Posters (Wednesday, February 12, 2014 15:30)

Electronic and Magnetic Properties of NpNi₅

A. Hen,^{1,2} E. Colineau,¹ R. Eloirdi,¹ J.-C. Griveau,¹ J.-P. Sanchez,⁴ A. B. Shick,^{1,5} <u>I. Halevy</u>,^{2,3} I. Orion,² and R. Caciuffo¹

¹European Commission, Joint Research Centre, Institute for Transuranium Elements, Postfach 2340, D-76125 Karlsruhe, Germany ²Ben Gurion University, IL84105 Beer-Sheva, Israel ³Nuclear Research Center Negev, P.O. Box 9001, IL84190 Beer-Sheva, Israel ⁴SPSMS, UMR-E CEA/UJF-Grenoble 1, INAC, FR-38054 Grenoble, France ⁵Academy of Sciences of the Czech Republic, Na Slovance 2, CZ-182 21 Prague, Czech Republic

The physical properties of binary actinides and transition metal alloys are of great importance for the safety assessment of nuclear fuels. Since transition metals are major components of the cladding material of fuel rods (stainless steel, HT-9 etc.), the physical properties of those compounds formed by accidental fuelcladding interactions could have limiting factors on the fabrication, life time operation and disposal of nuclear fuels. Binary compound of the form ReT₅ (Re = rare earth, T = Transition metal) has been in the focus of interest mainly because of their magnetic properties (small Re to T stoichiometric ratio, large spontaneous magnetization and high Curie temperature) and their ability to store large amount of hydrogen per formula unit (f. u.)

Reported in 1997 [1], NpNi₅ was synthesized and identified to have hexagonal (D₂d) CaCu₅ crystallographic structure, with room-temperature lattice parameters a = 8.3107(1) Å and c = 8.1058(1) Å. In the present study, NpNi₅ has been synthesized and characterized by means of powder x-Ray diffraction (**Fig.** 2 left panel), Superconducting – Quantum – Interference – Device magnetometry (SQUID, **Fig. 2** right panel), ²³⁷Np Mössbauer spectroscopy [2] (**Fig. 3** left panel) and specific heat measurements (**Fig. 3** right panel).

Magnetization curves indicate that NpNi₅ is a ferromagnet ($T_C \sim 16$ K), fit of the paramagnetic part to the Curie–Weiss law (C~1.7 emu·K/mol, $\theta_P \sim 14.6$ K) gives an effective moment $\mu_{eff} \sim 3.7 \mu_B$ per f.u. – no magnetization hysteresis was observed. The isomer shift ($\delta_{IS} \approx -11.1$ mm/s vs. NpAl₂) observed in Mössbauer spectra suggests a tetravalent Np state, but considering the influence of conduction electrons we determine a Np³⁺ (5f⁴ configuration) oxidation state. The hyperfine field determined by fitting of the spectra (~439T) gives an ordered moment at the Np site $\mu_{Np} \sim 2 \mu_B$ per Np ion (1 $\mu_B = 215$ T [3]). The magnetic transition is clearly visible in the temperature dependence of the specific heat, and a magnetic phase diagram as a function of temperature and external magnetic field was generated.

In this communication we will present the results obtained and discuss them in the light of theoretical considerations.



Fig. 2 Left: Powder x-ray diffraction pattern of NpNi₅, black dot – measured data, red line – fitted profile, blue line – difference profile, vertical tick – angular position of Bragg peaks. <u>Right</u>: Temperature dependence of the magnetic susceptibility, red line – fit to the Curie-Weiss law.



Fig. 3 Left: Mössbauer spectra taken in the ferromagnetic region, dots – measured data, lines – fits of hyperfine fields to the measured data, spectrum is composed of one hyperfine field as expected from the crystallography.

Right: Normalized heat capacity as a function of temperature in various external magnetic fields.

ACKNOWLEDGEMENTS

The high purity Np metals required for the fabrication of the compound were made available through a loan agreement between Lawrence Livermore National Laboratory and ITU, in the frame of a collaboration involving LLNL, Los Alamos National Laboratory and the US Department of Energy.

REFERENCES

[1] M. Akabori et al., Journal of Alloys and Compounds 257 (1997) 268-272.

[2] R.L. Mössbauer., Zeitschrift für Physik, Bd. 151, S. 124-143 (1958).

[3] B. D. Dunlap and G. M. Kalvius, in Handbook on the Physics and Chemistry of the Actinides,

edited by A. J. Freeman and G. H. Lander (North-Holland, Amsterdam, 1985), Vol. 2, p. 329.