1 Distinguishing between old and modern permafrost sources in the Northeast

- **Siberian land-shelf system with compound-specific** δ^2 **H analysis**
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- 21 **Keywords:** deuterium isotopes, yedoma, ice complex deposit, n-alkanoic acids, n-
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- 24

25 **Abstract**

- 26 Pleistocene ice complex permafrost deposits contain roughly a quarter of the organic 27 carbon (OC) stored in permafrost terrain. When permafrost thaws, its OC is 28 remobilized into the (aquatic) environment where it is available for degradation, 29 transport or burial. Aquatic or coastal environments contain sedimentary reservoirs 30 that can serve as archives of past climatic change. As permafrost thaw is increasing 31 throughout the Arctic, these reservoirs are important locations to assess the fate of 32 remobilized permafrost OC.
- 33 We here present compound-specific deuterium (δ^2H) analysis on leaf waxes as a tool 34 to distinguish between OC released from thawing Pleistocene permafrost (Ice 35 Complex Deposits; ICD) and from thawing Holocene permafrost (from near-surface 36 soils). Bulk geochemistry (%OC, δ^{13} C, %total nitrogen; TN) was analyzed as well as 37 the concentrations and δ^2H signatures of long-chain *n*-alkanes (C₂₁ to C₃₃) and 38 mid/long-chain *n*-alkanoic acids $(C_{16}$ to $C_{30})$ extracted from both ICD-PF samples 39 (n=9) and modern vegetation/O-horizon (Topsoil-PF) samples (n=9) from across the 40 northeast Siberian Arctic.
- 41 Results show that these Topsoil-PF samples have higher %OC, higher OC/TN values,
- 42 and more depleted $\delta^{13}C$ -OC values than ICD-PF samples, suggesting that these former
- 43 samples trace a fresher soil and/or vegetation source. Whereas the two investigated 44 sources differ on the bulk geochemical level, they are, however, virtually
- 45 indistinguishable when using leaf wax concentrations and ratios.

46 However, on the molecular-isotope level, leaf wax biomarker δ^2H values are 47 statistically different between Topsoil-PF and ICD-PF. For example, the mean δ^2H 48 value of C₂₉ *n*-alkane was -246±13‰ (mean±stdev) for Topsoil-PF and -280±12‰ 49 for ICD-PF. With a dynamic isotopic range (difference between two sources) of 34 to 50 50‰, the isotopic fingerprints of individual, abundant, biomarker molecules from 51 leaf waxes can thus serve as end-members to distinguish between these two sources. 52 We tested this molecular δ^2H tracer along with another source-distinguishing 53 approach, dual-carbon $(\delta^{13}C-\Delta^{14}C)$ isotope composition of bulk OC, for a surface 54 sediment transect in the Laptev Sea. Results show that general offshore patterns 55 along the shelf-slope transect are similar, but the source apportionment between the 56 approaches vary, which may highlight the advantages of either. This study indicates 57 that the application of δ^2 H leaf wax values has potential to serve as a complementary 58 quantitative measure of the source and differential fate of OC thawed out from 59 different permafrost compartments.

61 **1 Introduction**

62 Climate warming is causing permafrost soils to thaw, exposing its organic matter 63 (OM) to decomposition (e.g., Schuur et al., 2015; Zimov et al., 1993; Semiletov et al., 64 2012). Thaw will increase the hydrological connectivity of landscapes and will cause 65 release of OM into the aquatic environment (Walvoord et al., 2012; Vonk et al., 2015; 66 Anderson et al., 2011). Here, the OM can continue to decompose, generating 67 greenhouse gases (e.g., Semiletov et al., 1996a,b; Anderson et al., 2009; Shakhova et 68 al., 2015), or be destined for burial in inland and coastal sediments. These 69 sedimentary archives serve as long- and short-term reservoirs that attenuate 70 greenhouse gas emissions from thawing permafrost (Vonk and Gustafsson, 2013; 71 Semiletov et al., 2011).

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73 The release of OM from thawing permafrost into aquatic sediments varies over time 74 and space. A recent study showed that at the end of the last glacial, the surface active 75 laver of terrestrial permafrost released about 4.5 Tg organic carbon (OC) per vear 76 from just the Lena watershed onto the nearby shelf, whereas current annual OC 77 release is estimated to be only about a tenth of this (Tesi et al., 2016). In addition to 78 active layer material, OM from deeper and older permafrost sources can also thaw 79 and be released into the environment (Shakhova et al., 2007, 2014). This process 80 currently dominates the delivery of terrestrial material onto the East Siberian Arctic 81 shelf (Vonk et al., 2012; Semiletov et al., 1999) and is expected to increase due to 82 accelerating coastal erosion rates (Günther et al., 2013).

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84 Different permafrost OC stocks exhibit variable vulnerabilities to thaw remobilization 85 (Schuur et al., 2015). In addition to a subsea permafrost OC stock, soils and sediments 86 of the terrestrial northern permafrost zone store about 1300 ± 200 Pg OC, with 87 separate upscaling approaches applied for soil stocks (0-3m depth), deltaic sediments 88 (full depth) and Yedoma sediments (full depth) (Hugelius et al., 2014). Yedoma 89 sediments, a.k.a. Ice Complex Deposits (ICD) are polygenetic, ice-rich Pleistocene-90 aged deposits that are present in the unglaciated parts of Siberia and Alaska 91 (Schirrmeister et al., 2011). These deposits contain roughly a quarter of the OC stored 92 in permafrost terrain, but estimates vary from ca. 200-400 Pg C (Strauss et al., 2013; 93 Schuur et al., 2015). The presence of massive ice wedges in ICD causes landscapes to 94 collapse upon thaw, exposing deeper stocks of OC. This type of relatively abrupt thaw 95 is increasing in many parts of the arctic landscape (Schuur et al., 2015). At the same 96 time, deepening of the active layer causes gradual thaw that occurs across entire 97 landscapes (Shiklomanov et al., 2013).

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99 With a tool to detect and monitor different types of permafrost OM in coastal 100 environments, one could assess (historical and spatial) variability in permafrost 101 source input, degradation and thaw, as well as the relative degradation of different 102 permafrost types. For example, the relative release of OC from ICD versus topsoil 103 permafrost has earlier been distinguished and quantified through the use of dual-104 carbon isotopes ($\delta^{13}C$ and $\Delta^{14}C$) on bulk OC in the shelf environment of the Laptev 105 and East Siberian Sea. It was shown that topsoil permafrost OC dominates in 106 suspended particulate matter (Karlsson et al., 2011; 2016; Vonk et al., 2012) and ICD

107 permafrost OC dominates in the surface sediments (Vonk et al., 2012; Semiletov et al., 108 2011; 2012). Vonk et al. (2014) further showed that topsoil OC is actively degraded 109 during horizontal transport whereas ICD permafrost OC rapidly settles. Winterfeld et 110 al. (2015) showed, using dual-carbon isotopes on riverine material, that suspended 111 particulate OC in the Lena Delta mostly consists of Holocene material instead of 112 material from ICD permafrost.

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114 This $\delta^{13}C-\Delta^{14}C$ dual-carbon isotope approach carries the strong advantage that it 115 operates on the bulk OC level, thereby circumventing the "molecular-bulk upscaling" 116 challenge". This challenge relates to issues associated with upscaling from the 117 molecular isotope level to the bulk level. These issues relate to the relative 118 concentration (*n*-alkanes and *n*-alkanoic acids represent only a fraction of the total 119 OC) but also to processes such as selective degradation, differences in physical 120 association, or dispersion differences. However, the $\delta^{13}C-\Delta^{14}C$ approach also has 121 drawbacks, such as a weak distinction between the δ^{13} C end-member values of 122 Topsoil-PF versus ICD-PF. Also, the marine δ^{13} C end member values in coastal Arctic 123 shelf waters are uncertain and may be more depleted than at mid-latitudes due to 124 uptake of relatively depleted dissolved $CO₂$ values caused by cold polar water 125 (Meyers, 1997; Tesi et al. *this special issue*) or degradation of terrestrial matter 126 (Anderson et al., 2009; 2011; Semiletov et al., 2013; 2016), generating a potential 127 overlap between marine and topsoil $\delta^{13}C$ end-members.

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129 Here we propose a complementary tool to trace permafrost OC release into the 130 coastal environment based on molecular δ^2 H analysis on leaf waxes. We will evaluate 131 the performance of this tool using additional geochemical data as well as the bulk 132 $\delta^{13}C-\Delta^{14}C$ mixing approach. Isotopes in water molecules ($\delta^{2}H$ or $\delta^{18}O$) in glacial ice 133 cores as well as in massive ground ice in the northern hemisphere have been used for 134 reconstructing palaeotemperatures (e.g., Kotler and Burn, 2000; Johnson et al., 2001; 135 Opel et al., 2011; Meyer et al., 2015; Wetterich et al., 2016) as the isotopic value of 136 local precipitation is a function of local climate (Craig, 1961; Sachse et al., 2004; Smith 137 and Freeman, 2006). Higher plants use water as their primary source of hydrogen 138 during photosynthesis (Sternberg, 1988). The δ^2 H isotope values of leaf wax *n*-139 alkanoic acids or *n*-alkanes are therefore reflecting the δ^2 H isotopic value of local 140 precipitation (e.g., Sachse et al., 2004; Sessions et al., 1999), after correction for the 141 net fractionation during biosynthesis, and evapotranspiration (Leaney et al., 1985). 142 Global precipitation values can vary immensely (Dansgaard, 1964) with values up to $143 +50\%$ in Eastern Africa but approaching -200% near the North Pole (www.iaea.org) 144 or even below -400‰ in Antarctica (i.e. SLAP2 standard, Standard Light Antarctic 145 Precipitation, is -427.5‰). Additionally, the fractionation between source water and 146 plant wax molecules varies both in time and space, and can be up to -170% (Smith 147 and Freeman, 2006; Sachse et al., 2004; Polissar and Freeman, 2010) but appears 148 relatively small at higher latitudes (between -59 and -96‰; Shanahan et al., 2013; 149 Wilkie et al., 2013; Porter et al., 2016). Differences in δ^2 H signatures of leaf wax 150 molecules from terrestrial regions with different (past) climates could therefore 151 potentially be applied to derive the relative proportion of different types of thawing

152 permafrost in nearby coastal settings. Despite the plant fractionation associated with 153 kinetics and plant physiology (Sachse et al., 2012), we hypothesize that δ^2H 154 signatures of leaf wax *n*-alkanoic acids and *n*-alkanes are more depleted in OC from 155 permafrost deposits formed during the colder Pleistocene (generally correlating with 156 ²H-depleted precipitation), compared to more enriched values in OC from active layer

- 157 or surface permafrost formed during the warmer Holocene.
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159 This study investigates a source-specific δ^2H signature for both ICD permafrost and 160 recent, surface soil permafrost in Northeast Siberia. Furthermore, we explore the 161 possibilities of using these isotopic end-member values in regional source-162 apportionment calculations that aim to quantify the relative contribution of different 163 sources of permafrost OC. As permafrost thaw progresses, particularly in ice-rich 164 permafrost such as ICD, it is increasingly important to trace the fate of remobilized 165 and decomposing OC in the Arctic environment. Our proposed tool may be used to 166 trace these temporal and spatial differences in OC release from permafrost thaw, as 167 well as the extent of burial of OC in sedimentary reservoirs.

169 **2 Methods**

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171 **2.1 Sampling**

172 A total of 18 samples were collected throughout the Siberian Arctic. Recent surface 173 soils $(n=7)$ and vegetation $(n=2)$ samples were analyzed and (from here on) referred 174 to as the "topsoil" permafrost (Topsoil-PF) sample set, whereas ICD-PF samples were 175 obtained from ICD soil profiles $(n=7)$ and suspended particulates from ICD 176 formations $(n=2)$ (Fig. 1 and Table 1). Eight offshore sediments along a shelf-slope-177 continental rise transect in the Laptev Sea were collected in 2014, further marine 178 sampling details can be found in Bröder et al. (2016b).

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180 The Topsoil-PF samples represent O and A soil genetic horizons in sites with active 181 soil formation. The sites where chosen to represent typical soil and vegetation types 182 in the investigated permafrost landscapes, including both taiga and tundra sites. 183 Samples were collected by depth or soil horizon increments from open soil pits using 184 fixed volume sampling procedures.

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186 The ICD-PF samples were collected from vertical exposures that were excavated to 187 expose intact permafrost. Fixed-volume samples were collected by coring 188 horizontally into the frozen sediments to extract ICD-PF samples from consecutive 189 depths.

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191 For more details about sampling sites, including location, vegetation and soil types 192 see table 1 (terminology following the U.S.D.A. Soil Taxonomy; Soil Survey Staff, 193 2014). Sampling was done in late summer near the time of maximum annual active 194 layer depth, in July 2010 (ICD-8 and ICD-9; Vonk et al. (2013)) and August 2011 195 (Palmtag et al., 2015) for the Kolyma River region, in August 2012 for the lower Lena 196 River and Indigirka River (Siewert et al., 2015; Weiss et al., 2015) and in August 2013

- 197 for the upper Lena River (Siewert et al., 2016). For more detailed descriptions of 198 sample collection we refer to these references. The vegetation samples TS-8G (grass)
199 and TS-9G (grass) were obtained from the tundra near Medvezhka River and a birch and TS-9G (grass) were obtained from the tundra near Medvezhka River and a birch 200 forest near Y4 stream, respectively, in July 2012.
- 201

202 Samples ICD-8 and ICD-9 were obtained in July 2010 at the Duvannyi Yar ICD 203 exposure along the Kolyma River (Vonk et al., 2013). The particulate sediment 204 samples were taken from thaw streams that were freshly formed from thawing ICD 205 (transport time from thaw to sampling estimated to be less than 1h).

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207 **2.2 Analytical methods**

208 Freeze-dried samples were extracted using an ASE 200 accelerated solvent extractor 209 (Dionex Corporation, USA) using $DCM/MeOH$ (9:1 v/v) at 80 $°C$ (5x10⁶ Pa) 210 (Wiesenberg et al., 2004). After the extraction, solvent-rinsed activated copper and 211 anhydrous sodium sulfate were added to the extracts to remove sulfur and excess 212 water, respectively. After 24 h, extracts were filtered on pre-combusted glass wool 213 and concentrated with the rotary evaporator. Extracts were transferred into glass 214 tubes, evaporated to complete dryness and re-dissolved in 500 μ l of DCM. Lipid 215 fractionation was performed via column chromatography using amino-propyl Bond 216 Elut (500 mg/3 ml) to retain the acid fraction and Al_2O_3 to separate the hydrocarbon 217 and polar fractions (Vonk et al., 2010).

218

219 Prior to the analyses, saturated *n*-alkanes (hydrocarbon fraction) were further 220 purified using 10% AgNO₃ coated silica gel to retain the unsaturated fraction. The acid 221 fraction was methylated using a mixture of HCl, MilliQ water and methanol at 80° C 222 overnight to obtain the fatty acid methyl ester (FAME) fraction. Methylated acids 223 were extracted with hexane and further purified using 10% AgNO₃ coated silica gel. 224 The hydrocarbon and FAME fractions were quantified via gas chromatography mass 225 spectrometry $(GC-MS)$ in full scan mode $(50-650 \text{ m/z})$ using the response factors of 226 commercially available standards (Sigma-Aldrich). The GC was equipped with a 30 227 m×250 μ m DB5-ms (0.25 μ m thick film) capillary GC column. Initial GC oven 228 temperature was set at 60°C followed by a 10°C min⁻¹ ramp until a final temperature 229 of 310° C (hold time 10 min).

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231 The hydrogen-isotopic composition of hydrocarbon and FAME fractions was 232 measured with continuous-flow GC - isotope ratio - MS. Purified extracts were 233 concentrated and injected $(1-2 \mu)$ into a Thermo Trace Ultra GC equipped with a 234 30m×250 µm HP5 (0.25 µm thick film) capillary GC column. Oven conditions were 235 similar to the setting used for the quantification. The conversion of organic 236 biomarkers to elemental hydrogen was accomplished by high-temperature 237 conversion (HTC) at 1420° C (Thermo GC Isolink). After the HTC, H_2 was introduced 238 into the isotope ratio MS (Thermo Scientific™ Delta V™IRMS) for compound-specific 239 determination of δ^2 H values via a Thermo Conflo IV. Following a linearity test, we only 240 used peaks with amplitude (mass 2) between 1500 and 8000 mV for the evaluation. 241 The δ^2 H values were calibrated against saturated HMW *n*-alkanes using the reference 242 substance mix A4 (Biogeochemical Laboratories, Indiana University). The H_3 ⁺ factor 243 (Sessions et al., 2001) was determined every day and stayed constant $\left($ <3 ‰/V $\right)$ 244 throughout our analyses period. Each purified extract was injected three times. 245 FAMEs were further corrected to account for the methylation agent by comparing the 246 hydrogen abundance of lauric acid $(C_{12}-FA; i.e.$ 12 carbon atoms) as acid and 247 corresponding methyl ester. The average methylation correction for lauric acid was 248 23.97 \pm 3.9‰ (n=4). This correction was, normalized to chain length (i.e. increasing 249 chain lengths result in lower corrections), applied to all the FAMEs. δ^2 H values of *n*- 250 alkanes and FAMEs are reported as mean, standard deviation and weighted average.

251

252 Details of the analytical methods for extraction, work-up, and purification of the eight 253 offshore sediment samples for biomarker analysis that are included in our source-254 apportionment comparison (section 4.3) can be found in Bröder et al. (2016b). The 255 δ ²H analysis on the shelf sediments was performed in parallel with the ICD-PF and 256 Topsoil-PF samples, according to the method described above.

257

258 **2.3 Source apportionment**

259 The compound-specific δ^2H signatures in this study were used to differentiate 260 between the two major sources (end-members), Topsoil-PF and ICD-PF, using an 261 isotopic mass-balance model. We used a Markov chain Monte Carlo (MCMC) approach 262 to account for the end-member variability (Andersson et al., 2015; Bosch et al., 2015). 263 The end-members were represented by normal distributions, with mean and 264 standard deviations obtained from our analysis $(\delta^2H \text{ on TS and ICD samples})$ and 265 from literature $\delta^{13}C$ and $\Delta^{14}C$ on end-members)". For each Laptev Sea station, the 266 isotope signatures from three different terrestrial molecular markers (long-chain *n*-267 alkanes C_{27} , C_{29} and C_{31}) were used jointly to improve source apportionment 268 precision. The δ^2 H signatures for the two end-members were based on our Topsoil-269 PF and ICD-PF samples.

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271 The compound-specific δ^2 H-based source apportionment was compared to 272 $\Delta^{14}C/\delta^{13}C$ -based analysis of bulk OC using analogous MCMC techniques (e.g., Vonk et 273 al., 2012). The $\Delta^{14}C/\delta^{13}C$ -approach allows estimation of the relative contribution of a 274 third source, marine, which does not affect the presently investigated (terrestrial) 275 compounds. Accounting for the marine component to OC allows direct comparison of 276 the Holocene and Pleistocene contributions. All MCMC calculations were made using 277 Matlab scripts (ver. 2014b) using 200,000 iterations, a burn-in phase (initial search 278 period) of 10,000 and a data thinning of 10.

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280 The spatial extent of ICD in the Lena River Basin was calculated by overlaying 281 the extent of the drainage basin (from WRIBASIN: Watersheds of the World 282 published by the World Resources Institute, www.wri.org/publication/watersheds-283 world) with the extent of the Yedoma Region (digitized from Romanovsky, 1993) in 284 an equal area map projection. It was assumed that 30% of the Yedoma Region consists 285 of intact ICD (following Strauss et al., 2013).

287 **3 Results**

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289 **3.1 Bulk geochemistry**

290 The investigated Topsoil-PF and ICD-PF samples are, on a bulk geochemical level, 291 very different. Mean organic carbon contents (as $%OC$) and total nitrogen content (as 292 \ %TN) are 25 ± 12 and 1.1 ± 0.67 for Topsoil-PF samples, and 1.6 ± 0.31 and 0.17 ± 0.058 293 for ICD-PF samples, respectively (Table 1). This gives TOC/TN ratios of 25 ± 8.0 for 294 Topsoil-PF samples and 10 ± 2.6 for ICD-PF samples. Stable carbon isotopic values of 295 Topsoil-PF and ICD-PF samples are $-27.8\pm1.3\%$ and $-25.7\pm0.75\%$, respectively 296 (Table 1). Radiocarbon ages were unfortunately only available for a few ICD samples, 297 and ranged between 17 and 28 $14C$ ka (Table 1).

298

299 **3.2 Molecular geochemical composition**

300 Long-chain *n*-alkanes and *n*-alkanoic acids are abundant in epicuticular waxes and 301 therefore indicative for a source of higher plants (Eglinton and Hamilton, 1967). 302 Concentrations of individual long-chain *n*-alkanes in Topsoil-PF samples ranged from 303 1 to 340 μ g/gOC (C₂₁-C₃₃; Table 2) with an average chain length of 28 \pm 1.6. The sum 304 of high-molecular weight (HMW) *n*-alkanes $(>C_{21})$ for Topsoil-PF samples was 305 418 $_{280}^{612}$ µg/gOC (median with interquartile range) and the most abundant *n*-alkanes 306 added up to 214_{148}^{494} µg/gOC (sum of C₂₅-C₂₇-C₂₉-C₃₁) (Table 4, Fig. 2a). For ICD-PF 307 samples, the individual concentrations of long-chain *n*-alkanes were between 4 and 308 160 μ g/gOC, and the average chain length 27 ± 0.7 (Table 2). The sum of high-309 molecular weight *n*-alkanes, and most abundant *n*-alkanes were 698₆₃₀ µg/gOC and 310 347_{323}^{405} µg/gOC, respectively (Table 4, Fig. 2a). The carbon preference index (CPI), a 311 molecular ratio indicative for degradation status with values >5 typical for fresher 312 terrestrial material and values approaching 1 typical for more degraded samples 313 (Hedges and Prahl, 1993), showed values for Topsoil-PF samples of 7.3 ± 3.6 314 (average±standard deviation) and ICD-PF samples of 3.6 ± 0.8 (CPI C₂₃-C₃₁; Table 4, 315 Fig. 2c). The $C_{25}/(C_{25}+C_{29})$ ratio, indicative for the input of peat moss (*Sphagnum sp.*) 316 material (Vonk and Gustafsson, 2009; *Sphagnum* values 0.72, higher plants 0.07; Nott 317 et al., 2000) was 0.33 ± 0.22 (average tstandard deviation) and 0.34 ± 0.05 for Topsoil-318 PF and ICD-PF samples, respectively (Table 4). Another commonly used *Sphagnum* 319 proxy (Bush and McInerney, 2013), $C_{23}/(C_{23}+C_{29})$, resulted in a sharper contrast 320 between ICD-PF and Topsoil-PF samples $(0.39\pm0.13$ and 0.25 ± 0.23 , respectively; Fig. 321 2e and Table 4).

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323 Long-chain *n*-alkanoic acids $(C_{22}$ and above) were abundant in concentrations 324 between 0.122 and 2670μ g/gOC for individual homologues in topsoils, with the sum 325 of HMW *n*-alkanoic acids (>C₂₂) being 6397 $_{3167}^{7454}$ µg/gOC (median and IQR) and the 326 most abundant *n*-alkanoic acids (sum of C₂₂-C₂₄-C₂₆-C₂₈) adding up to 327 4700 $_{2670}^{6092}$ µg/gOC (Table 3, 4 and Fig. 2b). ICD-PF samples contained individual long-328 chain *n*-alkanoic acids in 2.17 and 18700 μg/gOC (Table 2), a sum of HMW *n*-alkanoic 329 acids of 8290 $_{6290}^{11430}$ µg/gOC, and the sum of most abundant, even *n*-alkanoic acids of 330 $\,$ 6630 $^{8790}_{5285}$ µg/gOC (Table 4). Topsoil-PF and ICD-PF samples had average chain lengths 331 of 24.1 ± 1.1 and 24.3 ± 0.59 , and CPI ($C_{22}-C_{28}$) values of 5.9 ± 2.7 (average \pm standard 332 deviation) and 5.0 ± 1.6 , respectively (Table 4). Shorter-chain *n*-alkanoic acids C_{16} and 333 C₁₈ are produced in basically all types of life in soils or aquatic environments, and are 334 not specific for higher plants. Topsoil-PF contained C_{16} and C_{18} homologues in 335 concentrations between 220 and 4600μ g/gOC, and ICD-PF samples between 200 and 336 10400 µg/gOC (Table 3).

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338 Degradation of organic matter involves the loss of functional groups, e.g. the loss of 339 carboxylic acids (Meyers and Ishiwatari, 1993). A high ratio of HMW *n*-alkanoic acids 340 over HMW *n*-alkanes in a sample therefore implies a relatively fresh, less degraded, 341 status (i.e. relatively more functional groups present). For Topsoil-PF samples, the 342 HMW *n*-alkanoic acid/HMW *n*-alkane ratio varied between 5.6 and 25 with an 343 average value of 13 ± 7.6 , whereas ICD-PF samples varied between 7.6 and 140 with 344 an average value of $29±43$ (Table 4, Fig. 2f).

345

346 **3.3 Molecular isotopic composition**

347 We measured δ^2 H values in long-chain *n*-alkanes and *n*-alkanoic acids between -119 348 and -313% (Fig. 3, Table 5). Mean values for HMW *n*-alkanes $(C_{25} - C_{27} - C_{29} - C_{31})$ were 349 between -201 and -247‰ for Topsoil-PF samples and between -221 and -297‰ for 350 ICD-PF samples, with consistently lower δ^2H for longer chain lengths. For HMW *n*-351 alkanoic acids $(C_{22}-C_{24}-C_{26}-C_{28})$ mean δ^2 H values were between -203 and -236‰ for 352 Topsoil-PF samples and between -261 and -278‰ for ICD-PF samples (Table 5). The 353 decrease in δ^2 H values with increasing chain length is less distinct for *n*-alkanoic acids 354 but one can observe a decrease of around $25-30\%$ from C_{22} to C_{26} (Fig. 3). For ICD-355 PF samples, the isotopic depletion for the average of the three most abundant n -356 alkanes is comparable to the average for *n*-alkanoic acids, whereas in Topsoil-PF 357 samples, the isotopic depletion for the three most abundant *n*-alkanes is a bit larger 358 than for *n*-alkanoic acids (Fig. 4).

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361

360 **4 Discussion**

362 **4.1 Using bulk geochemistry and molecular proxies**

363 Bulk geochemical and isotopic analysis, as well as analysis of molecular proxies 364 remained inconclusive in distinguishing between the two investigated sources in this 365 study. Topsoil-PF samples have a higher organic content, higher TOC/TN values 366 (representing fresh, higher plant material; Meyers, 1994) and more depleted $\delta^{13}C$ 367 values (indicative for terrestrial C3 plants; Meyers, 1997) than ICD-PF samples, 368 suggesting that these samples indeed trace a fresh soil and/or vegetation source 369 (Table 1). The $\delta^{13}C$ values of a larger ICD-PF and Topsoil-PF dataset have earlier been 370 summarized (Vonk et al., 2012 and references therein; Schirrmeister et al., 2011) 371 giving values of $-26.3\pm0.67\%$ (n=374) and $-28.2\pm2.0\%$ (n=30), respectively. Our 372 values (Table 1) are in a similar range. Despite the differences between these two 373 sources in their bulk geochemistry, it is hard to use these parameters for source 374 distinction as their variability is fairly high, and their behavior in the environment is 375 not conservative, but e.g. affected by degradation processes. On a molecular 376 geochemical level the two investigated sources are virtually indistinguishable as

 377 there is a considerable variation in molecular concentrations and proxy values (Fig. 378 2). Only one of the tested parameters, the CPI $C_{23}-C_{31}$ of *n*-alkanes, showed a 379 statistically significantly different value for the two investigated sources.

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4.2 Evaluation of molecular δ^2 **H values as a source end-member**

382 To alleviate the difficulty to distinguish between Topsoil-PF and ICD-PF with just bulk 383 and molecular geochemical characteristics, we explore the δ^2 H values of leaf wax 384 molecules (i.e. long chain *n*-alkanoic acids and *n*-alkanes) to differentiate between 385 their relative source contributions. The overall mean δ^2H of the four most abundant 386 *n*-alkanoic acids is -231±29‰ and -271±13‰ for Topsoil-PF and ICD-PF samples, 387 respectively. These values compare well with available literature (Fig. 5). Pautler et 388 al. (2014) measured δ^2 H values on C₂₉ *n*-alkanes in modern soils of the Yukon, Canada 389 of -252±9.1‰ (n=4) and aged soil δ^2 H values of -269±8.6‰ (n=13; 24-25 ¹⁴C-ka ago) 390 and -273 \pm 16.4‰ (n=9; for MIS 4, \sim 70 ¹⁴C-ka ago). Yang et al. (2011) also reported $S₂₉$ *n*-alkane δ ²H values for modern vegetation from Alaska and Arctic Canada with an 392 average value of $-252\pm43\%$ (n=8). Zech et al. (2011) reported values of C_{29} *n*-alkanes 393 collected from a permafrost exposure along the Tumara River in northeast Siberia, 394 with an average value of $-266\pm7.5\%$ (n=23) for glacial paleosoils and $-247\pm9.4\%$ 395 (n=17) for interglacial paleosoils. Our values for C_{29} *n*-alkanes for Topsoil-PF (-396 246±13‰; n=9) and ICD-PF (-280±12‰; n=9) are in a similar range (Fig. 5). For C_{28} 397 *n*-alkanoic acids, Wilkie et al. (2013) measured -252±8.7‰ (n=6) for modern 398 vegetation in northeast Siberia, whereas Porter et al. (2016) measured -269 \pm 2.7‰ 399 (n=7) for ca. 31 cal ka BP old soils in the Yukon. Compared to these studies, our values 400 for C_{28} *n*-alkanoic acids are somewhat more enriched for Topsoil-PF with -220 \pm 15‰ 401 (n=7) but roughly in the same range for ICD-PF with $-267\pm16\%$ (n=9).

402

403 The mean isotopic difference between the most abundant *n*-alkanoic acids of the two 404 investigated sources is around 40% (δ ²H values of -231 \pm 29‰ and -271 \pm 13‰ for 405 Topsoil-PF and ICD-PF samples, respectively). Despite the relatively large standard 406 deviations, the isotopic differences are statistically significant for each of the *n*-407 alkanoic acids individually $(C_{22}, C_{24}, C_{26}, C_{28};$ Fig. 3). The isotopic differences between 408 the two sources for the mean value of the four most abundant *n*-alkanes is 35% , with 409 a mean value of $-229\pm33\%$ and $-264\pm34\%$ for Topsoil-PF and ICD-PF samples, 410 respectively. Here, the individual *n*-alkane isotopic signatures are statistically 411 significantly different for C_{27} , C_{29} , C_{31} (Fig. 3) in Topsoil-PF and ICD-PF samples. The 412 selection and application of individual chain length δ^2 H values as end-members, in 413 contrast to mean chain length values, might be more appropriate for several reasons; 414 (i) to reduce variability (δ^2 H ranges for C₂₉ and C₃₁ *n*-alkanes and C₂₂ and C₂₄ *n*-415 alkanoic acids are relatively low; Fig. 3), (ii) to target the most abundant species $(C_{29}$ 416 and C_{31} *n*-alkanes are generally more abundant in soils and ICD-PF compared to 417 shorter chain lengths; Table 2), and (iii) to make use of the largest dynamic range 418 between source end-member values $(C_{31} n$ -alkane $\delta^2 H$ values of Topsoil-PF and ICD-419 PF differ by 50‰). Based on these arguments, the C_{28} *n*-alkanoic acid and the C_{29} or 420 C₃₁ *n*-alkanes are most appropriate to use for source-apportionment. The available

421 previous studies (Fig. 5) have also selected these chain lengths $(C_{28} n$ -alkanoic acid 422 and C_{29} *n*-alkanes) for proxy development.

423

424 The use of molecular δ^2 H values as tracers of terrestrial material in a marine or 425 coastal setting has the advantage that it avoids uncertainty issues related to definition 426 of the marine end-member. On the other hand, the inherent bulk-upscaling challenge 427 of any molecular proxy, is a disadvantage of the δ^2 H approach as it introduces 428 unknowns related to the molecular-bulk upscaling effort (e.g. taking into account 429 sorting and recalcitrance; discussed in depth in 4.3). We also want to emphasize that $430\,$ δ^2 H leaf wax values in the two studied end-member sets (Topsoil-PF vs. ICD-PF) 431 largely depend on the climate (warm vs. cold) and continentality (near the coast vs. 432 further inland) during plant formation, and associated differences in fractionation 433 mechanisms. Consequently, when δ^2H values in samples are used for source-434 apportionment, this may represent the fraction leaf wax produced in cold vs. warm 435 conditions (as well as degree of continentality), and not necessarily the fraction 436 Topsoil-PF vs. ICD-PF.

437

438 Finally, we realize that the amount of soil and ICD samples analyzed in this study is 439 limited, and want to point out that the results may change when more data are 440 analyzed in the near future. Additionally, studies have shown that the δ^2H signature 441 of ice within ICD permafrost deposits can range from roughly -150% to -260% 442 depending on the type of ice (wedge ice vs. pore or texture ice) as well as the period 443 of formation (different Pleistocene cold stages) (Opel et al., 2017 and references 444 therein). The source of water (i.e. type of ice) and age of the deposit will therefore 445 influence the *n*-alkane or *n*-alkanoic acid δ^2H signal. However, regardless of the 446 natural variability associated with the processes mentioned above, both ICD and 447 texture-ice isotopic compositions appear to reflect long-term climate changes (Opel 448 et al., 2017; Schwamborn et al., 2006; Dereviagin et al., 2013; Porter et al., 2016) 449 which, likely, were also captured in the *n*-alkane or *n*-alkanoic acid δ^2H signal. 450 Unfortunately, we do not have $14C$ -ages available for all ICD samples, so cross-451 referencing to published stratigraphies in the region is not possible. Coastal 452 sediments, however, will represent a mixture of material released from different 453 depths, outcrops, and stratigraphies within the catchment or coast. For source-454 apportionment applications, we reason that a growing body of leaf wax δ^2H end-455 member data from the ICD region can overcome the variability issues highlighted 456 above.

457

4.3 Comparison with 13C- 458 **14C source-apportionment: a case-study**

459 Bulk OC dual-carbon isotope data provide a quantitative apportionment tool to assess 460 the relative contributions of Topsoil-PF vs. ICD-PF. Here, we present a case-study of 461 a shelf-slope transect in the Laptev Sea (Fig. 1) where both these source-462 apportionment tools for the first time can be applied, compared and evaluated. The 463 shelf-slope transect of eight surface sediment samples stretches over 600 km from 464 the nearshore zone $(72.7°N, \le 10m$ water depth) to the continental rise $(78.9°N, \le 10m)$ 465 >3000m depth) (Table 6). More molecular and bulk geochemical characteristics of 466 these samples can be found in Bröder et al. (2016b).

467

468 The δ^{13} C- Δ^{14} C source-apportionment uses three end-members (marine, Topsoil-PF, 469 and ICD-PF). End-member values are based on previously published values (Tesi et 470 al., 2016); with a δ^{13} C value of -27.0±1.2‰ (n=38; Rodionow et al., 2006; Tesi et al., 471 2014; Gundelwein et al., 2007; Bird et al., 2002) for Topsoil-PF, and $-26.3\pm0.67\%$ 472 (n=374; Vonk et al., 2012; Schirrmeister et al., 2011) for ICD-PF. The Topsoil-PF $\Delta^{14}C$ 473 endmember was defined as $-232\pm147\%$ (n=29; Winterfeld et al., 2015; Jasinski et al., 474 1998; Kaiser et al., 2007; Höfle et al., 2013; Palmtag et al., 2015). For ICD-PF we used 475 a Δ^{14} C value of -940±84‰ (n=300; Vonk et al., 2012 and references therein). The 476 marine end-member value was $-21.0\pm 2.6\%$ (n=10; Panova et al., 2015) and $-$ 477 50.4±12‰ (n=10; Panova et al., 2015) for δ^{13} C and Δ^{14} C, respectively. Calculations 478 were made using a Markov chain Monte Carlo approach (see 2.3).

479

 480 For δ^2 H source-apportionment there is no need to include a marine end-member as 481 marine organisms do not produce long-chain *n*-alkanes or *n*-alkanoic acids. We were 482 unfortunately only able to analyze *n*-alkanes in the shelf-slope transect samples, and 483 no *n*-alkanoic acids, due to limitations in sample volume. We used the δ^2H values of 484 the C_{27} , C_{29} and C_{31} *n*-alkanes, individually. In other words, these three chain lengths 485 are taken as independent markers, providing an overdetermined system (i.e. two 486 sources defined with three different markers). This is more representative than using 487 the average (concentration-weighted) δ^2 H value for these *n*-alkanes as the end-488 member values for each chain length are different. For Topsoil-PF we used -489 215±39‰, -246±13‰, and -247±23‰ for C₂₇, C₂₉ and C₃₁ *n*-alkanes, and for ICD-PF 490 we applied $-259\pm18\%$ ₀, $-297\pm15\%$ ₀, and $-282\pm13\%$ for C_{27} , C_{29} and C_{31} *n*-alkanes, 491 respectively (see also Table 5). Afterwards, we averaged the three end-member 492 contributions derived from the three calculations for each station, thereby taking the 493 variability introduced by the end-members into account.

494

495 The source apportionment of OC from Topsoil-PF and ICD-PF to surface sediments 496 along the Laptev Sea transect differ between the bulk $\delta^{13}C-\Delta^{14}C$ and leaf wax δ^2H 497 approaches (Table 6). The former approach suggests Topsoil-PF contributions 498 between 21-70%, generally decreasing offshore, and, consequently, ICD-PF 499 contributions of 30-79%, generally increasing offshore. The latter (leaf wax δ^2H) 500 approach results in a more extreme division of sources with Topsoil-PF contributions 501 of 83-91% and ICD-PF contributions of 9-17%, with similar patterns nearshore and 502 offshore (Table 6). A contribution of 9-17% may seem more in line with the estimated 503 extent of ICD in the Lena River basin: 12% of the basin falls within the Yedoma Region 504 (as defined by Romanovsky, 1993) and about 3% consists of intact ICD (see section 505 2.3). However, the cross-shelf sites are also strongly influenced by coastal and/or 506 subsea erosion (Karlsson et al. 2011; Vonk et al., 2012; Semiletov et al., 2012; 2016) 507 so the catchment characteristics are only one part of the story. It is challenging to 508 interpret the differences between the two proxies but we elaborate below on 509 potential reasons.

510

511 Assumptions in the bulk $\delta^{13}C-\Delta^{14}C$ approach may affect these results. First, the 512 outcome of the bulk $\delta^{13}C-\Delta^{14}C$ approach is sensitive to the definition of the marine 513 end-member. Changes in the currently used δ^{13} C and Δ^{14} C value of the marine end-514 member of the East Siberian Arctic Shelf (n=10; Panova et al., 2015) would likely alter 515 the relative Topsoil-PF and ICD-PF contributions. The currently used standard 516 deviation for the δ^{13} C marine end-member is 2.6‰, which is much higher than the 517 values for the terrestrial end-members. Second, lateral transport time enroute the 518 shelf-slope transect (>600 kilometers) causing potentially significant aging of 519 sediments and its organic carbon is not accounted for in the source-apportionment. 520 Lateral transport time results in older surface OC ages on the shelf, compared to those 521 at the initial coastal deposition. Without correcting for this factor, the source-522 apportionment will generate lower contributions of the (younger) Topsoil-PF 523 component. In an attempt to estimate this effect, we recalculated (similar to Bröder 524 et al. 2016a) the relative source contributions of Topsoil-PF, ICD-PF (and marine) 525 with the bulk $\delta^{13}C-\Delta^{14}C$ approach with the assumption that the Topsoil-PF ¹⁴C age 526 would be subject to a cross-shelf lateral transport time of 5000 yrs. We assumed a 527 linear aging along the transect based on distance from the coast, with a maximum 528 value of 5000 yrs aging at station SW-01. This resulted in Topsoil-PF contributions 529 that were up to 20% higher (for the deepest stations) compared to the source-530 apportionment where lateral transport time was unaccounted for (Table 6; Fig. 6).

531

532 Assumptions in the leaf wax δ^2 H source-apportionment approach could potentially 533 also impact the outcomes, and hence differences with the bulk $\delta^{13}C-\Delta^{14}C$ results. First, 534 there is an inherent assumption related to the molecular to bulk level upscaling 535 challenge. We assume that the physical association of *n*-alkanes in different source 536 end-members (Topsoil-PF vs. ICD-PF) as well as their fractionation in the coastal 537 system is similar. However, previous research has shown that *n*-alkanes behave 538 rather differently upon their release into coastal waters; *n*-alkanes originating from 539 surface soil or vegetation debris are not bound to minerals and remain in suspension 540 during transport while being actively degraded, whereas *n*-alkanes originating in 541 deeper mineral soils settle quickly and are protected from extensive degradation 542 (Vonk et al., 2010). It is possible that most of the *n*-alkanes in the Laptev Sea sediment 543 transect originate in (deeper) mineral soils. An effect of physical association, as well 544 as the potential effect of hydrodynamic sorting patterns (Tesi et al., 2016) on the leaf 545 wax δ^2 H values of both sources could impact the source-apportionment. Another 546 factor that can introduce a bias in our leaf wax δ^2 H approach is a proton exchange of 547 the C-bound H-atoms in *n*-alkanes with environmental water, either from in situ 548 sources (soil water) or during transport (river or ocean water, or sediment pore 549 water). As there is no evidence for such exchange in young ≤ 1 million years), cold 550 sediments (Sessions et al., 2004) we suspect this process may be minimal in our 551 samples (and end-members).

552

553 When accounting for an estimated lateral transport time, the difference in estimates 554 of source contribution by the two different approaches (bulk $\delta^{13}C-\Delta^{14}C$ and leaf wax 555 δ ²H) increases offshore, from about a 25% difference near the coast to a 40% 556 difference at stations SW-01 and SW-03. This increasing offset between the results of 557 the two end-member mixing methods may be caused by several factors such as 558 variability in the marine end-member (e.g. due to changes in seasonal ice cover), a 559 selective degradation (of the topsoil OC) enroute that introduces a source bias or 560 isotopic fractionation, or remaining factors related to the lateral transport time 561 (incorrect assumption of 5000 years, non-linear aging along transect). These 562 differences highlight that both source-apportionment tools still could be fine-tuned 563 further by (i) increasing the sample size of sources to reduce end-member 564 uncertainties, (ii) continuous adjustments in end-member values and Markov chain 565 Monte Carlo calculations based on latest knowledge, and (iii) assuring regional 566 testing and verification of the method when applied to new environments.

567

568 **5 Conclusions**

569 Leaf wax δ^2 H values in samples from aquatic recipient environments can be used to 570 source-apportion the incoming terrestrial OC into two end-members; a Pleistocene 571 ICD permafrost source and a younger, Holocene, topsoil source. Mean isotopic values 572 of the C_{29} *n*-alkane, C_{31} *n*-alkane, and C_{28} *n*-alkanoic acid showed a dynamic, 573 statistically significant range of 34, 50 and 46‰ between Topsoil-PF and ICD-PF 574 samples, respectively, with ICD-PF samples being consistently more depleted 575 indicative of formation during the colder and drier Pleistocene.

576

577 A case-study where we tested two isotopic proxies (leaf wax δ^2 H and bulk δ^{13} C- Δ^{14} C) 578 to calculate the relative terrestrial source contribution of Topsoil-PF and ICD-PF 579 along a Laptey Sea surface sediment transect, showed that the two proxies vield 580 variable results but overall generate similar trends offshore. We reason that 581 variability is caused by factors such as lateral transport time, remaining uncertainties 582 in end-member definition, or environmental factors such as physical association.

583

584 Both methods (leaf wax δ^2H and bulk $\delta^{13}C-\Delta^{14}C$) bring along their inherent 585 disadvantages and advantages. The molecular approach has the distinct advantage 586 that it circumvents the uncertainties that are associated with marine end-member 587 definition in the case of bulk OC mixing model analysis. However, application of 588 molecular δ^2 H in source-apportionment studies brings along challenges related to the 589 molecular-bulk upscaling step. Bulk $\delta^{13}C-\Delta^{14}C$ source-apportionment, on the other 590 hand, has the advantage to operate on a bulk and perhaps more representative level, 591 but is hampered by remaining uncertainties associated with the marine end-member.

592

593 This study shows that δ^2H of leaf wax molecules has the potential to be used in 594 quantitative source-apportionment studies of thawing permafrost in coastal or 595 marine settings. It can serve as an alternative or complementary approach to the 596 commonly applied bulk $\delta^{13}C-\Delta^{14}C$ method. We recommend continuing data collection 597 and optimization of end-member definition and calibration. Refining the molecular 598 δ ²H proxy presented here will be beneficial in pinpointing the location and extent of 599 OC release from thawing permafrost in the coastal or fluvial environment. With 600 enhanced Arctic warming and associated intensification of permafrost thaw, 601 constraining the amount and fate of permafrost OC release will help to assess the 602 magnitude of the permafrost carbon feedback to climate warming.

603

604 **Data availability**

605 All data are available in Tables 1 through 6, as well as Supplementary Table S1.

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623

624 **Author contributions**

625 Land-based samples were collected by GH and JEV, ship-based samples were 626 collected by IS, OD, $\ddot{\text{O}}$ G, TT, LB, and JEV. Laboratory analysis was performed by LB, TT, 627 and HH. Markov chain Monte Carlo simulations were run by AA. The manuscript was 628 written by JEV with input of all co-authors.

629

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- 912 Map of coastal northeast Siberia showing the extent of ice complex permafrost (ICD; red)
913 overlaid with the location of ice complex (n=9; black diamonds) and topsoil samples
- 913 overlaid with the location of ice complex $(n=9; black \, diamonds)$ and topsoil samples $(n=9; green \, squares)$. The shelf-slope Laptev Sea transect is shown with yellow stars.
- $(n=9;$ green squares). The shelf-slope Laptev Sea transect is shown with yellow stars.
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920 Molecular concentrations and ratios of topsoil Holocene permafrost (green; n=9) and

921 deeper Pleistocene permafrost (blue; n=9) samples, with (a) the sum of odd *n*-alkanes

- 922 $C_{25}-C_{31}$, (b) the sum of even *n*-alkanoic acids $C_{22}-C_{28}$, (C) the Carbon Preference Index
- 923 (CPI) for *n*-alkanes C_{23} -C₃₁, (d), the CPI for *n*-alkanoic acids C_{22} -C₂₈, (e) the ratio of C₂₃
- 924 over $C_{23}+C_{29}$ *n*-alkanes, and (f) the sum of high-molecular weight (HMW) *n*-alkanoic
- 925 acids over HMW *n*-alkanes. The CPI is calculated as CPI_{i-n} = $\frac{1}{2} \sum (X_i + X_{i+2} + ... + X_n)/\sum$
- 926 $(X_{i-1}+X_{i+1}+...+X_{n-1}) + \frac{1}{2} \sum (X_i+X_{i+2}+...+X_n) / \sum (X_{i+1}+X_{i+3}+...+X_{n+1}),$ where X is
- 927 concentration. Stars indicate that the two compared values are statistically significant
- 928 (95% confidence). Note that panel a and b are reported as median with IQR (interquartile 929 range) and the other panels are reported as average±standard deviation.
- 930

 Molecular isotopic signature against chain length of long chain *n*-alkanoic acids (top) and *n*-alkanes (bottom) for Holocene topsoil samples (green) and Pleistocene ice complex

- samples (ICD; blue). Stars indicate that the two compared values are statistically
- significant (95% confidence). Standard deviations are represented as vertical bars, and
- are smaller than the sample circles when not visible.
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- 946 Concentration-weighted mean δ^2 H values of C₂₇-C₂₉-C₃₁ *n*-alkanes plotted against
- concentration-weighted mean δ^2 H values of C₂₄-C₂₆-C₂₈ *n*-alkanoic acids to illustrate the
948 fractionation differences between these two leaf wax markers. Dashed line indicates an
- fractionation differences between these two leaf wax markers. Dashed line indicates an
- identical fractionation.
-

- 954 Comparison of δ^2 H values of C_{28} *n*-alkanoic acid (left) and C_{29} *n*-alkane (right) in
- 955 modern (Topsoil-PF; green circles) and ICD-PF for this study (blue circles) and
- 956 available literature, with crosses from Zech et al. $(2011;$ glacial and interglacial
- 957 paleosoils from permafrost bluff exposure at Tumara River northeast Siberia), black
- 958 triangles from Yang et al. (2011; C3 plants and trees from Canada and Alaska), light
- 959 grey triangles from Wilkie et al. (2013; C3 plants from the El'gygytgyn lake basin,
- 960 Siberia), white triangles from Pautler et al. (2014; modern and paleosoils from the
- 961 Yukon territory, Canada) and dark grey triangles from Porter et al. (2016; muck
- 962 deposits from the Yukon territory, Canada).
- 963

966 967

 Contribution of OC from Topsoil-PF (green) and ICD-PF (blue) sources to surface sediments along a shelf-slope transect in the Laptev Sea (see also Bröder et al., 2016b for 971 further transect information), calculated with a $\delta^{13}C - \Delta^{14}C$ (triangles) and leaf wax δ^2H mixing model (circles). Stations are plotted against log water depth (m; see also Table 6)

- following the transect order from the coastal, nearshore, zone in the South (furthest left;
- TB-46, 6 m depth) towards the continental rise in the North (furthest right; SW-01, 3146
- 975 m depth). Topsoil Δ^{14} C end-member values are corrected for cross-shelf transport time (see section 4.2).
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Site characteristics and geochemical properties of eight topsoil and eight ice complex deposit samples. A table with more detailed sample descriptions can be found in Supplementary Table 1.

a vegetation/grass samples, labelled with "G"

b data from Vonk et al., 2013

Long-chain *n-*alkane concentrations (in µg/gOC) of topsoil Holocene samples (modern vegetation/O-horizon) and Pleistocene ice complex samples.

Long-chain *n-*alkanoic acids concentrations (in µg/gOC) of topsoil Holocene samples (modern vegetation/O-horizon) and Pleistocene ice complex samples.

Sum of most abundant long-chain *n-*alkanoic acids and *n*-alkanes (concentrations in µg/gOC), and characteristic ratios of *n*-alkanoic acids and *n*-alkanes of topsoil Holocene (modern vegetation/O-horizon) and Pleistocene ice complex samples.

a HMW; high-molecular weight

b CPI; carbon preference index for chain lengths $C_{22}-C_{28}$, for calculation see caption of Fig. 2.

c CPI; carbon preference index for chain lengths $C_{23}-C_{31}$, for calculation see caption of Fig. 2.

d2 H signatures (in ‰) of *n*-alkanoic acids and *n*-alkanes of topsoil Holocene (modern vegetation/O-horizon) and Pleistocene ice complex samples.

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Location, sampling depth and isotopic values of samples along a surface sediment transect in the Laptev Sea (data from Bröder et al., 2016b), with percentage topsoil (TS) and ice complex deposit (ICD) OC contributions to the samples based on source-apportionment calculations with δ^2 H leaf wax end-members versus δ^{13} C- Δ^{14} C end-members (end-member values are described in the text).

				Sample values						Source contributions			
IDa	Lat	Long	Depth	C_{27}	C_{29}	C_{31}	$C_{27-29-31}$ _b	$\delta^{13}C$	Δ ¹⁴ C	TS	ICD	TSc	ICD ^c
	N	°E	m	$\%$	$\%$ ₀	$\%$	$\%$ ₀	$\%$ ₀	$\%$ ₀	using $\delta^2 H$		using $\delta^{13}C - \Delta^{14}C$	
TB-46	72.700	130.180	6	-236.2	-237.4	-230.4	-235.0	-26.5	-436	89%	11%	63% (63%)	37% (37%)
$YS-9$	73.366	129.997	23	-233.7	-231.0	-227.8	-231.1	-26.1	-415	91%	8.9%	63% (65%)	37% (35%)
$YS-6$	74.724	130.016	32	-234.2	-241.0	-235.4	-236.8	-25.6	-465	86%	14%	51% (59%)	49% (41%)
$SW-24$	75.599	129.558	46	-229.3	-236.5	-243.5	-236.4	-24.8	-284	87%	13%	70% (72%)	30% (28%)
$SW-23$	76.171	129.333	56	-219.9	-243.3	-243.3	-236.0	-25.0	-333	83%	17%	65% (70%)	35% (30%)
SW-06	77.142	127.378	92	-219.5	-237.0	-241.4	-233.2	-23.2	-364	87%	13%	39% (53%)	61% (47%)
SW-03	78.238	126.150	2601	-221.1	-238.0	-247.7	-235.9	-22.6	-426	85%	15%	23% (42%)	77% (58%)
SW-01	78.942	125.243	3146	-223.8	-241.8	-246.0	-238.0	-22.3	-418	83%	17%	21% (42%)	79% (58%)

a Location, depth and bulk carbon isotope data from Bröder et al. (2016b)

b weighted average based on individual concentrations

c numbers in brackets are source contributions using the $\delta^{13}C-\Delta^{14}C$ approach but with additional corrections for cross-shelf lateral transport time of topsoil OC (similar as in Bröder et al., 2016a); we applied linear aging along the transect based on the distance from the coast, with a maximum aging of 5000 years for station SW-01.