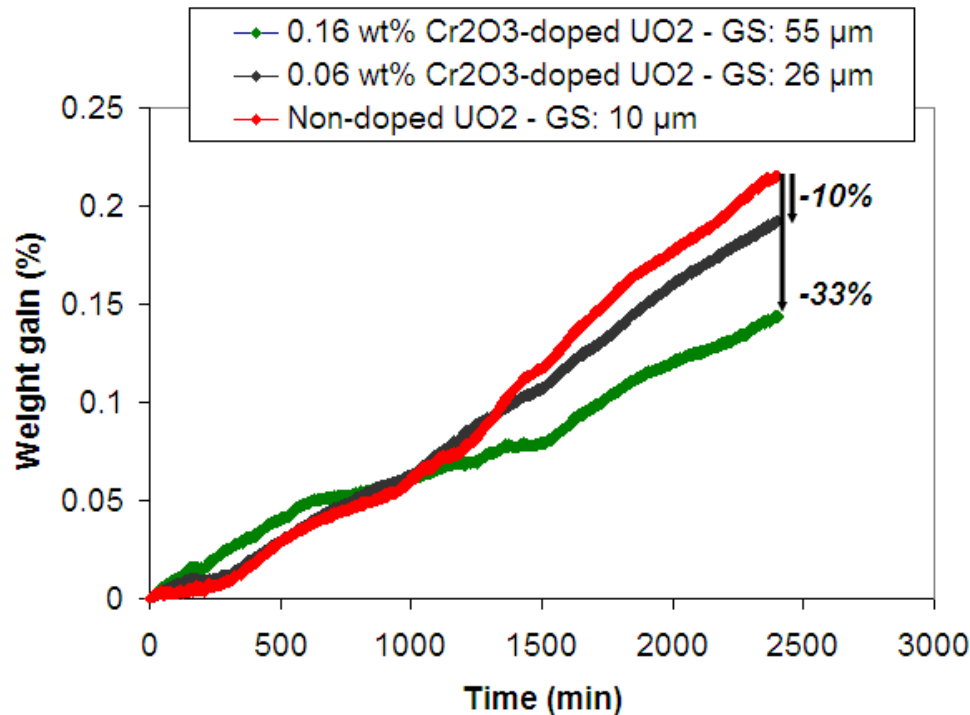
UO₂ samples
(Ar+0.01% O₂ – 380 °C)

Fuel Oxidation Behaviour

Influence of Pellet Characteristics (Cont.)

► Cr_2O_3 -doping distinctively enhances the resistance of fuel pellets against oxidation:

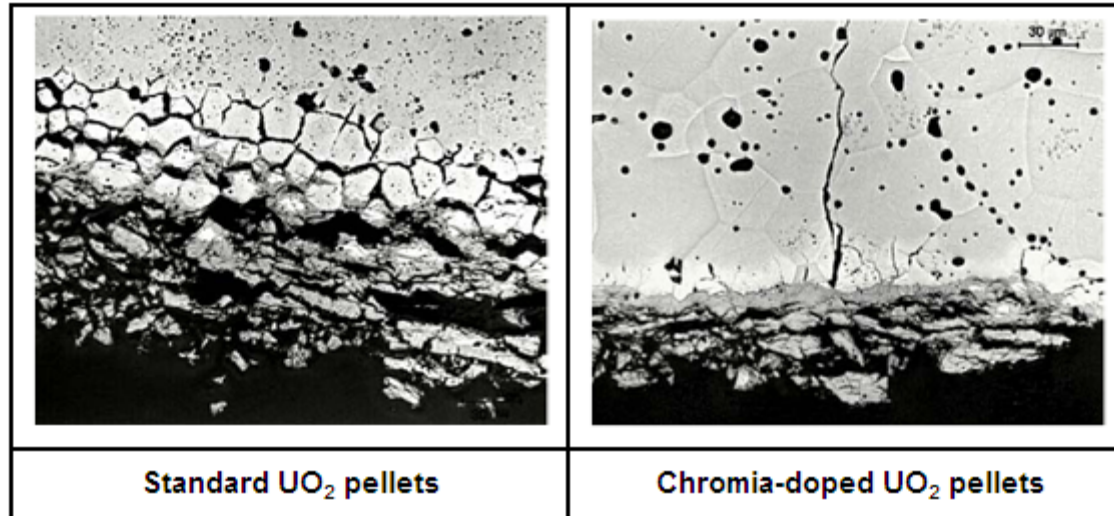
- ◆ Limited improvement when increasing the fuel density
- ◆ **More decisive effect due to fuel matrix grain size**



Fuel Oxidation Behaviour

Influence of Pellet Characteristics (Cont.)

- ▶ Oxidation mode for Cr_2O_3 -doped fuel pellets:




Ar+0.01% O₂ at 380 °C



No intergranular cracks formed in Cr_2O_3 -doped samples:

- ◆ Oxidation surface mode
- ◆ Oxidized layers offer a protection against oxygen diffusion and decelerate the oxidation rate
- ◆ Attacked layer thinner by a factor up to 2.5 compared to non-doped UO_2

FUEL



Assessment of AREVA Cr₂O₃-Doped Fuel Behaviour in a Defective Fuel Rod

Corrosion Behaviour

Fuel Corrosion Behaviour Experimental Means

► Test program to characterize:

- ◆ **The corrosion behaviour** of fuel pellets by autoclave leaching tests:

↳ **Simulation of in-reactor environments**

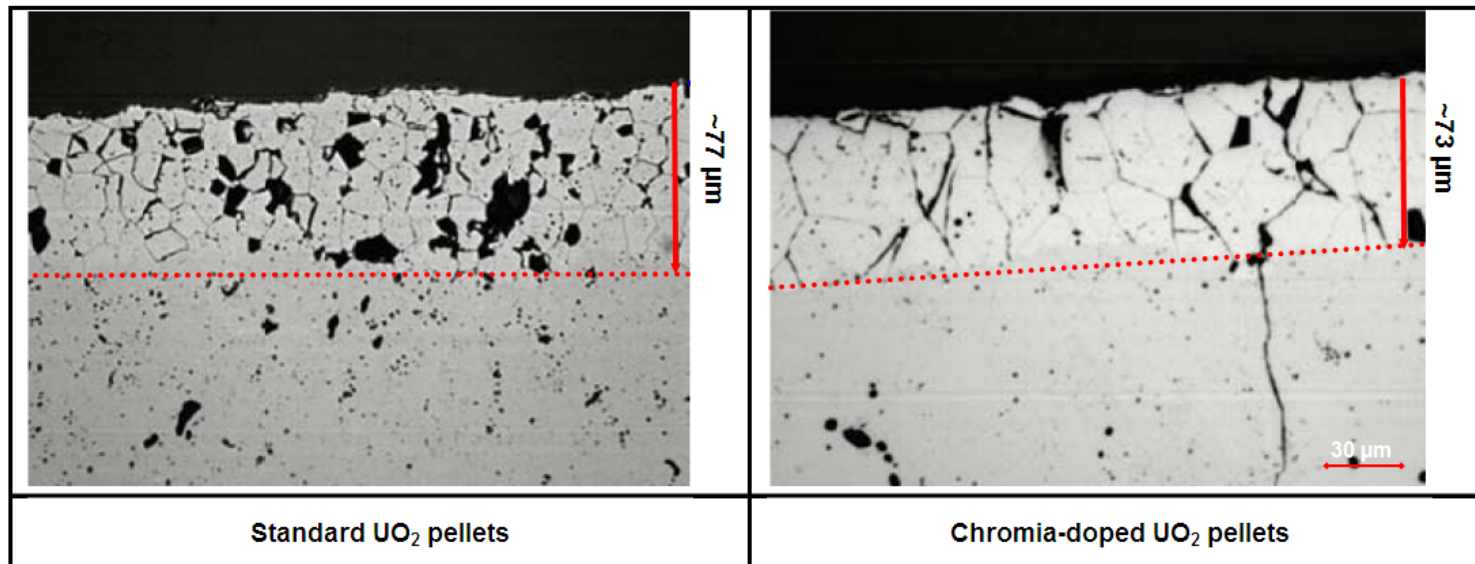
Study case	Pressure (bar)	Temperature (°C)	Water chemistry
BWR	70	290 (water)	70 ppm H ₂ O ₂
	70	360 (steam)	70 ppm H ₂ O ₂
PWR	180	360 (water)	650 ppm B by H ₃ BO ₃ , 2 ppm Li by LiOH (pH ~7.4)
	100	400 (steam)	No additives

- Comparison Cr₂O₃-doped UO₂ / (U-Gd)O₂ and non-doped UO₂ / (U-Gd)O₂ fuel pellets

- ◆ Follow-up as for thermogravimetry testing

Fuel Corrosion Behaviour PWR Conditions (*Water and Steam*)

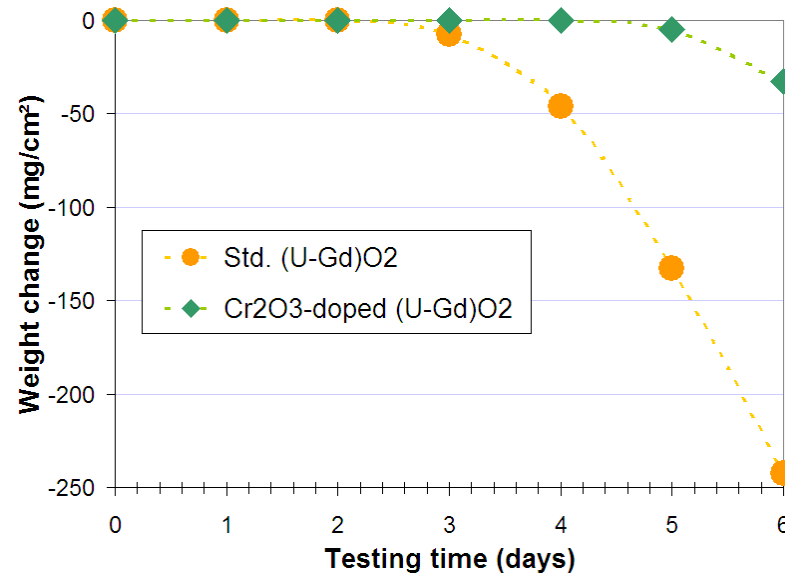
- ▶ No washout achieved due to the low O_2 concentration, but the corrosion process is initiated:
 - ◆ Non-doped UO_2 grain boundaries highly weakened
 - ◆ U_3O_7 formation as an intermediate step
 - ◆ Higher resistance of the Cr-doped fuel outside surface



[water at 360°C, 180 bar, pH 7.4 (650 ppm B – 1.5 ppm Li)]

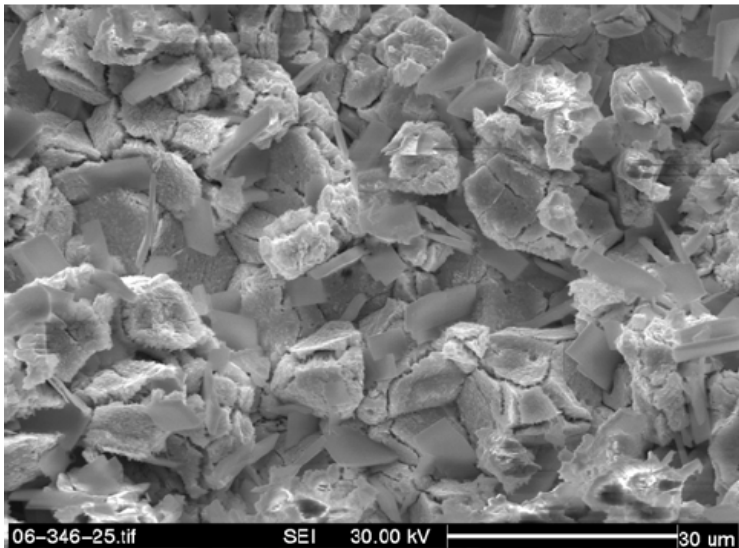
Fuel Corrosion Behaviour BWR Conditions

- ▶ In the most severe conditions investigated (360°C / 70 bars / 70 ppm H_2O_2)
 - ◆ Real washout of samples occurs
 - ◆ Significant differences revealed between non-doped and doped-samples:
 - **Mass loss reduced by a factor up to 5** for both configurations Cr_2O_3 -doped UO_2 vs. non-doped UO_2 and Cr_2O_3 -doped (U-Gd) O_2 vs. non-doped (U-Gd) O_2
 - **Mass loss of doped samples is delayed** compared to non-doped samples

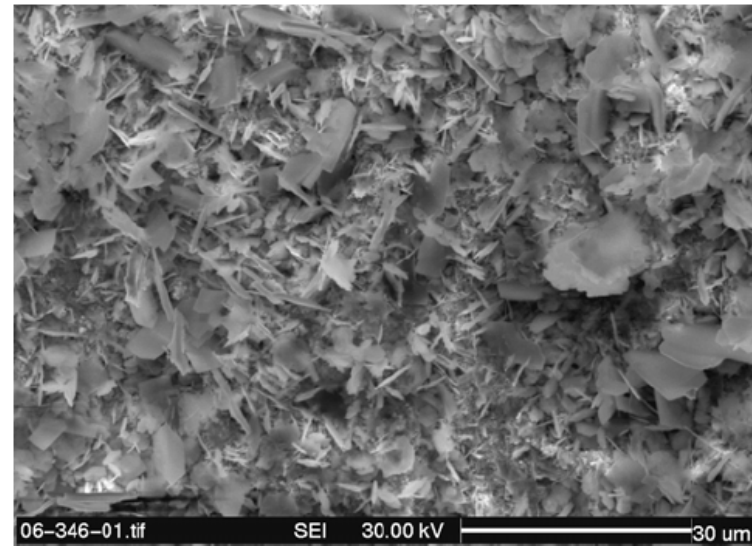


Fuel Corrosion Behaviour BWR Conditions (Cont.)

► Visual inspections on test-samples:



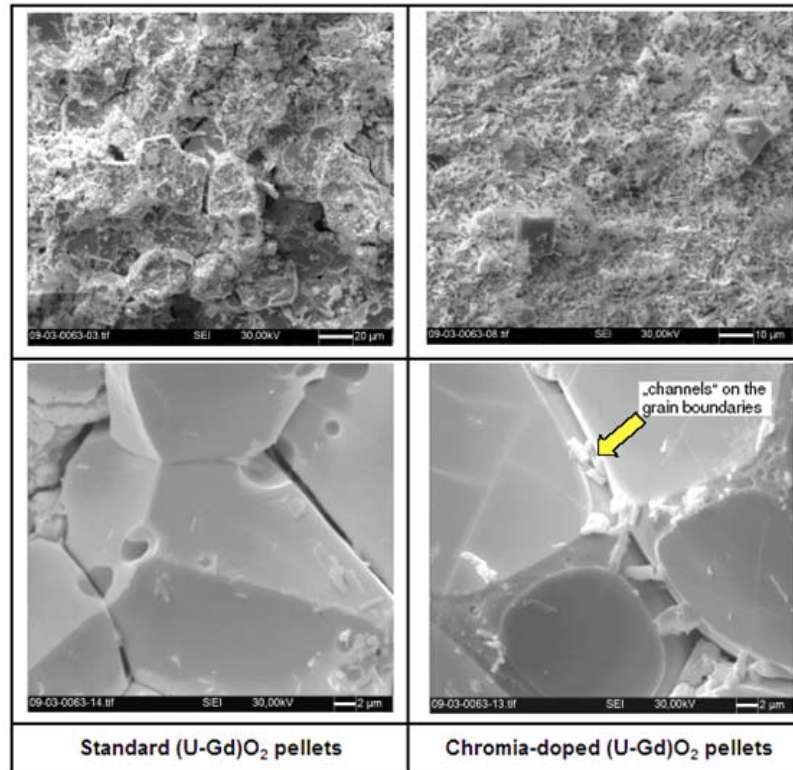
- ◆ **UO₂-based samples:** 'cauliflower' structure formed by cracked oxidized grains (volume expansion due to U₃O₈ conversion)



- ◆ **Cr₂O₃-doped based samples:** 'partially intact needle-like surface.'
- » **The outer layer protects the pellet meat from environment**

Fuel Corrosion Behaviour BWR Conditions (Cont.)

- ▶ Visual inspections on test-samples:



(top: hydrate regions bottom: leached broken-out areas)



Secondary phase residue on grain boundaries of Cr-doped (U-Gd)O₂ samples which possibly contributes to the better corrosion resistance



Conclusions

Conclusions

- ▶ The entire testing program demonstrates that **Cr₂O₃-doping enhances** the **corrosion resistance** of the fuel pellets
 - ◆ **Improved washout behaviour by a factor of 5** compared to non-doped UO₂ / (U-Gd)O₂ fuel pellets
- ▶ The **main driver of improvement is the grain size enlargement** of the fuel matrix
 - ◆ The optimum doping amount of **0.16 wt% Cr₂O₃** specified by AREVA for UO₂ and (U-Gd)O₂ fuels is especially efficient to obtain large grains of 50 and 40 μm respectively
- ▶▶
 - ◆ **Enhanced operational behaviour** desired to struggle against disintegration of fuel in case of defective rods and combined consequences to LWR primary coolant contaminations
 - ◆ The **AREVA Cr₂O₃-doped UO₂ and doped (U-Gd)O₂ fuels present high reliability and robustness features**
 - ◆ Both fuel types are **currently under irradiation** in commercial nuclear plants to meet future more demanding operating requirements

Thank you for your attention!

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